PERFORMANCE OF AN ANAEROBIC-AEROBIC REACTOR AND KINETIC STUDY OF ORGANIC MATTER REMOVAL OF CATTLE SLAUGHTERHOUSE EFFLUENT

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ABSTRACT: This paper sought to evaluate the behavior of an upflow Anaerobic-Aerobic Fixed Bed Reactor (AAFBR) in the treatment of cattle slaughterhouse effluent and determine apparent kinetic constants of the organic matter removal. The AAFBR was operated with no recirculation (Phase I) and with 50% of effluent recirculation (Phase II), with θ of 11h and 8h. In terms of pH, bicarbonate alkalinity and volatile acids, the results indicated the reactor ability to maintain favorable conditions for the biological processes involved in the organic matter removal in both operational phases. The average removal efficiencies of organic matter along the reactor height, expressed in terms of raw COD, were 49% and 68% in Phase I and 54% and 86% in Phase II for θ of 11h and 8h, respectively. The results of the filtered COD indicated removal efficiency of 52% and k = 0.0857h⁻¹ to θ of 11h and 42% and k = 0.0880h⁻¹ to θ of 8h in the Phase I. In Phase II, the removal efficiencies were 59% and 51% to θ of 11h and 8h, with k = 0.1238h⁻¹ and k = 0.1075 h⁻¹, respectively. The first order kinetic model showed good adjustment and described adequately the kinetics of organic matter removal for θ of 11h, with r² equal to 0.9734 and 0.9591 to the Phases I and II, respectively.

KEYWORDS: kinetic, cattle slaughterhouse, combined treatment system.

DESEMPENHO DE UM REATOR ANAERÓBIO-AERÓBIO E ESTUDO CINÉTICO DE REMOÇÃO DE MATÉRIA ORGÂNICA DE EFLUENTE BOVINO

RESUMO: O objetivo deste estudo foi avaliar o desempenho de um Reator Anaeróbio-Aeróbio de Leito Fixo (RAALF) com escoamento ascendente, no tratamento de efluente bruto de matadouro bovino, e obter constantes cinéticas aparentes da remoção de matéria orgânica. O RAALF foi operado sem recirculação (Etapa I) e com 50% de recirculação do efluente (Etapa II), com θ de 11 h e 8 h. Em termos de pH, alcalinidade a bicarbonato e ácidos voláteis, os resultados indicaram que o RAALF apresentou capacidade em manter condições favoráveis para os processos biológicos envolvidos na remoção de matéria orgânica, em ambas as etapas operacionais. As eficiências médias de remoção de matéria orgânica ao longo da altura do reator, expressa em termos de DQO bruta, foram de 49% e 68% na Etapa I e de 54% e 86% na Etapa II, para θ de 11 h e 8 h, respectivamente. Os resultados de DQO filtrada indicaram eficiências de remoção de 52% e k = 0,0857 h⁻¹ para θ de 11 h e 42% e k = 0,0880 h⁻¹ para θ de 8 h na Etapa I. Na Etapa II, as eficiências de remoção foram 59% e 51% para θ de 11 h e 8 h, com k = 0,1238 h⁻¹ e k = 0,1075 h⁻¹, respectivamente. O modelo cinético de primeira ordem indicou bom ajuste e descreveu adequadamente a cinética de remoção de matéria orgânica para θ de 11 h, com r² igual a 0,9734 e 0,9591 para as Etapas I e II, respectivamente.

PALAVRAS-CHAVE: cinética, matadouro bovino, sistema combinado de tratamento.

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INTRODUCTION

In slaughterhouses, water consumption can be quite variable, since it is affected by the species and size of the animal being slaughtered, by cleaning practices adopted, layout, by the slaughter method and also by the degree of industrial automation. Typical consumption values may vary from 1 to 15m³ of water per ton of slaughtered carcass (PACHECO & YAMANAKA, 2006; LÓPEZ-LÓPEZ et al., 2010; MÉNDEZ-ROMERO et al., 2011).

