FUNCTIONAL CHARACTERIZATION OF ACETYLATED BRAZIL NUT (Bertholletia excelsa HBK) KERNEL GLOBULIN

Cintia Maria Pinto RAMOS2, Pushkar Singh BORA2.*

SUMMARY

Defatted Brazil nut kernel flour, a rich source of high quality proteins, is presently being utilized in the formulation of animal feeds. One of the possible ways to improve its utilization for human consumption is through improvement in its functional properties. In the present study, changes in some of the functional properties of Brazil nut kernel globulin were evaluated after acetylation at 58.6, 66.2 and 75.3% levels. The solubility of acetylated globulin was improved above pH 6.0 but was reduced in the pH range of 3.0–4.0. Water and oil absorption capacity, as well as the viscosity increased with increase in the level of acetylation. Level of modification also influenced the emulsifying capacity: decreased at pH 3.0, but increased at pH 7.0 and 9.0. Highest emulsion activity (approximately 62.2%) was observed at pH 3.0 followed by pH 9.0 and pH 7.0 and least (about 11.8%) at pH 5.0. Emulsion stability also followed similar behavior as that of emulsion activity.

Keywords: Brazil nut kernelglobulin; chemical modification; acetylation; and functional properties.

1 – INTRODUCTION

The production of Brazil nut kernel is a result of extractive activities and has become one of the principal commercial products from the northern region of Brazil [15]. Major part of its production is exported to other parts of the country as well as to other countries. The broken, damaged or deformed nut kernels are sold at a very low price and are utilized principally by small-scale oil industries. The defatted kernel flour, in spite of being rich source of sulfur containing amino acids [24] is presently used in animal feed formulations. However, analyzing the results published by GLORIA & REGITANO-D’ARCE [6] on the functionality of Brazil nut kernel globulin may be helpful to evaluate its use in food formulations. Thus, the present work was undertaken to study the effect of different levels of acetylation on some of its functional properties.

2 – MATERIALS AND METHODS

2.1 – Material

Brazil nut (Bertholletia excelsa HBK) kernels were acquired from local market in Belém City in the State of Pará. The dried (50°C for 24 hrs) kernels were ground in a cyclone sample mill with 0.5-mm screen (UD Corporation, Boulder, CO, USA). The flour was defatted with hexane in a Soxhlet apparatus and was used for the preparation of globulin isolate.

2.2 – Preparation of globulin isolate

Brazil nut kernel globulin isolate was prepared by the method described by KOYORO & POWERS [13] with...
Acetylation of Brazil nut kernel globulin was carried out according to the method described by GRONINGER [8]. The pH of the globulin solution was adjusted in the range of 8.0 to 8.5 with 1N NaOH and cooled in an ice bath to reduce the temperature to 2 to 5°C. Acetic anhydride was added to globulin solution at a concentration of 0.1, 0.15 and 0.25g/g of protein. The pH of the solution was maintained with constant agitation. The reaction was considered to be complete when the pH of the solution was stabilized. Acetylated globulin were isoelectrically precipitated, centrifuged at 15.000g at 4°C for 20min, lyophilized and stored in glass vials at -20°C.

The extent of acetylation was measured as colored lysine ninhydrin derivatives produced in dimethylsulfoxide system. The difference in absorbance between dimethylsulfoxide derivatives of native and modified globulin at 580nm was used as an index of the extent of chemical modification [1].

Protein content was determined by biuret method [7]. Bovine serum albumin (Sigma) was used as a standard for calibration.

2.3 – Acetylation of globulin

2.4 – Functional properties

2.4.1 – Solubility

One hundred milligram lyophilized native and acetylated globulin were suspended in 20mL distilled water and the pH of the suspensions was adjusted from 2.0 to 12.0 using 0.1N HCl or NaOH solution. The suspensions were agitated with a magnetic stirrer for 30min at room temperature; the pH was checked and adjusted, then centrifuged at 4.300g for 30min. The quantity of protein dissolved in supernatant was determined.

2.4.2 – Water and oil absorption

Water and oil absorption were determined according to the method described by BEUCHAT [2]. Five hundred-milligram samples were dispersed in 5-mL distilled water or refined soy oil (Soya brand, Bunge alimentos S.A) and placed in 10mL graduated centrifuge tubes. The dispersions were stirred occasionally with a glass rod. After a holding period of 30min, dispersions were centrifuged at 3.000g for 10min and the volume of the released fluid was measured. Water and oil absorption was expressed as ml of water or oil retained per g of globulin.

