

# Article – Biological and Applied Sciences Isolation of Polyphenols from Soursop (Annona muricata L.) Leaves Using Green Chemistry Techniques and their Anticancer Effect

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Editor-in-Chie<u>f</u>: Paulo Vitor Farago Associate Editor:Jane Budel

Received: 2020.03.20; Accepted: 2020.12.18

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## HIGHLIGHTS

- Isolate, fractionate and characterize extracts obtained from soursop leaves.
- Use of emerging green technologies such as microwave-ultrasound hybridization.
- The extracts contain kaempferol, procyanidins, catechin, and quercetin.
- The total ethanolic extract demonstrates cytotoxic effect on HeLa cells.

**Abstract:** Cervical cancer is classified as the fourth most common malignancy in women. Natural compounds are a therapeutic alternative in cancer therapy. The aim of the study is to isolate, fractionate, and characterize extracts obtained from soursop leaves (*Annona muricata* L.) and determine their cytotoxic effect against HeLa cervical cancer cells and non-carcinogenic fibroblast 3T3 cells. The phytochemicals of soursop leaves were extracted through emerging green technologies such as the novel use of microwave-ultrasound hybridization and the use of environmentally friendly solvents (water and ethanol), in addition to the purification of extracts enriched in polyphenols by liquid chromatography with Amberlite XAD-16. Total aqueous and ethanolic extract were purified, as well as the fraction one of each extract. The extracts recovered from soursop leaves contained kaempferol and its isomers, procyanidins, catechin, and quercetin. The viability of the cells was determined with the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay. HeLa and 3T3 cells were exposed to concentrations of 25, 50, 75, 100, 150, 200, and 250 ppm of a solution of soursop leaf extract powder. The MTT assay showed that soursop leaf extracts were toxic to both cell lines in general, however, the ethanolic extract at 25 and 50 ppm demonstrated inhibition in cell viability against the HeLa

cancer line and low cytotoxicity for 3T3 fibroblast cells. In conclusion, the novel microwave-ultrasound hybridization technology allows the extraction of polyphenols that may have a potential cytotoxic effect on cancer cells.

## Keywords: green solvents; microwave-ultrasound; soursop; polyphenols; HeLa.

## INTRODUCTION

Each year, more than a half-million women are diagnosed with cervical cancer and results in more than 300,000 deaths worldwide, also, this cancer is classified as the fourth most common malignancy in women [1]. In 2012, 528,000 new cases were diagnosed, and 266,000 women died and almost 90 % of them were registered in countries with low or medium income and it is predicted that without urgent care, deaths due to cervical cancer could increase by almost 25 % over the next 10 years [2].

Cervical cancer is a malignant neoplasm that develops in the inner fibromuscular portion of the uterus, which is projected within the vagina, and it is related to human Papillomavirus (HPV), in addition to other factors of the local microenvironment, such as pH and the vaginal microbiota [3,4]. The treatment depends on the stage of cancer, and more than one therapy can be administered, such as radiotherapy, chemotherapy, or immunotherapy [5,6]. The most common drugs used to treat cervical cancer are cisplatin, carboplatin, paclitaxel, topotecan, and gemcitabine. However, different side effects have been associated with the use of these therapies [6]. In addition, the high cost of the treatments hinders their access to the population [7]. Therefore, it is essential to search for new treatments that are cheaper and that do not cause side effects. One of these alternatives is the use of bioactive compounds from natural sources and, therefore, take advantage of the biological wealth of developing countries [8]. Moreover, the process of obtaining bioactive compounds needs to incorporate "green extraction" methods, to achieve a faster extraction rate, more effective energy use, reduction in the number in processing steps, use of safer solvents, and avoiding waste production [9]. Currently, the use of "green" methodologies is being sought, which promotes the use of cleaner extraction protocols, with the proposal of protecting the consumer and the environment [10]. Ultrasound and microwave techniques have proven to be a viable alternative for the extraction of natural molecules. Likewise, the combination of analytical techniques is a strategy to save energy, time, and resources [9].

