Elemental concentration via portable x-ray fluorescence spectrometry: Assessing the impact of water content

Concentração elementar obtida por espectrometria de fluorescência de raios-x portátil: Efeito da umidade

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ABSTRACT

Portable X-ray fluorescence (pXRF) analysis can be considered one of the main recent advances for chemical characterization of earth materials. The water content of the samples can affect the pXRF performance. As a novelty, we aimed to establish relationships (linear regression) between the effect of water content on pXRF results and atomic number (Z) of the elements. Three certified reference materials (CRM) were investigated: OREAS 100a, OREAS 101a, and OREAS 101b. These materials were saturated (0.68 g g⁻¹) with distilled water and left to air-dry naturally. During the drying, the elemental concentrations (C) were determined at different water contents using a pXRF spectrometer. For each water content, the ratio \( C_{\text{wet}}/C_{\text{dry}} \) was determined and plotted against the water content. The attenuation coefficient (σ) was also determined. High σ values mean more influence of water content upon measurement element concentration. The obtained recovery rates allowed a qualitative determination. The concentration for the most elements reduced linearly with increasing water content. A predictable behavior of the water content on pXRF results as function of atomic number was not found. Elements identified by Lα spectral line with highest Z were more impacted by water content than elements identified by Kα line with lowest Z. Ti, Cr and Fe was not significantly influenced by water content, and Sr was the most impacted. Our findings contribute to decision-making before characterization earth materials via pXRF, obliging the use of dry samples for determination of impacted elements or by using moisture-corrected data.

Index terms: Proximal sensors; X-ray scattering; spectroscopy analysis.

INTRODUCTION

The total elemental composition of different materials can be obtained by X-ray fluorescence analysis (Jenkins, 1932; Potts, 2008). In recent years, technological advancements allowed the miniaturization of X-ray fluorescence equipment. Nowadays, there are many models of portable X-ray fluorescence equipment (pXRF)
available (Gazley; Fisher, 2014; Weindorf et al., 2014). The pXRF has been largely and successfully used for industrial, geological, agricultural and environmental purposes (Parsons et al., 2013; Widana; Priadi, 2015; McNulty et al., 2018; Yarborough et al., 2018). The pXRF analysis is fast, non-destructive, environmentally friendly, allowing the determination of several elements simultaneously (e.g., from Mg to U) in-situ or ex-situ within the order of seconds. The Soil Science has been greatly benefited from pXRF analysis (Ribeiro et al., 2017; Ribeiro et al., 2018; Santana et al., 2018; Silva et al., 2018; Mancini et al., 2019, Weindorf et al., 2014; Stockmann et al., 2016).

The principle of X-ray fluorescence analysis is the emission of secondary X-rays by the sample when it is irradiated by a source of primary X-rays (Kalnicky; Singhvi, 2001; Gazley; Fisher, 2014). This occurs when electrons are shifted from inner to outer shells. As a result, to fill inner-shell vacancies the outer electrons move to them. At this moment, energy is released in form of fluorescence. The emitted secondary X-rays are characteristic for each element (spectral signature). The emitted energy (keV) and wavelength are used to identify the element and the intensity of fluorescence allows its quantification (Margui; Grieken, 2013; Weindorf et al., 2014). It is reasonable to consider that the presence of water between the particles to be analyzed would act as a physical barrier for primary X-rays and/or scattering the secondary X-rays (Ge et al., 2005). As a consequence, the obtained concentration may be underestimated (Sahraoui; Hachicha, 2017; Schneider et al., 2015). The effect of water content on pXRF results is still quite divergent in the literature.

The water content during pXRF analysis is considered the main factor affecting the accuracy of derived element concentrations. There is not a consensus about which elements are more influenced by water content and how much water can influence the pXRF analysis. Elements with atomic number (Z) < 26 seem to be more influenced (Ge et al., 2005; Weindorf et al., 2014). However, Zr (Z = 40) and Pb (Z = 82) were also influenced by water content (Hürkamp et al., 2009; Shuttleworth et al., 2014; Hangen; Vieten, 2016). In some Brazilian soil samples the elements Si (Z = 14) and Al (Z = 13) were much more influenced by moisture than Ti (Z = 22) and Fe (Z = 26) (Ribeiro et al., 2018). Frequently, non-effect of water content on Fe quantification via pXRF has been reported (Berger et al., 2009; Stockmann et al., 2016). In general, water content between 5% and 15% does not seem to influence the pXRF results (Laiho; Perämäki, 2005; Piorek, 1997; Peinado et al., 2010). Considering the effect of water content on pXRF results, some correction factors have been proposed (Ge et al., 2005; Bastos et al., 2012; Ribeiro et al., 2018) in order to overcome such constraints. The procedures for pXRF analysis of soils and sediments are summarized in the US EPA 6200 method (US EPA, 2007) and by Weindorf and Chakraborty (2016). According to US EPA 6200 method the pXRF results are minimally impacted with moisture up to 0.20 g g⁻¹ (20% wt).

