Impact of Refrigeration on the Surface Hardness of Hybrid and Microfilled Composite Resins

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This \textit{in vitro} study evaluated the Knoop hardness of the composite resins Charisma\textsuperscript{®} (C) and Durafill VS\textsuperscript{®} (D) polymerized in 3 different conditions: at room temperature (A) (23 ± 1°C); refrigerated at 4 ± 1°C and immediately photo-activated after removal from the refrigerator (0); and, refrigerated at 4 ± 1°C and photo-activated after a bench time of 15 min at room temperature (15). One hundred and twenty specimens (4 mm diameter and 2 mm depth) were made using a stainless steel mold and following manufacturer’s instructions. All specimens were tested immediately after polymerization (1) and after 7 days of water storage in the dark at room temperature (7d). The data were subjected to ANOVA and post-hoc Tukey’s test ($\alpha=0.05$). On the top surface, CAI was statistically similar to C15I and DAI to D15I ($p>0.05$). On the bottom surface, CAI presented higher hardness values when compared to C0I and C15I ($p<0.05$). The D groups showed no significant differences ($p>0.05$) on the bottom surfaces for any tested polymerization condition. After 7 days of storage, the Knoop hardness decreased significantly ($p<0.05$) for groups C7d and D7d except for C07d which was not different from C0I at either surface ($p>0.05$). D07d showed higher Knoop hardness ($p<0.05$) values on the top surface when compared to the other groups.

Key Words: resin composite, refrigeration, Knoop hardness.

INTRODUCTION

The curing of methacrylate-based composite resins for dental applications is a major issue due to its importance regarding biocompatibility (1), bacterial growth associated with the release of unreacted monomers (2) and improvement of the material’s physical properties (3).

The balance of the components in the organic formulation has been proven essential during the polymerization process (4). Other factors controlling light-activated polymerization are irradiance, wavelength (4), and the time needed for polymerization to occur (5,6). Some studies have investigated the possibility that resin monomer flexibility and the temperature during polymerization may affect the conversion rate and the material properties (4).

Kloosterboer et al. (7) have shown that monomer mobility is higher with increased polymerization temperature. Trujillo et al. (8) have found that pre-heating the composite resin increases the immediate and the final conversion rates due to the increase of monomer mobility. Although relatively recent research has shown favorable results regarding the increase of the conversion rate with pre-heated composites (9), few studies have addressed the cooling of methacrylate resin materials (9,10). This is an important issue to be investigated because the refrigerated storage of composite resins is a common practice to increase their shelf-life and is suggested by some manufacturers.

Monomers seem to need a certain mobility to be properly converted into polymers (7,8). Since re-
frigeration could increase the material’s viscosity (4), thereby decreasing monomer mobility, an alteration of the properties of the polymerized composite resin could be anticipated. It would be expected in cases where the material is used for a clinical procedure immediately after removal from refrigeration.

Knoop hardness test is directly related to the degree of conversion (11). Bouschlicher et al. (12), showed a linear correlation between hardness and degree of conversion when investigating the differences between the type of composite resin and the depth of cure of the specimen.

The purpose of this study was to determine the impact of refrigeration on the Knoop hardness of two restorative composite resins at the top and bottom surfaces of the specimens. The resins were polymerized in three different conditions: at room temperature (23 ± 1°C); refrigerated at 4 ± 1°C and immediately photoactivated after removal from the refrigerator; and, refrigerated at 4 ± 1°C and photoactivated after a bench time of 15 min at room temperature. Measurements were made immediately after polymerization and after 7 days of storage. The hypotheses tested were that there are no differences in the mean Knoop hardness values of the composite resins regardless of the polymerization conditions or storage time before testing.

**MATERIAL AND METHODS**

Two commercially available composite resins were tested in this study: a hybrid composite (Charisma (C); Heraeus Kulzer GmbH & Co., Hanau, German) and a microfilled composite (Durafill VS (D); Heraeus Kulzer GmbH & Co.) (Table 1). The Knoop hardness test was carried out to determine the surface hardness of the composite resins when submitted to different temperatures prior to polymerization.

For the control groups (immediate and 7 days) for both materials (C and D), the composite resin tubes were kept at 23 ± 1°C and a relative humidity of 50 ± 5% up to the specimens preparation. For the other groups (refrigerated material), the composite resins were cooled to 4 ± 1°C for 24 h before preparing the specimens; after this time the composite resin was removed from the tubes, placed at the mold and polymerized after different bench time at room temperature (Table 2).

