Prenylated Coumarins, Chalcone and New Cinnamic Acid and Dihydrocinnamic Acid Derivatives from *Brosimum gaudichaudii*

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Três novos derivados naturais dos ácidos cinâmico e diidrocinâmico foram isolados das raízes de *Brosimum gaudichaudii*, além de mais quatorze substâncias naturais conhecidas (dez cumarinas, uma chalcona, os dois esteróides β -sitosterol e 3β -O- β -D-glicopiranosil- β -sitosterol e o triterpeno β -amirina). As estruturas destas substâncias foram estabelecidas com base na análise de dados espectrais, inclusive experiências de RMN 2D e espectros de massas. O ácido 4-hidroxi-3-prenilcinâmico pode ser postulado como precursor para todas substancias aromáticas. As novas substâncias naturais são derivadas do precursor das cumarinas que perdeu a possibilidade de formar o anel lactônico devido a O-metilação.

Three new natural cinnamic acid and dihydrocinnamic acid derivatives were isolated from the roots of *Brosimum gaudichaudii*, in addition to fourteen known substances (ten coumarins, one chalcone, β -sitosterol, 3β -O- β -D-glucopyranosylsitosterol and β -amyrin). The structures were established by spectral data, including analysis of 2D-NMR experiments and mass spectra. All aromatic compounds have 4-hydroxy-3-prenylcinnamic acid as a common precursor. The new substances are derived from a precursor of the coumarins, which through an O-methylation lost its ability to form the lactone ring.

Keywords: Brosimum gaudichaudii, Moraceae, Coumarins, Chalcone, Cinnamic acid and dihydrocinnamic acid derivatives

Introduction

In previous reports,¹⁻³ the isolation of the coumarins gaudichaudine (1), xanthyletin (2), luvangetin (3), psoralen (4), bergapten (5) and (+)-(2'S,3'R)-1'-hydroxy-marmesin (6) from extracts of the root bark of *Brosimum gaudichaudii* Trécul., Moraceae, was described.

In this paper, we report the isolation of eleven additional compounds, eight known substances, and three new cinnamic acid derivatives: 3-(7-methoxy-2,2-dimethyl-2*H*-6-chromenyl)-(*E*)-propenoic acid (**7**), 3-(7-methoxy-2,2-dimethyl-2*H*-6-chromenyl)propanoic acid (**8**) and 3-(6-methoxybenzo[*b*]furan-5-yl)propanoic acid (**9**). The known substances are the coumarins marmesin (**10**), 1',2'-dehydromarmesin (**11**), 8-methoxymarmesin (**12**) and 1'-hydroxy-3'-*O*- β -glucopyranosylmarmesin (**13**), the chalcone 2',4',4-trihydroxy-3',3-diprenylchalcone (**14**), the steroids β -sitosterol and 3 β -O- β -D-glucopyranosyl-

sitosterol (daucosterol) and the triterpene β -amyrin. All substances were isolated from the roots of this species.

The genus *Brosimum*, which is traditionally employed in the photochemotherapy of psoriasis,^{4,5} is known for the production of prenylated coumarins,⁶⁻⁸ including furanocoumarins. These linear furanocoumarins, also known as psoralens (*e.g.* **4** and **5**), are widely distributed in plants and have also been used internally and externally to promote skin pigmentation and skin tanning. Methoxysalen (xanthotoxin=8-methoxypsoralen) is used in medicine to facilitate skin repigmentation in patients affected by severe vitiligo.⁹

Results and Discussion

The known natural products marmesin (**10**), 1',2'dehydromarmesin (**11**), 8-methoxymarmesin (**12**), 1'hydroxy-3'-O- β -glucopyranosylmarmesin (**13**) and 2',4',4trihydroxy-3',3-diprenylchalcone (kanzonol C, **14**) were identified through their ¹H and ¹³C NMR spectral data,

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including the comparison with literature values for **10**,^{10,11} **11**,¹² **12**, named rutaretin methyl ether,¹³ **13**¹⁴ and **14**.¹⁵ The new peracetyl derivative **13a** was also used to confirm the structure of **13** and to obtain the complete ¹H and ¹³C NMR chemical shift assignments (see Experimental). The terpenoids β -sitosterol and β -amyrin were identified by direct comparison with authentic samples. 3β -O- β -Dglucopyranosylsitosterol was characterized by ¹H and ¹³C NMR spectral data, including the data of its peracetyl derivative, and comparison with literature values.¹⁶