Thus, we conducted a controlled laboratory experiment using three certified reference materials (CRM) to check for the pXRF results under influence of different water contents. The objectives were to identify elements, which, are more influenced by water content during the pXRF screening. As a novelty, this work aimed to establish relationships between atomic number and attenuation coefficient (σ) based on the Beer-Lambert Law. This parameter indicates how much a given element is influenced by water content. We hypothesized that the water content underestimate the concentration reported by pXRF and this effect decrease as the atomic number of elements increase. If so, the impact of water content as function of atomic number can be modelled.

**MATERIAL AND METHODS**

**Certified reference materials (CRM)**

For this study, three CRM’s for geochemical applications were selected: OREAS 100a, OREAS 101a, and OREAS 101b. These CRM’s are U-bearing materials prepared by Ore Research & Exploration Pty Ltd from geological exploration at Mont Gee, South Australia. The certified values were reported after characterization by different methods: lithium borate fusion for X-ray fluorescence analysis (XRF) and 4-acid digestion (HF, HCl, HNO₃ and HClO₄) for mass spectrometry inductively coupled plasma (ICP-OES/MS). The certified values were obtained at Ore Research & Exploration website (http://www.ore.com.au/oreas-reports/).

**Experimental setup and pXRF measurements**

Polystyrene chloride (PVC) rings (1 cm high and 5 cm diameter) situated at the bottom X-ray thin film (prolene film) were used to receive the CRM’s. The X-ray thin film was held by a rubber. Thus, the rings were packed (in triplicate) with 13 g of each CRM’s resulting in a CRM thickness of 9 mm. Distilled water (9 mL) was added to CRM rings until reaching complete saturation. The CRM rings were left under laboratory conditions to air-dry naturally. The pXRF measurements were performed on CRM samples...
at different water contents (from complete wet condition to air-dried condition). The water contents ranged from 0.68 to 0.02 g g\(^{-1}\). Before each pXRF measurement the samples were weighted to calculate the water content according to Equation 1:

\[
WC = \frac{W_{\text{wet}} - W_{\text{dry}}}{W_{\text{dry}}}
\]  

(1)

Where: WC is the water content (g g\(^{-1}\)); \(W_{\text{wet}}\) is the ring weight (g) containing wet CRM materials; \(W_{\text{dry}}\) is the ring weight (g) containing the equivalent dry CRM materials.

For pXRF measurements a handheld Bruker® spectrometer S1 Titan LE model was used. The operational conditions were: Rh X-ray tube (4W, 15-50 kV, and 5-100 µA); silicon drift detector (SDD) with resolution < 145 eV; Geochem application (Trace Mode); and screening time of 60 s. The dwell time (60 s) is automatically divided in two phases: i) phase 1 (first 30 s): 45 kV is applied and quantified the following elements: from Mn to U; phase 2 (last 30 s): the excitation energy is reduced to 15 kV and quantified the elements: from Mg to Ca. Using the Bruker® S1 Data Tool software, the pXRF report and spectral data were analyzed. The following elements were identified and quantified: i) using the Kα spectral line reference: Mg, Al, Si, P, Cl, K, Ca, Ti, Cr, Mn, Fe, and Co; ii) using the Lα spectral line reference: Cu, Rb, Sr, Zr, Nb, Ce, and Bi. Thus, in this work the studied elements were called as Kα and Lα elements. The pXRF results obtained using oven-dried (105 ºC) samples were used to calculate the recovery percentage taking into account the certified values for CRM’s.

The attenuation coefficient (\(\sigma\)) proposed by Ge et al. (2005) was calculated for each element and water content according to Equation 2:

\[
\frac{C_{\text{wet}}}{C_{\text{dry}}} = e^{-\sigma \cdot WC}
\]  

(2)

Where: \(C_{\text{wet}}\) is the concentration of the element at maximum water content; \(C_{\text{dry}}\) is the concentration of the element in the dry CRM material; \(\sigma\) is the attenuation coefficient (non-dimensional); and WC is the water content.

