ABSTRACT

Objectives: This study aimed to compare the micro-tensile bond strength of methacrylate resin systems to a silorane-based restorative system on dentin after 24 hours and six months water storage. Material and Methods: The restorative systems Adper Single Bond 2/Filtek Z350 (ASB), Clearfil SE Bond/Z350 (CF), Adper SE Plus/Z350 (ASEP) and P90 Adhesive System/Filtek P90 (P90) were applied on flat dentin surfaces of 20 third molars (n=5). The restored teeth were sectioned perpendicularly to the bonding interface to obtain sticks (0.8 mm²) to be tested after 24 hours (24 h) and 6 months (6 m) of water storage, in a universal testing machine at 0.5 mm/min. The data was analyzed via two-way Analysis of Variance/Bonferroni post hoc tests at 5% global significance. Results: Overall outcomes did not indicate a statistical difference for the resin systems (p=0.26) nor time (p=0.62). No interaction between material x time was detected (p=0.28). Mean standard-deviation in MPa at 24 h and 6 m were: ASB 31.38 (4.53) and 30.06 (1.95), CF 34.26 (3.47) and 32.75 (4.18), ASEP 29.54 (4.14) and 33.47 (2.47), P90 30.27 (2.03) and 31.34 (2.19). Conclusions: The silorane-based system showed a similar performance to methacrylate-based materials on dentin. All systems were stable in terms of bond strength up to 6 month of water storage.

Key words: Dentin. Bonding. Dental adhesives. Composite resins. Longevity

INTRODUCTION

The use of composite resins in restorations has increased over the years because it may be considered a minimally invasive technique and also for the continuous development of the material. Currently, most of the commercially available composites are a mixture of different methacrylate monomers that polymerize linearly producing about 3% polymerization shrinkage, depending on the volume of material. Under a clinical scenario, it can negatively compromise the longevity of resin-based restorations, resulting in unsatisfactory marginal adaptation, micro-leakage, caries recurrence and micro-cracks in the enamel, which are important factors related to failures in restorations.

In an attempt to minimize these shortcomings, strategies to control polymerization shrinkage have been developed, such as composites based on an organic matrix of siloxane and oxirane (silorane resins). Their polymerization is a slow reaction among camphorquinone, an iodonium salt, and an electron donor. The light spectrum activates camphorquinone, and the electron donor breaks the iodonium salt into an acid cation that, in turn, opens the oxirane rings. The monomers are proximal to each other and, due to the ring opening, the...
polymerization shrinkage is said to be less than 1%.

Studies have observed good marginal adaptation in enamel and dentin, although low bond strength has been described in dentinal cavities restored with silorane resin. Others did not find micro-leakage at the enamel-silorane interface, but the development of some tension at the Silorane-based composite was observed. The higher degree of subsurface polymerization of methacrylate-based compared to the silorane-based resin was also described.

Considering the potential advantages of Silorane-based resins, there are some questions yet to be investigated. Therefore, the aim of this study was to compare the micro-tensile bond strength (μTBS) of methacrylate-based systems to a silorane-based material after 24 hours and 6 months of water storage. The null hypotheses were that there was no difference in the bond strength among the restorative systems and there was no difference of the durability of the bond strength after six-months of water storage.

**MATERIAL AND METHODS**

**Experimental design**

It was an *in vitro study* involving two factors: restorative systems (in four levels) and time (in two levels). Bond strength was the main response variable to compare the factors. This research was approved by the Ethics Research Committee of the dental school (Protocol 0172/09).

**Specimen preparation**

Twenty recently extracted third molars were disinfected in 0.5% chloramine solution at 4°C for 7 days and stored in distilled water under refrigeration (4ºC) until the beginning of the experimental phase. The occlusal surfaces were removed with 150 silicon carbide sandpaper under refrigeration.