It has been shown that some plants are a source of antitumor compounds of medical importance, so it is important to conduct *in vitro/in vivo* research that supports their antineoplastic potential [11]. In species of the Annonaceae family, the presence of bioactive compounds with antioxidant, antitumor, immunosuppressive, anti-inflammatory, and antimicrobial activity among others has been reported [12]. *Annona muricata* L., widely known as soursop, is a tree 5-8 meters in height that has large, bright, and dark green leaves [13]. Different parts of *A. muricata* have been widely used in the traditional medicine of different countries for the treatment of various ailments and diseases [12]. *In vitro* studies have shown that plants contain compounds that block some essential metabolites of cancer cells and this induces the death of malignant cells [14-16]. Based on this background, the present study was established under the following objectives: Obtain phytochemicals of soursop (*Annona muricata*) leaves through emerging green technologies (hybridization of ultrasound and microwave) and environmentally friendly solvents, purify and separate rich-phytochemical extract fractions, and finally evaluate the cytotoxic activity of the purified fractions on HeLa and 3T3 cell lines.

## **MATERIAL AND METHODS**

## Plant material

Samples of dried soursop (*Annona muricata*) leaves were commercially acquired from the "Productos Rickland" company, which specializes in soursop products in Mexico and is located in Las Varas, Nayarit State, México. 1040 g of soursop leaves were purchased from the supplier. First, the leaves were selected, eliminating those that had some physical or microorganism damage. The leaves were then allowed to dry for three days at room temperature (22-26 °C) and 24 h in a Coldryer-NWT-5 brand dehydrator. Subsequently, the leaves were cut into small pieces, which were ground in a blender, and then, the powder was recovered.

#### **Extract preparation**

The extracts were obtained with two different solvents, water, and an ethanol/water mixture. To obtain the aqueous extract, 62.5 g of the powder of *A. muricata* leaves were weighed and mixed with 1000 mL of distilled water in a one litter reactor to obtain a 1:16 ratio mass: volume. For the ethanol extract, an ethanol/water mixture was made (700 mL of ethanol 96% in 300 mL of distilled water), and then 62.5 g of the crushed soursop leaves were mixed with the solution prepared in the one-liter reactor.

#### Phytochemical extraction

The mixtures of plant material and solvent previously prepared and contained in a 1 L reactor were placed in the "Ultrasonic Microwave Comparative Workstation (ATPI0, Nanjing ATPIO Instruments Manufacture Co., Ltd Company, China) equipment". The following conditions were used for Ultrasound: Power Radio 20, Ultrasonic on Relay 10, Ultrasonic Off Relay 3, Amplitude Transformer 25, and Set 20; and for Microwaves: Power Radio 800, Display Power 0, Set Time 70 °C and Holding Time 5. These conditions were defined based on the previous work of the research group for the extraction of polyphenolic compounds where an adequate extraction of the same and the conservation of biological properties were demonstrated [17,18]. Finally, the aqueous and the ethanolic extracts were filtered through a Whatman No.41 filter paper to remove the insoluble material and the liquid obtained was stored at 4 °C in an amber bottle.

#### Purification and fractionation of phytochemicals

Polyphenols extracts were obtained by using a liquid chromatography column with Amberlite XAD-16, the column was washed with distilled water, then the soursop extract was added, first elution was carried out with distilled water to discard undesirable compounds such as carbohydrates, lipids, and other impurities, and then ethanol was used to recover the phenolic fraction for 6 h. This procedure was performed for both extracts. The technique was repeated to obtain the fractions of the extracts. The fractions were collected at intervals of 1 hour for 6 h; the six fractions of each extract were stored independently. The samples of the complete chromatography and the fractions were placed in an oven at 45 °C for 12 h to remove the solvent and thus recover polyphenol powders. To avoid the degradation of phenolic compounds in the drying process, they were only heated up to 45 °C, since various articles have shown that the degradation process increases from 60-70 °C [19-21]. Finally, the polyphenols powder was stored in amber bottles at 4 °C to avoid compound degradation.