The attenuation coefficient (\(\sigma\)) can be calculated from the logarithmic form (Equation 3):

\[
\sigma = \frac{\ln C_{\text{wet}} - \ln C_{\text{dry}}}{WC}
\]  

(3)

For each element, the ratio \(C_{\text{wet}}/C_{\text{dry}}\) was plotted onto the Y axis and water content (WC) onto the X axis. Thus, linear and exponential models were adjusted to data using R-studio software and graphs were obtained using Sigma Plot software version 12.5. The attenuation coefficients (\(\sigma\)) were correlated to the atomic number of elements.

RESULTS AND DISCUSSION

Recovery rates of elements quantified by pXRF

In Table 1 the mean recovery rates of elements for each CRM using pXRF analysis are shown. The best recovery rate was obtained for Rb (100%) in the OREAS 100a. The worst recovery rates were obtained for Mg, P, Co, Ce, and Bi. For these elements the pXRF results were significantly higher than the certified reference values. Light elements (Si, K, and Ca) had low recovery rates (below 50%). XRF analysis becomes more inaccurate for the lightest elements since they have low fluorescence yield and can be influenced by fluorescence of heaviest elements (Magui; Grieken, 2013). The elements Al, Ti, Cr, Mn, Fe, Cu, Zn, Rb, Sr, Zr, and Nb had recovery rates between 60 and 200%. In addition, the recovery rates obtained in the three CRM’s were similar. A non-satisfactory recovery (e.g., far from 100%) may be due to: i) non-proper calibration of the equipment by the manufacturer; and ii) matrix effect causing spectral interference (e.g., S and Mo; Cl and Rh; As and Pb; V and Cr; Fe and Co) (Kalnicky; Singhvi, 2001).

Sinnesael et al. (2018) using a pXRF Bruker® Tracer IV model and seven CRM’s [CCH-1 (Liege University), COQ-1 (United States Geological Survey, USGS), CRM393 (Bureau of Analysed Samples, BAS), CRM512 (BAS), CRM513 (BAS), ECRM-782-1 (BAS) and SRM-1d (National Institute of Standards and Technology, NIST)], found the recovery rates (%) of Fe, Mn, Sp, and Ti of 57, 101, 115 and 379, respectively. NIST standard reference material (2711a), ERM® (European Reference Materials) CC141, and ISE (International Soil analytical Exchange) were used for quality assurance and control of pXRF analysis using an Olympus model Delta Professional (Caporale et al., 2018). These authors screened during 30 s (Geochem Application: Soil Mode) and provide contents of As, Ca, Cu, Cr, K, Fe, Mn, Ni,
Pb, Ti, Sr, V, and Zn. The recovery rates ranged from 82 to 116% and the correction factors \( \frac{\text{pXRF read}}{\text{certified value}} \) were determined. Using the NIST standard reference material (278), 12 elements (K, Ca, Ti, Mn, Fe, Zn, Rb, Sr, Zr, Nb, Pb, and Th) were identified and quantified using a pXRF Olympus Delta Premium (Ta X-ray tube), Obsidian-soil mode, and the screening time ranged from 5 to 100 s (Newlander et al., 2015). The highest relative standard errors (RSD - %) were obtained for Th (29), Zn (19), Pb (17), and Mn (10). The authors conclude that the accuracy of pXRF measurement is not influenced by the increase of screening time. However, the screening time can significantly influence the pXRF results, mainly the detection limit and repeatability (Gazley; Fisher, 2014). The different recovery rates found in the literature can be attributed to the intrinsic features of operational conditions and to the different characteristic of samples (soils, sediments, rocks, etc.).

Some agricultural soils were accurately characterized via pXRF analysis (Hu et al., 2014). The authors used a pXRF Niton® XT 960 model to assess the As, Pb, Cu, and Zn concentrations. The CRM GBW07401 was scanned for 90 s and the recoveries for As, Pb, and Cu ranged from 98 to 109%, and 117% for Zn. In our work the best recovery rate found for Zn was 183% (OREAS 100 a) (Table 1). As and Pb were below the detection limit.