The effluents from cattle slaughter are characterized by having high concentrations of organic matter that can range between 1100 and 10400mg.L⁻¹ in terms of COD, 600 to 6700mg.L -1 for BOD, from 90 to 3000mg.G⁻¹ for nitrogen, 300 and 6300mg.L⁻¹ for suspended solids, as well as pathogenic and non-pathogenic agents, among other pollutants. Its main components are feces, food waste, undigested fat, blood, suspended matter, urine, small pieces of meat, soluble proteins and excrement, which must be subjected to appropriate treatment before being discharged into receiving bodies in order to reduce and/or eliminate adverse effects to the aquatic environment and public health (PACHECO & YAMANAKA, 2006; DEBIK & COSKUN, 2009; ÜN et al., 2009; . CAO & MEHRVAR, 2011; DE NARDI et al., 2011).

CHAN et al. (2009) and RAJAKUMAR et al. (2012) state that highly polluting industrial effluents are treated preferentially by anaerobic systems, due to the high concentration of organic matter, lower production of excess sludge and the potential for power generation compared to conventional activated sludge processes. However, in practical applications, the anaerobic treatment becomes inefficient with a lower rate of microbial growth and lesser reduction of organic matter, hence the need for post-treatment.

Whereas the final effluent produced by the anaerobic treatment contains remaining organic matter, its combination with aerobic treatment becomes a suitable option to produce a quality final effluent to meet the standards required by the current environmental legislation. In this sense, new configurations of reactors have been studied with the aim of more efficient systems, with better utilization of the useful reactional volume, stability and easiness of operation. DOMINGUES (2005), CHERNICHARO (2006) and WANG et al. (2009) emphasize that the application of combined anaerobic-aerobic processes can provide significant improvements in the efficiency of wastewater treatment, as well as providing a number of advantages over conventional systems, among them, less power required for aeration in the aerobic phase, lower sludge production and lower costs of implementation and operation.

BODIK et al. (2003) state that the major difficulty of the application process in combined wastewater treatment is to find an optimal operating condition of the system, in order to achieve stability and efficiency of the system.

Therefore, the aim of this study is to evaluate the performance of an Anaerobic-Aerobic Fixed Bed Reactor (AAFBR) in the treatment of cattle slaughterhouse raw effluent, subjected to variations of hydraulic detention time (θ) and determine the kinetic constants of the organic matter degradation.

MATERIAL AND METHODS

The Anaerobic-Aerobic Fixed Bed Reactor (AAFBR) consisted of a tube of polymethyl methacrylate (acrylic), with an inner diameter of 90mm and length of 1000mm, formed by a feed chamber and a reaction bed, with a total storage volume of 4.75L, where 2.85L correspond to the anaerobic zone and 1.9L to the aerobic.

For feeding and recirculation of the AAFBR, Provitec[®] peristaltic pumps, model DM5000, were used. Aeration was provided by a Boyu[®] air compressor; model S2000A, with air injection into the aeration chamber through porous rock in the aerobic area. The reactor was kept at room temperature throughout the operation.

For immobilization of biomass of the AAFBR, two types of support material were used: expanded clay, with an average particle size ranging from 10mm to 20mm and cubic matrixes of polyurethane foam of 1cm of edge, bulk density of 23kg/m³ and a porosity of approximately 95%. The polyurethane foam of the anaerobic compartment of the AAFBR was inoculated by immersion, with approximately 15L of sludge from an anaerobic pond of the cattle slaughterhouse, over a period of ten days.

The substrate used to feed the AAFBR was composed of *in natura* raw agroindustrial effluent collected at the entrance of the cattle settling tank, whose physicochemical characteristics are presented in Table 1.

PARAMETERS		AFFLUENT	
	x	MIN	MAX
pH	7.2±0.6	6.2	8.5
Liquid temperatue (°C)	24±3	15	28
BA (mg CaCO ₃ L^{-1})	507 ± 289	119	1340
AV (mg L^{-1})	283±154	44	782
Raw COD (mg L^{-1})	1048±160	837	1524
Filtered COD (mg L^{-1})	888±209	242	1432
TKN (mg L^{-1})	321±136	110	515
N-amon (mg L^{-1})	191±49	97	323
$N-NO_2^{-1}$ (mg L ⁻¹)	0.04 ± 0.03	0.01	0.11
$N-NO_3 (mg L^{-1})$	0.10±0.0	0.10	0.10
Phosphorus (mg L^{-1})	$0.4{\pm}0.2$	0.0	0.6
TS (mg L^{-1})	2772±767	1384	4273
TSS (mg L^{-1})	706±434	166	2440