#### Identification of polyphenols by HPLC

For each sample, 1 mg of polyphenol powder obtained was mixed with 1 mL of ethanol (96%). The compounds obtained after chromatographies were analyzed by HPLC (VarianProStar, USA) with a diode array detector (280 nm). A mass spectrometer (MS) with a liquid chromatography ion trap (Varian 500-MS IT Mass Spectrometer, USA) equipped with an electrospray ion source was also used. Samples (5 µL) were injected into a Denali C18 column (3 µm, 150 mm × 2.1 mm, Grace, USA). The oven temperature was maintained at 30 °C. The eluents were formic acid (0.2%, v/v; solvent A) and acetonitrile (solvent B). The following gradient was applied: initial, 3% B; 0-5 min, 9% linear B; 5-15 min, 16% linear B; 15-45 min, 50% linear B. The column was washed and reconditioned. The flow rate was maintained at 0.2 mL/min and the elution was monitored at 245, 280, 320, and 550 nm. All eluent (0.2 mL/min) was injected into the mass spectrometer, without dividing. All MS experiments were performed in negative mode [M-H]<sup>-1</sup>. The data was collected and processed using the MS Workstation software (V 6.9). The samples were first analyzed in the full scan mode acquired in the range 50-2000m/z. MS/MS analyses were performed on a series of selected precursor ions. Use of standards was not necessary for this study since the identification of the compounds is carried out by molecular weight (m/z), supported by the HPLC/MS analysis and, finally, the compounds were compared using a database of bioactive compounds (WorkStation version 2.0 database, Varian, CA, USA).

#### **Cell culture**

The HeLa cell line (ATCC® CCL-2<sup>™</sup>), which are epithelial cells of the human cervix adenocarcinoma, and the 3T3 cell line (ATCC® CRL-1658<sup>™</sup>), fibroblasts from Swiss albino mouse embryonic tissues (3T3, NYU, USA); were grown in DMEM medium (Dulbecco Modified Eagles Minimal Essential Medium) supplemented with 100 U/mL penicillin, 100 mg/mL streptomycin, 0.1 mM non-essential amino acids, 10% fetal bovine serum, and 2 mM L-glutamine in 5% CO<sub>2</sub> at 37 °C.

### Cytotoxicity by MTT assay

1 mg of each extract was placed in a microtube and re-suspended in 1 mL of DMEM cell culture medium supplemented using vortex and sonication. Subsequently, dilutions of 25, 50, 75, 100, 150, 200, and 250 ppm were made. The cytotoxicity of the different concentrations of the purified molecules was determined by the 3-(4,5-dimethylthiazol-2yl)-2,5-diphenyltetrazolium bromide (MTT) assay. 7x10<sup>3</sup> cells per well were transferred to a 96-well plate and grown under the conditions described above for 24 h. Then, 200 µL of the culture medium with the extracts at different concentrations (25, 50, 75, 100, 150, 200, and 250 ppm) was added to each well and allowed to incubate for 24 h. After the incubation, morphological changes were examined with a Labomed inverted microscope at 100x. After this time, the supernatant was discarded and 100 µL of culture medium with 10 µL of the MTT solution (5 mg/mL) was added to each well, and the microplates were incubated for 4 h at 37 °C. Finally, the culture medium was discarded and 150 µL of DMSO was placed, and the absorbance was measured at 570 nm using a microplate reader (Multiskan, Thermo Scientific, USA). The results are expressed as the percentage of cell viability. A reduction in viability greater than 30% was considered a cytotoxic effect according to ISO 10993-5.

#### **Statistical analysis**

The data were expressed as mean ± standard error of the mean (SEM) of three replications. The results were analyzed using a two-way ANOVA followed by Sidak's test. p <0.05 were considered statistically significant. GraphPad Prism 6.01 software (GraphPad Software Inc., La Jolla, CA, USA) was employed for all analyses.

#### RESULTS

#### **Obtainment of leaf polyphenols**

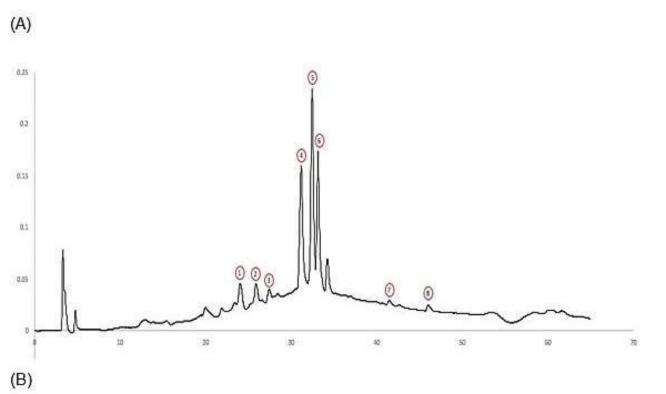
Table 1 shows the yield in grams of the powder obtained compared with 100% amount of vegetable raw material that was used to obtain the extracts. The percentage of yield was from 1.04-4.46%, this may be due to the solvent or the type of chromatographic column used.

