

# **SYNTHESIS AND ANTIBIOTIC ACTIVITY OF SOME $\beta$ -CARBOLINE ALKALOIDS.**

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## **SUMMARY**

*Nine  $\beta$ -carboline alkaloids were synthesized and screened for antibiotic activity. Six of the compounds tested showed inhibitory activity against one or more of the microorganisms assayed.*

## **INTRODUCTION**

In a earlier paper, it was reported the isolation from *Aniba sentaladora* Ducke (Lauraceae) of 1-(p-hydroxybenzyl)-6-methoxy- $\beta$ -carboline (9) the structure being confirmed by a two-step synthesis that involved 4 as an intermediate (Aguiar *et al.*, 1980).

This paper presents the investigation of the antibiotic activity of nine  $\beta$ -carboline alkaloids, all of them synthesized in our laboratory.

Pharmacological interest in  $\beta$ -carbolines arises from the fact that the hallucinogenic properties of the beverages, known as *ayahuasca*, *caapi* or *yagé*, used by the Indians in Amazonas basin are attributed partially to the presence of  $\beta$ -carbolines (Aguirre *et al.*, 1969). Harmine, harmaline and related compounds can inhibit the enzyme monoamine oxidase and this property has been related with some of their effects in the central nervous system (Udenfriend *et al.*, 1958).

However, the antibiotic activity of  $\beta$ -carbolines are poorly known since the publications are limited to the results obtained by McKenna and Towers (1981) which investigated the UV mediated cytotoxicity of  $\beta$ -carboline alkaloids using yeast and bacterial bioassay systems.

## **EXPERIMENTAL**

### **Synthesis**

1-(p-Hydroxybenzyl)-1,2,3,4-tetrahydro- $\beta$ -carboline-3-carboxylic acid (3). dL-Tryptophan (2.04g) and p-hydroxyphenylpyruvic acid (2.0g) in a mixture of 1N  $H_2SO_4$  (10 ml),

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EtOH (10 ml), and H<sub>2</sub>O (30 ml) were heated under reflux (21 hr). After evaporation of EtOH and addition of conc. NH<sub>4</sub>OH (10 ml) and active charcoal, the mixture was boiled (30 min), cooled, diluted with NH<sub>4</sub>OH (5 ml), filtered, washed with Et<sub>2</sub>O, concentrated and kept at 0° (24 hr). The yellow precipitate, a mixture of stereoisomers of 3 (1.7g) melted at 198-208° dec.  $\nu_{\text{max}}^{\text{KBr}} (\text{cm}^{-1})$ : 3400-2500, 1720, 1610 (b), 1510, 1450, 840.  $^1\text{H NMR} (\text{DMSO}-d_6)$ :  $\delta$  2.70-3.32 (m, CH<sub>2</sub>), 3.35-3.88 (m, CH<sub>2</sub>), 4.90 (b, NH, H-1, H-3), 6.40-7.90 (m-8H  $\phi$ ), 10.55 (br s, NH, disappear with D<sub>2</sub>O).

1-(p-Hydroxybenzyl)- $\beta$ -carboline (7). To a boiling solution of 3 (1.61g) in H<sub>2</sub>O (300ml) were added 10% aq. K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> (60 ml) and HOAc (10.2 ml). After reflux (30min) and cooling, 10% aq. Na<sub>2</sub>SO<sub>3</sub> was added. The mixture was made alkaline with aq. NaOH and extracted with Et<sub>2</sub>O. The Et<sub>2</sub>O solution was dried and evaporated. The residue was crystallized from EtOH-cyclohexane to 7 (0.56g), yellow crystals, mp 220-223° dec.  $\nu_{\text{max}}^{\text{KBr}} (\text{cm}^{-1})$ : 3400, 3220, 1610, 1600, 1590, 1560, 1510, 1500, 1450, 820, 735.  $\lambda_{\text{max}}^{\text{EtOH}} (\text{nm})$ : 242, 253 sh 293, 340, 358 sh ( $\epsilon$  37260, 31510, 15070, 19730, 8220, 87770).  $^1\text{H NMR} (\text{CD}_3\text{COCD}_3)$ :  $\delta$  4.41 (s, CH<sub>2</sub>) 6.70 (d, J=9 Hz, H-3', H-5'), 7.20 (d, J=9 Hz, H-2', H-6'), 7.40-8.40 (m, 6H), 10.3 (br s, NH, disappear with D<sub>2</sub>O). MS [m/z (%)] 274 (95%) M<sup>+</sup>, 273 (100), 272 (25), 256 (11), 137 (15), 136 (11), 128 (16). Found: C, 79.06; H, 5.35; N, 10.74. C<sub>18</sub>H<sub>14</sub>N<sub>2</sub> requires: C, 78.01; H, 5.14; N, 10.21%.