TABLE 1. Summary of the characteristics of the raw effluent used as substrate for the AAFBR.
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Legend: \overline{X} : arithmetic mean; MIN: minimum value; MAX: maximum value; BA: bicarbonate alkalinity; VA: volatile acids; COD: Chemical Oxygen Demand; TKN: Total Kjedhal Nitrogen; N-amon: ammoniacal nitrogen; N-NO₂⁻: nitrite; N-NO₃⁻: nitrate; TS: total solids; TSS: Total suspended solids.

To evaluate the operational performance of the AAFBR, the reactor was subjected to a variation of θ in 11h and 8h, with affluent average flow of 0.43L h⁻¹ and 0.60L h⁻¹ and organic load applied of 10.2g L⁻¹ d⁻¹ and 13.3g L⁻¹ d⁻¹, respectively.

Two operational steps were tested, with Step I corresponding to the combined operation (anaerobic-aerobic), with a continuous upward flow and without recirculation, and Step II, corresponding to the combined operation with recirculation of 50% of the affluent flow.

The behavior of the AAFBR was evaluated by monitoring the physical and chemical parameters: pH , bicarbonate alkalinity (BA) (mgCaCO₃.L⁻¹), volatile acids (VA) (mgHAc L⁻¹), chemical oxygen demand (COD) (mg L⁻¹) of raw and filtered samples, total solids (TS) (mg L⁻¹) and total suspended solids (TSS) (mg L⁻¹) in samples of the affluent (substrate) and effluent, in accordance with the methods described in APHA (2005). Spatial sampling profiles were performed, that is, along the reactor height, for each θ h evaluated, at the following sampling points: P0 (affluent reservoir of the AAFBR, equivalent to 0.0m); P1 (beginning of anaerobic area filled with expanded clay, 0.13m high); P2 (the end of the anaerobic area filled with polyurethane foam, equivalent to 0.84m high); and P4 (beginning of the compartment containing the treated effluent, equivalent to 0.93m high).

The kinetic coefficients of organic matter degradation (k), for the θ applied, were determined by fitting the data filtered of COD by first-order kinetic model, shown in eq.(1). The apparent kinetic parameters were obtained by fitting the decay curve of filtered COD, assuming a permanent basis of complete mixing and steady state.

$$\ln \frac{|\mathbf{C}|\mathbf{t}|}{|\mathbf{C}|\mathbf{u}|} = -k\mathbf{t} \tag{1}$$

where,

 $[C]_0$ - concentration in the affluent (mg L⁻¹); $[C]_t$ - concentration in the effluent (mg L⁻¹); *k* - reaction constan (h⁻¹), and t - time (h)

RESULTS AND DISCUSSION

AAFBR performance regarding the removal of organic matter

The variations in pH, bicarbonate alkalinity (BA) and volatile acids (VA) in the effluent samples over the height of AAFBR, referring to the operational steps I and II, with θ of 11h and 8h are shown in Figure 1.

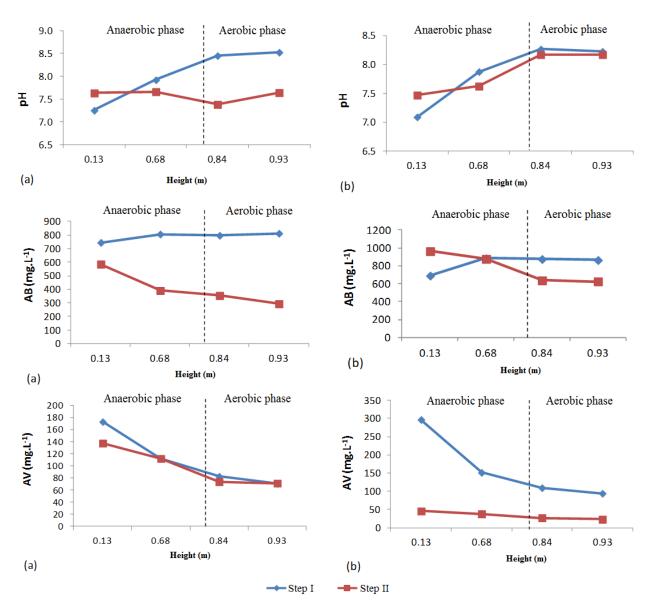


FIGURE 1. Variations in the values of pH, BA and VA along the height of the AAFBR for the operational Steps I and II for θ (a): 11h and (b): 8h.

