Investigation of Toxic Factors Affecting Cells of Rat Brains Exposed to 3-methylcatechol

George Emílio Sampaio Barreto\textsuperscript{1,2}, Elineusa Silva de Oliveira\textsuperscript{1}, Flávio Augusto Guerreiro Aragão de Villar\textsuperscript{1}, Gleide Souza dos Santos\textsuperscript{1}, Rute Maria Ferreira Lima\textsuperscript{1}, Silvia Lima Costa\textsuperscript{1}, Maria de Fátima Dias Costa\textsuperscript{1}, Eryvaldo Sócrates Tabosa Egito\textsuperscript{3} and Ramon Santos El-Bachá\textsuperscript{1*}

\textsuperscript{1}Laboratório de Neuroquímica e Biologia Celular; Departamento de Biofunção; Instituto de Ciências da Saúde; Universidade Federal da Bahia; Campus do Canela; 40110-100; Salvador - Bahia - Brasil. \textsuperscript{2}Programa de Pós-Graduação em Ciências da Saúde; Universidade Federal do Rio Grande do Norte - UFRN; Natal - Rio Grande do Norte - Brasil. \textsuperscript{3}Laboratório de Sistemas Dispersos; Departamento de Farmácia, UFRN, Natal - Rio Grande do Norte – Brasil.

ABSTRACT

The aim of this work was to study the effects of 3MC on the peroxidation of biomolecules in nuclear fractions and nonsynaptic mitochondrial respiration in organelles obtained from rat brains. The cytotoxicity towards rat primary astrocytes in vitro was also tested. 3MC at 1mM oxidized consuming oxygen at a rate of $1.98 \pm 0.19 \mu M/min$ and formed reactive quinones. At the same concentration, 3MC induced peroxidation of biomolecules in nuclear fractions obtained from rat brain homogenates and inhibited state 2 FADH\textsubscript{2}-linked respiration in nonsynaptic mitochondria. Furthermore, 3MC oxidized in the culture medium, leading to the formation of quinones. This toluene metabolite was cytotoxic to rat primary astrocytes. The concentration that killed 50% of cells after 72 h was 107 \mu M. The results of the study indicated a direct relationship between cytotoxicity and 3MC oxidation.

Key words: Astrocytes, brain, cytotoxicity, 3-methylcatechol, peroxidation, respiration

INTRODUCTION

Inert organic solvents such as benzene, toluene, ethylbenzene and xylene (BTEX) have been used by the chemical industry to prepare solid catalysts (Tashino et al., 2005), in polymerization reactions (Shikuma et al., 2005), to dissolve photosensitive compositions (Ishizuka and Kojima, 2005), and to synthesize pesticidal (Mori, 2005) and pharmaceutical compositions (Watanabe et al., 2005), among others. BTEX represent an estimated 35\% of fuel oils. Moreover, benzene and toluene are also gas phase tobacco smoke constituents (Xue et al., 2005). Since these compounds are recognized as hazardous fuels and solvents, methods for determination of aromatic hydrocarbons have been developed for toxicity assessment and confirmation of environmental contamination (Dijike, 2005). Inhalation of BTEX poses a health concern due to high volumes of production and release into the atmosphere. Biological transformation and volatilization are the major removal mechanisms used in cases of toluene contamination of soils and groundwater. Toluene is a substrate for wild or modified toluene dioxygenases expressed in prokaryotic cells.

\* Author for correspondence
leading to the formation of 3- and 4-methylcatechols (Arnold et al., 2005). In rats and humans, the hepatic clearance of toluene involves cytochrome P450 2E1 (Béliveau et al., 2005). Apparently the acute neurotoxicity of toluene results from specific receptor-mediated interactions between this solvent and ion channel proteins (Bushnell et al., 2005). Although the toxicity of toluene has been well established, the cytotoxicity of its metabolites has not yet been studied extensively. 3-Methylcatechol (3MC) is a minor metabolite of toluene in the human, but studies have reported that 3MC is cytotoxic. Inhibition of cell growth and a dose-related decrease in cell viability and cell protein content have been reported in HeLa cells exposed to 3MC (Shen, 1998). Furthermore, 3MC induced oxidative DNA damage in rat testis (Nakai et al., 2003).