In all groups, the temperature of the composite resin paste was verified before the polymerization with a digital multimeter (ET 2210; Minipa, São Paulo, SP, Brazil). The composite resins at room temperature (23°C) took 15 min to reach 4°C when placed inside the refrigerator. When started at 4°C, it was necessary the same 15 min out of the refrigerator for the resin paste to reach room temperature again. The same operator prepared all specimens and the 6 groups were divided as shown in Table 2.

For all groups cylindrical specimens were fabricated using a split stainless steel mold with a 4 mm diameter and 2 mm depth. A Mylar matrix was pressed against the composite resin surface using a 1 mm glass slide in order to obtain a smooth and flat surface. Visible light-activation was accomplished with a quartz-tungsten-halogen light-curing unit (Optilux VCL 501; Demetron/Kerr, Danbury, CT, USA) with irradiance of 490 mW/cm², as measured with a curing radiometer. The activation time was 20 s, as recommended by the manufacturer. The energy density emitted was calculated by multiplying the irradiance by the emission time. Therefore, all specimens were exposed to an energy density of 9.8 J/cm².

After removing the specimens from the mold, the top was distinguished from the bottom and a line was draw at

<table>
<thead>
<tr>
<th>Table 1. Tested materials.</th>
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<tbody>
<tr>
<td><strong>Composite resin</strong></td>
</tr>
<tr>
<td><strong>Charisma</strong></td>
</tr>
<tr>
<td><strong>Durafill VS</strong></td>
</tr>
</tbody>
</table>

Source: Manufacturer.
the center of each top and bottom surface. This provided one hemi-surface for immediate hardness testing and another for testing after 7 days. A microhardness tester (HVS 1000; NDT Instrument Co, Beijing, China) was used with a load of 50 g applied for 15 s. Each specimen was tested immediately after polymerization and after a 7-day storage in water at 23 ± 1°C. Seven indentations were made on the top and 7 indentations were made on the bottom of each specimen at each period. The distance between each indentation was ≥4 larger diagonals of the indenter (4D). This precaution was taken to prevent areas of plastic deformation produced by neighboring indentations from altering the hardness values (Fig. 1). The larger diagonal was measured with an optical microscope (Olympus BX 51; Olympus Corp, Tokyo, Japan) at 50-fold magnification, and using an image-analysis software (Image-Pro Express 5.0; MediaCybernetics Inc, Silver Spring, MD, USA). The values (in µm) were converted into Knoop hardness number values using the following equation:

$$KHN = \frac{14229 \times P}{d^2}$$

Where:
P=applied load, in kgf;
d=larger diagonal, in µm;
KHN=Knoop Hardness Number, in kgf/mm².

The average of the 7 indentations was calculated and this value was used as the KHN for each specimen (n=20). Values were obtained from the top and bottom surfaces for each group, and the the bottom/top ratio (B/T) was determined.

The data were analyzed by multiple-way analysis of variance. When statistical differences were verified between the treatments, the means were individually analyzed with Tukey’s HSD test at 5% significance level.

**RESULTS**

The hardness values for the top surface of groups C\(_A\) and C\(_{15}\) did not differ significantly from each other (p>0.05). C\(_{15}\) did not differ from C\(_0\) (p>0.05), but C\(_0\) was significantly lower than C\(_A\) (p<0.05). For the bottom surfaces, C\(_0\) and C\(_{15}\) showed lower mean values when compared to C\(_A\) (p<0.05). The mean values for C\(_{A}\) and C\(_{0}\) did not differ significantly for either the top or the bottom surfaces (p<0.05). After storage, no
significant reduction in hardness was observed at the top and at the bottom surfaces of C₀ (Table 3).

For D at the top surface, there were no statistically significant differences between the mean values for Dₐ and D₁₅ (p>0.05). D₁₅ did not differ from D₀ (p>0.05), but D₀ yielded statistically lower mean values than Dₐ (p<0.05). Still at the top surface, microfill resin composite had also showed no statistically significant difference (p<0.05) between Dₐ₀ₜₐ and Dₖₐₜₐ. D₀ₜₐ showed higher hardness values when compared to both groups. After storage, significant reductions (p<0.05) in hardness of the top and the bottom surfaces were observed when comparing immediate and 7-day tests within the groups (Table 4).

The B/T ratio did not present statistically significant differences with the tested temperatures between the immediate and the 7-day tests for C (p>0.05). However, statistically significant differences (p<0.05) were observed for D, which presented lower values after 7 days of storage, though without significant differences between the tested temperatures (p>0.05).