Regarding the pH, bicarbonate alkalinity and volatile acids, results indicate that AAFBR presented an ability to maintain favorable conditions for the biological processes involved in the removal of organic matter in both operating steps, since the alkalinity is sufficient to maintain the pH greater than 6.6 and the concentration of volatile acids below 250mg L⁻¹. Values of pH of the effluent observed in Step I were 8.0 ± 0.6 and $7.9 \pm 0.6 \theta$ for 11h and 8h, respectively, and 7.6 ± 0.1 and $7.9 \pm \theta 0.4$ for 11h and 8h, respectively in Step II.

The bicarbonate alkalinity ranged from 745 ± 382 to $812 \pm 66mgCaCO_3 L^{-1}$ for θ of 11h and 689 ± 404 and $863 \pm 497 mgCaCO_3 L^{-1}$ for θ 8h, during Step I along the height of the AAFBR. During Step II, with recirculation, a reduction of the bicarbonate alkalinity along the height of the AAFBR of 586 ± 75 to $294 \pm 220 mgCaCO_3 L^{-1}$ for θ of 11 h and 963 ± 150 to $625 \pm 263 mgCaCO_3 L^{-1}$ for θ of 8 h. ABREU & ZAIAT (2008) evaluated the performance of an anaerobic-aerobic reactor filled with polyurethane foam for the treatment of sewage, operated with θ of 12h, and also confirmed the reduction of bicarbonate alkalinity of $143 \pm 38mgCaCO_3 L^{-1}$ in the affluent and $58 \pm 62mgCaCO_3 L^{-1}$ in the effluent when the reactor was operated in combined condition.

The concentration of VA during Step I ranged from 172 ± 46 to 71 ± 5 mgHAc L⁻¹ in θ 11h, and 297 \pm 94 and 177 \pm 52mgHAc L⁻¹ in θ 8h. In Step II, these values were 137 ± 85 and 71 ± 32 mgHAc L⁻¹ for θ 11h 46 \pm 6:23 \pm 6mgHAc L⁻¹ for θ 8h. From Figure 1, it is possible to see that the value of VA concentration in the effluent was lesser than the affluent over the height of the reactor to the operating conditions tested, characterizing balance of the organic matter removal process, and confirming the operational stability of the system.

NETTO (2011), when operating an anaerobic-aerobic reactor for sewage treatment in real scale, monitored the concentration of VA to evaluate the stability of the reactor and observed in his results similar behavior to the reported in this study, whose results showed a gradual reduction with average values of 56 ± 14 mgHAc L⁻¹ for the affluent and 17 ± 2 mgHAc L⁻¹ for the effluent with recirculation rate (r) of zero; 25 ± 7 mgHAc L⁻¹ for r = 1.5 and 21 ± 4 mgHAc/L for r = 3.0.

Therefore, the results showed stability of bicarbonate alkalinity during step I, after the step of anaerobic digestion and a gradual decrease in the concentration of volatile acids in the height of the reactor, as can be seen in Figure 1, indicating that there was oxidation of the organic matter in the anaerobic phase, with balance between acidogenic bacteria and methanogenic archaea, without accumulation of volatile acids inside the AAFBR. It is noteworthy that even with lower values of bicarbonate alkalinity in Step II, this variation did not indicate disturbance in the buffering system condition.

Figure 2 shows the variations in mean concentrations of TS, TSS and COD of raw and filtered samples along the height of AAFBR, referring to Steps I and II, θ 11h and 8h, respectively, of effluent samples.

According to the results of the average concentration of the TS along the height of the AAFBR, it is possible to find that the system was efficient in terms of total solids removal during Step I, when the AAFBR was operated without recirculation, with removal efficiencies of 54% and 84% for θ 11h and 8h, respectively, and in Step II efficiencies were 38 and 86%.

