Introduction

Fossil fuels, which meet most of the world today, are being depleted rapidly. Also, their combustion products are causing global environmental problems, such as the greenhouse effect, ozone layer depletion, acid rain… etc. These changes to the earth’s ecosystem pose great danger to our environment and life on earth. Engineers and scientists agree that the solution to all of these global problems would be to replace the existing fossil fuel systems with the more eco-friendly fuel systems. Hydrogen presents one such great alternative solution. It is a very efficient and clean fuel; its combustion produces no greenhouse gasses, no ozone layer depleting chemicals, and little or no acid rain ingredients and pollution.

Hydrogen was used as fuel in internal combustion engines during the late 1800’s by Picardo who was able to achieve high thermal efficiencies with hydrogen powered internal combustion engine (Richardo, 1923).

Hydrogen proved promising alternative to gasoline owing to two major factors: the almost near-zero engine-out emissions and higher thermal efficiencies which in some cases exceeded those values for conventional petrol-fuelled SI engines (Swain et al., 1981; Tang et al., 2002; and Das, 1991).

These unique combustion characteristics of hydrogen that allow clean and efficient operation at low engine loads present difficulties at high engine loads. These problems resulted from the detonation of the hydrogen-air mixture as a result of low ignition energies. This factor triggers frequent uncontrolled combustion events, accompanied with high combustion temperatures of mixtures near stoichiometric ratios leading to increased NOx production and thermal as well as mechanical stresses on the engine components.

According to the best estimates, the world’s fossil fuel (coal, petroleum and natural gas) production, which meets about 80% of our energy requirements today, will start to decline in 20-30 years time. The petroleum and natural gas production is already decreasing in parts of the world like United States. On the other hand, since the demand for energy is ever increasing as the nations of the world try to improve their living standards; alternative energy sources are being considered. Research regarding their conversion into usable forms of energy is being accelerated.

Reviewing the literature one finds voluminous amount of literature and reviews discussing the use of hydrogen as an automotive fuel, some of which dates back to more than one hundred years. Some of the greatest and comprehensive reviews compiled in this area are those written by Veziroglu (2007), Bal and Wietschel (2008), White et al. (2006), and Das (1990). They not only discussed the major effect of utilizing hydrogen as H2ICE fuel, but also they discussed the major difficulties, technical problems and economy of utilizing hydrogen as fuel for IC engines.

Further, Billings et al. (1966) have modified and tested a hydrogen-powered vehicle. The modification included the installation of a gaseous-type carburetor modified to accommodate hydrogen, accompanied by a water injection system to cool the combustion flame and to quench the hot spots within the cylinder which can cause backflashing. Further, the modification to overcome the backflashing problem included the installation of stainless steel spark plugs and the decreasing of the spark gap. Finally, they compared the performance of hydrogen powered engine with that powered by gasoline fuel; they found out that the hydrogen driven engine was an average of 21% more efficient than the corresponding gasoline-powered one.

The effect of compression ratio, surface/volume ratio of combustion chamber, and the boost pressure on thermal efficiency and exhaust emissions were investigate by Yasuo et al. (2000), as they used a hybrid electric vehicle with internal combustion engine fueled with hydrogen. They found that a decrease in the surface/volume ratio may result in a maximum increase in the indicated thermal efficiency by 44%. NOx emission was found to be less than 10 ppm.

This work was carried out at the University of Jordan and represents the first attempt to implement hydrogen in IC engines. It aimed to study a locally designed hydrogen regulator to overcome a problem that arise from the utilization of hydrogen in internal combustion engine, this problems lies in the fact that the flow of hydrogen into the engine might be accompanied by pulses leading to problem during the operation of the engine. It is hoped that this first step will be accompanied by further research in this area and also the introduction of this regulator will eliminate such pulses and hence the engine will run in smooth conditions. Having installed the regulator, the relative change in the performance of the engine due to the utilization of hydrogen as fuel instead of gasoline will be measured.
The Gaseous Regulator

The regulator shown in Fig. 1 is made up of stainless steel coated with a chrome layer that would prevent any chemical attach on the body of the regulator. It is located between the engine and the compressed hydrogen cylinder. It is attached to a throat placed at the inlet of the engine, as shown in Fig. 2.

