Bond strength of three luting agents to zirconia ceramic - Influence of surface treatment and thermocycling

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ABSTRACT

Objectives: This in vitro study aimed to evaluate the influence of different surface treatments, 3 luting agents and thermocycling on microtensile bond strength (μTBS) to zirconia ceramic. Material and Methods: A total of 18 blocks (5x5x4 mm) were fabricated from zirconia ceramic (ICE Zirkonia) and duplicated into composite blocks (Alphadent). Ceramic blocks were divided into 3 groups (n=6) according to the following surface treatments: airborne-particle abrasion (AA), silica-coating, (SC) (CoJet) and silica coating followed by silane application (SCSI) (ESPE Sil). Each group was divided into 3 subgroups (n=2) according to the 3 luting agents used. Resin-modified glass-ionomer cement (RMGIC, Ketac Cem Plus), self-adhesive resin cement (UN, RelyX Unicem) and adhesive resin cement (ML, MultiLink Automix) were used for bonding composite and zirconia blocks. Each bonding assembly was cut into microbars (10 mm long and 1±0.1 mm²). Seven specimens of each subgroup were stored in water bath at 37ºC for 1 week. The other 7 specimens were stored in water bath at 37ºC for 30 days then thermocycled (TC) for 7,500 cycles. μTBS values were recorded for each specimen using a universal testing machine. Statistical analyses were performed using a 3-way ANOVA model followed by serial 1-way ANOVAs. Comparison of means was performed with Tukey’s HSD test at (α=0.05). Results: μTBS ranged from 16.8 to 31.8 MPa after 1 week and from 7.3 to 16.4 MPa after 30 days of storage in water and thermocycling. Artificial aging significantly decreased μTBS (p<0.05). Considering surface treatment, SCSI significantly increased μTBS (p<0.05) compared to SC and AA. Resin cements (UN and ML) demonstrated significantly higher μTBS (p<0.05) compared to RMGIC cement. Conclusions: Silica coating followed by silane application together with adhesive resin cements significantly increased μTBS, while thermocycling significantly decreased μTBS.

Key words: Ceramics. Surface treatment. Resins. Bond strength.

INTRODUCTION

High mechanical properties, chemical stability, and biocompatibility make zirconia an attractive core material for fabrication of all-ceramic restorations.7,15-19 Bonding techniques of all-ceramic restorations are dependent on chemical compositions of each ceramic system.10,11 Hydrofluoric acid etching and silanization are mandatory steps to achieve a durable resin bonding to silica ceramics.12, On the other hand zirconia ceramic requires alternative techniques for long-term durable resin bonding.10,22 Therefore several surface treatments are used to improve bonding to zirconia ceramic, such as selective infiltration etching (SIE) technique,1 laser etching,10, alumina coating,11 silica ceramic coating,13,14 tribochemical silica coating or airborne-particle abrasion.1,8,9,14,20,22,25 Several ceramic primers have been introduced into the dental market recently to enhance chemical bonding to zirconia ceramic, such as primers containing a phosphonic acid monomer, 6-MHPA.
(6-methacryloxyhexylphosphonoacetate), (AZ, primer) or 3-trimethoxysilylpropyl methacrylate, MDP, ethanol (Clearfil Ceramic Primer) and organophosphate monomer, carboxylic acid monomer and other monomers (Z-Prime Plus)\textsuperscript{12,14,16}. However, there are obvious problems in obtaining a durable bonding to zirconia ceramics\textsuperscript{7,14}. Conventional luting agents, such as glass ionomer cement (GIC), could be used for cementation of zirconia ceramic full-coverage restorations\textsuperscript{5,9,14}. However adhesive cementation is preferred in case of compromised retention and resin-bonded fixed dental prosthesis\textsuperscript{3,5,14}. Self-adhesive resin cements have been introduced into the dental market to simplify bonding procedures\textsuperscript{21}. However bond strength results to zirconia ceramics in the literatures using different categories of luting agents are very controversial\textsuperscript{2,9,15,22,25}. Moreover, chemical composition of zirconia ceramic and intaglio surface morphology are unique for each commercial system\textsuperscript{4,5}. Therefore, conclusions drawn considering bonding to one zirconia ceramic system may not be applicable to other systems\textsuperscript{4,5}.