It can be observed similarity in the behavior of the variation of the concentration of TS and TSS for both θ tested and the two operating steps, as shown in Figure 2. The removal efficiencies of TSS were 89% and 92% in Step I to θ 11h and 8h, respectively, and 65 and 91% in Step II.

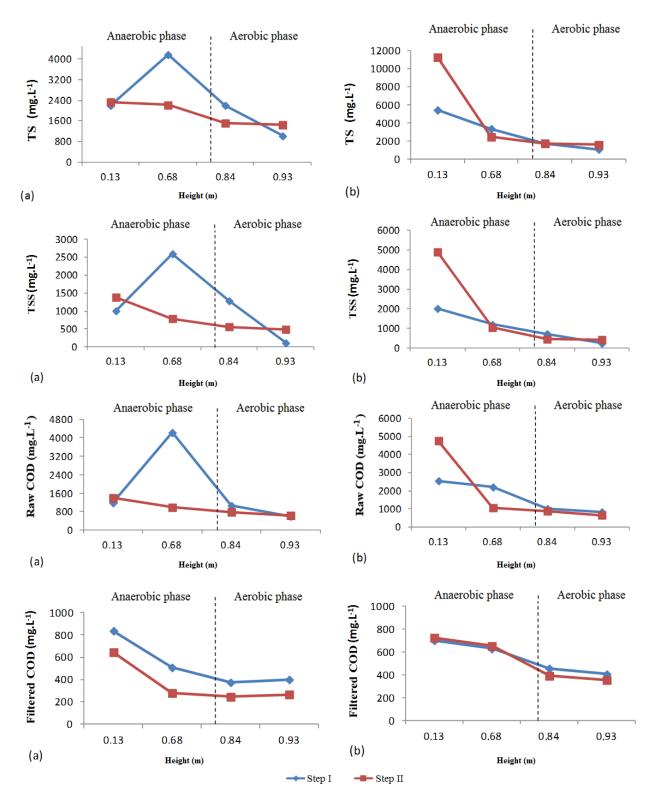


FIGURE 2. Variations in the values of TS, TSS and raw and filtered COD along the height of the AAFBR for the operational steps I and II for θ (a): 11h and (b): 8h.

The analysis of the results suggests that Step I was advantageous in terms of removal of solids (total and suspended), compared to Step II, and the efficiencies achieved in this step may have been harmed by the increase in the upflow speed, promoted by recirculation of the effluent. A peak concentration of TS and TSS can be seen in Step I, for θ 11h, which can be attributed to excessive drag of solids, due to the detachment of biomass at the time of sample collection.

The average removal efficiencies of organic matter along the height of the reactor, expressed in terms of raw COD were 49% and 68% in Step I and 54% and 86% in Step II, for θ 11h and 8h, respectively. These results indicated apparent operational stability of the reactor, which can be justified by the increase liquid-solid mass transfer speed promoted by increased speed, due to the recirculation of the effluent.

The results of the COD filtered samples indicated removal efficiencies of 52% and 42% in Step I for θ 11h and 8h, respectively, and 59% and 51% in Step II. There is a reduction in removal efficiency when decreasing θ from 11h to 8h, in both operating conditions. MARTÍN et al. (2010) examined the kinetics of psychrophilic anaerobic digestion of synthetic sewage, using an upflow filter, formed by an acrylic column, with a useful volume of 1.35L, with immobilized biomass. Their results show that in θ of 10h and 17h, removal efficiency of organic matter was almost constant and independent of θ , with an average of 80%. However, when the θ was lowered to 7h and 3h, the efficiency decreased to 77% and 65%, with a corresponding increase in the proportion of volatile/alkalinity acids.

Comparing the values of COD in raw and filtered samples, the results indicate that most of the organic matter removal, approximately 10% and 60% of Step I, 43% and 81% in Step II for θ 11h and 8h, respectively, occurred in the anaerobic compartment. The low yield occurred at θ of 11h can be attributed to the interference of the acclimatization period, which may have been insufficient to allow increased removal of organic matter under anaerobic condition.