When more fuel is required by the engine, butterfly valve is opened and more air passes through the throat causing more pressure drop at the regulator outlet. As a result, the pressure inside the regulator decreases causing the diaphragm to deflect downward and hence the connecting arm pushes the needle downward to allow more fuel to flow into the engine.

The reverse action occurs when less fuel is required by the engine or when the vehicle is suddenly brought to rest or the accelerator paddle is released.

Further, the regulator will eliminate the occurrence of backfire or fire spreading into the supply lines through the following mechanisms:

1. The inlets and outlets of the regulator are designed based on the quenching diameter of hydrogen.
2. The supply lines will always convey hydrogen ONLY IF suction pressure acts on the diaphragm, and
3. The spring will help shut the supply lines if the suction pressure is lost either due to fire or sudden brake or closure of the throttle valve in the inlet manifold.

Flame Trapper

This part was used in the experiment as additional safety to conduct this experiment. It is called the flame trapper. It is located between the hydrogen cylinder and the regulator and its main function is to quench the flame in the case of backfire. It consists of water bath, as shown in Fig. 3, half filled with water and the hydrogen is introduced at the bottom of the bath and accumulated at the top. The regulator inlet is directly connected to the top of this bath. In case of a flashback, the flame will be quenched inside the flame trapper before it reaches the cylinder.

The Engine

The engine used in this study was a basic TD43 internal combustion engine, with a compression ratio that can be varied between 5.1 and 18.1. It is water-cooled, single-cylinder, four-stroke spark-ignition engine that has a swept volume of 580 cm$^3$. The carburetor used is fitted with a variable fuel flow type of jet. An electrical dynamometer is directly coupled to the engine for starting and for the measurement of the brake power. Experimental values of torque, brake power, inlet and outlet cooling water temperatures, cooling water flow rate, exhaust temperature and the air flow rate are obtained by direct readings from engine control panel. The gasoline flow rate was measured with the help of a 16 mL burette, while Hydrogen gas was measured with the help of gas flow meter calibrated for hydrogen.
To be able to use the engine with gaseous fuel, a special mixer which consists of a throat and butterfly valve only is fitted in place of the carburetor to control the flow of fuel. The gaseous fuel is introduced through a hole at the tip of the throat. The engine testing conditions were as shown in Table 1, below.

Table 1. Engine test conditions.

<table>
<thead>
<tr>
<th>Engine Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Engine speed</td>
<td>1500-2500 rpm</td>
</tr>
<tr>
<td>Compression Ratio</td>
<td>9:1 (fixed)</td>
</tr>
<tr>
<td>Spark Advance</td>
<td>Variable</td>
</tr>
<tr>
<td>Throttle Position</td>
<td>Wide Open (WOT)</td>
</tr>
</tbody>
</table>

Experimental Procedure

As an initial step, the regulator was tested for any leaks. This test was performed after it was installed between the engine and the hydrogen cylinder. During two hours of testing under a pressure of two atmospheres, no leak was detected.

Having carried out the leak test, the next step was to estimate the performance of the engine with hydrogen as fuel and compare its relative performance with gasoline.

Results

It is to be mentioned here that during the experimental work carried out, the engine was running smoothly with neither knocking nor leak of hydrogen.

Figure 4 shows the volumetric efficiency for both gasoline and hydrogen fuels. First thing noticed is the reduction in the engine volumetric efficiency with hydrogen fuel. Similar results were shown by others like Furuhama et al. (1978). This is expected since the inlet manifold is usually supplied with some heat to help vaporize the liquid (gasoline) fuel. Therefore, hydrogen gas expands more than liquid gasoline inside the engine, leading to a drop in the volumetric efficiency of a hydrogen fired engine. Further, the displacement of air by the large volume (expanded) hydrogen occurs due to heated inlet manifold. It was found by researchers that a stoichiometric mixture of Hydrogen-Air fuel contains approximately 30% hydrogen by volume, whereas for gasoline-air stoichiometric mixture the percentage is 20% of fully vaporized gasoline by volume.

