Hyperspectral remote sensing as an alternative to estimate soil attributes

Sensoriamento remoto hiperspectral como alternativa na estimativa de atributos do Solo

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ABSTRACT - Minimizing environmental impacts and increasing crop productivity depend mainly on the knowledge of chemical, physical and mineralogical characteristics of the soil attributes. However, traditional methods are time-consuming and costly. The objective of this study was to determine and validate a method to quantify soil attributes using UV-Vis-NIR Spectroscopy as an alternative to conventional methods of soil analyses. The work comprised two main phases: (1) creation and calibration of statistical models to determine the soil attributes derived from spectral data extracted from soil samples collected in area 1, (2) validation of statistical models in area 2 and correlations between the estimated and observed values (conventional method) for each soil attribute. The equations of the attributes Fe$_2$O$_3$, Al$_2$O$_3$, and clay reached $R^2 > 0.80$ and may be applied to a different database than the one that was used to generate the equations, provided that they belong to the same study site.

Key words: Reflectance. Soil analysis. Remote Sensing.

RESUMO - A minimização de impactos ambientais e o aumento da produtividade agrícola dependem, principalmente, do conhecimento de características químicas, físicas e mineralógicas do solo. Os métodos tradicionais utilizados para este fim consomem muito tempo e são de elevado custo financeiro. O objetivo deste trabalho foi o de testar e validar um método de quantificação de atributos do solo pela energia eletromagnética refletida e detectada por espectrorradiometro no visível e infravermelho. Este procedimento surge como apoio ao método convencional de análise de solo. Para sua execução, as etapas de trabalho compreenderam duas fases: (1) criação e calibração de modelos estatísticos de determinação dos atributos do solo obtidos a partir de dados espectrais (obtidos por sensor em laboratório 450-2500 nm) extraídos de amostras de terra em uma área; (2) validação dos modelos estatísticos numa outra área desconhecida e correlações entre os valores estimados e determinados (método convencional). Concluiu-se que as equações dos atributos Fe$_2$O$_3$, Al$_2$O$_3$ e Argila, atingiram $R^2 > 0.80$ podendo ser aplicadas a uma base de dados diferente daquela que foi utilizada na geração das equações, desde que pertença a mesma região de estudo.

Palavras-Chave: Reflectância. Solos-análise. Sensoriamento Remoto.
INTRODUCTION

The use of traditional laboratory analyses is the most reliable method to determine soil attributes. Many studies and tests have been conducted to obtain the standard methods. However, these methods are time-consuming and costly (CHANG et al., 2001).

In Brazil, more than 1 million samples of soil chemical analyses are performed a year. This requires a large amount of reagents and generates a lot of residuum (RAJ et al., 2001), therefore, the development of more effective techniques to determine soil attributes is necessary (CHANG et al., 2001). Reflectance spectroscopy is an alternative method to estimate soil attributes namely clay and cation exchange capacity (CEC) (LAGACHERIE et al., 2008; STEVENS et al., 2006; VISCARRA-ROSSEL et al., 2006).

The reflectance in the near infrared spectroscopy (NIRS) or near infrared analysis (NIRA) is a nondestructive analysis technique that studies the interactions between the incident light and the surface of a body identifying its components (STARK; LUCHTER; MARGOSHES, 1986). This technique has been widely used industrially due to its simplicity, speed and because it requires little or no sample preparation (CHANG et al., 2001).

NIRA was developed 40 years ago as a technique of quick analysis of grain moisture (BEN-GERA; NORRIS, 1968). Although initiated years ago, the use of remote sensing of soil surface spectra has increased significantly in the last decade due to the advancement of sensing technology. Currently, NIRA methods are applied in the pharmaceutical, petrochemical, and various areas industries (WORKMAN JUNIOR, 1996).

Since the end of the 1990s, NIRA has been used operationally in laboratories, enabling quick and accurate determinations of most features of soil samples (LAGACHERIE et al., 2008; VISCARRA-ROSSEL et al., 2006) as, for example, in studies to determine soil moisture (LOBELL; ASNER, 2002), organic carbon, electrical conductivity, CEC, pH, N, C, P, S, Ca, Mg, Na, K, Fe and Mn (BARNES et al., 2003; CHANG et al., 2001; DUNN et al., 2002; NANNI; DEMATTÊ, 2006; SHEPHERD; WALSH, 2002).