2.4.3 – Relative viscosity

Relative viscosity of 1, 2 and 3% native and acetylated globulin solutions in 0.5M NaCl, 50mM potassium phosphate buffer, pH 7.2 at room temperature (29°C) was determined before and after heating to 90°C for 15min in an Ostwald viscometer.

2.4.4 – Emulsifying properties

Emulsifying capacity was determined using the methods of KATO et al. [11] and HUNG & ZAYAS [10]. Native and acetylated globulin dispersions (1.0mg/mL) were adjusted to pH 3.0, 5.0, 7.0 and 9.0. Fifty-ml volumes of dispersions were transferred to the glass cup of emulsification apparatus [18] containing at its bottom a pair of electrodes connected to a galvanometer. The dispersions were agitated at 10,000rpm with a constant flux (22mL/min) of refined soy oil. The quantity of oil necessary for the inversion of emulsion (characterized by a sudden drop in electric current) was measured. The result was expressed as mL oil/100mg globulin.

Emulsifying activity and stability of globulin were determined by the method of YASUMATSU, et al. [23]. Twenty ml portions of globulin solution (25mg/mL) of varying pH (3.0, 5.0, 7.0 and 9.0) were homogenized with 20mL refined soy oil at a speed of 5 in a scale varying from 1 to 10 of Omni Sorvall mixer, for 1min. The emulsions were centrifuged at 1.100g for 5min. The height of emulsified layer and that of the total contents in the tube was measured. The emulsifying activity was calculated as:

\[
\text{Height of emulsified layer in the tube} \times 100 \div \text{Height of the total contents in the tube}
\]

To determine the emulsion stability, the emulsions were heated at 80°C for 30min and centrifuged again. Emulsion stability was calculated using the same formula as above.

Five repetitions were carried out for the determination of functional properties except that of solubility for which three repetitions were made.

2.5 – Statistical analysis

Statistical analysis of the results was done with Statistics for Windows 5.0 [20] and the t-test was used to determine significance of differences between means. Trends were considered significant when means of compared sets differed at p >0.05.
3 – RESULTS AND DISCUSSION

The extent of modification expressed as percentage of acetylated e-amino groups of lysine of Brazil nut globulin obtained with different concentrations of acetic anhydride was as follows:

<table>
<thead>
<tr>
<th>Acetic anhydride (g/g globulin)</th>
<th>% Acetylation of amino groups of lysine</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.00</td>
<td>0.0</td>
</tr>
<tr>
<td>0.10</td>
<td>58.6</td>
</tr>
<tr>
<td>0.15</td>
<td>66.2</td>
</tr>
<tr>
<td>0.25</td>
<td>75.3</td>
</tr>
</tbody>
</table>

The solubility profile of native and acetylated Brazil nut kernel globulin in the pH range of 2.0 to 12.0 is shown in Figure 1. The isoelectric pH of the native globulin was at 5.0 which is in agreement with the isoelectric pH (4.0 to 5.0) of most of the plant proteins [22], while for acetylated globulin independent of their level of acetylation, it was at pH 4.0. The shift reflects an increase in the net negative charge as a result of replacing the positively charged amino groups of lysine with neutrally charged hydrophobic acetyl groups. According to the SHEEN [7], SITOHY, SHAROBEEM & ABDEL-GHANI [10], DUA et al. [4] AND GRUENER & ISMOND [9] acylated proteins exhibit a shift in their isoelectric pH thus resulting in an increased solubility at neutral to alkaline pH. In comparison to native globulin, acetylated globulin in our study have also shown enhanced solubility at pH above 6.0 but was greatly reduced at pH 3.0 to 4.0. Among different acetylated globulin samples, very little difference in solubility was observed.

![FIGURE 1. Solubility of native and acetylated brazil nut kernel globulin as a function of pH](image)

The data on the water absorption capacities of acetylated globulin is shown in Table 1. Relative to native globulin, there was a marked improvement (approximately 70 to 85%) in water absorption capacity of acetylated globulin. Statistically, significant difference at the level of p > 0.05 among globulin samples with increase in the level of acetylation was observed. Similarly, improvement in the water absorption capacity of proteins by acetylation has also been reported for mung bean [5], rapeseed [4] and oat protein isolates [16]. SHEEN [17] and PONNAMPALAM et al. [16] also observed an increase in water absorption with increase in the extent acetylation for tobacco leaf and oat proteins.

