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Isolation of Natural Inhibitors of Papain Obtained from *Carica papaya* Latex

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ABSTRACT

Studies were carried out to natural papain inhibitor from papaya latex. Fresh latex from green fruits of Carica papaya was collected and immediately transported in ice bath to the lab, from which three fractions with inhibitor effect of esterase papain activity were isolated by latex dialysis, Sephadex G-25 gel filtration and ionic exchange chromatography in SP-Sephadex C-25. The isolated fractions, identified as inhibitors I and II, showed a negative reaction with ninhydrin; however, the fraction identified as P-III showed positive reaction with ninhydrin. Kine tics data showed non-competitive inhibition (inhibitor I) and uncompetitive (inhibitors II and P-III).

Key words: Esterase activity, natural papain inhibitor, protease inhibitor

INTRODUCTION

The papaya tree (*Carica papaya*) is cultivated in tropical countries and the fruit is appreciated by its flavor as well as by its easy digestion and nutritive value. The latex from green fruits contains a mixture of cysteine endopeptidase (Dubois et al., 1988), such as papain EC 3.4.22.2 (Mitchel et al., 1970), chymopapains (EC 3.4.22.6) A and B (Watson et al., 1990), papaya endopeptidase II, papaya endopeptidase IV (Barret and Buttle, 1985; Ritonja et al, 1989), and Omega endopeptidase (Dubois et al., 1988). Papain is a proteolytic enzyme widely used in both food industry and research laboratories. The occurrence and purification of natural inhibitors of proteolytic enzymes have been described in the literature for a long time (Learmonth, 1951; Whitaker and Bender, 1965; Heinrikson and Kézdy, 1976;

Kopitar et al., 1978). Hanada et al. (1978) isolated thiol proteinase inhibitor (E-64) from a culture of Aspergillus japonicus TPR-64. The purification and some characteristics of thiol proteinase inhibitors, such as papain, ficin and bromelin, were described in the literature (Järvinen, 1979; Rele et al., 1980; Vartak et al., 1980; Sasaki et al., 1981; Tanigushi et al., 1981; Norioka et al., 1982). Tamai et al. (1981) found an inhibition of the papain activity by analogous of E-64 optically active (EP-475). They showed that the SH groups of the papain decreased parallel to the lost of the incubated activity when with quantities analogous corresponding to the EP-475, suggesting that this inhibitor linked to the active thiol group in the proportion mol per mol. The synthesis of proteinase inhibitors is known to be induced in some plants by mechanical injury and by insect attack or microorganisms (Ryan and An,

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1988). A number of gymnosperm and angiosperm seeds were examined and it was found that they had a wide variation in trypsin inhibitor content. Papain inhibitor content of the same seeds showed smaller variation. Fernandes et al. (1991) suggested that the former was associated to physiological/metabolic processes while the latter was associated to defense mechanism. Hines et al. (1992) made a screening for cysteine proteinase in 44 vegetables. Papain was inhibited by soybean hypocotyls and cotyledon extracts. Protopeptides released during the activation of some proteinase zymogens may act as efficient enzyme inhibitors. The mechanism of the interaction between proteolytic enzymes and protein inhibitors were presented by Mosolov (1994). Such zymogens can be considered as covalently enzyme-inhibitor complexes. A review about the structural basis of the endoproteinase-protein inhibitor interaction was recently published by Bode and Huber (2000). These authors described that living organisms used protein inhibitors as a major tool to regulate the proteolytic activity of proteinases .In the procedure according to Kimmel and Smith (1954) six fractions were obtained from the purification of dried latex by using essentially salt fractionating in the presence of cysteine. Alternative purification has been reported. Monti et al. (2000) described a method of crystallizing papain from fresh papaya latex without using high concentration of salts or thiol-containing substances such as cysteine and dithiothreitol. Crystallized papain was obtained by this method in the form of a complex with natural inhibitor existent in latex, which could be removed by dialysis. This work shows results from partial purification of papain inhibitors and kinetics studies.

MATERIALS AND METHODS

Inhibitors extraction

Papain crystallization studies from fresh latex of *Carica papaya* were described by Monti et al. (2000), where the presence of natural inhibitors complexed to papain was detected. To extract the inhibitors in this work, the latex extract had its pH adjusted to 9.0, centrifuged and the supernatant was dialyzed in distilled water (19 mL of extract to 50 mL of water) at 4°C during 72 h. The water of the dialysis with the inhibitors was lyophilized and the dried material obtained was re-dissolved in 2.5 mL of water milli-Q.

