Antioxidant activity during storage of apples subjected to irradiation

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ABSTRACT
Nowadays, the number of studies about fruit conservation as well as the benefits of consumption of foods rich in antioxidant compounds was increased. This study aimed to quantify antioxidant compounds and their activity and to evaluate the physical and chemical changes during the post-harvest storage of Eva’s apple cultivars that were subjected to gamma irradiation at doses of 0; 0.5; 1.0 and 1.5 kGy. The antioxidant activity was measured by the 2,2-diphenyl-1-picrylhydrazyl (DPPH) and the β-carotene/linolenic acid methods. Levels of total phenolics, ascorbic acid, color, and soluble solids were evaluated. The results showed that, regardless of the irradiation dose, there was an increase in % SRL during post-harvest storage of Eva’s apples. Low irradiation doses were able to preserve the phenolic compounds, maintain the ascorbic acid levels and avoid an increase in the soluble solids content.

Index terms: Postharvest; storage; antioxidants; food quality.

INTRODUCTION
Pomiculture has shown a gradual and linear growth, and the search for disease-resistant varieties with good agronomic and technological characteristics has made available a wide variety of cultivars to producers. However, little is known about the behavior of the Eva’s apple cultivar, and studies to understand its behavior are needed. This cultivar has a low requirement for cold temperature storage and has therefore been successfully grown in southeastern Brazil (Hauagge; Tsuneta, 1999).

Antioxidants are naturally present in fruits; each plant species contains high concentrations of a specific group of antioxidants. Apples, which contain health-beneficial compounds such fiber and vitamins, are one of the best sources of antioxidants and phenolic compounds, which vary among different apple cultivars (Napolitano et al., 2004; Lata; Przerdzka; Binkiwska et al., 2005; Pieniz et al., 2009).

The antioxidant activity of a food is a result of the action of each of its antioxidants, which can interact and produce synergistic or inhibitory effects (Lee et al., 2003). Among the various food components with antioxidant activity, the most important are the phenolic compounds and ascorbic acid. Phenolic compounds (or polyphenols), which consist of a heterogeneous group of substances found in different concentrations in plant foods, are of great interest due to their antioxidant potential (Sacbert; Williamson, 2000; Lopes et al., 2012).

The antioxidant activity of bioactive compounds is related to their structure and concentration in the plant food. In turn, the concentration of these bioactive compounds is largely influenced by genetic factors, ripeness degree, environmental and processing, storage
time and packaging method (Amakura et al., 2000; Kaur; Kapoor, 2001; Kiralp; Toppare, 2006; Lima et al., 2009; Queiroz et al., 2011).

Irradiation has emerged as an alternative method for the preservation of fruits and vegetables without the use of chemical preservatives (Hussain et al., 2012). During irradiation, foods are exposed to a controlled amount of ionizing radiation for a specific amount of time. This preservation method, which has been studied to improve food safety since the 1950s, offers advantages to the industry, retailers and consumers (Park et al., 2010).

The determination of the optimal dose of irradiation for each product is very important because, if the dose is higher than necessary can promote unfavorable changes in the quality of irradiated products (Azelmatt et al., 2006). Gamma irradiation is effective in reducing bacterial and rust contamination, inactivating pathogens present in fresh produce, and decreasing the fruit ripening and decomposition processes (Sendra et al., 1996; Niemira et al., 2003; Baskaran; Devi; Nayak, 2007; Guimarães et al., 2012). Irradiation can inhibit senescence, ensure food safety, and expand product shelf life (Farkas, 1998; Deruiter; Dwyer, 2002; Bari et al., 2004; Kamat et al., 2005; Shashidhar et al., 2007; Niemira, 2008). Moreover, since the microstructural and rheological properties within the values associated with quality, gamma irradiation does not change the determining characteristics for product acceptance (Guimarães et al., 2013). Thus, this preservation method is effective during the post-harvest storage of many perishable products. Moreover, the combination of different preservation methods can result in improvements in the product. For example, gamma irradiation and a cooling treatment can reduce the level of microbial contamination and improve product shelf life (Hussain et al., 2010).