The effects of 3MC on the peroxidation of biomolecules in nuclear fractions obtained from rat brains were examined in order to clarify whether this compound could be toxic to the central nervous system. The effects of this molecule on FADH$_2$-linked basal respiration in nonsynaptic mitochondria were also evaluated in order to determine whether 3MC would be capable of interfering with brain nonsynaptic mitochondrial function. Furthermore, 3MC-induced cytotoxicity to rat primary astrocytes in vitro was investigated using a test that measured the activity of mitochondrial succinate dehydrogenase.

**MATERIALS AND METHODS**

**Materials**

All reagents used in this study were of analytical grade. 3-Methylcatechol was obtained from Aldrich, São Paulo, São Paulo, and solutions were freshly prepared in 0.01 M HCl in order to inhibit its autoxidation. HCl was obtained from Nuclear, Diadema, São Paulo, and phosphate salts from Reagen, Rio de Janeiro, Rio de Janeiro. All the solutions were prepared in type I clinical laboratory reagent water.

**Measurement of 3-methylcatechol autoxidation in a cell-free system**

Neutral aqueous solutions of 3MC, like other phenolic compounds, undergo spontaneous oxidative decomposition (Rosenau et al., 2005). Autoxidation constitutes a multi-step reaction process resulting in reactive oxygen species (ROS) and the formation of quinone derivatives. 3MC has a maximum light-absorption at 273 nm in aqueous solution at pH 3. No autoxidation has been observed in acidic conditions. However, samples autoxidize in neutral conditions (pH 7.4), and a new peak appeared at 320 nm and increased with time. In the present study, the autoxidation rates of 1 mM 3MC were measured at 320 nm in 1 mM HCl, 50 mM phosphate buffer (pH 7.4) in the absence or in the presence of 17 U superoxide dismutase (SOD; EC 1.15.1.1). 3MC oxidation was also measured by means of a Clark-type oxygen electrode connected to an oxygen monitoring system.

**Animals**

One-day-old and adult Wistar rats weighing 250-350 g were obtained from the Department of Physiology of the Health Sciences Institute, Federal University of Bahia (Salvador, BA, Brazil). All the experimental protocols were conducted according to the regulations established by the Ethical Committee for animal experimentation of the Federal University of Bahia.

**Isolation of nuclear fraction**

Nuclear fraction was isolated from the brains of adult rats. To isolate nuclei, three brains were homogenized on ice using a ground-glass pestle in 25 ml of 0.32 M sucrose, 1 mM ethylenediamine tetraacetic acid (EDTA), and 50 mM phosphate buffer (pH 7.4). Cells were lysed with 8 up-and-down pestle strokes. Tissue homogenate was centrifuged at 400 g for 10 minutes at 4 °C and the supernatant was stocked at 4 °C. The pellet was resuspended in the same buffer and centrifuged again under the same conditions. Supernatants were combined and centrifuged at 1,500 g for 10 minutes at 4 °C. The nuclei were resuspended in 0.2 M EDTA, 20% (v/v) glycerol, 0.1 M phosphate buffer (pH 7.4). Protein concentrations were determined according to the procedures established by Lowry et al. (Lowry et al., 1951).

**Peroxidation assay**

Nuclear fractions (1 mg protein/ml) were incubated with 1 mM 3MC in 1 mM HCl, and 50 mM phosphate buffer (pH 7.4) at room temperature for 17 h. Negative controls were
incubated without 3MC, and positive controls were incubated with 5 μM Fe₂SO₄ and 0.5 mM ascorbate. The formation of thiobarbituric acid (TBA)-reactive substances (TBARS) was used to assess peroxidation (Slater, 1984; Dobliniski et al., 2003). A great variety of oxidized substances form pink TBA complexes, such as malonaldehyde, oxidized sugars and amino acids (Esterbauer and Cheeseman, 1990). All these compounds can be found in nuclear fractions submitted to oxidative stress. After the addition of 2 ml of 0.67% (w/v) TBA, 15% (w/v) trichloroacetic acid in 0.25 M HCl, samples were boiled in capped tubes for 10 minutes and cooled in fresh water. The reaction mixture was centrifuged at 1,500 g for 10 minutes. The optical density (OD) was measured at 535 nm against a blank without TBA.