Acetylation also improved oil absorption capacity of Brazil nut kernel globulin. Similar to water absorption capacity, significant difference in the oil absorption capacity among samples of different level of acetylation was observed. The oil absorption of 58.6, 66.2 and 75.3% acetylated globulin samples increased about 74.6, 81.5 and 93.8% in comparison to native globulin. EL-ADAWY [5], GRUENER & ISMOND [9], DUA et al. [4] and PONNAMPALAM et al. [16] have reported improvement in oil absorption capacity of various acetylated protein isolates. However, contrary to our study with acetylated Brazil nut kernel globulin, DUA et al. [4] and PONNAMPALAM et al. [16] observed decrease in oil absorption with increase in the level of acetylation.

The relative viscosity of acetylated Brazil nut globulin is also shown in Table 1. Significant difference among the relative viscosity of different acetylated samples was observed. The relative viscosity of the acetylated Brazil nut kernel globulin solutions increased with the level of acetylation. For 58.6, 66.2 and 75.3% acetylated globulin samples, in most of the cases, there was a progressive increase in relative viscosity of 1.0, 2.0 and 3.0% solutions at room temperature before and after heating at 90°C for 15min. In comparison to unheated samples, significant increase in relative viscosity was observed in all heated samples.

Emulsifying properties of native and modified Brazil nut kernel globulin as a function of pH is shown in Table 2. Emulsifying capacity of the samples varied with the pH as well as with the degree of acetylation. At pH 3.0 native globulin, due to its better solubility than acetylated globulin, possessed higher emulsifying capacity. However, the extent of acetylation greatly influenced emulsifying capacity, higher the extent of acetylation, lower was the emulsifying capacity. At pH 5.0, all samples presented minimum but almost similar emulsifying capacities. Contrary to pH 3.0, emulsifying capacity of acetylated globulin improved at pH 7.0 and 9.0, which increased with the degree of acetylation. No statistical difference was observed in emulsion activity and stability of acetylated globulin, independent of their degree of acetylation. However, with respect to pH, best emulsion activity and stability was observed at pH 3.0. Difference in emulsifying properties of acetylated proteins has appeared in scientific literature. GRUENER & ISMOND [9] and PONNAMPALAM et al. [16] reported an increase in emulsion activity and stability, while according to DUA et al. [4] acetylation adversely affected emulsifying properties of rapeseed proteins.
TABLE 1. Water and oil absorption capacities, and viscosity of native and acetylated Brazil nut kernel globulins

<table>
<thead>
<tr>
<th>Sample</th>
<th>WAC (ml/g protein)</th>
<th>OAC (ml/g protein)</th>
<th>Relative viscosity of globulin solution at room temperature (29°C)</th>
<th>Before heating</th>
<th>After heating at 90°C for 15 min (1%) (3%) (5%) (1%) (3%) (5%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Native globulin</td>
<td>1.96±0.04 a</td>
<td>1.47±0.02 a</td>
<td>pH 3.0 pH 5.0 pH 7.0 pH 9.0</td>
<td>52±1.2 a D</td>
<td>1.42±0.03 a C 1.11±0.03 a B 1.18±0.02 a B 1.51±0.03 a D</td>
</tr>
<tr>
<td>58.6% acetylated globulin</td>
<td>3.34±0.02 b</td>
<td>2.55±0.04 b</td>
<td>pH 3.0 pH 5.0 pH 7.0 pH 9.0</td>
<td>66.2±1.2 b D</td>
<td>1.74±0.03 b C 1.37±0.04 b B 1.53±0.05 b D 1.86±0.04 b E</td>
</tr>
<tr>
<td>66.2% acetylated globulin</td>
<td>3.51±0.01 c</td>
<td>2.65±0.03 b</td>
<td>pH 3.0 pH 5.0 pH 7.0 pH 9.0</td>
<td>75.3±1.0 c D</td>
<td>1.91±0.02 c B 1.59±0.05 c B 1.84±0.04 c D 2.01±0.03 c E</td>
</tr>
<tr>
<td>75.3% acetylated globulin</td>
<td>3.65±0.01 d</td>
<td>2.83±0.05 c</td>
<td>pH 3.0 pH 5.0 pH 7.0 pH 9.0</td>
<td>86±0.02 d D</td>
<td>1.99±0.04 d C 1.86±0.02 d B 2.06±0.05 d CD 2.12±0.04 d D</td>
</tr>
</tbody>
</table>