Filtration in Sephadex G-25

Samples of 2.5 mL of the inhibitors solution, obtained as described above, were submitted to filtration columns of Sephadex G-25 (120 x 1.5 cm) using as diluent a 0.05 M ammonium bicarbonate buffer with flow of 30 mL.h⁻¹ and fractions of 3.0 mL/tube were taken in Frac-200 collector (Amersham-Biosciences). The inhibitory activity was detected in the elute after lyophilization up to the elimination of all ammonium bicarbonate. The material obtained was re-dissolved in water and samples were assayed with papain in a 0.1 M sodium phosphate buffer pH 8.0. For each experiment carried out, a control tube was made, containing just the enzyme in its proper buffer.

High voltage electrophoresis

High voltage electrophoresis was made at pH 2.1 using the inhibitors I, II, III in Whatman paper 3 MM, 57 cm length and 10 cm in one of the edges; at pH 3.5 and 6.5 the samples were applied 28.5 cm from the edge. The electrophoresis was carried out in electrophoresis Savant flask, with water circulation, with difference potential of 3,000 V, using acetic acid-formic acid buffer pH 2.1 and acetic acid-pyridine pH 3.5 and 6.5. After the runs at pH 2.1 and 3.5, the paper was dried and pulverized with 0.2% ninhydrin + 1% pyridine in acetone whereas in electrophoresis at pH 6.5 just 0.2% of ninhydrin in acetone was used. The unveiling, that is, the appearance of the violet color was carried out at 60° C. The bands of positive ninhydrin electrophoresis were eluted from the paper with water and tested to verify to which of them would correspond to the inhibitors.

Chromatography in SP-Sephadex C-25

Samples of 0.2 mL of the solutions containing inhibitors, obtained by filtrations in gel Sephadex G-25, were applied in columns (35 x 0.8 cm) containing gel SP-Sephadex C-25, pre-equilibrated with 0.05 M formic acid buffer pH 2.6, collecting fractions of 1 mL/tube. Elution were promoted using a 0.1 M ammonium formiate buffer pH 3.5. Fractions of 1 mL/tube were collected and were analyzed in spectrophotometer Ultrospec 1000 (Amershan-Bioscences) at 250 nm. The fractions with papain inhibitory activity were mixed and lyophilized.

Papain activity in absence and presence of natural inhibitors I, II, and III

Papain activity was determined according to Kirsh and Igelströn (1966) with N-carbobenzoxiglycil pnitrophenyl ester (Z-Gly-pNP) and also using α -Nbenzoyl-L-arginin ethyl ester (BAEE) as substracts (Jacobsen et al., 1957). The reactions were carried out in a thermostatic cuvette at 25° C. A Beckman model SS-2 potentiometer sensitive to variations of 0.01 units in pH was used and Tris was added with a Nimetrics microsyringe capable of measuring amounts up to 0.1 μ L. The solution was thoroughly mixed by constant shaking and protected against atmospheric molecular oxygen by constant bubbling with nitrogen. Papain activity 3x10⁻⁶ M or 5x10⁻⁵ M was incubated at 25° C with volumes of inhibitors I, II and III separately and in duplicates, in increasing order (A₂₅₀ varying between 0.009 and 0.1) during 45 minutes under nitrogenous atmosphere. After this time, samples were taken and the esterase activity was measured at different substrate concentrations, according to the description above. The enzyme used in these experiments was always activated with DTE and filtrated in Sephadex G-25 as described by Monti et al. (2000). The stability was also verified as the time went on, pre-incubating at the beginning and at the end of each experiment. The initial rate of reaction was calculated directly from the linear portion of the curves for substrate hydrolysis in function of time, and the catalytic constant for BAEE was determined by the ratio between maximum velocity and active enzyme concentration. Km and Vmax for papain and papain plus natural inhibitors were determined according to the method of Lineweaver and Burk, (1934).

papain activity (Table1) and the results indicated a decrease up to 75% of the activity, using small inhibitors aliquots. Samples of the inhibitors were submitted to filtration in Sephadex G-25. The filtration revealed the presence of four peaks through the lectures at 250 nm and the enzyme was strongly inhibited with the elutes of the fraction 93 up to 110 (Fig. 1). Three columns were made and the total volume of these fractions was lyophilized, re-dissolved in 2.5 mL of water milli-Q and re-chromatographed in the same conditions. This second chromatography confirmed the presence of inhibitors in fractions 86 up to 111 (Fig. 2). They were mixed, lyophilized and re-dissolved in 2.5 mL of water milli-Q.