The multiplicity of each carbon of the components present in the mixture of the cinnamic acid derivatives 3-(7-methoxy-2,2-dimethyl-2H-6-chromenyl)-(E)-propenoic acid (7), 3-(7-methoxy-2,2-dimethyl-2H-6-chromenyl) propanoic acid (8) and 3-(6-methoxybenzo[b]furan-5yl)propanoic acid (9) was deduced from the analysis of the HBBD- and DEPT-¹³C NMR (125 MHz) spectra (Table 1). This analysis in combination with the ¹H NMR and low resolution mass spectra (m/z 260, 66.2 %, [M]⁺ of 7, $C_{15}H_{16}O_4; m/z 262, 39.5\%, [M]^+ \text{ of } \mathbf{8}, C_{15}H_{18}O_4; 220, 60.8\%,$ $[M]^{+}$ of 9, $C_{12}H_{12}O_4$) allowed the deduction of the expanded molecular formulas: (C)₅(COOH)(CH)₆(CH₃)₂ (OMe)(O) for 7, $(C)_{5}(COOH)(CH)_{4}(CH_{5})_{5}(CH_{3})_{5}(OMe)(O)$ for 8 and $(C)_4(COOH)(CH)_4(CH_2)_2(OMe)(O)$ for 9 (Table 1). On the basis of the relative intensities of the singlet signals corresponding to H-5 (7/8: $\delta_{\rm H}$ 7.13/6.74) and H-4 (9 $\delta_{\rm H}$ 7.33) in the ¹H NMR spectrum (500 MHz) the percentages of 7 (43.9 %), 8 (13.8 %) and 9 (42.3 %) in the mixture were calculated. The composition of the mixture was confirmed by GC/EIMS analysis: 7 yielded a peak at Rt = 33.95 min, with the molecular ion at m/z 260, **8** at Rt = 29.33 min, with the molecular ion at m/z 262, and **9** at Rt = 25.40 min, with the molecular ion at m/z 220 (Scheme 1).

The presence of a 6-substituted 7-methoxy-2,2dimethyl-2*H*-chromenyl moiety $[(C)_5(CH)_4(CH_3)_2(OMe)O = C_{12}H_{13}O_2]$ in the compounds **7** (43.9 %) and **8** (13.8 %) was recognized by signals corresponding to H-8 [**7**/**8**: δ_{H}



Scheme 1. Proposed fragmentation patterns for 7, 8 and 9 (only peaks classified as principals)

	7		8		9	
	δ_{c}	$\delta_{_{ m H}}$	$\delta_{ m c}$	$\delta_{_{ m H}}$	$\delta_{ m c}$	$\delta_{_{ m H}}$
С						
1'	172.82	-	178.93	-	178.93	-
2	77.50	-	77.28	-	-	-
5	-	-	-	-	124.75	-
6	114.31	-	120.56	-	155.93	-
7	160.16	-	158.28	-	-	-
8	-	-	-	-	154.91	-
9	157.05	-	152.83	-	119.91	-
10	115.89	-	113.68	-	-	-
СН						
2	-	-	-	-	143.79	7.49 (d, 2.0)
2'	114.50	6.34 (d, 16.0)	-	-	-	-
3	128.55	5.51 (d, 9.8)	126.61	5.42 (d, 9.8)	106.25	6.64 (d, 2.0)
3'	142.28	7.97 (d, 16.0)	-	-	-	-
4	121.34	6.26 (d, 9.8)	121.91	6.22 (d, 9.8)	121.31	7.33 (s)
5	127.10	7.13 (s)	127.47	6.74 (s)	-	-
7	-	-	-	-	94.14	6.99 (s)
8	99.86	6.36 (s)	99.54	6.32 (s)	-	-
CH ₂						
2'	-	-	34.23	2.60 (t, 8.0)	34.33	2.67 (t, 7.6)
3'	-	-	25.27	2.82 (t, 8.0)	26.32	3.00 (t, 7.6)
CH ₃						
4', 5'	28.37	1.43 (s)	28.05	1.40 (s)	-	-
MeO-6	-	-	-	-	55.59	3.83 (s)
MeO-7	56.70	3.84 (s)	55.38	3.75 (s)	-	-

Table 1. ¹H (400 MHz) and ¹³C (125 MHz) NMR spectral data for the mixture of **7**, **8** and **9**, in CDCl_3 . Chemical shifts in δ (δ_H and δ_C) and coupling constants (*J*, in parentheses) in Hz*