Organic matter remotion kinects

GHANIYARI-BENIS et al. (2010) point out that kinetic studies are useful to reproduce the empirical behavior of the degradation process of organic matter and understand the metabolic pathways of biodegradation and simultaneously save time and money. However, the development of a model of degradation of organic matter is complex and presents difficulties in view of the large number of variables, including the hydraulic detention time, substrate concentration and configuration of the reactors that affect and interfere with the anaerobic system and aerobic degradation.

The decay curves of organic matter, expressed in terms of filtered COD along the height of AAFBR, operated with θ of 11h, with and without recirculation, to a first order kinetic model, are presented in Figure 4.

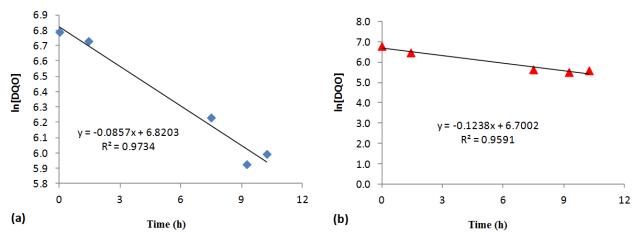


FIGURE 4. Decay curves of organic matter consumption along the height of the AAFBR for θ of 11h: (a) without recirculation; (b) with recirculation.

The decay curves of organic matter, expressed in terms of filtered COD along the height of the AAFBR, operated with θ 8h, with or without effluent recirculation, to a first order kinetic model, are presented in Figure 5.

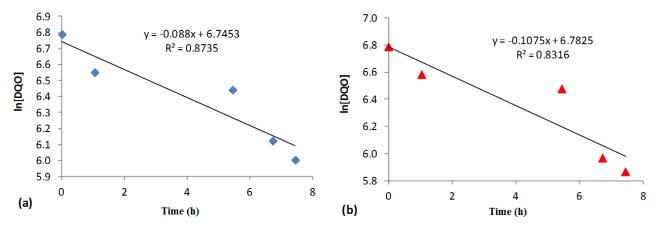


FIGURE 5. Decay curves of organic matter consumption along the height of the AAFBR for θ of 8h: (a) without recirculation; (b) with recirculation.

Table 2 shows the first order kinetics constant values which represent the consumption of organic matter, in terms of filtered COD and their respective correlation coefficients.

θ (h)	Operating condition	k (h ⁻¹)	$v_r (mg L^{-1} h^{-1})$	r^2
11	Step I	-0.0857	-0.584	0.9734
	Step II	-0.1238	-0.829	0.9591
8	Step I	-0.0880	-0.682	0.8735
	Step II	-0.1075	-0.729	0.8316

TABLE 2. Average parameters estimated from the first order kinetic model considering θ of 11h and 8h.

From the results shown in Table 2, it can be seen that when the AAFBR was operated with recirculation of the effluent in step II, in both θ , the coefficients of organic matter removal (k) are greater. This increase can be explained by the reduced mass transfer resistance caused by the recirculation of the effluent. This behavior can be seen in the larger removal efficiencies of filtered COD at this step.

FAZOLO (2003) evaluated kinetics parameters and mass transfer of a fixed bed radial reactor, at pilot scale, with immobilized biomass in polyurethane foam, fed with effluent from a Horizontal-Flow Anaerobic Immobilized Biomass (HAIB) treating sanitary sewage and concluded that the resistance to mass transfer significantly affects the overall conversion speed in the system. The author has found values of first-order removal kinetics constants that ranged from 1.58h⁻¹ to 2.78h⁻¹. Both authors observed a tendency toward the apparent first-order kinetics constants with the increasing of air flow rate, due to the decrease of the external resistance to mass transfer as the air flow rate was increased.

CASTILLO et al. (1999) studied the kinetics of an anaerobic-aerobic combined system consisting of an UASB reactor followed by two RBC aerobic reactors (rotating biological contactor) in series, with air diffusers, treating domestic sewage. The authors found that the first-order kinetic model was suitable to describe the degradation of organic matter in the combined system and obtained kinetic constants (k) of $0.31h^{-1}$ for a temperature of 19° C and $0.20h^{-1}$ for a temperature of 12.5° C in the anaerobic unit.

