Molybdenum Oxide Coatings Deposited on Plasma Nitrided Surfaces

Renan Matos Monção^a, Miguel Rubira Danelon^b, Larissa Solano de Almeida^c,

Luciana Sgarbi Rossino^{5,c}, Fernanda Roberta Marciano^d, Thércio Henrique de Carvalho Costa^e* 💿,

Michelle Cequeira Feitor^e, Ruben Maribondo do Nascimento^e,

Rômulo Ribeiro Magalhães de Sousa^a 💿

^aUniversidade Federal do Piauí, Departamento de Engenharia Mecânica, Teresina, PI, Brasil. ^bFaculdade de Tecnologia de Sorocaba (FATEC), Sorocaba, SP, Brasil. ^cUniversidade Federal de São Carlos (UFSCar), Sorocaba, SP, Brasil. ^dUniversidade Federal do Piauí, Departmento de Física, Teresina, PI, Brasil. ^eUniversidade Federal do Rio Grande do Norte, Programa de Pós-graduação em Engenharia Mecânica, Natal, RN, Brasil.

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Molybdenum trioxide (MoO₃) thin films are generally deposit on metallic surfaces to increase their tribological performance. In this manuscript, MoO_x coatings were for the first time deposited on plasma nitrided surfaces. Vickers microhardness tests, X-ray diffractometry, and micro abrasive wear tests characterized the samples. It was investigated the effect of temperature in hardness and tribological properties of austenitic stainless steel surfaces. When the temperature of plasma nitriding increases, the coating thickness and the nitrogen diffusion on the treated surface also increases, favoring the hardness of the coatings. After MoO_x deposition, the predominant MoO₃ phase with monoclinic and orthorhombic unit cells was observed. MoO_x coatings after plasma nitriding demonstrated the highest wear resistance.

Keywords: molybdenum oxide, coatings, plasma nitriding; micro abrasive wear.

1. Introduction

Molybdenum trioxide (MoO₃) thin films exhibit highly important optical and electronic properties, permitting their use as gas sensors, lithium-ion batteries, electrochromic device applications, and so on¹. Orthorhombic (α -MoO₃) and monoclinic (β -MoO₃) are the major phases; α -MoO₃ is the thermodynamically stable and β -MoO₃ the metastable². Sputtering, chemical vapor deposition, and pulsed laser deposition has been the main techniques for the synthesis of these films. Moreover, heated substrates can extent the crystallinity of the as-deposited thin films³. To achieve the ideal stoichiometry, the annealing temperature is above 400 °C in oxygen-bearing atmospheres⁴. However, the effect of substrate temperature on the stoichiometry and phase purity needs to be discussed in more detail¹.

Yang⁵ demonstrated the importance of retaining a stable and sufficiently thick MoO_3 surface layer to maintain the beneficial effect of Mo on the tribological performance of the coatings. Mo-containing nitride coatings (MoTiN and MoAlTiN) deposited by cathodic arc evaporation were investigated in different conditions [5]. The coatings demonstrated lower friction coefficients and high wear resistance (against WC-6Co and Si₃N₄ balls) compared to their corresponding Mo-free TiN, AlTiN monolithic coatings. However, these coatings showed less improvements in tribological properties when Al₂O₃ ball was used⁵.

Besides the wide variety of deposition processes and techniques, MoO_3 is generally deposited directly over the metal surface³. The deposition of MoO_3 on plasma nitrided surfaces has never been performed. Plasma nitriding has been extensively used to improve the hardness, wear-resistance, and the tribological stability of metallic surfaces⁶. On temperatures below 450 °C, plasma nitriding is established for conventional coarse-grained austenitic stainless steels and improves corrosion and wear resistance⁷.

In this manuscript, MoO_x coatings were for the first time deposited on plasma nitrided surfaces. It was investigated the effect of temperature in hardness and tribological properties of austenitic stainless steel surfaces.

2. Experimental Procedure

AISI 316 stainless steel substrates (10 mm x 10 mm x 6 mm) were polished to a mirror-like surface. A molybdenum cage was used for MoO_x deposition. This cage was first polished before deposition and, after, it was sonicated in acetone and dried in hot air jet.

A pulsed source reactor (SDS Plasma) with a maximum voltage of 800 V, using a cylindrical chamber (400 mm diameter and 700 mm height) was used for nitriding and

^{*}e-mail: thercioc@hotmail.com

 MoO_x deposition. The reactor set up was presented in a previous manuscript⁸. For the conventional plasma nitriding, the samples were placed on an insulating alumina disk, without any cage. Using this setup, the plasma is formed directly on the surface of the samples. For MoO_x deposition, the molybdenum cage used is in the cathodic potential, producing a hollow cathode effect⁹.