DISCUSSION

The results suggest influence of refrigeration on Knoop hardness values for the hybrid (C) and microfilled (D) composite resins tested in this study.

Both resins presented changes in the hardness values verified right after the polymerization (I) between the specimens prepared at room temperature (A) and just after removal of the resin from refrigeration at 4°C (₀).

The strong influence of temperature in the polymerization process of the composite resins found by Daronch et al. (9), who showed an increase in the degree of conversion at a 60°C and a decrease at 3°C, does not seem to apply when using hardness testing and temperatures of 4 ± 1°C and 23 ± 1°C.

With a bench time of 15 min after removal from the refrigerator, the temperature of both composites was 23°C. The statistically similar top surface hardness values for the groups tested at room temperature were 23°C. The statistically similar top surface hardness values for the groups tested at room temperature and 15 min after removal from the refrigerator was an expected finding. The hardness on the top surface was higher than the hardness on the bottom at all time periods, indicating difficulty in polymerizing the bottom surface. This can be partially attributed to the fact that proximity of the top surfaces to the light source and their receiving a higher energy density (5,6,13,14).

The temperature increase caused by halogen light-curing units, as reported by Soh et al. (13), might have helped the polymerization kinetics at the top surface by heating (4). However, the temperature increase may not have been enough to aid the polymerization at the bottom surface due to the isolating characteristics of the composite resin and the great distance from the light source (15,16).

<p>| Table 3. Mean Knoop hardness number (KHN) and standard deviations (sd) for the composite Charisma tested at the top and bottom surfaces, immediately after polymerization and after 7 days of water storage. |</p>
<table>
<thead>
<tr>
<th>Top₁</th>
<th>CA</th>
<th>C₀</th>
<th>C₁₅</th>
</tr>
</thead>
<tbody>
<tr>
<td>CA</td>
<td>33.24 (5.35) aA</td>
<td>30.46 (3.57) bA</td>
<td>31.88 (3.97) abA</td>
</tr>
<tr>
<td>C₀</td>
<td>28.65 (4.41) aB</td>
<td>29.79 (5.21) aA</td>
<td>28.41 (4.73) aB</td>
</tr>
<tr>
<td>C₁₅</td>
<td>17.60 (4.29) aC</td>
<td>15.46 (3.31) bC</td>
<td>16.10 (3.15) bC</td>
</tr>
<tr>
<td>Bottom₁</td>
<td>14.51 (3.16) aD</td>
<td>14.99 (4.77) aC</td>
<td>14.49 (2.94) aD</td>
</tr>
<tr>
<td>Bottom₁₅</td>
<td>19.16 (2.87) aA</td>
<td>18.02 (1.84) bA</td>
<td>18.58 (2.61) abA</td>
</tr>
<tr>
<td>Top₂₅</td>
<td>15.42 (2.15) aB</td>
<td>16.32 (3.0) bB</td>
<td>15.07 (3.09) aB</td>
</tr>
<tr>
<td>Bottom₂₅</td>
<td>9.43 (1.65) aC</td>
<td>9.34 (2.67) aC</td>
<td>9.21 (3.98) aC</td>
</tr>
<tr>
<td>Bottom₂₅</td>
<td>7.25 (1.74) aD</td>
<td>7.61 (2.53) aD</td>
<td>6.83 (1.80) aD</td>
</tr>
</tbody>
</table>

r= Immediate; rₗ= 7 days; Means followed by the same lowercase letters in rows and uppercase letters in columns indicate no statistically significant difference at the 95% confidence level (Tukey’s test, p>0.05).

<p>| Table 4. Mean Knoop hardness number (KHN) and standard deviations (sd) for the composite Durafill VS tested at the top and bottom surfaces, immediately after polymerization and after 7 days of water storage. |</p>
<table>
<thead>
<tr>
<th>Top₁</th>
<th>DA</th>
<th>D₀</th>
<th>D₁₅</th>
</tr>
</thead>
<tbody>
<tr>
<td>DA</td>
<td>19.16 (2.87) aA</td>
<td>18.02 (1.84) bA</td>
<td>18.58 (2.61) abA</td>
</tr>
<tr>
<td>D₀</td>
<td>15.42 (2.15) aB</td>
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</tr>
<tr>
<td>D₁₅</td>
<td>9.43 (1.65) aC</td>
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<tr>
<td>Bottom₁</td>
<td>7.25 (1.74) aD</td>
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<td>6.83 (1.80) aD</td>
</tr>
</tbody>
</table>