Nonsynaptic mitochondrial isolation

Nonsynaptic mitochondria were isolated from the brains of adult rats by differential centrifugation. Brains were homogenized using a ground-glass pestle in 0.2 M mannitol, 75 mM sucrose, 1 mM ethylene glycol-bis[β-aminoethyl ether]-N,N,N',N'-tetraacetic acid (EGTA), 0.05% (w/v) bovine serum albumin (BSA), and 1 mM Tris buffer (pH 7.4). Homogenates were incubated in the absence (controls) or in the presence of 1 mM 3-methylcatechol for 0, 15 or 30 minutes at 37 °C prior to centrifugation. Two independent experiments were performed. Whole cells, nuclei, cytoskeletons and plasma membranes were removed by centrifugation at 550 g for 10 minutes at 4 °C, followed by centrifugation of the supernatant at 7,100 g for 10 minutes at 4 °C. The mitochondrial pellet was resuspended in the same buffer and washed twice at 6,400 g for 10 minutes at 4 °C in order to eliminate the excess 3MC, which could have interfered with oxygen uptake measurements. Finally, nonsynaptic mitochondria were resuspended in 1 ml of the same isolation buffer. Protein concentration was measured in the same way as that described for nuclear fractions.

Oxygen uptake

Oxygen consumption was measured at 37 °C in a closed chamber containing a Clark type oxygen electrode connected to a monitor. Isolated nonsynaptic mitochondria were suspended in 3 ml of 10 mM KCl, 0.2 mM EDTA, 0.25 M mannitol, 0.025% (w/v) BSA, 10 mM Tris, and 5 mM phosphate buffer (pH 7.4) at a final concentration of 0.4 mg protein/ml. Succinate was added to a final concentration of 10 mM to induce nonsynaptic mitochondrial basal respiration (state 2). Oxygen uptake in resting conditions was monitored for 12 minutes. Four assays were analyzed for each nonsynaptic mitochondrial fraction.

Cell cultures

Cultures of isolated cortical astrocytes were prepared as previously described (Booher and Sensenbrenner, 1972; Silva et al., 2007). Briefly, astrocytes were prepared from the neocortex of 1-day-old Wistar rat brains, maintained in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal calf serum, 2 mM L-glutamine, penicillin (100 IU/ml), and streptomycin (100 μg/ml). Cells were grown in a humidified atmosphere of 5% CO₂ and 95% air at 37 °C, and the culture medium was replaced three times a week. Cultures were used after they had attained confluence. At the time of the experiment, cells were seeded at a density of 8 × 10⁴ cells/cm² in a 96-well plate and treated 72 hours later.

Treatment of cells with 3MC

Sterile dilutions of 3MC were made in 0.01 M HCl. Logarithm dilutions in the range of 60 to 3,000 μM were used to examine the cytotoxic effect of 3MC on astrocytes and to determine the EC₅₀, which is the concentration of 3MC that kills 50% of cells. Eight replicates for each dose were used per 96-well plate. Cultures were exposed to 3MC for 72 hours.

Cell viability and 3MC autoxidation

Cell viability was assessed using the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay as described previously (Hansen et al., 1989; Andrade et al., 2005). In brief, this colorimetric assay measures the reduction of MTT by mitochondrial succinate dehydrogenase in intact cells. Because reduction of MTT can only occur in metabolically active cells, the level of activity is a measure of cell viability. The 3MC oxidation in the medium was monitored spectrophotometrically at 405 nm based on quinones and melanin-like pigment formation (Lai and Yu, 1997).
Statistical Analysis
Results were expressed as the mean ± SD for each group and Student’s t-test was used to evaluate statistical differences. Results were expressed as median, 25th, and 75th percentiles whenever the normality test failed. In this case, the Mann-Whitney U test was used to assess statistical differences. Differences were considered to be statistically significant at P < 0.05. A nonlinear regression was performed to fit concentration-response curves. Statistical comparisons between many groups were performed by analysis of variance (ANOVA) followed by the post hoc Student-Newman-Keuls test. Data were analyzed using the Kruskal-Wallis non-parametric ANOVA with the post hoc Dunnett’s method when the normality test failed.