In recent years, several works worldwide have used reflectance spectroscopy to quantify soil attributes for various purposes. Kuang and Mouazen (2012) quantified soil attributes (clay and C) from different regions in Europe using soil reflectance. Viscarra Rossel et al. (2011) and Viscarra Rossel and Webster (2012) quantified soil attributes in Australia for pedological purposes. In the same year, Leone et al. (2012) used reflectance spectroscopy to analyze soil clay from the Mediterranean in southern Italy, while Gholizade et al. (2013) evaluated the reflectance in the Vis-Nir to predict soil chemical attributes in lowlands. Cezar et al. (2013) used this method to quantify clay in regions of Paraná State, Brazil, in an agricultural area. Volkani Bil吉利 et al. (2010), in turn, assessed C contents using spectroscopy, but in a semi-arid region of Turkey. Most works validate their models within their own modeling areas, which is biased. The question lies whether these models work in different regions. It is necessary, therefore, to test these models in regions outside the place where they were generated.

Regardless of the method for processing the spectral data, estimated values of soil attributes are highly correlated with the data obtained from conventional methods. Thus, this study correlates specific information about chemical, physical and mineralogical features of soil samples with their respective spectral data, obtaining equations of standardized multiple regression that can be applied to determine the same sample attributes of unknown features as an alternative to conventional soil analyses.

MATERIAL AND METHODS

The study site (Figure 1) covers approximately 28,000 ha in the northeastern region of the São Paulo State, in several municipalities such as São Carlos, Araraquara, Ibaté, among others. Altitudes in the region range from 450-900 m and the climate is Cwb, temperate with dry winter. The average annual temperature is 20°C and rainfall rates range from 1,000 to 1,800 mm annually.

The lithology of the site is represented mainly by the Serra Geral formations, Botucatu and Pirambóia (São Bento Group) and coverage of the Serra de Santana (Taubaté Group).

To obtain and validate the equations to determine soil attributes, the study site was divided into two subareas. Area 1 covered 15,000 ha and area 2 covered 13,000 ha. In area 1, soil samples were collected to obtain the spectral data and the analytical results using conventional methods, which were used as basis for calibration of models. In area 2, soil samples were also collected and their analyses were initially regarded as “unknown”. The reflectance spectra were obtained from these samples. The data of the spectrum were introduced to the quantification of soil attributes. Thus, the contents of samples were determined from soil samples of area 2, based on prediction models generated in area 1. To validate the process, laboratory analyses were subsequently carried out in samples of area 2, “unknown samples”, and compared with the results obtained by the sensor.
Hyperspectral remote sensing as an alternative to estimate soil attributes

Figure 1 - Scheme of distribution of sampling points in area 1 (calibration phase) and area 2 (validation stage)

Phase 1 - Obtaining database for the development of phase 2

In area 1, 319 points were allocated based on the toposequence method and the points were sampled with an auger at 0-20, 40-60 and 80-100 cm, designated layers A, B and C, respectively. The samples were analyzed physically for sand (coarse and fine), silt and clay contents according to methodology of Camargo, Klant and Kauffman (1987). They were analyzed chemically to determine the pH (CaCl$_2$, H$_2$O and KCl) and contents of OM, P, K, Mg, Ca, AI and H+Al, according to the methodology described by Raij et al. (2001). Based on these data, the sum of the bases (SB), the cationic exchange capacity (CEC), saturation by bases (V%) and saturation by Al (m%) were obtained. The total Fe$_2$O$_3$, SiO$_2$, Al$_2$O$_3$, in addition to the Ki index, totaling 19 dependent variables.