Type of extract	Vegetal raw	Type of	Amount (g) recovered	Yield (%)	
	material (g)	chromatography	from extract		
Aqueous	62.5	Complete	1.2	1.92	
Aqueous	62.5	Fractionated	0.65	1.04	
Ethanolic	62.5	Complete	2.65	4.22	
Ethanolic	62.5	Fractionated	2.79	4.46	

Table 1. Percentage yield of the extract obtained from soursop leave

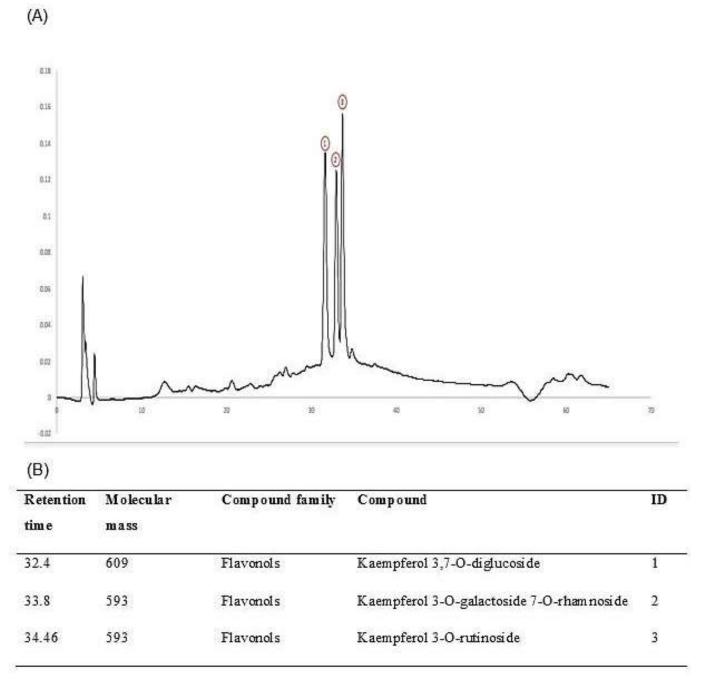
#### Characterization of Polyphenols by HPLC

The HPLC scans, retention time, molecular mass, the compound family and the compounds that were detected in the extracts are shown in Figures 1-4.

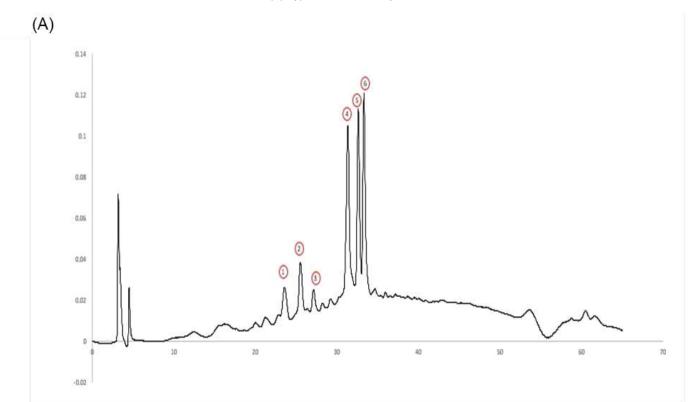


Retention time	Molecular mass	Compund family	Compound	
24.81	576.9	Proanthocyani din dim ers	Procyanidin dim er B1	1
26.78	289	Catechins	(+)-Catechin	2
28.21	864.9	Proanthocyanidin trimers	Procyanidin trimer C1	3
31.96	609	Flavonols	Kaempferol 3,7-O-diglucoside	4
33.3	593	Flavonols	Kaempferol 3-O-galactoside 7-O-rhamnoside	5
33.89	593	Flavonols	Kaempferol 3-O-rutinoside	6
42.36	300.9	Flavonols	Quercetin	7
46.91	284.9	Flavonols	Kaempferol	8

**Figure 1**. (A) Chromatogram of the aqueous extract of the soursop leaf of fraction one of the chromatography. (B) Compounds detected in the aqueous extract of the soursop leaf of fraction one of the chromatography.