*Multiplicity of carbon signals deduced by comparative analysis of HBBD- and DEPT-¹³C NMR spectra. Homonuclear 2D ¹H-¹H-COSY and heteronuclear HMQC and HMBC (Table 2) spectra were also used in these assignments. Chemical shifts and coupling constants (*J*) obtained from 1D ¹H NMR spectrum

	7		8		9	
	² <i>J</i> _{CH}	³ <i>J</i> _{CH}	$^{2}J_{\rm CH}$	³ <i>J</i> _{CH}	² <i>J</i> _{CH}	³ <i>J</i> _{CH}
С						
1'	Н-2'	H-3'	2H-2'	2H-3'	2H-2'	2H-3'
2	H-3 3H-4'/3H-5'	H-4	H-3 3H-4'/3H-5'	H-4	-	-
5	-	-	-	-	2H-3'	H-7 2H-2'
6	H-3'		2H-3'	H-8 2H-2		'2H-3' MeO-6
7	H-8 H	H-5 H-3' MeO-7	H-8	2H-3' MeO-7	-	-
8	-	-	-	-	H-7	H-2' H-3' H-4
9	H-8	H-4 H-5	H-8	H-5	H-4	H-3 H-8
10	H-4	H-3	H-5	H-3 H-8	-	-
СН						
3					H-2	
3'						H-4
4		H-5				
5		H-3' H-4		2H-3'	-	-
CH ₂						
2'			2H-3'		2H-3'	
3'			2H-2'	H-5	2H-2'	
CH ₃						
4', 5'		H-3				

Table 2. Heteronuclerar long-range coupling of ¹H-¹³C for the mixture of 7, 8 and 9 observed in HMBC spectrum, in CDCl₃*

*Multiplicity of carbon signals deduced by comparative analysis of HBBD- and DEPT-¹³C NMR spectra. Homonuclear 2D ¹H-¹H-COSY and heteronuclear HMQC (Table 1) spectra were also used in these assignments

6.36/6.32 (s)], H-5 [**7**/**8**: $\delta_{\rm H}$ 7.13/6.74 (s)], H-4 [**7**/**8**: $\delta_{\rm H}$ 6.26/6.22 (d, J 9.8 Hz)], H-3 [7/8: $\delta_{\rm H}$ 5.51/5.42 (d, J 9.8 Hz)], the methyl groups at C-2 [7/8: $\delta_{\rm H}$ 1.43/1.40 (s)] and the methoxyl at C-7 [7/8: $\delta_{\rm H}$ 3.84/3.75 (s)] observed in the ¹H NMR spectrum of the mixture (Table 1). This deduction was confirmed by the ¹³C NMR spectrum (Table 1). The location of a methoxy group at carbon atom C-7 of the chromenyl moiety was confirmed by HMBC correlations after the unambiguous assignment of the chemical shifts of the directly bound CH pairs by cross-peaks observed in the HMQC spectrum (Table 1), such as hydrogen atoms H-5 [**7**/**8**: $\delta_{\rm H}$ 7.13/6.74 (s)] and H-8 [**7**/**8**: $\delta_{\rm H}$ 6.36/6.32 (s)] and the corresponding carbon atoms C-5 (7/8: δ_{C} 127.10/ 127.47) and C-8 (7/8: δ_c 99.86/99.54). In the HMBC spectrum (Table 2) the following correlations established the location of the methoxyl group: a) correlation of H-8 $[7/8: \delta_{\mu} 6.36/6.32 \text{ (s)}]$ with both oxygenated carbons C-7 (7/8: $\delta_{\rm C}$ 160.16/158.28, $^2J_{\rm CH}$) and C-9 (7/8: $\delta_{\rm C}$ 157.05/ 152.83, ${}^{2}J_{CH}$); b) correlation of H-5 [**7**/**8**: δ_{H} 7.13/6.74 (s)] with both C-7 and C-9; c) correlation of MeO-7 [7/8: δ_{μ} 3.84/3.75 (s)] with C-7; and d) correlation of H-4 [7/8: $\delta_{\rm H}$ 6.26/6.22 (d, J 9.8 Hz)] with C-9, along with other data summarized in Table 2. Thus, the difference of 2 daltons observed between the molecular ions of 7 (m/z 260, [M]^{.+}, $C_{15}H_{14}O_{4}$) and of **8** (*m*/*z* 262, ([M]⁻⁺, $C_{15}H_{18}O_{4}$) was attributed to the presence of 6-propenoic acid $(C_3H_5O_2 = C_{15}H_{18}O_4 - C_{15}H_{18}O_4)$ $C_{12}H_{13}O_2$ and 6-propanoic acid ($C_3H_3O_2 = C_{15}H_{16}O_4$ - $C_{12}H_{13}O_2$ moieties, respectively. The signals corresponding to these moieties, observed in the ¹H NMR spectrum (Table 1), were attributed to 2H-3'/H-3' [8/7: $\delta_{\rm H}$ 2.82 (t, J 8.0 Hz)/7.97 (d, J 16.0 Hz, E-configuration) and 2H-2'/H-2' [8/7: $\delta_{\rm H}$ 2.60 (t, J 8.0 Hz)/6.34 (d, J 16.0 Hz, Econfiguration). Through correlations in the ¹H-¹³C-HMQC- ${}^{1}J_{CH}$ spectrum (Table 1) the corresponding carbon signals were identified: CH₂-3'/CH-3' (8/7: δ_{C} 25.27/142.28) and CH₂-2'/CH-2' (8/7: δ_{C} 34.23/114.50). Thus, the structures of these two new benzopyran derivatives were determined as 3-(7-methoxy-2,2-dimethyl-2H-6-chromenyl)-(E)propenoic acid (7) and 3-(7-methoxy-2,2-dimethyl-2H-6chromenyl)-propanoic acid (8). The complete ${}^{1}H$ and ${}^{13}C$ chemical shift assignments of these natural products, as derived from the homonuclear 2D 1H-1H-COSY and heteronuclear 2D 1H-13C shift-correlated (HMQC and HMBC) experiments, are summarized in Table 2.