**Material**

<table>
<thead>
<tr>
<th>Material (Batch number)</th>
<th>Composition</th>
<th>Application</th>
</tr>
</thead>
<tbody>
<tr>
<td>Adper Single Bond 2 (8RL)</td>
<td>Bis-GMA, HEMA, diurethane dimethacrylate, polycalenoic acid copolymer, camphorquinone, water, ethanol, 1.3 glycerol dimethacrylate, 10% silica particles (5 nm)</td>
<td>37% H₃PO₄ (15 s); Wash (15 s); Air stream (10 s); Adhesive (2 coats); Air stream (10 s); Light cure (10 s at 550 mW/cm²); Composite restoration</td>
</tr>
<tr>
<td>Clearfil SE Bond Primer (00788A) Bond (01146A)</td>
<td>Primer: MDP, HEMA, camphorquinone, hydrophilic dimethacrylates, N, N-diethanol p-toluidine, ethanol, water. Bond: MDP, bis-GMA, HEMA, aliphatic hydrophobic dimethacrylates camphorquinone, N, N-diethanol p-toluidine, silica</td>
<td>Apply primer (20 s); Air stream (10 s); Apply adhesive; Air stream (10 s); Light cure (10 s at 550 mW/cm²); Composite restoration</td>
</tr>
<tr>
<td>Adper SE PLUS Primer (8BE) Adesivo (8BB)</td>
<td>Primer: water (80%), HEMA (20%), rose bengel, surfactant. Adhesive: methacrylate resins (UDMA, TEGDMA, TMPTMA, HEMA phosphate, MHP (acidic monomers), zircon and photoinitiators</td>
<td>Apply primer; Adhesive (20 s); Air stream (10 s); 2ª coat of adhesive; Air stream (10 s); Light cure (10 s at 550 mW/cm²); Composite restoration</td>
</tr>
<tr>
<td>P90 Adhesive System Primer (4763P) Bond (4763B)</td>
<td>Primer: phosphate methacrylates, Vitrebond monomer, Bis-GMA, HEMA, ethanol, silica, stabilizers, camphorquinone. Bond: hydrophobic dimethacrylate, phosphate methacrylates, TEGDMA, silica, camphorquinone, stabilizers</td>
<td>Apply primer (15 s); Air stream (10 s); Light cure (10 s at 550 mW/cm²); Apply bond; Air stream (10 s); Light cure (10 s at 550 mW/cm²); Composite restoration</td>
</tr>
<tr>
<td>Filtek Z350 (N12715)</td>
<td>Bis-GMA, Bis-EMA, UDMA, TEGDMA, zirconia/silica, silica, camphoroquinone</td>
<td>2 mm thick increment, light curing (20 s at 550 mW/cm²)</td>
</tr>
<tr>
<td>Filtek P90 (N183458)</td>
<td>Siloxane, oxirane, camphoroquinone, iodonium salt, electron donor, quartz, triur, fluoride, stabilizers, pigments</td>
<td>2 mm thick increment, light curing (20 s at 550 mW/cm²)</td>
</tr>
</tbody>
</table>

HEMA (2-hydroxyethylmethacrylate); UDMA (urethane dimethacrylate); TEGDMA (triethyleneglycol dimethacrylate); TMPTMA (trimethylolpropane trimethacrylate); MHP (phosphate methacrylate); Bis-GMA (bisphenol glycidyl methacrylate); MDP (10-methacryloyloxydecyl dihydrogen phosphate); Bis-EMA (bisphenol ethyl methacrylate).

**Figure 1** - Resin materials; composition and application mode
testing machine (EMIC, São José dos Pinhais, PR, Brazil) at 0.5 mm/min speed. The other half was stored in distilled water at 37°C for 6 months and then tested.

The failure modes were observed in a scanning electron microscope (JSM T220A, JEOL, Peabody, MA, USA) and classified into adhesive/mixed, cohesive in resin and cohesive in dentin. The percentage of failure modes was calculated for each group. The bond strength data (MPa) was treated by two-way Analysis of Variance and Bonferroni post hoc tests at 5% global significance.

RESULTS

The uniformity of measures relating to the cross-sections (mm²) of the sticks was analyzed by one-way Analysis of Variance. The values ranged from 0.86 to 0.88 mm² and differences among experimental groups were not observed (p>0.05). The absence of differences in mean values of areas suggests that there was no interference of this aspect on the bond strength to the dentin.

Table 1 shows the results of the bond strength (MPa) to the dentin over time. There was no difference for the resin materials (p=0.26) nor the time (p=0.62). No interaction between material and time was detected either (p=0.28).

Table 2 shows the percentage of failure mode and premature failures in the different experimental conditions. The adhesive/mixed failure was prevalent in all groups. After 6 months, an increase of the cohesive resin and premature failures for all groups was observed.

DISCUSSION

This study aimed at analyzing the bond strength of silorane-based and methacrylate based restorative systems to dentin over 6 months. The results of the present investigation did not show a higher bond strength for the Filtek Silorane restorative system to dentin, comparatively to the methacrylate-based resins, since their bond strength was similar at 24 hours and 6 months. For this reason, the null hypothesis must be accepted.