The present study aimed to quantify antioxidants and their activity in the Eva’s apple cultivar and to evaluate the physical and chemical changes that occur during the post-harvest storage of an Eva’s apple cultivar that was subjected to gamma irradiation.

MATERIAL AND METHODS

Sample preparation

Eva’s apples were harvested during an optimal marketing period (November) in a commercial garden at Barbacena, Minas Gerais. The samples were harvested and selected for their marketable appearance and absence of injuries or diseases. Following harvest, the apples were sanitized for 10 minutes with 100 mg L⁻¹ sodium hypochlorite.

Experimental design

The study was carried out using fruits from the 2009/2010 season and different doses of gamma irradiation. Harvested and sanitized apples were randomly divided into four groups and stored in cleaned Styrofoam boxes at 8 ± 3 °C. The Styrofoam boxes containing the fruits were sent to the National Nuclear Energy Center (CDTN) at Belo Horizonte, where each group was subjected to a different dose of gamma irradiation (0, 0.5, 1.0 or 1.5 kGy). The source used was Co⁶⁰ and the fruit were exposed to pre-defined times according to the dose tested. After the irradiation treatments, the fruits were stored at cold temperatures (0.5 ± 0.5 °C) for 135 days; the antioxidant amounts and activities were assessed every 45 days during the post-harvest storage. We used a completely randomized design (4 x 3), with five replicates per treatment, and each treatment consisted of five fruits.

Measurements

For the assessment of antioxidant activity (DPPH and β-carotene/linolenic acid methods) and the total phenolic compound concentration, 10 g of each sample was homogenized at room temperature in 20 mL 50% methyl alcohol. After an hour, the mixture was centrifuged (14,000 rpm for 17 minutes), and the supernatant was collected. The pellet was then homogenized in 20 mL of 70% acetone; following one hour, it was re-centrifuged (14,000 rpm for 17 minutes). The supernatants obtained from the first and last centrifugations were pooled, and the volume was adjusted to 50 mL with distilled water (Rufino et al., 2007).

The antioxidant activity was determined based on the extinction of absorption by the 2,2-diphenyl-1-picril hydrazyl (DPPH) radical (Rufino et al., 2007). Approximately 0.5 mL of 60 μM DPPH was added to each extract. The control sample consisted of 0.5 mL of methanol with 0.5 mL of 60 μM DPPH. After 30 minutes, absorbance was measured in a Beckman 640 B spectrophotometer at 515 nm, and the results were expressed as percentage of free radical scavenging (% SRL) according to Equation 1:

\[
% \text{ SRL} = \frac{(A_c - A_m)}{A_c} \times 100
\]  

where \( A_c \) = control sample absorbance and \( A_m \) = sample absorbance.

The determination of the total antioxidant activity by the β-carotene/linolenic acid method was performed using 0.4 mL extract to which 5.0 mL of an emulsion
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(containing 40 µL linoleic acid 99%, 530 µg Tween 40 90% and 20 mg β-carotene 93%, all from Sigma) was added. The absorbance was promptly measured in a spectrophotometer at 470 nm. Tubes were then incubated at 40 °C. After 120 minutes (during which oxidation reactions take place), the absorbance was measured again. The results were expressed as percentage of oxidation inhibition. The absorbance reduction in the absence of antioxidants (Equation 2) was considered to be 100%.

The absorbance reduction of the system without antioxidants (Equation 2) was taken as 100% (Rufino et al., 2006).

Absorbance reduction = \( \text{Abs}_{\text{initial}} - \text{Abs}_{\text{final}} \)  \tag{2}

The oxidation percentage was calculated by dividing the reduction in the sample absorbance by the reduction in the system absorbance (Equation 3). To obtain the protection percentage, the oxidation percentage of each sample was subtracted from 100 (Equation 4).

\[
\text{Oxidation \%} = \left( \frac{\text{Reduction Abs}}{\text{Reduction Abs}} \right) \times 100 \tag{3}
\]

\[
\text{Protection \%} = 100 - \text{(Oxidation \%)} \tag{4}
\]

For the determination of total phenolic compound concentration, 2.5 mL of 10% Folin-Ciocalteu and 2 mL of 4% sodium carbonate solutions were added to 0.5 mL of sample extract. The tubes were shaken and then kept in the dark for 2 hours. The blue color that appeared as a result of the Folin-Ciocalteu reagent being reduced by the phenolic compounds was measured in a spectrophotometer at 750 nm. The phenolic concentration of the samples was determined from a gallic acid standard curve. The results were expressed as mg of gallic acid equivalent per 100 g of sample (mg EAG.100g⁻¹) (Waterhouse, 2002).