The remaining signals observed in ¹H and ¹³C NMR (Tables 1 and 2) and mass (m/z 220, 60.8%, [M]⁺, C₁₂H₁₂O₄) spectra were used to establish structure **9** [(C)₄(COOH) (CH)₄(CH₂)₂(OMe)(O)] for the third component (42.3 %) present in the mixture. The additional signals reported in the ¹H NMR spectrum (Table 1) were used to characterize the 5-substituted 6-methoxybenzofuran unit [$\partial_{\rm H}$ 6.99 (s),

H-7; 7.33 (s), H-4; 6.64 (d, J 2.0 Hz), H-3; 7.49 (d, J 2.0 Hz), H-2] and the linked propanoic acid moiety $[\delta_{\rm H} 3.00$ (t, J 7.6 Hz, 2H-3') and $\delta_{\rm H} 2.67$ (t, J 7.6 Hz, 2H-2')] of structure **9**. Through the correlations in the HMQC spectrum (Table 1) the corresponding signals of the carbon atoms were found at $\delta_{\rm C}$ 94.14 (C-7), 121.31 (C-4), 106.25 (C-3), 143.79 (C-2), 26.32 (C-3') and 34.33 (C-2'). Additional analysis of the ¹H and ¹³C NMR spectra, including the homonuclear 2D ¹H-¹H-COSY and heteronuclear 2D ¹H-¹³C shift-correlated spectra (Table 2), was used to confirm structure **9** and to assign all ¹H and ¹³C chemical shifts unambiguously. Thus, the structure of the third component was defined as 3-(6-methoxybenzo[*b*]furan-5-yl)propanoic acid (**9**), a new natural product.

Compound **7** (chromenylacrylic acid) was reported as intermediate product in the synthesis of 7-methoxy-2,2dimethyl-6-vinylchromone (anhydroencecalinol), a natural product isolated from *Flourensia cernua*.¹⁷ The natural products **9a** and **9b** similar to **9** were isolated from *Ruta graveolens*.¹⁸

The new natural compounds have structures closely related to those of the isolated coumarins. In fact, all the isolated aromatic compounds including the chalcone 14 can be joined in a biosynthetic sequence involving the common precursor 4-hydroxy-3-prenylcinnamic acid, which. by a hydroxylation at the 6-position should furnish 2,4-dihydroxy-5-prenylcinnamic acid. Subsequently, 2,4dihydroxy-5-prenylcinnamic acid can undergo a cyclization leading to the coumarins through 7-hydroxy-6-prenylcoumarin (7-hydroxy-6-prenylcoumarin $\rightarrow 2$ and 10; $2 \rightarrow 3$; $10 \rightarrow 6$ and 12; $6 \rightarrow 1$, 4, 11 and 13; $4 \rightarrow 5$) or, alternatively, an O-methylation at the 2-hydroxyl giving 4-hydroxy-2-methoxy-5-prenylcinnamic acid, which makes the formation of the coumarin lactone ring impossible and may be postulated as the direct precursor of the isolated new natural compounds 7, 8 and 9.