Recently, a new zirconia ceramic (ICE Zirconia, ZirkonZahn) has been introduced in the dental market. According to its manufacturer, this zirconia ceramic has bending strength over 1,400 MPa and could be used for fabrication of 16-unit fixed dental prosthesis. However, no independent data considering bonding to this zirconia ceramic has been published yet.

Several studies have evaluated bond strength of adhesives in vitro in terms of shear\textsuperscript{4,5,12}, tensile\textsuperscript{27,28} and microtensile bond strength (μTBS)\textsuperscript{2,26,30}. However, μTBS test is considered as the most accurate one\textsuperscript{26,30}. Moreover, aging and thermocycling are two important factors that have been shown to decrease the bond strength in in vitro studies\textsuperscript{4,5}. The purpose of this study was to investigate in vitro the influence of different surface treatments, storage in water and thermocycling, on the μTBS of 3 luting agents to this zirconia ceramic. The null hypotheses of the study were (1) a durable bonding to the zirconia ceramic would be achieved regardless of the surface treatments, (2) self-adhesive resin cement would provide a durable bonding to this zirconia ceramic similar to multistep adhesive resin cement, and (3) resin-modified GIC (RMGIC) would provide a durable bonding to this zirconia ceramic compared to both adhesive resin cements.

**MATERIAL AND METHODS**

A total of 18 fully-sintered zirconia blocks (5x5x4 mm) in dimension were used for this study. Each ceramic block was duplicated in light-polymerized hybrid, type 2 restorative composite resin (Alphadent composite, shade A2) using vinyl polysiloxane material (President, Coltène Whaledent, Altstätten, Switzerland) (Figure 1). Composite resin was applied in increments 1-2 mm thickness and carefully condensed with ST Instrument with plastic working end (OptrastSculpt, Ivoclar Vivadent, Schaan, Liechtenstein). Each increment was light-polymerized for 40 s at 5 mm distance and an intensity of irradiation 130 mW/cm\textsuperscript{2} (FutoLux 2, Carlo De Gorgi, Milano, Italy).

**Surface treatment before bonding**

Ceramic blocks were divided into 3 groups (n=6) according to surface treatment as follow: Group 1: airborne-particle abrasion (AA), using Al\textsubscript{2}O\textsubscript{3} (50-μm aluminum oxide particles, Pluradent, Offenbach, Germany) at 0.28 MPa for 13 s at a distance of 10 mm\textsuperscript{4} (Ney Blastmate II, Ney, CA, USA). Zirconia ceramic blocks were ultrasonically cleaned in distilled water for 3 min and dried with oil-free air stream; Group 2: Silica coating (SC), airborne-particle abrasion using 50-μm Al\textsubscript{2}O\textsubscript{3} particles at 0.28 MPa for 13 s at a distance of 10 mm\textsuperscript{2} followed by airborne-particle abrasion with 30-μm SiO\textsubscript{2} particles (CoJet sand, 3M ESPE, Seefeld, Germany) at a pressure of 0.25 MPa for 20 s at a distance of 10 mm\textsuperscript{26} (Ney Blastmate II). Zirconia ceramic blocks were ultrasonically cleaned in distilled water for 3 min and dried with oil-free air stream; Group 3 Silica coating and silane application (SCSI), same as Group SC followed by silane application\textsuperscript{26} (ESPE Sil, 3-methacryloxypropyltrimethoxysilane in ethanol, 3M ESPE). Five minutes were allowed to elapse for silane reaction\textsuperscript{26}.

**Bonding zirconia and composite blocks**

Each main group was divided into 3 subgroups (n=2) according to the following 3 luting agents, Ketac Cem Plus, RelyX Unicem and MultiLink Automix. Bonding procedures were performed according to the manufacturers’ recommendations for each luting agent:

Ketac Cem Plus (GI): self-curing, radiopaque, fluoride-releasing, RMGIC containing BisGMA and HEMA (3M ESPE). Equal amounts of past A and B were extruded on waxed paper pad, mixed for 20 s using a plastic spatula until a uniform color was achieved. The mix was applied to the intaglio surfaces of the zirconia ceramic blocks.