RESULTS
The oxygen consumption during autoxidation of 1 mM 3MC was estimated quantitatively. The rate of oxygen consumption was 1.98 ± 0.19 µM.min⁻¹ (mean ± SD, n = 15). Fig. 1A showed that the addition of 17 U SOD significantly inhibited oxygen consumption by 56%. The formation of quinones during the autoxidation of 3MC increased the OD at 320 nm. The rate of increase in OD was 0.0021 ± 0.0002 min⁻¹ (mean ± SD, n = 5). The rate of quinone formation decreased significantly by 16% in the presence of 17 U SOD (Fig. 1B). These data suggested that 3MC spontaneously reacted with oxygen in physiological conditions producing superoxide and reactive quinones. Hence, the possibility has to be considered that exposure of cells to 3MC could produce oxidative damage.

Figure 1 – (A) Oxygen consumption during autoxidation of 1 mM 3-methylcatechol in 1 mM HCl, 50 mM phosphate buffer (pH 7.4) at 37 °C. 3MC was tested alone (3MC, n = 15) or in the presence of 17 U superoxide dismutase (+SOD, n = 3). Each measurement refers to the mean ± SD, (***), p < 0.0001. (B) Formation of quinones during autoxidation of 1 mM 3MC in 1 mM HCl, 50 mM phosphate buffer (pH 7.4). 3MC was tested alone (3MC, n = 5) or in the presence of 17 U superoxide dismutase (+SOD, n = 4). Each value represents the mean ± SD, (*), p < 0.05.
The effects of 3MC on the peroxidation of biomolecules found in nuclear fractions of rat brain cells were evaluated. The formation of TBARS increased significantly compared to the spontaneous peroxidation in negative controls (58.4%, P < 0.001; Fig. 2A). The peroxidation in positive controls treated with 5 µM FeSO₄ and 500 µM ascorbate for 17 h increased by 153.2% compared to negative control values (P < 0.02; Fig. 2B). These data showed that a 17 h exposure of nuclear fractions to 1 mM 3MC led to an oxidative stress.

Mitochondrial Complex II receives FADH₂ directly from succinate dehydrogenase, and is, therefore, mainly dependent on the mitochondrial tricarboxylic acid cycle. In this study, succinate was used to induce nonsynaptic mitochondrial basal respiration (state 2). The rate of oxygen consumption in nonsynaptic mitochondria isolated from rat brain homogenates during state 2 FADH₂-linked respiration was 2.5 ± 0.3 nmol.min⁻¹.mg⁻¹ protein⁻¹. This rate decreased by 40% when nonsynaptic mitochondria were isolated after the addition of 3MC (1 mM) to brain homogenates immediately prior to centrifugation (Fig. 3). In another group, nonsynaptic mitochondria were isolated from brain homogenates after incubation at 37 °C for 15 minutes. The rate of oxygen consumption during state 2 FADH₂-linked respiration in this group was 3.3 ± 0.2 nmol.min⁻¹.mg⁻¹ protein⁻¹. This rate decreased by 51% when nonsynaptic
mitochondrial fractions were obtained after the addition of 3MC (1 mM) to brain homogenates and incubated at 37 °C for 15 minutes prior to centrifugation. Under the same resting conditions, the mean rate of FADH$_2$-linked oxygen consumption was 2.8 ± 0.1 nmol.min$^{-1}$mg protein$^{-1}$ when nonsynaptic mitochondrial fractions were obtained after an incubation of brain homogenates with 1 mM 3MC at 37 °C for 30 minutes. Thus, nonsynaptic mitochondrial basal respiration was inhibited by 51% when 1 mM 3MC was added to brain homogenates and incubated at 37 °C for 30 minutes prior to isolation of nonsynaptic mitochondria. These data suggest that cellular oxidative function may be impaired in the presence of 3MC.