The spectral data were prepared according to the methodology proposed by Nanni and Demattê (2006). A matrix was generated to correlate the 22 bands and the 13 heights with the real values of the soil attributes, from which multiple regression equations were generated statistically that considered the determined soil attributes in conventional laboratory as dependent variables and the spectral data as independent variables. The equations estimated the following physical and chemical soil attributes, obtained from the correlation matrix: sand, clay, pH in H$_2$O, pH in KCl, OM, P, K, C, Mg, H+Al, Al, sum of bases (SB), CEC, Al by saturation (m), saturation by bases (V), Fe$_2$O$_3$, SiO$_2$, Al$_2$O$_3$, in addition to the Ki index, totaling 19 dependent variables.

Statistical Package for the Social Sciences was a software used to process the multiple linear regression analysis. The stepwise method was chosen to derive the estimation of equations for soil attributes based on the spectral data. According to Glantz and Slinker (1990), the stepwise method allows the user to establish more reliable regression models, because the process of entering dependent variables is gradual and sequential, one variable is entered at a time into a regression equation, improving the estimation ability of the method. This method is particularly useful in situations of a large dataset with many independent variables. It tracks and identifies, in this large dataset, a small subset of variables that correlates and determines the value of a dependent variable.

Besides the coefficient of determination $R^2$, other statistical parameters, such as RMSE (Root Mean Square Error) and EM (Error of the Mean), were also used to indicate the effectiveness of the estimation models of soil attributes. The RMSE, also defined as RMSD (Root Mean Square Deviation) (GLANTZ; SLINKER, 1990), provided an idea of deviations in relation to line 1:1 (BROWN et al., 2006; KOBAYASHI; SALAM, 2000; PROBERT et al., 1998; WOLSCHICK et al., 2007) and the percentage of error of the mean (EM) reported, on average, in terms of the observed value distancing from the estimated value in percentage.

Phase 2 - Obtaining the database for the development of phase 2

Phase 2 was developed by applying the equations obtained in phase 1 into the database obtained from area 2. In area 2, observation points in the field were allocated by the toposequence method and 225 samples were collected at 0-20, 40-60 and 80-100 cm, the samples were returned to the laboratory to obtain the spectral data.
designated as A, B and C layers, respectively. Similar to phase 1, the data were submitted to the same physical and chemical analysis in the laboratory and afterwards, it was used to obtain the corresponding spectral data, which were subsequently processed to complete the data matrix.

**Application of the equations obtained in phase 1 and data validation**

After the data matrix from area 2 were organized, it received the equations obtained from data of area 1, respectively for each physical, chemical and mineralogical attributed studied. This procedure generated a table with analytical results for each of the 225 soil samples similarly to a conventional routine analysis, however obtained from spectral data.

The observed data (spectral analytical results) were correlated with the real data (conventional analytical results) and plotted on graphs of dispersion. Associated with these graphs, the coefficients $R^2$, RMSD and EM were analyzed, which allowed determining the correlation between the conventional and spectral results.

**RESULTS AND DISCUSSION**

The results obtained for clay and sand contents indicate a high textural variability of the study site (Table 1) in agreement with the results obtained by Sousa Junior, Demattê and Araújo (2011). Other attributes such as P, K, $m\%$ and Al also showed high variability with the values of coefficient of variation (CV) above 100.

The equations for clay, sand and Mg (Table 2) showed the highest number of independent variables 10, 09, and 09 respectively. The smallest equation refers to K, with two variables. All other 19 equations showed at least one variable of height in their compositions.

The attributes with the best indices of determination were the $Fe_2O_3$, $Al_2O_3$, clay and sand, with $R^2$ values above 0.7, which is considered a satisfactory value according to Coleman et al. (1991). Although the Ki index showed a $R^2$ lower than 0.7 (0.61), it can be considered satisfactory due to low values of RMSD and ME (less than 15), showing that, on average, the observed value differed slightly from the estimated values, indicating a good relationship between the soil weathering rate and spectral reflectance.