RelyX Unicem Aplicap (UN): dual-cure, self-adhesive resin cement, containing phosphoric acid monomer and methacrylate monomers (3M ESPE). The capsule was activated for 4 s, and then mixed in amalgamator (Silver Mix 80, Carlo De Giorgi, Milano, Italy) for 10 s. The capsule was inserted into the applier and cement was dispensed directly onto the intaglio surfaces of the zirconia ceramic blocks.

MultiLink Automix (ML): self-curing, transparent, two-past adhesive resin cement, containing...
dimethacrylate and HEMA (Ivoclar Vivadent). Equal amounts of adhesive resin cement were extruded, mixed for 20 s and applied to the intaglio surfaces of the zirconia ceramic blocks. Finally composite blocks were bonded to intaglio surfaces of zirconia ceramic blocks. Excess luting cement was removed using disposable minibrush (Ivoclar Vivadent). The bonding assembly was kept under a static load of 40 N for 5 min (Articolo 719/00, #1L01; Carlo De Giorgi), Combination of 3 surface treatments and 3 luting agents resulted in 9 test groups as follows: GIAA, GISC, GISCSI, UNAA, UNSC, UNSCSI, MLAA, MLSC and MLSCSI.

Preparation of microbars
Each block was then bonded with cyanoacrylate glue (Uhu®, batch 40267647, Uhu, Bühl, Germany) to a metal base that was fixed to a cutting machine2,26 (Isomet 1000, Buehler Ltd., Lake Bluff, IL, USA). Cutting was created under wet condition. The first 0.5-mm-thick slice was discarded2,26. Three slices ±0.1 mm in thickness were obtained. Each slice was rotated 90° and bonded to another metal base again2,26. The first slice 0.5 mm was also disregarded. Other 3 additional cuts ±0.1 mm in thickness were made2,26. This procedure was repeated for the other 2 slices. A total of 9 non-trimmed bar specimens (10 mm long and ±0.1 mm2) bonded surface area were obtained from each block2. Microscopic examination (Wild M420, Heerbrugg, Switzerland) at 20x magnification of the bar specimens revealed that only 15 specimens from group GISC and 14 specimens from group GISCSI were free from microcracks. Therefore for standardization of the test groups, only 14 specimens from each subgroup were used to complete the test. Seven specimens from each subgroup were stored in water bath at 37°C for 1 week. While the other 7 specimens were stored in water bath at 37°C for one month followed by thermocycling (TC) for 7,500 cycles. Each cycle was consists of 1 minute in 5°C cold bath and 1 minute in 55°C hot bath with a dwell time of 30 s. Specimens were dried and glued parallel to the long axis of an adapted caliper using cyanoacrylate glue (Uhu®, batch 40267647, Uhu). This apparatus was fixed to the universal testing machine (Type 500, Lloyd Instrument, Farnham, UK).

<table>
<thead>
<tr>
<th>Materials</th>
<th>Lot/Batch No</th>
<th>Manufacturer</th>
</tr>
</thead>
<tbody>
<tr>
<td>Zirconium-oxide ceramic with the following composition [Y_2O_3, 4.95-5.26, Al_2O_3, 0.15-0.35, SiO_2, Max 0.02, Fe_2O_3 max. 0.01, Na_2O_3 Max.0.04]</td>
<td>CE0476</td>
<td>ICE Zirkonia, ZirkonZahn, Gais, Italy</td>
</tr>
<tr>
<td>Transparent, two past, self curing adhesive resin cement dimethacrylate and HEMA, with barium glass silica and filler base/catalyst past in Clicker</td>
<td>K49940</td>
<td>Multilink Automix; Ivoclar Vivadent, Schaan, Liechtenstein</td>
</tr>
<tr>
<td>Self curing, radiopaque, fluoride-releasing, resin modified glass ionomer cement Paste A/Paste B in Clicker Paste A, fluoroaluminosilicate (FAS) glass, proprietary reducing agent, HEMA, water, opacifying agent Paste B, methacrylated polycarboxylic acid, BisGMA, HEMA, water, potassium persulfate, zirconia silica filler</td>
<td>56930/PF8ML</td>
<td>Ketac Cem Plus,3M ESPE, Seefeld, Germany</td>
</tr>
<tr>
<td>Dual-curing, universal self adhesive resin cement in capsule, Liquid, methacrylate monomers containing phosphoric acid groups, methacrylate monomers, Initiator components, Stabilizers Powder, alkaline (basic) fillers, silanated fillers, initiator components, pigments</td>
<td>325464</td>
<td>RelyX Unicem Aplicap, 3M ESPE, Seefeld, Germany</td>
</tr>
<tr>
<td>Silane coupling agent 3-methacryloxypropyltrimethoxysilane in ethanol</td>
<td>323995</td>
<td>ESPE Sil, 3M ESPE</td>
</tr>
<tr>
<td>50 μm Al_2O_3 particles for airborne-particle abrasion 30 μm SiO_2</td>
<td>14540</td>
<td>Alumina particles Pluraldent, Offenbach, Germany</td>
</tr>
<tr>
<td>Visible light activated, radiopaque hybrid, type 2 composite resin</td>
<td>P121FJ</td>
<td>Alpha.Dent composite, Dental technologies, Illinois, USA</td>
</tr>
</tbody>
</table>