The inhibitors obtained in the re-chromatography were analyzed by electrophoresis at pH 6.5. This experiment unveiled a positive ninhydrin dot in the original point where the sample was applied. At pH 3.5 a small migration occurred to the negative pole, although without a clear separation. Moreover at pH 2.1, there was a separation of the three substances that migrated towards the negative pole, confirmed by duplicate (Fig. 3).

Based on the data obtained by electrophoresis, chromatography experiments in SP-Sephadex C-25 were made. These results revealed the separation of three peaks with papain inhibitory activity. Inhibitors I and II did not link to the resin and were eluted separately soon after the void volume of the equilibrium buffer, but the inhibitor III was adsorbed on the resin and was eluted with the 0.1 M ammonium formiate buffer pH 3.5 (Fig. 4). Similar results were obtained in three identical columns. The fractions containing the inhibitors I, II, III were re-chromatographed separately in the same conditions mentioned above, and lyophilized and re-dissolved in water milli-Q.

RESULTS

The inhibitors obtained by dialysis were analyzed regarding to their inhibitory power in esterase

 Table 1 - Esterase papain activity measured at different inhibitors concentrations.

| Inhibitors (µL) | Kcat.s ⁻¹ /mol protein | Kcat.s ⁻¹ /mol SH | Inhibition (%) |
|-----------------|-----------------------------------|------------------------------|----------------|
| 0 | 3.94 | 6.68 | - |
| 10 | 1.97 | 3.34 | 50.0 |
| 20 | 1.52 | 2.58 | 61.4 |
| 40 | 1.15 | 1.95 | 70.9 |
| 60 | 0.98 | 1.65 | 75.3 |

Kcat was determined using the substrate Z-Gly-pNP 2 x 10^{-4} M, enzyme concentration 2x 10^{-8} M in 0.1 M sodium phosphate buffer in 0.001 M EDTA, pH 7.0; ionic strength 0.3 in acetonitrile 6.7%; temperature 25°C.



Figure 1 - Gel filtration in Sephadex G-25 (120 x 1.5) eluted with 0.05 M ammonium bicarbonate buffer, fractions of 3 mL/tube.



Figure 2 - Gel filtration in Sephadex G-25 (120 x 1.5 cm) eluted with H₂O; flux: 20 mL/h; Fractions: 3 mL.



Figure 3 - Paper high voltage electrophoresis at pH 2.1 of the sample obtained by Sephadex - G25 filtration; samples 1µL in triplicate. P: Standard (A-arginine; L-lysine; H-histidine; N-neutral; As-aspartic acid; G-glutamic acid); 1: inhibitor I; 2: inhibitor II; 3: inhibitor III.

Inhibitor III showed positive reaction towards ninhydrin confirming its peptidic nature, while inhibitors I and II showed negative reaction. Due to the fact that inhibitor III had a peptidic nature, it was submitted to high voltage electrophoresis at pH 2.1, revealing the presence of at least two ninhydrin positive fractions (Fig. 5). In similar electrophoresis without staining, the bands were eluted from the paper separately and submitted to inhibition tests; the higher mobility band was designed as Inhibitor P-III. This inhibitor was characterized and compared to inhibitors I and II partially purified. Then to verify the three natural inhibitors light absorption spectrum, a scanning spectrophotometry was made at 220 and 440 nm of length wave. Inhibitor P-III showed a maximum absorption peak at 275 nm, while inhibitors I and II showed light absorption at 250 nm, but did not show a defined peak.



Figure 4 - Chromatography in SP-Sephadex C-25 (35 x 0.8 cm) equilibrated with 0.5 M ammonium formiate buffer, pH 2.6. Flux: 20 mL/h; fractions: 1 mL. The inhibitor III was eluted with 0.1M ammonium formiate buffer, pH 3.5. Samples obtained in fractions 86-111 from Fig. 2.



Figure 5 - Paper high voltage electrophoresis at pH 2.1 of inhibitor III.

Papain kinetic studies were carried out using carbobenzoxyglycyl-p-nitrophenyl ester (Z-gly-pNP) as substrate and the inhibitors solutions in the necessary concentrations to inhibit 50% of the esterase papain activity, determined at 250 nm (Fig. 6). This procedure was necessary because the molecular mass of the inhibitors have not been determined yet.

Km and Vmax values were determined using α -Nbenzoyl-L-arginine ethyl ester (BAEE) as substrate in the absence or presence of inhibitors, as showed on Table 2. Inhibitor II showed higher inhibitory power of the esterase papain activity followed by inhibitors I and P-III. The results allowed to determine that inhibitor I showed noncompetitive inhibition but on the other hand allowed to determine that inhibitors II and P-III showed uncompetitive inhibition (Table 2).