The fact that all isolated aromatic compounds bear a prenyl-unit or prenyl-derived substituent at the carbon corresponding to the 3-position in cinnamic acid, led to the conclusion that 4-hydroxy-3-prenylcinnamic acid should be the common precursor, implicating that the prenylation should precede the formation of the lactone ring. This contrasts with some previous studies on the biosynthesis of furanocoumarins, in which the prenylation was considered to occur only after the coumarin lactone ring was formed.^{19,20} In the most recent study on this subject it was shown that umbelliferone in *Apium graveolens* is prenylated and that this prenylation is achieved via the novel mevalonate independent pathway.¹⁹ It remains to be investigated if in *Brosimum* the biosynthesis is different, as was suggested by the results presented here.

Experimental Section

General

NMR spectra in CDCl₃ or CD₃SOCD₃ solvents were recorded at 300, 400, 500 and 600 MHz for ¹H and 75, 100, 125 and 150 MHz for ¹³C on a Varian Unit Plus 300, Bruker Avance DPX 400, Avance 500 or Avance 600 spectrometers, respectively, using TMS as int. standard or by reference to solvent signals CHCl₃ at $\delta_{\rm H}$ 7.26 or CD₂HSOCD₃ at $\delta_{\rm H}$ 2.49 and ${}^{13}CDCl_3$ at δ_C 77.00 or ${}^{13}CD_3$ SO ${}^{13}CD_3$ at δ_C 39.50; LRMS was obtained on a GS/MS-QP 5000 or HP 5989 mass spectrometer. The ¹³C multiplicity was deduced by comparative analysis of the HBBD- and DEPT-13C NMR spectra. Homonuclear ¹H connectivity was determined by ¹H-¹H-COSY spectra. Heteronuclear ¹H and ¹³C connectivity was deduced by ¹³C-¹H-COSY-¹J_{CH} [spin-spin coupling of carbon and hydrogen via one bond (${}^{1}J_{CH}$ 138.0 Hz) and ${}^{13}C$ -¹H-COSY-ⁿ J_{CH} [n = 2 and 3, spin-spin interaction of carbon and hydrogen via two $({}^{2}J_{CH})$ and three $({}^{3}J_{CH})$ bonds, optimized for "J_{CH} of 8.0 Hz]. IR spectra with KBr plates were obtained on a FT-IR Perkin Elmer 1600/1605 spectrometer. Silica gel 60 (70-230 mesh, Merck) and neutral alumine oxide (BDH Laboratory Supplies) were used for column chromatography and silica gel 60 F_{254} plates (Merck) for TLC.

Plant material

The roots of a specimen of *Brosimum gaudichaudii* Trécul., Moraceae family were collected in Araguari, Minas Gerais State, Brazil, in July, 1994, and compared with voucher specimen deposited at the Reserva Florestal da Companhia Vale do Rio Doce (CVRD), Espírito Santo State, Brazil.

Extraction and isolation

Air-dried and powdered root bark (1.89 kg) was successively extracted at room temperature with CH_2Cl_2 followed by MeOH.

The residue (24.6 g) obtained from the CH_2Cl_2 extract was submitted to CC (silica gel) eluted with a gradient of hexane/CH₂Cl₂/MeOH resulting in 12 frs. The fr. 5-8 (5.4 g) was rechromatographed to produce a mixture of psoralen (4), bergapten (5); fr. 9-12 (9.2 g) also was rechromatographed using a gradient of hexane/CH₂Cl₂/MeOH to produce 42 fractions Fr. 9 furnished 4 (42.0 mg); 10-2 yielded a mixture of 4 and 5 (4.5 mg) and 2',4',4trihydroxy-3',3-diprenylchalcone (14, 11.0 mg); 33-38 (2.3 g) afforded 1'-hydroxymarmesin (6, 30.0 mg) and a mixture of 1',2'-dehydromarmesin (11) and marmesin (10), after rechromatographed on CC (neutral alumine oxide) eluted with increasing polarity gradient of CH₂Cl₂-Et₂O/MeOH.