Figure 1- Materials used in the study
### Microtensile bond strength test

Specimens were loaded in tension to failure (Figure 2) at a crosshead speed of 1 mm/min. Microtensile bond strength μTBS values were recorded for each specimen in MPa using the formula:

$$\sigma = \frac{L}{A}$$

where 'L' is the load at failure (N) and 'A' is the bonded area 1±0.1 mm².

Statistical analysis was conducted using the SPSS statistical software, version 16.0 (SPSS Inc, Chicago, Ill, USA). Statistical analyses were performed with 3-way ANOVA model followed by serial 1-way ANOVAs. Post Hoc Tukey-HSD test at α=0.05 was performed to test statistical significance between the groups.

### Microscopic examination of the debonded specimens

The fractured interfaces of the debonded specimens were examined with a light microscope (Wild Makroskop M 420) at x20 magnification to determine the failure pattern, which was assigned to cohesive failure within resin cement or composite resin²⁸, adhesive at ceramic/cement interface²⁸ or

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**Table 1** - Mean±standard deviations (SD), minimum and maximum microtensile bond strength in MPa after 1 week and 1 month storage in water with thermocycling of test groups and P values

<table>
<thead>
<tr>
<th>Test groups</th>
<th>One week</th>
<th>One month/Thermocycling</th>
<th>P values</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Mean±SD</td>
<td>Min.</td>
<td>Mean±SD</td>
</tr>
<tr>
<td>-------------</td>
<td>----------</td>
<td>------</td>
<td>---------</td>
</tr>
<tr>
<td>GIAA</td>
<td>18±4.3</td>
<td>11.4</td>
<td>7.3±3.5</td>
</tr>
<tr>
<td></td>
<td>23.5</td>
<td></td>
<td>12.8</td>
</tr>
<tr>
<td>GISC</td>
<td>17.3±3.9</td>
<td>11.6</td>
<td>0</td>
</tr>
<tr>
<td></td>
<td>22</td>
<td></td>
<td></td>
</tr>
<tr>
<td>GISCSI</td>
<td>16.8±3.7</td>
<td>13.3</td>
<td>0</td>
</tr>
<tr>
<td></td>
<td>23</td>
<td></td>
<td></td>
</tr>
<tr>
<td>MLAA</td>
<td>18.7±2.5</td>
<td>14.8</td>
<td>10.2±2.4</td>
</tr>
<tr>
<td></td>
<td>22.7</td>
<td></td>
<td>13.7</td>
</tr>
<tr>
<td>MLSC</td>
<td>20.4±2.2</td>
<td>17.1</td>
<td>9.7±3</td>
</tr>
<tr>
<td></td>
<td>23.7</td>
<td></td>
<td>15</td>
</tr>
<tr>
<td>MLSCSI</td>
<td>31.8±3.4</td>
<td>26.6</td>
<td>15.4±2.8</td>
</tr>
<tr>
<td></td>
<td>36.3</td>
<td></td>
<td>18.6</td>
</tr>
<tr>
<td>UNAA</td>
<td>19.1±4.4</td>
<td>11.9</td>
<td>9.2±3.9</td>
</tr>
<tr>
<td></td>
<td>23.8</td>
<td></td>
<td>18.6</td>
</tr>
<tr>
<td>UNSC</td>
<td>21.6±3.3</td>
<td>17.3</td>
<td>13.1±3.8</td>
</tr>
<tr>
<td></td>
<td>26.2</td>
<td></td>
<td>19.2</td>
</tr>
<tr>
<td>UNSCSI</td>
<td>28.6±6.2</td>
<td>19.2</td>
<td>16.4±4.2</td>
</tr>
<tr>
<td></td>
<td>36</td>
<td></td>
<td>24</td>
</tr>
</tbody>
</table>