Figure 6 - Estimation of the inhibition of the esterase papain activity by natural inhibitors I, II and P-III.

| Inhibitor | Km (mM) | Vmax (µmols L ⁻¹ s ⁻¹) | Kcat s ⁻¹ |
|-----------|---------|---|----------------------|
| - | 17.9 | 22.20 | 28.50 |
| Ι | 17.9 | 6.41 | 8.04 |
| - | 20.0 | 22.73 | 28.10 |
| Π | 4.02 | 4.05 | 4.73 |
| - | 20.8 | 23.80 | 29.00 |
| III | 11.4 | 12.80 | 15.80 |

Table 2 - Km and Vmax determination using BAEE as substrate.

Conditions: BAEE in KCl 0.3 M, pH kept at 6.0 with tris at 25° C. Enzyme concentration from 1 to $2x10^{-6}$ M, activated with DTE and filtered in Sephadex G-25.

DISCUSSION

Protease inhibitors are important elements of plant defense in response to insect predation and some of them act in infections by nematodes. Protease inhibitors of proteic nature are generally categorized according to the class of protease that they inhibit. Papain is inhibited by phytocistatins inhibitors (Koiwa et al., 1997). Nevertheless, different forms of inhibitors endoproteinase interaction were studied, unveiling even a new type of papain inhibitor, a kind of tiroglobuline (Polgar, 1984). The chymopapain aminoacid sequence was thoroughly studied, but some reports described isoforms of latex proteins (Polgar, 1984; Barret and Buttle, 1985; Brocklehurst and Salih, 1985). It could be possible that these isoforms were a result of different steps of chymopapain manufacturing during the latex coagulation. The occurrence of some peptides in Carica papaya latex, collected in various incisions times of green fruits, were analyzed. SDS-PAGE showed variation in the composition of these peptides during latex coagulation and that these variations could let loose the enzyme inactivation present up to two seconds after the fruit injury (Moutim et al., 1999). A different method for papain precipitation was used and, through a systematic kinetic study, the presence of natural inhibitors complexed to papain was shown (Monti et al., 2000), in accordance with the results of Moutim et al. (1999). In the present work, three fractions with inhibitory effect of the esterase activity of the papain were isolated from fresh latex by filtration in gel and ionic exchange chromatography. Inhibitors of proteolytic enzymes have been reported from other vegetable sources and also from animals. However, the method used for papain precipitation and consequently the fractions with inhibitory activity was different from those described in the literature because the inhibitors isolation after fresh latex collection of Carica papaya was immediately brought to the lab and its pH was adjusted to 9.0 and after centrifugation was dialyzed with distilled water. The dialysis water was lyophilized and the powder obtained corresponded to the starting inhibitory material for the separation. Similarly, a proteinase inhibitor (PEPI) was isolated and purified from passion fruit (Passiflora edulis Sim) juice by DEAE-Sephacel and trypsin-Sepharose 4B column chromatography. PEPI inhibited trypsin and chymotrypsin activities but did not inhibit papain activity (Hashiguschi et al., 1993). The fractions

with papain inhibitory activity obtained in the present study identified as inhibitor I and II, showed negative reaction with ninhydrin, but on the other hand fraction of P-III showed positive reaction with ninhydrin, confirming the peptide binding. Through kinetic studies it was shown that inhibitor I showed non-competitive inhibition, inhibitors II and P-III showed incompetitive inhibition.

CONCLUSIONS

Three fractions with inhibitory effect of the esterase activity of the papain were isolated from fresh latex. The fraction identified as P-III showed positive reaction with ninhydrin, confirming the peptide nature. The inhibitor I was identified with non-competitive inhibition, while II and P-III exhibited incompetitive inhibition.

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RESUMO

Este trabalho apresenta novos dados sobre inibidores naturais de papaína. O látex fresco de frutos verdes de Carica papaya foi coletado pela manhã em plantações da região de Araraquara, SP, Brasil e imediatamente transportado ao laboratório em banho de gelo. Três frações com efeito inibitório da atividade esterásica da papaína foram isoladas a partir do látex fresco, através de diálise, filtração em Sephadex G-25 e cromatografia em C-25. SP-Sephadex As frações isoladas identificadas como inibidores I e II, mostraram reação negativa à ninidrina; entretanto, a fração identificada como P-III mostrou reação positiva. Dados cinéticos revelaram inibição não-competitiva (inibidor I) e incompetitiva (inibidores II e P-III).

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