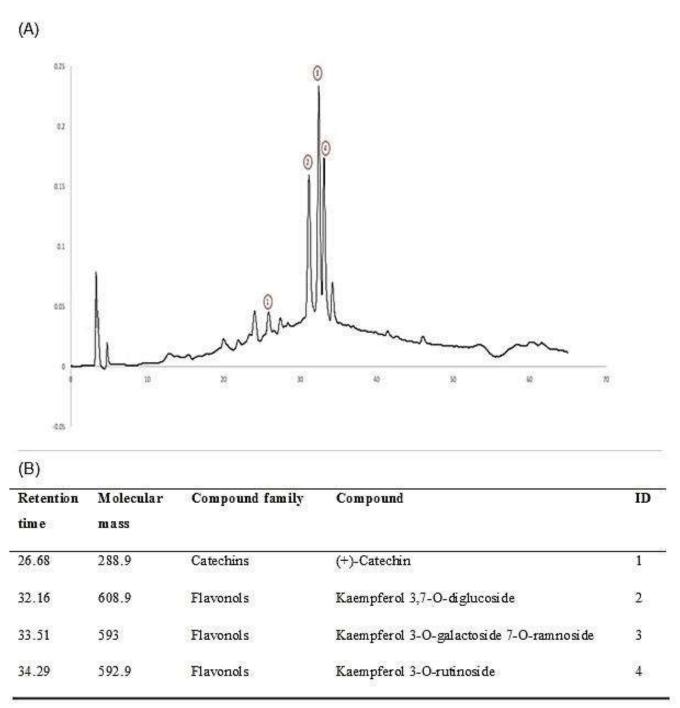
**Figure 2.** (A) Chromatogram of the aqueous extract of the soursop leaf of the complete chromatography. (B) Compounds detected in the aqueous extract of the soursop leaf of the complete chromatography.



(B)

Retention Molecular		Compound family	Compound	
time	mass			
24.29	576.8	Proanthocyanidin dimers	Procyanidin dimer B1	1
26.35	288.9	Catechins	(+)-Catechin	2
27.86	864.8	Proanthocyanidin trimers	Procyanidin trimer C1	3
32.13	608.9	Flavonols	Kaempferol 3,7-O-diglucoside	4
33.5	592.9	Flavonols	Kaempferol 3-O-galactoside 7-O-ramnoside	5
34.06	592.9	Flavonols	Kaempferol 3-O-rutinoside	6

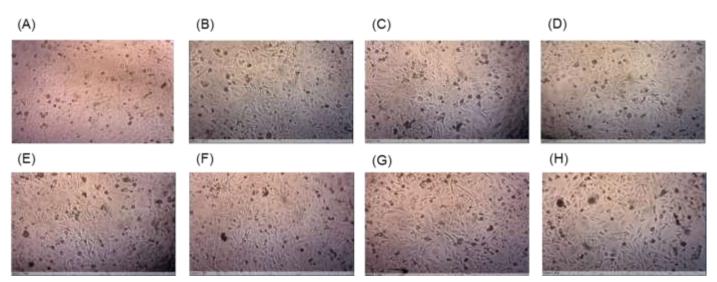
**Figure 3**. (A) Chromatogram of the ethanolic extract of the soursop leaf of the fraction one of the chromatography. (B) Compounds detected in the ethanolic extract of the soursop leaf of fraction one of the chromatography.



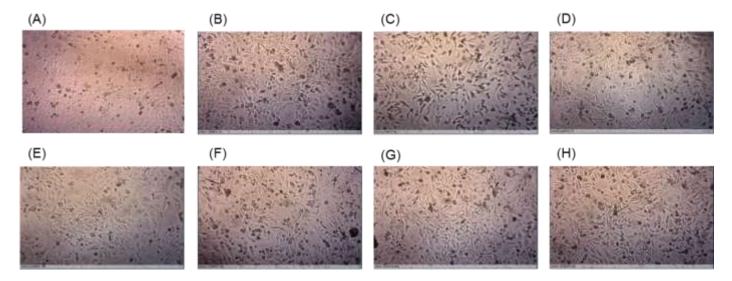
**Figure 4.** (A) Chromatogram of the ethanolic extract of the soursop leaf of the complete chromatography. (B) Compounds detected in the ethanolic extract of the soursop leaf of the complete chromatography.