**Group codes**
- GI: Ketac Cem Plus
- AA: Air borne particle abrasion
- ML: Multilink Automix
- SC: Silica coating
- UN: RelyX Unicem
- SCSI: Silica coating and silane application
"Test groups Failure Pattern" | One week | One month/Thermalcycling
--- | --- | ---
| Mixed | Adhesive | Cohesive | Mixed | Adhesive | Cohesive |
MLSCSI | 4 | 3 | 0 | 2 | 5 | 0 |
MLSC | 2 | 3 | 2 | 1 | 6 | 0 |
MLAA | 3 | 3 | 1 | 0 | 7 | 0 |
UNSCSI | 3 | 4 | 0 | 2 | 5 | 0 |
UNSC | 6 | 0 | 1 | 2 | 5 | 0 |
UNAA | 3 | 4 | 0 | 2 | 5 | 0 |
GISCSI | 1 | 5 | 1 | 0 | 7 | 0 |
GIS | 3 | 3 | 1 | 0 | 7 | 0 |
GIAA | 3 | 4 | 0 | 1 | 6 | 0 |

Group codes
GI: Ketac Cem Plus | AA: Air borne particle abrasion
ML: Multilink Automix | SC: Silica coating
UN: RelyX Unicem | SCSI: Silica coating and silane application

**Figure 3-** Failure pattern of all test groups after 1 week and 1 month storage in water

mixed adhesive/cohesive modes\(^{28}\). Representative specimens for each failure pattern were examined using a scanning electron microscope (SEM; XL 30 CP; Philips, Eindhoven, Netherlands) with an acceleration voltage of 15 kV and a working distance of 10 mm.

**RESULTS**

Mean ± standard deviation, minimum and maximum \(\mu\)TBS without and with thermocycling of test groups in MPa and P values are summarized in (Table 1). \(\mu\)TBS means were compared across all test groups using 3-way ANOVA model including the following factors (luting agent, surface treatment, storage time and interaction). The overall F-test was highly significant (\(p<0.0001\)), indicating differences in mean \(\mu\)TBS across at least one of the 3 factors. All individual factors were significant, (\(p<0.001\)).

The interaction between surface treatment and luting agent and surface treatment and storage time were significant (\(p<0.001\)). However luting agent and storage time was not significant (\(p=0.208\)).

Multiple comparisons with Post Hoc Tukey-HSD test at \(\alpha=0.05\) revealed that, considering 3 surface treatment performed, SCSI significantly increased \(\mu\)TBS (\(p<0.05\)) compared to SC and AA. However, there was no statistically significant difference between SC and AA (\(p=0.26\)).

Considering 3 luting agents used, ML and UN resin cements showed significantly higher \(\mu\)TBS than the GIC cement (\(p<0.001\)). However, there was no significant difference in the \(\mu\)TBS of the 2 resin cements (\(p=0.37\)).

Considering aging condition, 30-day water storage and thermocycling significantly decreased \(\mu\)TBS compared to 1-week water storage (\(p<0.05\)). Groups GISC and GISCSI were spontaneously debonded.

