ABSTRACT

The aim of this study was to evaluate the features of dentinal tubules occlusion following application of three commercially available desensitizing agents: potassium oxalate-based / Oxa-Gel (OX), HEMA and glutaraldehyde-based / Gluma Desensitizer (GD) and acidulated phosphate fluoride-based / Nupro Gel (AF). Buccal cervical areas of twenty-four extracted human third molars were smoothed and wet-polished with SiC papers and diamond pastes, in order to simulate the clinical aspect of hypersensitive dentin cervical surfaces. The teeth were randomly divided into four groups (n=6), according to the dentin surface treatments: G1: untreated; G2: OX; G3: GD; G4: AF. Specimens were fractured in the lingual-buccal direction and prepared for SEM analysis. OX promoted tubule occlusion by crystal-like deposits in the lumen of the tubules. While GL created a thin layer over the dentin surface, AF application produced precipitates that occluded the tubules. According to the SEM analysis, all desensitizing agents were able to occlude the dentinal tubules.

UNITERMS: Dentin hypersensitivity; Desensitizing agents; Dentinal tubule occlusion.

INTRODUCTION

Previous studies have shown that hypersensitive areas may present exposed dentinal tubules. The hydrodynamic theory assumes that a stimulus applied on the dentin surface causes movement of tubular fluid, which in turn activates mechanoreceptor nerves, eliciting pain and discomfort.

According to this theory, if the functional radius of open dentinal tubules decreases, then the permeability is also decreased, reducing dentin sensitivity. Thus, treatments for hypersensitivity should occlude dentinal tubules and prevent nerve sensitivity. Therefore, some products have been indicated to treat dentin hypersensitivity due to their ability to occlude tubules.
Dentinal tubules can be obliterated on the surface and/or occluded within the tubule orifices. However, superficial occlusion of tubules can be removed by daily tooth brushing, dissolution of the precipitate promoted by saliva or consumption of acidic beverages, leading to short-term desensitizing effects. Effective treatment with long-term results has been related to intratubular deposition, which reduces the fluid flow rate or totally seals the tubule lumen.

Several previous studies have only assessed the effects of selected desensitizing agents on the dentin surface, which do not support any evidence about the features of these products on tubule occlusion and as such does not determine their potential effectiveness in reducing hypersensitivity over time. Since several desensitizing agents have been shown to promote significant tubule occlusion in vitro, these may also have the ability to be clinically effective. Therefore, the purpose of this study was to evaluate the ability of three selected desensitizing agents on dentinal tubule occlusion and morphological tubule changes using scanning electron microscopy (SEM).

MATERIALS AND METHODS

The specimens were prepared from twenty-four freshly extracted sound human third molars stored in 10% formalin (pH 7.0) at room temperature. The teeth were obtained after informed consent of the patients and under the protocol (38/2003) analyzed and approved by the Ethics Committee of Piracicaba Dental School / UNICAMP, Brazil. The teeth were cleaned of gross debris and placed in deionized water for twenty-four hours before beginning the experiment.

The teeth were sectioned in mesiodistal direction using a water-cooled diamond saw. From each buccal surface, a sectioned sample (5mm length x 5mm width x 3.5mm depth) was obtained including the cervical area. The exposed dentin surfaces were wet-polished with 1000- and 1200-grit aluminum oxide abrasive paper and diamond pastes (6mm, 3mm, 1mm and ¼mm), simulating dentin hypersensitivite in cervical regions. Specimens were ultrasonicated for 12 min and rinsed with deionized water.

The specimens were randomly assigned to four groups (n=6), according to the dentin surface treatments followed, as recommended by the manufacturer’s instructions:

- Group 1- untreated surface (control).
- Group 2- Oxa-Gel applied for 30s.
- Group 3- Gluma Desensitizer applied for 30s.
- Group 4- Nupro Gel applied for 4min.

Table 1 describes the composition of tested desensitizing agents. Specimens were fixed in Karnovisk solution for 1 hour and fractured with a sharp blade placed perpendicular to the buccal surface. Each fractured sample was sputter coated (MED 010, Balzers) with a thin gold layer and examined under SEM (VP 435, Leo). Photomicrographs of representative dentin surface areas were taken at magnification between 6,000x and 15,000x.

RESULTS

Photomicrographs for untreated, control group and desensitizing agents applied on dentin surface are shown in Figures 1 to 4. Examined untreated areas presented a smooth appearance and presented opened tubules orifices. Moreover, no smear layer covering dentin and no smear plugs closing tubules were observed (Figure 1a and 1b).