#### **Cell viability test**

To test the anticancer effect of the phytochemicals isolated from the soursop leaves, the HeLa carcinogenic cell line and the 3T3 fibroblast cell line were used as control. Figures 5, 6, 7, and 8 show light photomicrographs of HeLa cells treated with different extracts and concentrations obtained from soursop leaves. Figures 5 and 6 show fusiform cells with classic morphology, whereby the fraction one and the total aqueous extract thereof have no effect on the morphology of the HeLa cancer cells.

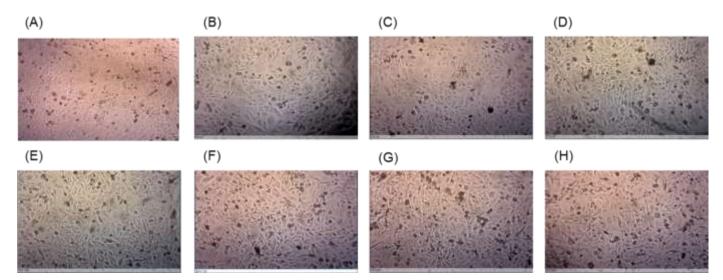


**Figure 5.** Representative light photomicrographs showing HeLa cells exposed to different concentrations of the fraction one of the aqueous extract fraction one from soursop leaves using the inverted microscope at 100x. (A) Control, (B) 25 ppm, (C) 50 ppm, (D) 75 ppm, (e) 100 ppm, (F) 150 ppm, (G) 200 ppm, and (H) 250 ppm.

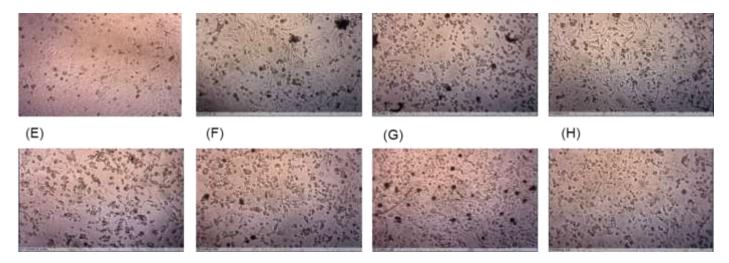


**Figure 6.** Representative light photomicrographs showing HeLa cells exposed to different concentrations of the total aqueous extract from soursop leaves using the inverted microscope at 100x. (A) Control, (B) 25 ppm, (C) 50 ppm, (D) 75 ppm, (E) 100 ppm, (F) 150 ppm, (G) 200 ppm, and (H) 250 ppm.

In Figure 7, fusiform cells with classic morphology are observed, so that fraction one of the ethanolic extract has no effect on cell morphology. In contrast, Figure 8 shows the effect of the total ethanolic extract on HeLa cells, where the presence of cells with a round morphology that is characteristic of a cell death process is observed, likewise this process generated spaces without cells compared to untreated cells, this from the 50 ppm concentration.

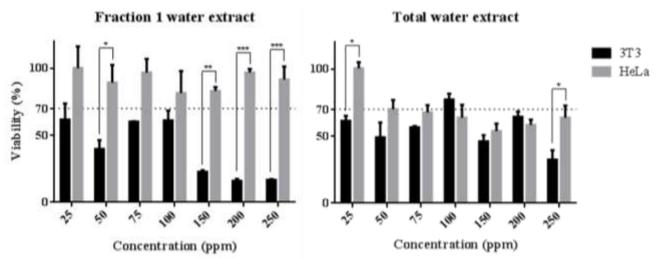


**Figure 7.** Representative light photomicrographs show HeLa cells exposed to different concentrations of the fraction one of the ethanolic extract from soursop leaves using the inverted microscope at 100x. (A) Control, (B) 25 ppm, (C) 50 ppm, (D) 75 ppm, (E) 100 ppm, (F) 150 ppm, (G) 200 ppm, and (H) 250 ppm.



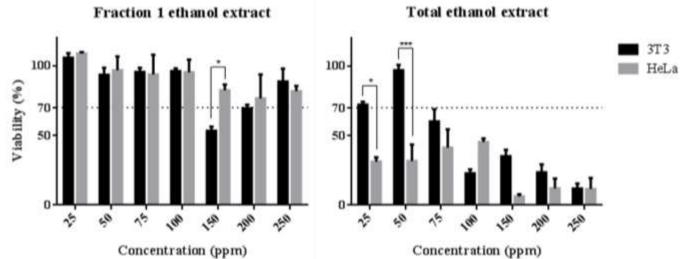
**Figure 8.** Representative light photomicrographs show HeLa cells exposed to different concentrations of the total ethanolic extract from soursop leaves using the inverted microscope at 100x. (A) Control, (B) 25 ppm, (C) 50 ppm, (D) 75 ppm, (E) 100 ppm, (F) 150 ppm, (G) 200 ppm, and (H) 250 ppm.