Before plasma nitriding and MoO_x deposition, the samples were pre-sputtered in H₂: Ar (1:1) atmosphere, pressure 150 Pa, and temperature 350 °C for 1 h to eliminate contaminants such as oxides and/or grease/fat adsorbed on the surface. The samples were divided into eight groups, and the parameters for plasma nitriding and MoO_3 deposition are described in Table 1.

Vickers microhardness tests (ASTM E-384) were performed (Insize microhardness tester, model ISH-TDV 1000) with a diamond indenter using 50 gf and loading of 15 s.

The phase compositions were identified using a highresolution Malvern Panalytical Empyrean X-ray diffractometer (XRD, CoK α radiation, $\lambda = 0.1789$ nm) using Bragg-Brentano reflection geometry. Diffraction spectra were taken from 10° to 105° with a step of 0.013°.

The microstructural analyzes were performed according to ASTM E 395-00 norm. Transverse sections were cut, polished to a mirror-like surface, and attacked with aqua regia (15 mL nitric acid + 20 mL hydrochloric acid + 30 mL distilled water). The BEL Photonics Technical (MTM-1A) optical microscope performed the cross-sectional images. The semi-quantitative analyses were evaluated using X-ray dispersive energy spectroscopy (EDS) on the Hitachi TM-3000 bench-scanning microscope to determine the elements over the sample (point position and line profile).

The micro-abrasive wear tests (MAWTs) were carried out using a fixed ball microwear device. The 52100 steel ball with 25.4 mm of diameter was used. Each MAWT was performed using 40 Hz of frequency and a normal load of 8.0 N for 10 min, without any abrasive or liquid lubricant. The friction coefficient was determined during the test using a load cell¹⁰. After tests, all the samples were analyzed by optical microscope Leica, model MC170 HD, with camera and software analysis to measure the diameters and crater radio. The wear volume (V) was calculated according to Rutherford and Hutchings's methodology¹¹. All MAWTs were carried out in duplicate to ensure good reproducibility.

3. Results and Discussion

Figure 1 shows the cross-sectional optical microscopy from the samples. It's possible to see the microstructures of each treated surface (regular and homogeneous topography). The thickness of each coating is shown in Table 2.

When the temperature increases, the coating thickness also increases¹². In CPN400 sample (Figure 1a), it's seen

Table 1. Main parameters used for plasma nitriding and MoO_v deposition.

Samples	Conventional plasma nitriding (Gas flow: 50% H ₂ + 25% N ₂ + 25% Ar)			MoO_x deposition (Gas flow: 50% H ₂ + 50% O ₂)		
	Pressure	Time	Temperature	Pressure	Time	Temperature
AISI 316	-	-	-	-	-	-
CPN400	300 Pa	4 h	400 °C	-	-	-
CPN450	300 Pa	4 h	450 °C	-	-	-
CPN500	300 Pa	4 h	500 °C	-	-	-
MoO _x	-	-	-	100 Pa	4 h	400 °C
CPN400+MoO _x	300 Pa	4 h	400 °C	100 Pa	4 h	400 °C
CPN450+MoO _x	300 Pa	4 h	450 °C	100 Pa	4 h	400 °C
CPN500+MoO _x	300 Pa	4 h	500 °C	100 Pa	4 h	400 °C



Figure 1. Cross-sectional optical microscopy from (a) CPN400, (b) CPN450, (c) CPN500, (d) MoO_x , (e) CPN400+ MoO_x , (f) CPN450+ MoO_x and (g) CPN500+ MoO_x samples.

the formation of a thin and homogeneous layer, with white color. The expanded austenite and iron nitride alloyed layer formed is dense and has the precipitation of free white surface layers¹³. When the temperature increases (CPN450 sample, Figure 1b), starts the precipitation of a darkness phase (chrome nitride) in the formed layer, which becomes predominant at 500 °C (CPN500 sample, Figure 1c).

At lower temperatures, there's the formation of a thinner nitrided layer with nitrogen saturation on the surface, whose wt%N drops abruptly along with the thickness of the formed layer (Table 3). As the temperature increases, the diffusion coefficient increases, favoring nitrogen diffusion in stainless steel, which increases the layer thickness and wt%N along with the treated layer (Table 3). Thus, high temperatures, such as 500 °C, provide lower wt%N on the surface of the layer, which remains at greater depths, resulting in a thicker and harder layer and the depth of the formed layer. This directly reflects the hardness values of the formed layer (Figure 2). With increasing temperature and greater nitrogen diffusion on the treated surface, in addition to the increase in thickness, the formation of hard phases such as chromium nitride is favored, increasing the hardness of the layer¹². Table 4 shows the surface microhardness values

Table 2. Minimum, maximum and mean thickness values from the coatings. Each mean corresponds to the average of 5 different measurements.