3MC was probably cytotoxic towards cells from the central nervous system since peroxidation of biomolecules and inhibition of state 2 FADH$_2$-linked nonsynaptic mitochondrial respiration were induced by this compound in vitro. The formation of quinones and cell viability was then assayed in cultures of isolated primary astrocytes to determine the cytotoxicity of 3MC. Cell viability was measured by the use of MTT, which was a substrate for mitochondrial succinate dehydrogenase in intact cells. 3-Methylcatechol oxidation led to a concentration-dependent formation of quinones in the medium of rat astrocyte cultures treated with this molecule for 72 h (Fig. 4). Optical densities measured at 405 nm in the medium of control cells that were not treated with 3MC were significantly different from all other groups, 0.012 ± 0.005 (mean ± SD; n = 7). The formation of quinones was well-fitted (R$^2$ = 0.981) to the equation [1]:

$$\text{OD} = 0.07 + \{0.10/[1 + 10^{3.34 - 2.33 \log [3MC]}]\} \quad [1]$$

in which OD corresponds to optical density at 405 nm and [3MC] is the 3MC concentration. When the level of quinones increased with the concentration of catechol, an induction in cell death was detected (Fig. 4). Control cells reduced MTT and the median OD at 560 nm calculated for this group (median, 0.500; 25th percentile, 0.480; 75th percentile, 0.517; n = 7) was considered as 100% viability. The cell viability of groups treated with concentrations of 3MC > 100 µM differed significantly from that of the controls. The EC$_{50}$ of 3MC on rat astrocytes corresponded to 107 µM after 72 hours. Equation [2] represents the non-linear regression calculated from these data (R$^2$ = 0.9962):

$$V = 3.33 + \{66.67 / [1 + 10^{-46.40 + 22.70 \log [3MC]}]\} \quad [2]$$

in which V is the cell viability normalized to values measured under control conditions and [3MC] corresponds to 3MC concentration.
DISCUSSION

Deliberate or involuntary inhalation of organic solvents causes toxic effects and behavioral changes. Thinner is a neurotoxic mixture containing 60-70% toluene, which has been shown to induce an increase in lipid peroxidation in the hippocampus, cerebellum and cortex of rats chronically exposed to this solvent (Baydas et al., 2005). Toluene permeabilizes mitochondria (Giulivi et al., 1998) and induced the formation of ROS in rat brain synaptosome fractions (Myhre and Fonnum, 2001), but at concentrations up to 178 µg/ml did not produce cytotoxic effects towards CD3-CD28-stimulated human peripheral blood mononuclear cells (Wichmann et al., 2005). However, 1 mM toluene increased caspase-9 activity in LLC-PK1 immortalized renal proximal tubular epithelial cells derived from the pig, after being treated for 96 hours (Al-Ghamdi et al., 2004). Toluene was also toxic towards primary cultured hippocampus neurons at concentrations above 3 mM after 24 hours (Yan et al., 2004).

The measurement of oxygen consumption during the autoxidation of 1 mM 3MC in 1 mM HCl, 50 mM phosphate buffer (pH 7.4) showed that 1.98 µM oxygen reacted with 3MC per minute leading to the formation of superoxide, 3-methyl-o-benзоsemiquinone and 3-methyl-o-benzoquinone, which absorbed light at 320 nm. These results agreed with data obtained during the electrochemical oxidation of 3MC (Fakhari et al., 2005). Scheme 1 shows the reaction of 3MC with oxygen.

Figure 4 - Concentration-response curves for quinone formation and 3MC-induced cytotoxicity in rat astrocyte cultures after 72 hours. The formation of quinones during the autoxidation of 3MC was measured by colorimetry at 405 nm (●). Data on autoxidation are represented by means ± SD, n = 8. The mean O. D. of the control group was 0.012. 3MC autoxidation led to a significant formation of quinones at all concentrations tested (P < 0.05). Statistical significance was analyzed using one-way ANOVA followed by the Student-Newman-Keuls test. 3MC also alters the viability of these cells (■). Viable cells were quantified colorimetrically using the MTT assay following 72 h in the presence of 3MC. Data were normalized to the values measured under control conditions (median of OD at 560 nm = 0.5). 3MC caused a significant reduction in cell viability at concentrations > 100 µM. Results are shown as medians ± 25th and 75th percentiles, n = 8. Statistical significance was analyzed using the Kruskal-Wallis one-way ANOVA followed by the Dunnett’s method.
For the first time it was shown that 3MC induced the peroxidation of biomolecules in rat brain nuclear fractions. 4-Methylcatechol at concentrations up to 100 µM failed to induce lipid peroxidation of rat liver microsomes (Boots et al., 2002). However, the formation of quinones induced a thiol arylation of proteins, which inactivated some enzymes. Several catechols derived from arenes with a heterocyclic group at low concentrations inhibited the lipid peroxidation induced by free radicals in rat brain homogenates (Misawa et al., 2005), but the toxicity of these compounds at higher concentrations was not tested.