In general, aqueous and ethanol extracts presented higher toxicity for 3T3 cells. Fraction one of the aqueous extracts had a cytotoxic effect on the 3T3 cell line at all concentrations used, and also did not show a cytotoxic effect on HeLa cells. With respect to the total aqueous extract, only the concentration of 100 ppm did not present a cytotoxic effect on 3T3 cells, whereas, for HeLa cells, only the concentration of 25 ppm did not present a cytotoxic effect according to ISO 10993-5. In addition, doses starting at 100 ppm show a significant decrease in the viability of HeLa cells with respect to the initial concentration of 25 ppm (\*p < 0.05) (Figure 9).



**Figure 9.** Cytotoxic effect of different concentrations of fraction one and complete aqueous extract on HeLa and 3T3 cell lines. The viability percentage of each cell line is presented in the graph as mean  $\pm$  SEM. \*p < 0.05, \*\*p < 0.01, \*\*\*p < 0.001.

Fraction one of the ethanolic extract does not affect the viability of HeLa cells at any concentration used, however, concentrations of 150 and 200 showed a cytotoxic effect on 3T3 control cells according to ISO 10993-5 and only the concentration of 150 ppm showed a significant decrease in viability with respect to HeLa cells (\*p < 0.05). The total ethanolic extract had a non-selective cytotoxic effect on both cell lines, except for concentrations of 25 and 50 ppm, which only affected the viability of HeLa cells without affecting 3T3 cells, this effect being more marked for the concentration of 50 ppm (\*\*\*p < 0.001) (Figure 10).



**Figure 10.** Cytotoxic effect of different concentrations of fraction one and complete ethanolic extract on HeLa and 3T3 cell lines. The viability percentage of each cell line is presented in the graph as mean  $\pm$  SEM. \*p < 0.05, \*\*p < 0.01, \*\*\*p < 0.001.

#### DISCUSSION

Medicinal plants can provide a useful or complementary alternative to conventional therapies used to treat cancer [11]. *A. muricata* leaves have been used in traditional medicine for the treatment of various inflammatory diseases as well as in the treatment of cancer [22]. This is due to the high content of flavonoids present in *A. muricata* leaves, in comparison to its roots and stalks contents, in addition, leaves are the most accessible source for obtaining polyphenols [16]. In the present work, the extraction of polyphenolic compounds was carried out through the use of green methodologies such as combined extraction by ultrasound-microwave. This extraction process consumes less energy compared to conventional methods, in addition, the processing time and temperatures are minimized, which is useful for the extraction of thermolabile compounds, such as polyphenols [9, 10, 23]. Likewise, the yield of the polyphenols obtained was higher when using an ethanol extraction compared to the aqueous extraction. This may be due to the fact that other compounds could be extracted in addition to the polyphenols, such as proteins and carbohydrates, which were more soluble in water compared to ethanol or methanol [24].

The polyphenolic compounds of A. muricata leaf extracts were identified by the HPLC-ESI-MS technique. These compounds were observed at a wavelength of 280 nm. The compounds detected were: guercetin, procyanidins, kaempferol, and catechin. It has been reported that some flavonoids such as isoflavones, catechins, flavones, and flavonol glycosides are absorbed between 260 and 290 nm [25]. In general, the same compounds were observed in the different extracts, with the exception of the isomers of kaempferol, procyanidin, and quercetin. This result may be due to the polarity of the solvents used, highlighting that the water is highly polar, while the mixture of ethanol and water is medium or moderately polar. The difference of the isomers or the compounds found could be due to the polyphenols present in the A. muricata leaf solubilize less in ethanol solvents, than those found in aqueous extractions, since, the flavonoids present in the soursop are more polar and more soluble in polar solvents and less soluble in moderately polar solvents. This is consistent with other studies in which the total polyphenol content of the soursop leaves was shown to be higher when polar solvents were used during the extraction process [26, 27]. When comparing the two chromatographies, the complete and the fractionated one, there were fewer compounds in the first one than in the fractionated one, regardless of the extraction conditions. The possible cause for more compounds in the fraction one is because the compounds are very large in size and have more fluid desorption, therefore these molecules are generally obtained first in purification because they are part of the group of extracted molecules that do not enter the chromatographic packing by rapidly passing through the stationary phase of the column, such as procyanidins and their isomers.