The MeOH extract (40.5 g) was partitioned using CH₂Cl₂/MeOH:H₂O (2:1) to furnish hydroalcoholic and dichloromethane solutions and a precipitate (4.1 g). This precipitate was chromatographed on CC (neutral aluminum oxide) eluted with CH₂Cl₂ and MeOH to afford a mixture of 4 and 5 (3.7 g) and 15 (6.0 mg). The residue obtained of the CH₂Cl₂ solution was partitioned using hexane/MeOH (1:1) yielding a hexane fraction and a precipitate (0.6 g, a)mixture of 4 and 5); the residue (4.0 g) obtained of the MeOH solution was submitted to CC (neutral alumine oxide) using CH₂Cl₂, Et₂O, EtOAc and MeOH to furnish 4 fractions. Fr. 1-3 (2.0 g) was rechromatographed on CC (silica gel) with a gradient of hexane/CH₂Cl₂/Et₂O/MeOH yielding 22 frs. Fr. 1-10 (1.3 g) was submitted to filtration on sephadex LH-20 eluted with MeOH/CHCl₂ (50-70 %) affording a mixture of 4 and 5 (32.0 mg) and 14 (10 mg); 15-20 (160.0 mg) also submitted to filtration on sephadex LH-20 eluted with MeOH followed by CC (silica gel) to yield a mixture of 10 and 11 (44.0 mg); 21-22 (160 mg) furnish 6. The fr. 4 was submitted a CC (silica gel) with gradient of CH2Cl2/(CH2)2O/MeOH to furnish 25 fractions Fr. 20-25 was washed with MeOH to afford $3-O-\beta$ glucopyranosilsitosterol (20.0 mg).

The hydroalcoholic fraction was partitioned with EtOAc/*n*-BuOH. The EtOAc fraction (660.0 mg) was chromatographed on CC (silica gel) with a gradient of CH₂Cl₂/(CH₃)₂O/MeOH to furnish a mixture of **4** and **5**. The residue (7.1 g) of the *n*-BuOH fraction was washed with MeOH to furnish 1'-hydroxy-3'-O- β -glucopyranosilmarmesin (**13**, 630.0 mg). The remaining solution was evaporated and the residue obtained (6.3 g) was submitted to partition with EtOAc-EtOH/H₂O (1:1). The EtOAc fraction furnished 3-O- β -glucopyranosilsitosterol (480.0 mg).

Air-dried and powdered root wood (2.2 kg) was extracted with MeOH at room temperature. The residue (19.8 g) obtained was partitioned with CH₂Cl₂/H₂O (1:1). The CH₂Cl₂ fraction (17.8 g) was submitted to on CC (silica gel) with a gradient of hexane/CH₂Cl₂/MeOH yielding 35 fractions Fr. 1-14 (1.9 g) showed rich in fatty acid, sitosterol and β -amyrin. The fractions 23-33 (4.4 g) were submitted to washing with acetone to furnish a residue (676.0 mg); before concentration, the remaining solution furnish a residue (1.7 g) that was submitted to filtration on sephadex LH-20 eluted with MeOH to furnish 25 fractions. Fr. 6-10 was chromatographed on CC (silica gel) to furnish 6 (87.0 mg), a mixture of **10**, **11** (5.0 mg) and 8-methoxymarmesin **12** (84.0 mg); 11-22 (1.3 g) was resubmitted to filtration on sephadex LH-20 eluted with MeOH to furnish a mixture of cinnamic acids derivatives **7**, **8** and **9** (40.0 mg). The residue (676.0 mg) was chromatographed on CC (silica gel) eluted with gradient $CH_2Cl_2/Et_2O/MeOH$ followed by filtration on sephadex LH-20 eluted with MeOH to furnish **6** (53.0 mg), **12** (61.0 mg) and $3-O-\beta$ -glucopyranosylsitosterol (80.0 mg).