The ascorbic acid concentration was determined by a colorimetric method using 2,4-dinitrophenylhydrazine and a Beckman 640 B spectrophotometer equipped with a computerized system; the results were expressed as mg of ascorbic acid per 100g⁻¹ of pulp (Strohecker; Henning, 1967).

The background color of the epidermis was measured on opposite sides of the fruit using a Minolta CR 400 colorimeter, set to a CIE L*, a*, b* mode. The coordinate L* is as lighter or darker the sample, with values ranging from 0 (completely black) to 100 (pure white). Soluble solids were assessed using a digital handheld pocket refractometer (Atago, Pocket PAL-1), the results were expressed as °Brix according to the standard method (Association of Official Analytical Chemists - AOAC, 2005).

**Statistical analyses**

The results were analyzed by one-way analysis of variance, and the means were compared using Tukey test at the 5% probability level using the program R Development Core Team software (R Development Core Team, 2010).

**RESULTS AND DISCUSSION**

Before the gamma irradiation, five samples from six fruits were analyzed for baseline characteristics. The fruits had a firmness of 47.36 N (unshelled), a soluble solid content of 11.5 °Brix, a titratable acidity of 0.436% malic acid, an epidermis background color L* value equal to 77.29, a Chroma of 40.3° and a Hue angle of 99.78°H.

Foods that naturally contain antioxidants are interesting because of their nutritional potential and therapeutic effects (Rufino et al., 2009; Lopes et al., 2012). The antioxidant activity, assessed by the DPPH method, revealed a significant increase in the percentage of free radical sequestration during storage to control fruits and treated with doses of 0.5 and 1.5 kGy. However, in fruits that were treated with 1.0 kGy, there was no change in antioxidant activity during storage (Table 1). A decrease in the protection percentage, measured by the β-carotene/linolenic acid method, was obtained in the control and 0.5 kGy-irradiated fruits during storage (Table 1).

The data suggest an increase in free radical sequestration during storage, independent of the irradiation treatment. Similar results were obtained by Camargo et al. (2011) in IAC-Tatu’s peanut cultivars. However, the authors of this study obtained a higher antioxidant activity in IAC-Runner 886 peanuts that were treated with a 15.0 kGy dose of irradiation (Camargo et al., 2011). Authors who studied irradiated juices at doses of 1.0, 3.0 and 5.0 kGy noted either no change or a significant increase in antioxidant activity, which is essential for the product’s preservation (Lee et al., 2009). Similarly, in a study with irradiated alfalfa seeds, the authors noted that irradiation had no effect on the nutritional value but was able to increase the antioxidant activity (Fan; Thayer; Sokorai, 2004). Moreover, Queiroz et al (2011) working with cashews, found that storage did not affect the antioxidant activity of this fruit.
Table 1: Total antioxidant activity (TAA) obtained by two different methods in apples cv. Eva stored for 45, 90 or 135 days (0.5 ± 0.5 °C) and subjected to different gamma irradiation doses at the beginning of the storage.

<table>
<thead>
<tr>
<th>Irradiation dose (kGy)</th>
<th>TAA – DPPH (% SRL)</th>
<th>TAA - β-carotene/linolenic acid (% Protection)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Storage (days)</td>
<td></td>
</tr>
<tr>
<td></td>
<td>45</td>
<td>90</td>
</tr>
<tr>
<td>0</td>
<td>45.4Ab</td>
<td>57.6Aa</td>
</tr>
<tr>
<td>0.5</td>
<td>27.6Bb</td>
<td>31.9Dab</td>
</tr>
<tr>
<td>1.0</td>
<td>41.3Aa</td>
<td>42.7Ca</td>
</tr>
<tr>
<td>1.5</td>
<td>38.5Ac</td>
<td>48.1Bb</td>
</tr>
</tbody>
</table>

Letters represent a significant difference in the row (lowercase letters) or in the column (uppercase letters) by the Tukey test at 5% error probability.