The results of the filtered COD samples of the present study indicate removal efficiencies of 52% for θ of 11h and k of 0.0857h⁻¹, and 42% and of k 0.0880h⁻¹ for θ of 8h for Step I. In Step II, the efficiencies were 59% and 51% for θ of 11h and 8h, with values of $k = 0.1238h^{-1}$ and 0.1075h⁻¹, respectively. It is observed that the higher the apparent kinetics coefficient (k) of the first order for degradation of organic matter, the higher the removal efficiency of COD, so there is a directly proportional to the performance of the AAFBR, as the greater the value of (k), the higher the reaction rate and hence the more efficient the system.

PINHEIRO (2006) investigated the influence of varying the recirculation rate of the liquid phase and increase the substrate concentration on the stability, efficiency and kinetics of an anaerobic reactor, operated in sequential batch with recirculation of the liquid phase, and obtained removal efficiency of 79% for filtered samples with affluent COD of 1,000 mg L⁻¹ and first-order kinetics constant of 1.14 h⁻¹, whereas for COD of 1,500 mg L⁻¹, the efficiency was 82% and the kinetics parameter was $1.31h^{-1}$.

The obtained values of r^2 (Table 2) confirmed that the first order kinetic model was more efficient to describe the behavior of AAFBR regarding the degradation of organic matter to the θ of 11h with $r^2 = 0.9734$ in Step I and $r^2 = 0.9591$ in Step II, than for the θ of 8h, with r^2 equal to 0.8735 and 0.8316 for Steps I and II, respectively.

ZAIAT & FORESTI (1997) presented a methodology for estimating both apparent and intrinsic kinetics parameters of a horizontal-flow anaerobic immobilized biomass in polyurethane foam, operated with θ of 8h, treating glucose-based effluent. COD and volatile acids data were used to estimate the kinetics parameters. The authors stated that the first-order kinetic model showed the best fit to the experimental data and the apparent kinetics parameter resulted in 0.233h⁻¹ with a correlation coefficient of 0.998.

GHANIYARI-BENIS et al. (2010) performed kinetics modeling of anaerobic degradation of a chambered hybrid anaerobic reactor, operated at mesophilic temperature in treating synthetic sewage. The results obtained for the kinetics constant (k) was 0.60 h^{-1} assuming a hydrodynamic behavior of reactors in series and steady state conditions. The authors concluded that the first-order kinetic model was adequate to reproduce the degradation of organic matter with $r^2 = 0.9995$ when operated with θ of 8h and $r^2 = 0.9996$ for θ of 16h.

LUIZ (2007) conducted an operational evaluation of three upflow anaerobic fixed bed reactors containing immobilized biomass of blast-furnace slag, polyurethane foam and gravel, in the treatment of wastewater from processing coffee fruit. The author reports that the reactor filled with polyurethane foam showed better removal efficiency of organic matter in terms of filtered COD samples (83%), compared to the other media materials, and confirmed that the first-order kinetic model for this operating condition was able to describe the behavior of upflow anaerobic reactors, with values of k equal to 0.0854 h^{-1} ; 0.0795 h^{-1} and 0.1666 h^{-1} .

CONCLUSIONS

Considering the results of raw COD samples, it can be concluded that the reduction of θ 11h to 8h promoted an increase in the removal efficiency of organic matter in both operational steps.

Regarding pH, bicarbonate alkalinity and volatile acids, the results indicate that AAFBR showed an ability to maintain favorable conditions for the biological processes involved in the removal of organic matter in both operational steps. The bicarbonate alkalinity and gradual decrease in the concentration of volatile acids along the height of the reactor showed a buffering system, balance between the acidogenic bacteria and methanogenic archaea in the anaerobic phase and the consumption and generation of alkalinity in the processes of nitrification and denitrification of the aerobic phase.

The first-order kinetic model indicated good fit and adequately described the kinetics of removal of organic matter of AAFBR for θ of 11h, with r² equal to 0.9734 and 0.9591 for Steps I and II, respectively. In conclusion, from the values of the kinetics coefficients of the first order for degradation of organic matter, the higher the values of k) found, the higher the removal efficiency of COD.

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