Mixture of marmesin (10) and 1',2'-dehydromarmesin (11)

IR ν_{max} /cm⁻¹: 3478, 1707, 1628, 1570, 1482, 1446, 1399, 1369, 1266, 1180, 1134 (KBr); ¹H (300 MHz) and ¹³C NMR (75), in CDCl₃, in accordance with literature data;^{10,12} GC/EIMS *m*/*z* (int rel) **10**: EIMS *m*/*z* (int rel): 246 ([M⁺], 7.5), 228 ([M – H₂O], 2), 213 ([M – H₂O – Me⁻], 13), 188 ([M – Me₂C=O], 37), 187 ([M – Me₂C=O – H⁻] and/or ([M – Me₂C=O – H⁻] and/or ([M – Me₂C=O – H⁻], 75), 160 ([M – Me₂C=O – CO], 25), 159 ([M – Me₂C=O – H⁻ - CO] and/or ([M – Me₂C-OH - CO], 12), 59 ([Me₂C⁺=OH], 100)**11**: 244 ([M⁺⁺], 9), 229 ([M – Me⁻], 40), 187 (M – Me – CH₂=C=O], 12), 43 (Me-C⁺=O, 100).

3'-hydroxy-4'-O-b-glucopiranosylmarmesin (13)

Amorphous powder; Spectral data are in agreement with literature data.¹⁴

Peracetyl derivative 13

Natural product 13 (80.0 mg) was treated with Ac₂O (9.0 mL) and dry pyridine (1.0 mL) at room temperature. After the usual workup, the crude peracetyl derivative was chromatographed on a silica gel column eluting with increasing polarity of CH₂Cl₂/EtOAc/MeOH to furnish the acetate 13a (19.3 mg). Amorphous power, mp 113-118 °C; IR ν_{max} /cm⁻¹: 1750, 1630, 1574, 1487, 1438, 1374, 1227, 1126, 1041 (KBr); ¹H NMR (300 MHz, CDCl₂): δ_{μ} 6.27 (d, J 9.3 Hz, H-3), 7.63 (d, J 9.3 Hz, H-4), 7.57 (s, H-5), 6.84 (s, H-8), 6.27 (d, J 6.3 Hz, H-1'), 4.50 (d, J 6.3 Hz, H-2'), 1.53 (s, 3H-4'), 1.46 (s, 3H-5'), 4.89 (d, J7.8 Hz, H-1"), 4.98 (dd, J7.8, 9.6 Hz, H-2"), 5.28 (t, J9.6 Hz, H-3"), 5.03 (t, J 9.6 Hz, H-4"), 3.74 (m), 4.21 (dd, J 12.3, 5.7 Hz, H-6"a), 4.10 (dd, J 12.3, 2.7 Hz, H-6"b), 2.02 (s, Ac), 2.03 (s, Ac), 2.04 (s, Ac), 2.05 (s, Ac) and 2.06 (s, Ac); HBBD- and DEPT-¹³C NMR (75 MHz, CDCl₂): δ_{c} 160.58 (C-2), 113.04 (CH-3), 143.52 (CH-4), 126.43 (CH-5), 123.64 (C-6), 162.03 (C-7), 98.75 (CH-8), 156.88 (C-9), 113.47 (C-10), 71.20 (CH-1'), 90.35 (CH-2'), 78.81 (C-3'), 23.21 (CH₃-4'), 22.76 (CH₃-5'), 95.12 (CH-1"), 71.35 (CH-2"), 72.65 (CH-3"), 68.45 (CH-4"), 71.49 (CH-5"), 62.06 (CH₂-6"); EIMS m/z (int rel): 634 ([M]⁺⁺, 4), 574 ([M-AcOH, 5), 331 ([glucopyranosylAc₄]⁺,15), 287 ([M-OglucopyranosylAc₄, 8), 271 ([glucopyranosylAc₄-AcOH]⁺, 5), 245 ([M-Me₂C- OglucopyranosylAc₄,15), 227 ([M-AcOH-OglucopyranosylAc₄, 100).

8-Methoxymarmesin (12)

¹H (300 MHz) and ¹³C NMR (75 MHz), in CDCl₃, in accordance with literature data;¹³ EIMS m/z (int rel) 276 ([M]⁺,100), 258 ([M-H₂O], 23), 246 ([M-CH₂O], 25), 243 ([M-H₂O-Me], 50), 229 ([M-CH₂O-HO⁻], 20), 218 ([M-Me₂C=O], 90), 217 ([M-Me₂C=O - H] and/or [M-Me₂C - OH], 85), 190 ([M-Me₂C=O - CO], 83), 175 ([M-Me₂C=O - Me⁻], 22), 59 ([Me₂C⁺=OH], 68).