Regarding the polymerization of composites,
The stress polymerization continues to be a concern and studies have pointed out that the cavity configuration (C-Factor) may influence the process. A direct relationship between the C-factor and shrinkage was described, which means that the higher the ratio of bonded to un-bonded surfaces, the greater the shrinkage and its stress. This situation is worse considering the application of composites in cavities that restrict the mobility of the polymers. But in the present study, the bonding procedure was performed on a flat dentin surface, and this favorable circumstance may have explained the similarity among the bond strength of all groups, since the C-factor is low.

This similar performance observed for all groups may also be influenced by the characteristics of associated dentin bonding systems, since it was reported that the acidity of the primer should interfere on dentinal demineralization and adhesion. In the present study, this was not confirmed, because the bond strength to dentin was similar among the groups. The silorane composite is used with a self-etching primer system, with pH 2.7. Adper SE Plus, when mixed reach a pH level of approximately 1.5 and the pH level of the Clearfil SE Bond is 2. But the acidities of the materials did not influence the bonding to dentin in the present study.

Probably other aspects related to the composition of the materials may explain the similarities. The stress generated by each component of the restorative systems (Filtek P90, Filtek Z350, Filtek Z350 flow, composites, primers and adhesives) was tested individually. A higher contraction stress was observed for the adhesive of the Filtek Silorane system compared to the methacrylates. The authors explained the results based on the viscosity of the adhesive, since the Filtek P-90 is filled with 5 to 10% of silanized silica. But according to the manufacturer, neither the adhesive nor the primer of the Silorane Adhesive system contains silorane in its composition, and this may suggest the same mechanisms of polymerization among the restorative adhesive systems, resulting in a similar bond strength over time. A similar contraction stress was also observed between the silorane-based and the methacrylate-based composites. This subject was also confirmed in a clinical study that had shown no difference in stress behavior between low-shrinkage and methacrylate-based materials.

Another investigation observed the hybrid layer formed by the silorane adhesive system and found similar thicknesses between the silorane and methacrylate-based adhesives. The adhesion of the Silorane Adhesive system to the dentin is ensured by the presence of hydrophilic monomers into the silorane primer, whereas the bond contains hydrophobic bi-functional monomers that match the silorane composite. The bond is placed in the cured silorane primer prior to being cured itself, but a Raman spectra showed an intervening zone of 1 micrometer of mixed spectral intensities between the primer and adhesive of the silorane. According to the authors, this area should be explained due to the remaining oxygen inhibition layer on the primer surface and that should be the weak link in the bonding process of the silorane systems. By observing the percentage of premature failures after 6 months and the increase of cohesive fractures, some degradation of the silorane-based resin might have occurred, although it was not detected by the bond strength test. So, long-term studies should be considered in order to evaluate the bonding stability of silorane-resins.

Thinking about the etch-and-rinse approach, it is well known that the discrepancy between the depth of demineralization and resin infiltration to the dentin may leave unprotected collagen fibrils beneath the hybrid layer prone to degradation. Phosphoric acid is a strong acid (pH=0.03-0.05) used as a conditioning step of ASB, sufficiently acidic to activate gelatinolytic and collagenolytic activity; however, it is also able to denature the MMP activity. This can also explain why no differences were observed between 24 h and 6 m for the Adper Single Bond 2.

On the contrary, the pH level of self-etch adhesives is insufficient to destroy MMPs, thus, it can be argued that demineralized collagen fibrils by two-step self-etch adhesives were not affected because the majority of them were covered by resin and thus, protected from enzymatic and hydrolytic action. As shown previously, the use of a more hydrophobic adhesive coat over the primer makes the adhesive more resistant to water sorption.

A long-term clinical study showed that methacrylate-based composite restorations worked well during 56 months on posterior restorations, due to the low failure rate that was related to secondary caries and loss of anatomic structure. Another one year clinical study (Baracco, 2012) showed a tendency to degradation of marginal adaptation of silorane-based and methacrylate-based composites, comparatively to the baseline data. The authors stated that although the Filtek Silorane had shown acceptable performance after one-year of clinical service, they did not find advantages of using silorane-based over the methacrylate-based composites. So, additional studies are therefore needed to clarify the performance of low-shrinkage composites to dental substrates.

**CONCLUSION**

Under the limitations of the methodology employed, it may be concluded that the bond...
strength of both methacylate and silorane resin materials to dentin was not influenced by 6-months of water storage.

REFERENCES