SEM observation of the selected desensitizing agents presented different modes of tubular occlusion. Figures 2A and 2B are micrographs of dentinal surfaces treated with oxalate potassium-based desensitizing agent. Precipitation of oxalate crystals was observed within the tubules. Some crystals had approximately the same diameter of the tubules. A thin layer of approximately 1µm thick was observed covering the treated surfaces with Gluma Desensitizer (Figure 3). The majority of dentinal tubules were obliterated with a coat that covered the surface and infiltrated into tubules as plugs. Figures 4A and 4B showed the fractured dentin surface treated with acidulated phosphate fluoride. Precipitates were observed closing most dentinal tubules;
however, they were not attached to the tubules walls. A mild mineral removal of peritubular dentin was observed at the dentinal tubules entrance.

**DISCUSSION**

This study evaluated the occlusion of dentinal tubules by three desensitizing agents. The active ingredient of Oxagel, Desensitizer and Nupro Gel are potassium oxalate,

**FIGURE 2**- SEM micrograph of potassium oxalate-treated dentin. Note the crystal formation extending into the tubules. Areas between parentheses exhibit large formation of crystals within the dentin tubules. (A and B: x 7,000)

**FIGURE 3**- SEM micrograph of HEMA and glutaraldehyde-treated dentin. A thin layer can be seen over the dentin (between arrows), occluding the entrance of dentinal tubules. (A: x 10,000; B: x 15,000)

**FIGURE 4**- SEM micrograph of acidulated phosphate fluoride-treated dentin. Note the precipitates within the tubules and below the surface (white arrow), and the funnel shape dentinal tubules opening (black arrows). (A: x 10,000; B: x 15,000)
Calcium fluoride might occlude most dentinal tubules, its precipitation on and within the tubules. Even though fluoride, the raising of ionized calcium concentration in the tubular fluid and that liberated during the fluoride gel was able to etch peritubular dentin and form substantial crystal-like deposits within the tubule lumen. The gel was able to penetrate into the tubule and occlude a large portion of the tubule diameters. Besides, the tube occluding properties of potassium oxalate might be combined with the inhibitory property of potassium on intradental nerves. This theory is based on the fact that the increased potassium ion concentration ([K⁺]) in the extracellular fluids into the tubules may cause a sustained depolarization of the nerves. This will result in inactivation of action potential generation through a mechanism such as axonal accommodation.

Gluma Desensitizer is an aqueous solution containing 5% glutaraldehyde and 35% HEMA. Because glutaraldehyde is a biological fixative, it has been suggested that the dentinal tubules are occluded as an effect of reaction with plasma proteins from dentinal fluid. Although studies displayed transverse septa in the lumen of dentinal tubules as a result of glutaraldehyde action, no septum was observed in this study. The presence of a thin layer over treated dentin was noted penetrating and occluding dentinal tubules. The coat was up to 1-µm thick and its composition was probably resinous. HEMA is a hydrophilic monomer compound of dentin bonding agents with ability to infiltrate into acid-etched and moist dental hard tissues.

Nupro Gel is a commercial product containing 1.23% fluoride and 0.1M phosphoric acid. With the acidic property, the fluoride gel was able to etch peritubular dentin and form a funnelled portion of the tubule at the dentin surface. Ionized calcium in the tubular fluid and that liberated during the acid attack react with the active ingredient of the gel, sodium fluoride. The raising of ionized calcium concentration exceeded the solubility product constant for calcium fluoride, precipitating on and within the tubules. Even though calcium fluoride might occlude most dentinal tubules, its use as a desensitizing agent has been questioned due to its solubility in saliva.

This present SEM examination provided understanding of the potential occluding effect of three in-office desensitizing agents. The different mechanisms of dentinal tubules occlusion probably result in different effectiveness of hypersensitivity reduction (i.e. degrees of pain relief). However, the longevity of the precipitates or resins in dentinal tubules and their ability to resist acid challenge over time is unknown, although oxalate precipitates appear to wash out under challenge in the clinical environment. Therefore, the clinical effectiveness of these materials will depend in part on the dissolution resistance or solubility level of precipitates or resins.

CONCLUSION

The results of the present study confirmed by SEM analysis of fractured dentin samples, demonstrated that Oxa-Gel, Gluma Desensitizer and Nupro Gel were able to occlude dentinal tubules by different modes. However, further research is required to provide evidence of the durability of occlusion of these desensitizing agents under simulated clinical conditions together with functional studies designed to assess their ability to reduce fluid flow (hydraulic conductance) through dentin.

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