2',4',4-trihydroxy-3',3-diprenylchalcone (14)

Oil; IR ν_{max} /cm⁻¹: 3396, 1626, 1491, 1369, 1245, 1104, 978 (KBr); ¹H (300 MHz) and ¹³C (75 MHz) NMR, in CDCl₃, in accordance with literature data.¹⁵

Mixture of the cinnamic acid derivatives 3-(7-methoxy-2,2-dimethyl-2H-6-chromenyl-6)-(E)-propenoic acid (7), 3-(7-methoxy-2,2-dimethyl-2H-6-chromenyl-6)propanoic acid (8) and 3-(6-methoxybenzo[b]furan-5-yl)propanoic acid (9). Amorphous solid; ¹H and ¹³C NMR: Tables 1 and 2. EIMS: Scheme 1.

3β -O- β -D-glucopyranosylsitosterol

¹H and ¹³C NMR spectral data, including of the peracetyl derivative, in agreement with literature data.¹⁶ The natural product 3β -O- β -D-glucopyranosylsitosterol (16.0 mg) was treated with Ac₂O (9.0 mL) and dry pyridine (1.0 mL) at room temperature. After the usual workup, the crude peracetyl derivative was chromatographed on a silica gel column eluting with increasing polarity gradient of CH₂Cl₂/EtOAc/MeOH to furnish the peracetyl derivative (19.3 mg), amorphous powder, mp 160-164 °C.

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References

- Vilegas, J. H. Y.; Lanças, F. M.; Vilegas, W.; Pozetti, G. L.; *Phytochem. Anal.* **1993**, *4*, 230.
- Vilegas, W.; Pozetti, G. L.; Vilegas, J. H.Y.; J. Nat. Prod. 1993, 56, 416.
- Vieira, I. J. C.; Mathias, L.; Monteiro, V. F. F.; Braz-Filho R.; Rodrigues-Filho, E.; *Nat. Prod. Lett.* **1999**, *13*, 47.
- Gupta, A. K.; Anderson, T. F.; J. Am. Acad. Dermat. Part I 1987, 17, 703.
- Parrish, J. A.; Fitzpatrick, T. B.; Tanenbaum, L.; Pathak, M. A.; New England J. Med. 1974, 291, 1207.
- Braz-Filho, R.; Magalhães, A. F.; Gottlieb, O. R.; An. Acad. Brasil. Ciênc. 1970, 42 (Supl.), 139.
- Braz-Filho, R.; Magalhães, A. F.; Gottlieb, O. R.; An. Acad. Brasil. Ciênc. 1971, 43, 585.
- Braz-Filho, R.; Magalhães, A. F.; Gottlieb, O. R.; *Phytochemistry* 1972, 11, 3307.
- Dewick, P.; Medicinal Natural Products A Biosynthetic Approach; John Wiley: New York, 1997, p. 134.

- Breitmaier, E.; Voelter, W.; Carbon-13 NMR Spectroscopy: High-Resolution Methods and Applications in Organic Chemistry and Biochemistry, 3rd ed., VCH: Weinheim, 1987, p. 447.
- Grande, M.; Aguado, M. T.; Mancheño, B.; Piera, F.; *Phy-tochemistry* **1986**, *25*, 505.
- Quader, M. A.; El Turbi, J. A.; Armstrong, J. A.; Gray, A. I.; Waterman, P. G.; *Phytochemistry* **1992**, *31*, 3083.
- 13. Sharma, B. R.; Rattan, R. K.; Sharma P.; *Phytochemistry* **1980**, *19*, 1556.
- 14. Lemmich, J.; Havelund, S.; Thastrup, O.; *Phytochemistry* **1983**, 22, 553.
- Christensen, S. B.; Ming, C.; Anderson, L.; Hjorne, U.; Olsen, C. E.; Cornett, C.; Theander, T. G.; Kharazmi, A.; *Planta Med.* 1994, 60, 121.
- Pessoa, O. D. L.; De Lemos, T. L. G.; De Carvalho, M. G.; Braz-Filho, R.; *Phytochemistry* **1995**, 40, 1777.
- Naidu, M.V.; Rao, G.S.K.; *Indian J. Chem.*, Sect. B 1980, 19B, 313.
- Chen, C–C.; Huang, Y–L.; Huang, F–I.; Wang, C-W.; Ou, J– C.; J. Nat. Prod. 2001, 64, 990.
- Stanjek, V.; Piel, J.; Boland, W.; *Phytochemistry* **1999**, *50*, 1141.
- 20. Brown, S.A.; Steck, W.; Phytochemistry 1973, 12, 1315.

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