The analysis of the antioxidant activity at different irradiation doses, were found that after 45 days of storage, the control fruits and those treated with the highest irradiation doses had the highest percentage of free radical sequestration and protection percentage as determined by the β-carotene/linolenic acid method. After 90 days of post-harvest storage, the same results were obtained. After 135 days of refrigerated storage (0.5 ± 0.5 °C), the control fruits and those treated with the highest irradiation (1.5 kGy) showed the best results with respect to % SRL and % protection.

The antioxidant activity, as determined by the β-carotene/linolenic acid method, measures the ability of an antioxidant to protect the sample from lipid oxidation; therefore, the method determines the ability to inhibit free radicals generated during the peroxidation of linoleic acid. The β-carotene/linolenic acid method differs from the DPPH method in that it is based on the electron transfer from an antioxidant to a free radical (Duarte-Almeida et al., 2006).

A study that assessed the antioxidant activity of Brazilian fruits revealed that plums and Spondias tuberosa have intermediate levels of oxidation inhibition when assessed by the β-carotene/linolenic acid method (Rufino et al., 2010). Similarly, Eva’s apples have an intermediate level of oxidation inhibition because they showed, a protection ranging from 60 to 73%.

The phenolic compound concentrations are shown in Table 2. Similar to the control fruit, total phenolics decreased in the samples that had been subjected to the 1.0 kGy irradiation dose. Already the fruits subjected to the 0.5 kGy dose showed an increase in total phenolic concentration during storage; we did not detect a significant difference in the samples subjected to the highest dose (1.5 kGy). However, the fruits that had not been irradiated had a more pronounced decrease in total phenolic concentration at the end of the post-harvest storage. Moreover, the fruit responses were inversely proportional to the applied dose; better responses were obtained in the fruits that had been subjected to the lowest doses of irradiation.

In general, the total phenolic compounds concentration remains relatively stable during storage, but some individual compounds may vary (Awad; Jager, 2000; Awad; Jager, 2003). Moreover, processing conditions and prolonged storage promote both chemical and enzymatic oxidation of phenolic compounds, contributing to its reduction (Lee et al., 2003). Studies involving bioactive compounds and antioxidant activity in fruits have shown that both Myrciaria cauliflora and Euterpe oleracea Mart. have intermediate levels of these bioactive compounds (100 to 500 EAG per 100 g of pulp). Thus, the data suggest that the Eva’s apple cultivar has intermediate levels of phenolic compounds.

Ascorbic acid decreased with storage time, except in the 0.5 kGy-irradiated fruits (Table 2). The treatment with the highest irradiation dose resulted in higher ascorbic acid concentrations at 45 and 90 days of storage. In addition, at 45 days, the 1.0 kGy and 1.5 kGy irradiation doses resulted in similar ascorbic acid concentrations. The 0.5 kGy and control groups had the highest of ascorbic acid concentration, suggesting that low doses of irradiation during the post-harvest storage are more effective in maintaining ascorbic acid levels. Lima et al (2009) found that ascorbic acid content was higher in samples of “buriti fruit” subjected to irradiation. Although ascorbic acid is important to human health, this compound represents a small fraction of the apple’s antioxidant activity (Lee et al., 2003; Boyer; Brown; Liu, 2004).

The background color (L* value) and soluble solids (SS) content are shown in Table 3. There were no statistically significant differences in L* values according to irradiation doses and storage time.
A slight increase in SS content was obtained in the control and 1.5 kGy-irradiated fruits. No significant differences were obtained for the other irradiation doses. However, the irradiated fruits had a lower SS content after 90 days of cold storage, indicating lower metabolic activity in these fruits, most likely due to the applied treatment.

**CONCLUSIONS**

Gamma irradiation applied after harvest of Eva’s apple cultivar is able to maintain the ascorbic acid content, color and fruit quality. The highest irradiation doses showed the best results when it comes to percentage of sequestration of free radicals and percentage of protection, and the dose of 1.5 kGy was able to keep the phenolic compounds.

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**REFERENCES**