This result was also shown by many other researchers and was corrected by certain modifications such as separating the inlet from the outlet manifold, using hydrogen fuel injectors, cooling the inlet manifold for the hydrogen engine.

Figure 5 indicates the relation between the brake power and the speed of both the hydrogen and gasoline fired engines at different engine speeds and ignition timing. As shown in the figures, the brake power generated by hydrogen is less than that generated when gasoline is used. This is expected since the volume efficiency of hydrogen-fuelled engine is lower than that for gasoline-powered engines.

Further noticed is that the peak brake power values for hydrogen tends to occur at spark timings less (i.e. retarded) than those for gasoline. This means that with hydrogen as fuel, the engine spark timing must be retarded (brought near to top dead center) compared with gasoline. This is the direct result of the faster burning of hydrogen compared with gasoline.

Similar trend is also present in Fig. 6, which shows the reduction of engine torque when using hydrogen compared with conventional gasoline fuel. Same reason can be said as that for the brake power. Further, this figure suggests that the MBT (maximum brake torque) angle for hydrogen is near the top dead center compared with gasoline.
Figure 6. Variation of brake torque (N-m) with engine speed at various ignition timing.

Figure 7. Variation of brake thermal efficiency (%) with engine speed at various ignition timing.

The thermal efficiency of the hydrogen powered and gasoline powered engines is shown in Fig. 7. As shown in this figure, the hydrogen powered engine has higher efficiency than that of gasoline and this is in agreement with all previous results. Obviously and as shown, the thermal efficiency for both cases has the trend of variation with speed. These values can be further improved if the compression ratio of the engine is increased. This is due to the higher RON (Research Octane Number) of hydrogen of about 120 (Swain et al., 1981 and Topinka, 2004) compared with gasoline of about 91-99 (Heywood, 1988). In addition to this factor, the low lean-flammability limit of hydrogen (about 0.1 ≤ Φ ≤ 7.1) compared with gasoline (about 0.7 ≤ Φ ≤ 4) with Φ is the equivalence ratio.

Those two factors help attain higher values of thermal efficiency. However, in this study the compression ratio was kept constant for the sake of comparison only.

Further, it is shown that the peak thermal efficiency for hydrogen occurs at retarded spark timings corresponding to higher engine powers compared with gasoline.

The variation specific fuel consumption with speed is shown in Fig. 8. It is very clear that the utilization of hydrogen as a fuel leads to significant drop in the specific fuel consumption (on mass basis), which is considered one of the main advantages of using hydrogen as a fuel in internal combustion engines.

Figure 8. Variation of brake specific fuel consumption (kg/kw-hr) with engine speed at various ignition timing.

As shown in the figure, the specific fuel consumption of a hydrogen powered engine is almost independent of speed, for the speed test range in this work, while it increases at certain speed for a gasoline fired engine. Further, the specific fuel consumption of a hydrogen engine is almost one third of that for a gasoline fired engine.

However, since fuel is sold on volume basis, Fig. 9 shows that the picture is totally different if the comparison is based on volume
basis. This figure shows the variation of the brake specific energy consumption in m³/kw-hr, rather than the ordinary kg/kw-hr. Due to the lower density of hydrogen, the energy density on volume basis that was consumed by the engine was much higher than that for gasoline (keeping in mind that the engine developed less output compared with gasoline).

**Figure 9.** Variation of brake specific energy consumption (m³/kw-hr) with engine speed at various ignition timing.

**Conclusions**

1. From this work it may be concluded that the utilization of a new designed gas regulator in a hydrogen powered engine was very promising and the engine operated in a very smooth fashion.
2. The thermal efficiency of the engine is better when hydrogen fuel is used instead of gasoline.
3. The spark timing for hydrogen engine must be retarded to give better performance.
4. The engine tends to develop less power (and torque) with hydrogen fuel than gasoline engine.
5. Though the BSFC looks interesting, however, comparison based on BSEC shows more energy (per volume) needed to be supplied with hydrogen fuel.

**References**


Billings, R.S. The hydrogen engine, A solution to pollution. 17th International Science Fair, Dallas, Texas, 1966.