**Effect of water content on elemental concentration obtained via pXRF**

In Table 2 the significance and \( R^2 \) values (coefficient of determination) for regression analysis performed between elemental concentration \( \frac{C_{\text{wet}}}{C_{\text{dry}}} \) ratio and water content are shown. The linear and exponential regression models were significant showing similar \( R^2 \) values for the three CRM’s. Only the elements having recoveries between 60 and 200% were considered. These

| Table 1: Mean recovery rates (%) of elements identified and quantified by pXRF Bruker® spectrometer S1 Titan LE model. |
|---|---|---|---|---|
| Element | Z | Spectral line | keV | LOD (mg kg\(^{-1}\)) |
| Mg | 12 | Ka | 1.254 | 9,000 |
| Al | 13 | Ka | 1.486 | 1,150 |
| Si | 14 | Ka | 1.740 | 500 |
| P | 15 | Ka | 2.010 | 170 |
| K | 19 | Ka | 3.314 | 55 |
| Ca | 20 | Ka | 3.692 | 50 |
| Ti | 22 | Ka | 4.512 | 21 |
| Cr | 24 | Ka | 5.415 | 30 |
| Mn | 25 | Ka | 5.900 | 30 |
| Fe | 26 | Ka | 6.405 | 19 |
| Co | 27 | Ka | 6.931 | 2 |
| Cu | 29 | La | 0.928 | 5 |
| Zn | 30 | La | 1.012 | 4 |
| Rb | 37 | La | 1.692 | 3 |
| Sr | 38 | La | 1.806 | 4 |
| Zr | 40 | La | 2.044 | 4 |
| Nb | 41 | La | 2.169 | 4 |
| Ce | 58 | La | 4.439 | 35 |
| Bi | 83 | La | 10.839 | 17 |

\( \text{Recovery rate of 100%: pXRF result is equal to certified reference value; Recovery rate above 100%: pXRF result is higher than certified reference value; Recovery rate below 100%: pXRF result is lower than certified reference value. LOD: limit of detection.} \)
comprise Al, Ti, Cr, Mn, Fe, Cu, Zn, Rb, Sr, Zr, and Nb. The quantification of Ti (Z=22), Cr (Z=24), and Fe (Z=26) was not influenced by water content. The tested regression models were not significant for these elements, except for Fe (OREAS 100a linear model), but with low $R^2$ (0.26). For the other elements, a predictable behavior was identified, indicating the effect of water content on elemental concentration obtained by pXRF, as can be seen in Figure 1 indicating a diluting effect of the water content upon element concentration.

For each element, the slopes of regression lines were similar for the CRM’s (Figure 1). The Lα elements (Cu, Zn, Rb, Sr, Zr, and Nb) showed a steeper slope than the Kα elements (Al, Mn, and Fe). The slope ranged from 0.65 (Zn) to 0.75 (Zr) in the Lα group, and from 0.20 (Fe) to 0.33 (Al) in the Kα group. The mean slope for the Lα group was 2.7 times higher than for the Kα group. Taking the slope as a parameter to describe how much the elements concentrations were impacted by the water content, in the Kα group it decreased as follows: Al > Mn > Fe. The effect of water content on the quantification of Fe using pXRF can be considered quite insignificant. A significant effect was observed only for OREAS 100a, but with very low slope and $R^2$ values of regression. For the elements of the Lα group the slope declined as follows: Nb > Zr > Rb > Cu > Sr > Zn.

Based on the Beer-Lambert Law, the $\sigma$ parameter (Equations 2 and 3) was presumed to assess the effect of water content on elemental concentration obtained via pXRF analysis (Ge et al., 2005). In our work, $\sigma$ was calculated for each element according to Equations 2 and 3 and correlated to atomic number (Z) and X-ray fluorescence energy values (Figure 2). There was not a consistent and predictable relationship between $\sigma$ and atomic number corroborating other works (Sahraoui; Hachicha, 2017; Schneider et al., 2015). Conversely, some works related a consistent relationship between atomic number and the effect of water content on signal reduction, underestimating the results (Ge et al., 2005; Kido et al., 2006).