The results also showed for the first time that 3MC significantly inhibited state 2 FADH$_2$-linked respiration in rat brain nonsynaptic mitochondria. In a previous work, 1,2-dihydroxybenzene (catechol) also inhibited state 2 FADH$_2$-linked respiration of rat liver mitochondria (Barreto et al., 2005). These data suggest that catechols may interfere with the respiratory chain. However, further studies need to be carried out to fully clarify the effects of 3MC on mitochondrial function.

As 3MC is able to produce superoxide and reactive quinones, induce peroxidation of biomolecules and inhibit FADH$_2$-linked respiration, these processes could account for cytotoxic effects. This study demonstrated that cell viability decreased in cultures of isolated rat cortical astrocytes after treatment with 3MC. The
decrease in cell viability was accompanied by the production of quinones in the culture medium. The \( EC_{50} \) was 107 \( \mu \)M after 72 h. These data correlate with the observed cytotoxicity of 3MC to bluegill sunfish BF-2 fibroblast cells (Shen et al., 2000) indicating that this metabolite of toluene represents a hazardous environmental pollutant. The average amount of 3MC in one cigarette is 38 \( \mu \)g, and 63% (24 \( \mu \)g) is partitioned in the lipid phase of a compartmentalized system (Smith and Hansch, 2000). This means that theoretically this molecule can be stored in lipids that are present in the brain. Despite the fact that 3MC is present in cigarette smoke condensate, it is listed in the Registry of Toxic Effects of Chemical Substances database under category 6 because there are insufficient data in the scientific literature with respect to its toxicity (Smith and Hansch, 2000). Therefore, this study contributed to update data about 3MC toxicity in vitro.

Other catechols are also cytotoxic compounds. In a previous study (Pereira et al., 2004), it was demonstrated that catechol was cytotoxic to human glioblastoma cells via the production of superoxide and reactive quinones. Catechol at 230 \( \mu \)M killed 50% of these cells after 72 h. In another work (El-Bachá et al., 2001), the neurotoxicity of apomorphine, a catechol (potent dopamine agonist), was studied. Results showed that the toxicity of this drug was also related to the autoxidation and formation of ROS and quinones. Apomorphine at 200 \( \mu \)M killed 50% of glial C6 cells after 48 h (El-Bachá et al., 2001). The toxicity of 3MC, one of the major aromatic intermediates of aerobic transformation of toluene in the environment, was evaluated in the organelles and cells of rat brains. This toluene metabolite can elicit peroxidation of biomolecules, inhibition of state 2 FADH\(_2\)-linked respiration in nonsynaptic mitochondria and death of astrocytes. The results of the study indicated a direct relationship between cytotoxicity and 3MC oxidation. Since the cytotoxic effects induced by 3MC involved an oxidative process, this model could be technologically used for screening the protective and antioxidant activity of biomolecules obtained from natural products.

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**RESUMO**

O 3-metilcatecol (3MC) é um metabólito do tolueno. Para esclarecer se o 3MC seria tóxico para o sistema nervoso central, examinou-se seus efeitos sobre a peroxidação de biomoléculas em frações nucleares e a respiração mitocondrial em organelas obtidas de cérebros de ratos. Também se testou a citotoxicidade para astrócitos primários de ratos. O 3MC a 1mM oxida-se consumindo oxigênio a uma taxa de 1,98 ± 0,19 \( \mu \)M.min\(^{-1}\), formando quinonas reativas. Nessa mesma concentração o 3MC peroxidou biomoléculas nas frações nucleares. Esse composto também inibiu o estado 2 da respiração mitocondrial associada ao FADH\(_2\). Além disso, o 3MC também se oxida em meio de cultura levando à formação de quinonas. Esse metabólito do tolueno foi citotóxico para astrócitos de ratos. A concentração que matou 50% das células após 72 horas foi 107 \( \mu \)M. Os resultados desse estudo indicam uma relação direta entre a citotoxicidade e a oxidação do 3MC.

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