r= Immediate; rₗ= 7 days; Means followed by the same lowercase letters in rows and uppercase letters in columns indicate no statistically significant difference at the 95% confidence level (Tukey’s test, p>0.05).
Studies have reported that the hardness values at the bottom surface should be between 80 and 90% of the hardness at the top surfaces in order to indicate a proper polymerization (11,12,17). In this study, the exposure duration recommended by the manufacturer resulted in bottom-to-top-surface Knoop hardness ratios between 47 to 54% for the tested resins. This is in agreement with the findings of Cohen et al. (18), who reported the need for exposing the specimens from a 5- to 20-fold longer time than that indicated by the manufacturer to achieve 80% bottom-surface hardness with respect to the top. Caldas et al. (6) have also suggested using a longer curing time, in order to increase the energy density at the bottom of the layer and increase the degree of conversion.

Ideally, the manufacturers should inform the needed energy density and not only the curing time for an appropriate conversion at the bottom of the recommended maximum layer (19). The B/T ratio was not altered for C in the tests carried out immediately and 7 days after polymerization. However, for D, this ratio decreased after 7 days, which suggests that although the values at the top and bottom have decreased with water storage, the bottom presented a more accentuated decrease. This could be explained by the bottom being less polymerized and thus, more susceptible to the influence of water during storage.

After 7 days for C, in both tested surfaces, the differences seen with the refrigerated material within groups were not observed, thus accepting the null hypothesis. A possible explanation could be that the initial difference was impacted by residual stress resulting from polymerization due to the effect of the temperature generated by the curing unit, mainly on the composite at the room temperature. Stress relaxation may have contributed to the hardness values being not statistically different for the 3 tested temperature conditions after 7 days of storage.

However, D0 showed the smallest decrease in hardness at the top surface after 7 days, differing from the others and rejecting the null hypothesis.

The microfilled resin composite with a filler load of 40 vol% and hence more resin matrix than the hybrid composite was even more sensitive to the phenomenon of residual stress and posterior relaxation at the top surface. The composite resin polymerized at the room temperature presented lower hardness mean values than the resin polymerized at 4ºC after 7 days. A possible explanation would be that the hardness at the top surface of both refrigerated resin composites was less affected by the temperature of the curing unit, generating less stress during its polymerization. It may be speculated that faster polymerization occurred at room temperature was influenced by the heat generated by the halogen lamp, which led to the entrapment of low molecular weight monomers among long polymer chains, hindering the formation of the polymer network. Therefore, hardness measurement immediately after the testing would not express the actual hardness of the material because the residual stress generated during the polymerization process accounts for reaching increased values.

These results suggest that the refrigeration allowed a behavior of the composite resin similar to the “soft start polymerization method” because, the low temperature could have proportioned less monomer mobility, decreasing the reaction velocity and providing an extension of the pre-gel phase without decreasing hardness values (20).

Despite the promising results from this preliminary study, pointing to a decrease in the stress generated during polymerization with refrigerated composite resins, further research has been conducted keeping the molds at 37ºC to simulate the intraoral conditions in a better manner, although the insulating properties of composites is known.

According to the results, and within the limitations of this in vitro study, it may be concluded that for the hybrid composite resin, only the immediate hardness values were impacted by the temperature at the top and bottom surfaces. After 7 days for C, no statistically significant difference was observed regardless of the temperature. For D4ºC, the top hardness value was higher than D0ºC. However, the D0ºC hardness value was higher than the D4ºC values after 7 days. The B/T ratio did not differ for the hybrid composite resin among the tested groups. The B/T ratio did not differ for the microfilled composite resin with the different temperatures, but decreased after 7 days of storage.

RESUMO

O objetivo do presente estudo in vitro foi avaliar a dureza Knoop das resinas compostas, uma resina híbrida (Charisma - C) e uma de micropartículas (Durafill VS - D) quando polimerizadas à temperatura ambiente (A) (23 ± 1ºC); refrigeradas (4 ± 1ºC) e fotoativadas imediatamente após a remoção do refrigerador (D0ºC); e refrigeradas (4 ± 1ºC), com tempo de espera de 15 min após a remoção do refrigerador para fotoativação (D4ºC). Cento e vinte corpos-de-prova com 4 mm de diâmetro por 2 mm de altura
Impact of refrigeration on composite hardness

ACKNOWLEDGEMENTS

The authors would like to thank Heraeus Kulzer for supplying the composite resins and the Physics’ Department of the Federal University of Parana, especially Professors Carlos Maurício Lepienski and Paulo César Soares Júnior for the technical and intellectual support during this research.

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Accepted February 5, 2009