The aqueous and ethanolic extracts of the soursop leaves were tested against the HeLa and 3T3 cell lines. Fraction one of the aqueous extracts and fraction one of ethanol extract did not have a cytotoxic effect in HeLa cells at any concentration used, however, for 3T3 cells, fraction one of the aqueous extracts was cytotoxic for this cell line. Regarding the total extracts, the aqueous extract had a non-selective cytotoxic effect, decreasing the viability of both cell lines at the different concentrations used. With respect to the total ethanolic extract, the results showed similar behavior in both cells types, without presenting a selective effect of cell death, with the exception of concentrations of 25 and 50 ppm. These concentrations had a cytotoxic effect according to ISO 10993-5 only for HeLa cells (69 and 68% cell inhibition) without affecting the viability of 3T3 cells (28 and 3% cell inhibition). Several studies have demonstrated the anti-cancer effect of soursop leaves both in vitro and in vivo [14, 16, 26, 28-30]. In general, aqueous and ethanolic extracts of A. muricata leaves have shown a cytotoxic effect on different cancer cell lines [31]. The proposed mechanism of soursop leaf extracts is to induced caspase-3 mediated apoptosis (mitochondrial route) and inhibition of cell growth by promoting arrest in the G0/G1 phase of the cell cycle [12, 32, 33]. In the case of cervical cancer, Jeno and coauthors [34] reported that HeLa cells exposed to 75 ppm of the crude extract of A. muricata showed 80% cellular inhibition. Also, Astirin and coauthors [35] demonstrated that the soursop leaf extract, which uses chloroform as a solvent, caused apoptosis on HeLa cells in a higher percentage compared to the aqueous extract. In contrast, in the present study, only the polyphenols of the total ethanolic extract were those that had an anticancer effect against the HeLa cell line without affecting the 3T3 control cells. The compounds found in the total ethanol extract were (+) catechin and kaempferol. These flavonols can reduce the risk of cancer, according to other studies [36, 37]. Catechin has been shown to induce apoptosis on SiHa cervical cancer cells by increase the expression of pro-apoptotic genes such as caspase-3, -8, and -9 [38]. Kaempferol is one of the most common dietary flavonols and exerts anticancer activity through several pathways, including induction of apoptosis, G2/M cell cycle arrest, and caspase 3-dependent apoptosis [39-41]. Regarding cervical cancer, Tu, and col. [42] found that kaempferol inhibits the growth and proliferation of SiHa cells in a time and dose-dependent manner, and induces apoptosis due to the disruption of mitochondrial membrane potential. In another study, kaempferol has been shown to suppress the growth of HeLa cells as compared with HFF cells (normal cells), also, reported that kaempferol effectively induced apoptosis via the up-regulation of pro-apoptotic genes such as p53, p21, caspase-3, and -9 [43].

Therefore, the cytotoxic effect of the polyphenols obtained from the ethanolic extract described in this work could be due to the synergistic activity of catechin and kampeferol. However, more studies are needed to verify which molecules are responsible for the cytotoxic effect and to elucidate the mechanism of action of the extracts from *Annona muricata* leaves.

## CONCLUSION

The extraction process of polyphenols through the combination of ultrasound and microwave is novel, efficient, fast, and easy to carry out. In addition, the purification by chromatography with Amberlite XAD-16 proved to be a successful methodology for obtaining polyphenolic compounds from soursop leaves. The polyphenols of the total ethanolic extract of the *Annona muricata* leaves at doses of 25 and 50 ppm were cytotoxic against the HeLa cell line according to ISO 10993-5, without affecting the viability of the 3T3 line, so it could be an alternative cancer therapy.

**Funding:** This study had financial support from The Secretary of Agriculture, Fishing and Livestock-Mexico, through the Project: FON.SEC. SAGARPA-CONACYT CV-2015-4-266936.

Conflicts of Interest: The authors declare no conflict of interest.

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