1-(p-OAc-benzyl)- $\beta$ -carboline (8), mp 169-171° (MeOH/H<sub>2</sub>O). Found: C, 75.61; H, 5.18 N, 8.76. C<sub>20</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub> requires: C, 75.93; H, 5.10; N, 8.85%.  $\nu_{\text{max}}^{\text{KBr}} (\text{cm}^{-1})$ : 1755, 1620, 1600 1560, 1500, 1450, 1200, 900, 820, 800, 770, 745.  $^1\text{H NMR} (\text{CDCl}_3)$ :  $\delta$  2.30 (s, CH<sub>3</sub>), 4.35 (s, CH<sub>2</sub>), 6.82 (d, J=9 Hz, H-3' and H-5'), 7.00-7.50 (m, H-6, H-7, H-2', H-6'), 7.85 (d J=6 Hz, H-4), 8.10 (dd, J=8 and 2 Hz, H-5), 8.40 (d, J=6 Hz, H-3), 9.15 (br s, NH, disappear with D<sub>2</sub>O).

### Determination of antibiotic activity

Compounds 1 to 9 were dissolved in dimethylsulfoxide (DMSO) and tested for antibiotic activity against seven microorganisms as listed in Table 1.

The activity was assayed by dipping antibiotic assay (AA) discs into the test sample draining and then transferring the discs to the surface of an agar plate (Trypticase-soy agar) previously seeded with the test organism.

For quantitative bioassays, disc containing respectively 500  $\mu\text{g}$ , 100  $\mu\text{g}$ , 50  $\mu\text{g}$ , 25  $\mu\text{g}$  and 10  $\mu\text{g}$  of the samples were used. A solvent blank AA disc was included in all bioassays and discs containing 100  $\mu\text{g}$  of streptomycin sulfate and 100  $\mu\text{g}$  of amphotericin B were included in plates inoculated respectively with bacteria or yeast. The plates were incubated at 37°C and examined after 24 and 48 hours for growth.

## RESULTS AND DISCUSSION

### Synthesis of the compounds

Compounds 1 and 2 (known as synthetic products) were synthesized by condensation of dL-tryptophan with acetaldehyde and phenyl-acetaldehyde. Oxidation/decarboxylation

of these two compounds with  $K_2Cr_2O_7/CH_3COOH/H_2O$  led to compounds 5 (known as natural and synthetic product) and 6 (known as synthetic product) (Snyder *et al.*, 1948). Compound 3 was synthesized using the same procedure as for 9 that is, by condensation of dL-tryptophan with p-hydroxyphenylpyruvic acid (Aguilar *et al.*, 1980). Oxidation/decarboxylation with  $K_2Cr_2O_7/CH_3COOH/H_2O$  led to compound 7. Compound 8 was obtained by acetylation of 7. Compounds 3, 7 and 8 are new products.

### Bioassays

From the nine alkaloids tested, only compounds 1, 8 and 9 were totally inactive. Compound 5 (harman) exhibited growth inhibitory activity against all the microorganisms assayed. Compounds 3 and 4 showed activity only in concentrations above 500  $\mu$ g per disc.

A summary of the results are presented in Table 1, where the minimum inhibitory concentrations are expressed in  $\mu$ g per disc.

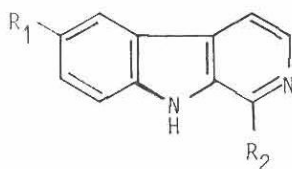
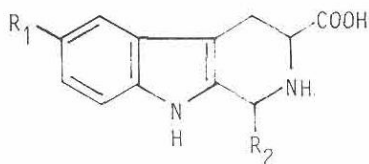
It is difficult to correlate structure and activity. It seems that activity is reduced when the piridine ring loses the aromaticity. An exception is compound 2 which still maintains the activity.

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### RESUMO

Neve alcalóides  $\beta$ -carbolínicos foram sintetizados e avaliados quanto à sua ação antibiótica. Seis dos compostos inibiram o crescimento de um ou mais dos microorganismos ensaiados.



- 1  $R_1 = H, R_2 = Me$
- 2  $R_1 = H, R_2 = Benzyl$
- 3  $R_1 = H, R_2 = p-OH-Benzyl$
- 4  $R_1 = OMe, R_2 = p-OH-Benzyl$

- 5  $R_1 = H, R_2 = Me$
- 6  $R_1 = H, R_2 = Benzyl$
- 7  $R_1 = H, R_2 = p-OH-Benzyl$
- 8  $R_1 = H, R_2 = p-OAc-Benzyl$
- 9  $R_1 = OMe, R_2 = p-OH-Benzyl$

**Table 1 - Antibiotic activity of  $\beta$ -carbolines** (minimum inhibitory concentrations in  $\mu\text{g}$  per disc)

Organism	Compound								
	1	2	3	4	5	6	7	8	9
<i>C. albicans</i>	-	250	+	+	50	250	250	-	-
<i>M. smegmatis</i>	-	25	-	+	100	50	100	-	-
<i>E. coli</i>	-	-	-	-	50	-	-	-	-
<i>S. epidermidis</i>	-	+	-	-	10	-	-	-	-
<i>S. aureus</i>	-	-	-	-	25	-	-	-	-
<i>K. pneumoniae</i>	-	-	-	-	25	-	-	-	-
<i>S. paratyphosa</i>	-	-	-	-	50	-	-	-	-

- no inhibition

+ inhibition only concentrations above 500  $\mu\text{g}$  per disc.

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