Fracture patterns of all test groups are summarized in (Figure 3). After 1 week of storage in water, the debonded specimens showed mainly mixed failure pattern (Figure 4). Some specimens showed adhesive and cohesive failure (Figure 5). After 30 days of storage in water and thermocycling, the failure pattern was mainly adhesive with remnants of the luting cements still adhered to zirconia ceramic surface (Figure 6).
DISCUSSION

Clinically, restorations are subjected to repeated thermal stress and mechanical fatigue due to masticatory forces. Therefore, one limitation of this study is that specimens were subjected only to thermal stress without mechanical fatigue. In addition, storage in water during 1 month might be too short to allow water saturation of the luting cements. It is likely that hydrolytic effects might affect the bond strength negatively after longer time of storage in water.

Microtensile bond strength test is more accurate than shear and tensile tests. Because the small dimensions and small interfacial bonding zone of the specimens result in a more uniform distribution of the applied stresses. Airborne-particle abrasion and silica coating are surface treatments recommended by most of the manufactures of zirconia ceramics and luting agents to improve bonding to zirconia ceramics. Therefore, these surface treatments were tested in this study.

A range of 10-13 MPa was suggested as the minimum range for acceptable clinical bonding. Therefore the results of this study clearly indicated that, silica coating and silane application together with resin cements would ensure a durable resin bonding to this zirconia ceramic 15-16 Mpa.

Considering surface treatment, airborne-particle abrasion produced an activated microroughened zirconia surface, increased the bonding area and modifying the surface energy and wettability. In case of silica coating, alumina particles modified with silica acid were sprayed under pressure. This tribochemical reaction produces a high temperature contact area that can hold the silica layer on the ceramic surface. These particles formed a base for micromechanical interlocking. Because no silane was applied before bonding, bond strength was directly correlated to the quality of micromechanical interlocking with the silica coating layer. However, bond strength to zirconia ceramic after silica coating was not improved in comparison to airborne-particle abrasion.

The results of this study are in agreement with the findings of several studies, which reported that bond strength to zirconia ceramics was not improved after silica coating compared to airborne-particle abrasion. Oyagüe, et al. (2009) and Kern (2009) reported that airborne-particle abrasion of zirconia ceramics produces a certain roughness, but only limited undercuts were produced, thus not improving bonding to zirconia. Matinlinna, et al. (2006), after energy-dispersive x-ray analysis of silica-coated zirconia ceramic, reported that silica-coverage originating from the coating particles appears not to have become embedded onto the hard zirconia surface, consequently bond strength was not improved after silica coating. Moreover, ultrasonic cleaning in distilled water might removed a significant amount of silica coating layer, consequently bond strength of resin cements to silica-coated zirconia ceramic was decreased, as reported by Nishigawa, et al. (2008). However, the results of this study were contradicting to the results of other studies, which reported that silica coating improved bond strength to zirconia ceramics compared to airborne-particle abrasion. This difference in the results could be attributed to the fact that, Atsu, et al. (2006) used Al2O3 for airborne-particle abrasion followed by Al2O3 particles modified by silica. Panavia luting cement together with different ceramic primers were used for bonding. On the
other hand Lüthy, et al.\textsuperscript{15} (2006) used the Rocetac system for silica coating, which combines 110 mm Al\textsubscript{2}O\textsubscript{3} for airborne-particle abrasion followed by 110-mm Al\textsubscript{2}O\textsubscript{3} particles modified by silica. Rounded rods were bonded to zirconia ceramics instead of restorative composite resin. Moreover both studies used shear test. In the present study, 50-mm Al\textsubscript{2}O\textsubscript{3} was used for airborne-particle abrasion followed by 30-mm Al\textsubscript{2}O\textsubscript{3} particles modified by silica and a μTBS test were employed.

Silica coating and silane application significantly increased bond strength compared to airborne-particle abrasion or silica coating alone, as reported in several studies\textsuperscript{2,3,10}. Silane coupling agent did not promote adequate bonding to zirconia ceramics, as these ceramics contain minimal or no silica content\textsuperscript{9,12,14}. Therefore, silica coating of zirconia ceramics is a prerequisite for durable siloxane bonding, as it leave a physically and chemically active outer surface layer\textsuperscript{18}. Silane coupling agent wets the adherent, increases its surface energy and makes it accessible for effective bonding\textsuperscript{18}. Moreover, it is capable of forming covalent bond at silica coated ceramic/resin cement interface through formation of silanol groups\textsuperscript{18,26,29}. Therefore, after silica coating and silanization, bond strength was based on both micromechanical interlocking plus chemical adhesion due to silane application\textsuperscript{18,26,29}.