Samples	Minimum (µm)	Maximum (µm)	Mean (µm)
CPN400	11.86	13.79	12.52 ± 0.84
CPN450	16.54	19.58	17.81 + 1.21
CPN500	50.46	51.84	51.23 ± 0.71
MoO ₃	15.44	16.13	15.78 ± 0.25
CPN400+MoO _x	13.51	14.89	14.12 + 0.60
CPN450+MoO _x	33.92	35.57	34.69 ± 0.66
CPN500+MoO _x	63.14	65.44	64.57 ± 0.84

Table 3. Quantity of nitrogen (by wt%) at the interface points of the nitride samples. The position of the point is determined from the surface edge (point 1) towards the substrate with 0.006 coordinate advances.

Points	CPN400	CPN450	CPN500
Point 1	4.6	7.3	6.4
Point 2	0.7	0.4	4.4
Point 3	0.0	0.0	5.6
Point 4	0.0	0.0	5.2

Table 4. Surface microhardness values of AISI 316 stainless steel after plasma nitriding and MoO₃ deposition.

Samples	Vickers Hardness (HV)
AISI 316	230.0
CPN400	1421.4
CPN450	1514.8
CPN500	1812.3
MoO ₃	868.0
CPN400+MoO _x	995.3
CPN450+MoO _x	865.4
CPN500+MoO _x	816.6

according to the treatment received. Diamond indenter generally induces a massive plastic deformation, causing fracture¹⁴. Surface hardness up to 1800 HV was achieved from plasma nitriding. The surface hardness is related to treatment parameters. Increasing temperature increases surface hardness¹³.

Figure 3 shows the line profile of the CPN400+MoO_x sample using EDS. The microstructural analysis determined the elements over the sample, showing the duplex treatment (nitriding + MoO_x deposition).

Figure 4 shows the XRD diffractograms of the samples according to the treatment received. The formation of chromium nitride in the coating appeared more intensely at 450 °C and 500 °C. High temperatures favor the formation of chromium nitride in the formed layer, whose phase is hard and fragile, presenting low resistance to wear and corrosion^{12, 15}. It is expected that the formation of precipitates in the diffusion zone increases surface hardness and leads to superior fatigue resistance¹⁶. Ion nitriding causes surface hardening due to diffusion of Cr_N and Fe_N into ferrous and non-ferrous metallic surfaces, promoting a diffuse zone and oxide and compound layers¹⁷. After MoO_x deposition (Figure 4b), the predominant formation of the MoO₂ phase with the monoclinic and orthorhombic unit cell was observed, with Mo₈O₂₃ and the combination of molybdenum and substrate, forming the Fe_{9.7}Mo_{0.3} phase.

Figure 5 shows the results of the micro-abrasive wear tests on the samples. There is an increase in wear resistance for all



Figure 2. Microhardness values as a function of the layer thickness and the treatment received.



Figure 3. Line profile of CPN400+MoO_x sample by EDS.



Figure 4. X-ray diffractograms after (a) nitriding and (b) nitriding + MoO_x deposition.



Figure 5. Wear volume calculated after micro-abrasive tests.



Figure 6. Friction coefficients of the samples.



Figure 7. Wear crater with EDS analysis of (a-b) stainless steel, (c-d) CPN400, (e-f) MoO₂ and (g-h) CPN400 + MoO₂

treated materials compared to stainless steel, demonstrating the effectiveness of the treatment performed. Also, it is observed that the wear resistance is not directly related to the hardness of the formed layer^{18, 19}. Although the MoO_x sample did not have the highest hardness value among the treated materials, this was the condition of the highest wear resistance. The wear volume was almost the same for all nitriding + MoO_x deposition samples. The worst wear resistance was on the CPN400 sample, due to the small layer thickness formed.

Figure 6 shows that the friction coefficients of the treated samples are similar. The increase in the stainless steel friction coefficient at the end of the test is justified by the greater wear volume suffered by this material, which promotes the formation of oxidized wear debris over the crater as illustrated in Figure 7(a) and confirmed in Figure 7(b), increasing the material removed and friction coefficient throughout the test. As observed in Figure 7(c-d), the formation of the wear debris of the nitrided material is lower than all treated materials, justifying your lower friction coefficient. It is possible to observe, in Figure 7(e-f) and (g-h) significant wear oxidized debris formation in the materials with MoO_x films that remain around the crater.

4. Conclusion

In this manuscript, MO_x coatings were for the first time deposited on plasma nitrided surfaces. When the temperature of the plasma nitriding increases, the coating thickness also increases. The temperature also increases the nitrogen diffusion on the treated surface, favoring the hardness of the coatings. After MO_x deposition, the predominant MO_3 phase with monoclinic and orthorhombic unit cells was observed. MO_x coatings after plasma nitriding demonstrated the highest wear resistance. This manuscript showed that the combined treatments (nitriding + MoO_x deposition) increased the surface hardness and the wear resistance of the metallic substrates, amplifying their use as tribological protective coatings.

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