Parsons et al. (2013) suggested that low-Z elements are predominantly influenced by water content. In contrast, in our study the Kα elements (Al, Ti, Cr, Mn, and Fe) showed lower $\sigma$ compared to Lα elements (Figure 2). This indicates that elements with higher atomic number (e.g., $^{41}$Nb) may be more influenced by water content than lighter elements (e.g., $^{13}$Al). Screening the surface of undisturbed samples of tropical soils, Si and Al (light elements) were significantly more influenced by increase of water content than Fe and Ti (heaviest elements) (Ribeiro et al., 2018). In each group (Kα or Lα), a consistent and predictable relationship between $\sigma$ and atomic number or fluorescence energy values (keV) was not detected. Thus, the effect of water content on elemental concentration obtained via pXRF might be related to a combination of X-ray fluorescence energy line (keV) and the intrinsic characteristic of the element. Kalnicky and Singhvi (2001) suggested that elements with X-ray energy values smaller than 5 keV are more impacted by water content. This corroborates to our findings. The Lα

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<th>Element</th>
<th>OREAS 100a</th>
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<tr>
<td>Mn</td>
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<td>0.31*</td>
<td>0.29*</td>
<td>0.26*</td>
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<tr>
<td>Fe</td>
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<tr>
<td>Cu</td>
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<td>0.90*</td>
<td>0.90*</td>
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<tr>
<td>Zn</td>
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<td>0.57*</td>
<td>0.48*</td>
<td>0.45*</td>
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<td>Rb</td>
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<tr>
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<tr>
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Table 2: $R^2$-values of the regression analysis between $C_{wet}/C_{dry}$ ratio and water content (ranging from 0.68 g g$^{-1}$ to air-dry). All regressions are significant at the p-value < 0.05.
Figure 1: Effect of water content on $C_{wet}/C_{dry}$ ratio for Al, Ti, Cr, Mn, Fe, Cu, Zn, Rb, Sr, Zr, and Nb of the three CRMs (OREAS 100a, OREAS 101a, and OREAS 101b) using pXRF; ns = not significant.
group, which is more impacted than the Kα group, has keV values ranging from 0.928 to 2.169. In the Kα group, Cr, Mn, and Fe have characteristic X-ray energy values higher than 5 keV and Ti close to 5 keV, respectively. Only Al has a keV-value of 1.486.

In a similar study, Schneider et al. (2015) studied the effect of water content on determination of 11 elements. Fe, Cu, Mn, Pb, Rb, Sr, and Zn showed the lowest σ values (ranging from 0.87 to 0.99), and Ba, Cr, and Sn the highest values (1.41, 1.28 and 1.03, respectively). Shuttleworth et al. (2014) assessed Pb in contaminated Histosols, comparing acid digestion method (US EPA 3051a) followed by inductively coupled plasma optical emission spectrometry (ICP-OES) to pXRF analysis (in-situ and ex-situ). The authors found a significant difference between wet and dry samples on their results. However, the moisture-corrected data and in-situ pXRF analysis showed a strong correlation to ICP-OES results. Sahraoui and Hachicha (2017) assessed the concentrations of Ca, Mg, Cd, Cr, Cu, Ni, Zn, P, Fe, Mn and Pb via pXRF in 60 soil samples (Tunisia) at different moisture: in-situ, oven-dried, 0.40 g g⁻¹ (40% wt) and complete saturated. They found the elements may be differently impacted according to the water content, and the magnitude of the water content effect on pXRF results seems to be also related to its own content. For instance, Ribeiro et al. (2018) modelled the effect of water content on the determination of SiO₂ and Al₂O₃ in tropical soils adjusting a power function. The effect of water content is more pronounced in the wettest samples (soil moisture > 0.20 g g⁻¹). These findings corroborate to the recommendations of US EPA 6200 method in which it is reported that the pXRF results may be minimally impacted by moisture between 5 to 20%.

![Figure 2: Attenuation coefficient (σ) for Al, Ti, Cr, Mn, Fe, Cu, Zn, Rb, Sr, Zr, and Nb specified for Kα (blue) and Lα (red) X-rays characteristic energies, calculated for the maximum water content (0.68 g g⁻¹).](image-url)
CONCLUSIONS

The qualitative elemental concentrations of the most assessed elements by pXRF significantly reduced linearly with increasing water content. Conversely to our hypothesis, it was not possible a modelling of the impact of water content, assessed by the attenuation coefficient, as function of atomic number. In this work, elements with highest atomic number and identified by Lα spectral line (Cu, Zn, Rb, Sr, Zr and Nb) were the most impacted by water content. Ti, Cr and Fe were not influenced by water, and Sr was the most impacted element. Our findings contribute to pXRF users to identify the elements impacted or non-impacted by water content, contributing to decision-making before characterization of earth materials. Dry samples are required for the determination of the impacted elements via pXRF or by using moisture-corrected data.

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