Variations in chemical composition, wetting capacity, viscosity and mechanical properties for each luting cement could be responsible for variations in the bonding capacity to zirconia ceramics\textsuperscript{9,15,17,20,28}. According to the manufacturer, RelyX Unicem contains methacrylate monomers, adhesive phosphate monomer and silanated fillers in its chemical composition. Adhesive phosphate monomer enhanced self bonding to zirconia ceramics especially after silica coating and silane application as reported in several studies\textsuperscript{12,15,20}. Ketac Cem Plus contains BisGMA, HEMA and zirconia filler. Several studies have reported that bond strengths of BisGMA containing resin cements were dramatically decreased after artificial aging\textsuperscript{12,15} due to their remarkable weak mechanical properties.

Multilink Automix contains no adhesive phosphate monomer. However, it contains dimethacrylate, HEMA and silica filler. Dimethacrylate and HEMA improved its mechanical properties. The improved mechanical properties could be responsible for high bond strength results of Multilink Automix. Nothdurft, et al.\textsuperscript{20} (2009) found no significance difference in the mean bond strength of Multilink Automix 21.2 MPa and RelyX Unicem 23.1 MPa to zirconia ceramics after different surface treatments.

RelyX Unicem and Multilink Automix have better mechanical properties than Ketac Cem Plus. According to the manufacturers flexural strength for RelyX Unicem (75 MPa) and Multilink Automix (70 MPa) are in the same range. However it is higher than that of Ketac Cem Plus (31.6 MPa). Flexural strength is an indicator for the mechanical properties of each luting agent. Luting cements with high mechanical properties are more resistant to aging conditions\textsuperscript{15}. Therefore, variations in the mechanical properties of the 3 luting cements used could be another contributing factor for bond strength results as reported in other studies\textsuperscript{15,17}.

In \textit{in vitro} studies, water storage and thermocycling are two important factors that decrease the bond strength\textsuperscript{4,26,27}. Therefore, 1 month storage in water and thermal cycling for 7,500 cycles was used as aging regime to simulate clinical conditions. After 30 days of storage in water and thermocycling, μTBS were significantly decreased in all groups. This decrease in bond strength might be due to degradation of the luting cement itself\textsuperscript{4,17,27} and the hydrolytic effect of water at the luting cement/ceramic interface\textsuperscript{9,27}.

Moreover, mismatch between the coefficient of thermal expansion of the bonded specimens (zirconia ceramic, luting agent and composite resin) could result in hoop stress during thermocycling\textsuperscript{27}. Another factor could be the fact that silanized surfaces were unstable in contact with moisture as reported by Derand, et al.\textsuperscript{9} (2005). Accumulation of negative effect of water, thermocycling and instability of silane could be responsible for the decrease of bond strength for all test groups.

The initial high bond strength results were reflected on the failure pattern of debonded specimens as examined by optical reflection microscope and confirmed by scanning electron microscopy. All groups showed mainly mixed failure pattern or adhesive failure while cohesive failure was minimal. Cohesive failure within composite resin could be due to the initiation of microcracks during cutting of the specimens\textsuperscript{24}. Therefore, in this study, specimens with apparent microcracks under stereomicroscopic examination were discarded.

After 30 days of storage in water and thermocycling, the failure pattern was mainly adhesive indicating a decrease in the bond strength due to the hydrolytic effect of water, hoop stress due to thermocycling and degradation of the luting resin itself.

The general outcome of this study suggests that although conventional cements could be used for cementation of zirconia ceramics, resin cements are preferred for long-term bond durability. Bonding techniques based on micromechanical interlocking and chemical adhesion using silanes are preferred than other techniques.
CONCLUSIONS

Within the limitations of this study, the following conclusions were drawn:

Silica coating and silane application significantly improved μTBS compared to silica coating or airborne-particle abrasion.

There was no significant difference in the bond strength after using self-adhesive resin cement or multistep adhesive resin cement.

Both adhesive resin cements significantly increased μTBS compared to RMGIC.

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