Mean in each column and row followed by different (small and capital letters, respectively) were significantly different (p > 0.05 level)

WAC – Water absorption capacity, OAC – Oil absorption capacity

TABLE 2. Emulsifying properties of native and acetylated Brazil nut Kernel globulin

<table>
<thead>
<tr>
<th>Emulsifying Property</th>
<th>Globulin Samples</th>
<th>pH 3.0</th>
<th>pH 5.0</th>
<th>pH 7.0</th>
<th>pH 9.0</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Native Globulin</td>
<td>93.9±1.2 a A 38.8±0.8 a B 48.4±1.0 a C 52.4±1.2 a D</td>
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<tr>
<td></td>
<td>58.6% acetylated</td>
<td>86.5±0.9 b a 38.8±0.1 a B 66.7±1.1 b C 96.0±1.2 b D</td>
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<tr>
<td></td>
<td>globulin</td>
<td>74.4±1.0 c A 39.6±0.9 a B 74.2±1.2 c A 99.5±1.6 c C</td>
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<tr>
<td></td>
<td>66.2% acetylated</td>
<td>66.7±1.3 d A 41.4±0.2 b B 78.6±1.5 c D 106.5±1.4 d D</td>
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<tr>
<td></td>
<td>globulin</td>
<td>63.8±1.4 a A 11.8±0.9 a B 51.5±0.9 a C 53.2±1.2 a C</td>
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<tr>
<td></td>
<td>58.6% acetylated</td>
<td>62.3±1.1 a A 11.8±1.0 a B 51.9±1.6 a C 54.0±1.3 a C</td>
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<tr>
<td></td>
<td>globulin</td>
<td>62.3±0.8 a A 11.7±0.9 a B 52.0±1.1 a C 54.5±1.1 a D</td>
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<tr>
<td></td>
<td>66.2% acetylated</td>
<td>62.2±1.1 a A 11.9±1.1 a B 52.1±1.2 a C 54.8±1.0 a D</td>
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<tr>
<td></td>
<td>globulin</td>
<td>62.8±1.4 a A 11.7±0.9 a B 51.2±1.0 a C 52.6±1.1 a C</td>
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<tr>
<td></td>
<td>58.6% acetylated</td>
<td>62.2±1.2 a A 11.7±0.8 a B 51.0±1.0 a C 52.9±1.2 a C</td>
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<tr>
<td></td>
<td>globulin</td>
<td>62.2±0.9 a A 11.8±0.8 a B 51.3±0.8 a C 54.5±1.0 a D</td>
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<td></td>
<td>66.2% acetylated</td>
<td>62.2±0.9 a A 11.8±1.0 a B 51.5±1.1 a C 54.6±1.1 a D</td>
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<td></td>
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</tr>
</tbody>
</table>

Mean in each column and row followed by different (small and capital letters, respectively) were significantly different (p > 0.05 level)

4 – CONCLUSIONS

Acetylation modified the functional properties of the Brazil nut kernel globulin. In comparison with the native globulin, the solubility was appreciably enhanced above pH 6.0, while the solubility was decreased at pH range of 3.0 to 4.0. Acetylated globulin showed a shift in their isoelectric pH from 5.0 of that of native globulin to 4.0. The solubility behavior of acetylated globulin suggests their use in the formulation of high protein drinks with their pH near neutrality. Water and oil absorption capacity of acetylated globulin was also improved showing their potential as an ingredient in the formulation of sausage type of products. Small increase in relative viscosity of modified globulin also suggestive of their use in the formulation of high protein soups. High emulsifying capacity, emulsion activity and stability at pH around 7.0 to 9.0 is also indicative of the use of Brazil nut kernel acetylated globulin as an emulsifying agent.

5 – REFERENCES

[6] GLORIA, M. M. & REGITANO-D’ARCE, M. A B. Concen-

Functional characterization of acetylated Brazil nut (Bertholetia excelsa HBK) kernel globulin, Bora & Ramos


6 – ACKNOWLEDGEMENT

One of the authors, Ramos, C. M. P; thanks Conselho Nacional de Pesquisa e Desenvolvimento Tecnológico (CNPq), Brazil for financial Assistance during postgraduate study.