**Table 1 - Descriptive statistics of the database (three depths) of soil attributes.** Average values of the entire dataset of 319 samples

<table>
<thead>
<tr>
<th>Attributes</th>
<th>Unit</th>
<th>n(1)</th>
<th>Min</th>
<th>Max</th>
<th>Average</th>
<th>SD(2)</th>
<th>CV(3)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sand</td>
<td>g Kg(^{-1})</td>
<td>871</td>
<td>60.00</td>
<td>914.00</td>
<td>637.64</td>
<td>197.80</td>
<td>31.02</td>
</tr>
<tr>
<td>Clay</td>
<td>g Kg(^{-1})</td>
<td>871</td>
<td>72.00</td>
<td>745.00</td>
<td>279.56</td>
<td>127.14</td>
<td>45.48</td>
</tr>
<tr>
<td>pH $H_2O$</td>
<td>-</td>
<td>871</td>
<td>3.90</td>
<td>6.90</td>
<td>5.32</td>
<td>0.50</td>
<td>9.42</td>
</tr>
<tr>
<td>pH $KCl$</td>
<td>-</td>
<td>871</td>
<td>3.60</td>
<td>6.20</td>
<td>4.64</td>
<td>0.53</td>
<td>11.45</td>
</tr>
<tr>
<td>OM</td>
<td>g Kg(^{-1})</td>
<td>871</td>
<td>2.00</td>
<td>33.00</td>
<td>10.55</td>
<td>5.06</td>
<td>47.93</td>
</tr>
<tr>
<td>P</td>
<td>mg Kg(^{-1})</td>
<td>871</td>
<td>1.00</td>
<td>144.00</td>
<td>6.06</td>
<td>11.78</td>
<td>194.50</td>
</tr>
<tr>
<td>K</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>0.10</td>
<td>10.60</td>
<td>0.75</td>
<td>1.05</td>
<td>140.60</td>
</tr>
<tr>
<td>Ca</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>2.00</td>
<td>67.00</td>
<td>13.24</td>
<td>10.12</td>
<td>76.46</td>
</tr>
<tr>
<td>Mg</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>1.00</td>
<td>29.00</td>
<td>3.89</td>
<td>3.39</td>
<td>87.11</td>
</tr>
<tr>
<td>H + Al</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>10.00</td>
<td>98.00</td>
<td>26.61</td>
<td>9.75</td>
<td>36.64</td>
</tr>
<tr>
<td>Al</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>0.00</td>
<td>11.00</td>
<td>1.95</td>
<td>1.96</td>
<td>100.47</td>
</tr>
<tr>
<td>S</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>3.00</td>
<td>94.00</td>
<td>17.89</td>
<td>13.31</td>
<td>74.39</td>
</tr>
<tr>
<td>CEC</td>
<td>mmol Kg(^{-1})</td>
<td>871</td>
<td>17.00</td>
<td>120.00</td>
<td>44.50</td>
<td>16.41</td>
<td>36.87</td>
</tr>
<tr>
<td>V</td>
<td>%</td>
<td>871</td>
<td>8.00</td>
<td>82.00</td>
<td>37.66</td>
<td>16.31</td>
<td>43.31</td>
</tr>
<tr>
<td>m</td>
<td>%</td>
<td>871</td>
<td>0.00</td>
<td>63.00</td>
<td>15.07</td>
<td>16.22</td>
<td>107.58</td>
</tr>
<tr>
<td>$Fe_2O_3$</td>
<td>g Kg(^{-1})</td>
<td>64</td>
<td>27.20</td>
<td>303.20</td>
<td>103.68</td>
<td>70.22</td>
<td>67.73</td>
</tr>
<tr>
<td>$Al_2O_3$</td>
<td>g Kg(^{-1})</td>
<td>64</td>
<td>26.00</td>
<td>219.00</td>
<td>122.14</td>
<td>43.35</td>
<td>35.49</td>
</tr>
<tr>
<td>$SiO_2$</td>
<td>g Kg(^{-1})</td>
<td>64</td>
<td>25.00</td>
<td>182.00</td>
<td>84.45</td>
<td>29.02</td>
<td>34.36</td>
</tr>
<tr>
<td>Ki</td>
<td>-</td>
<td>64</td>
<td>0.57</td>
<td>1.73</td>
<td>1.22</td>
<td>0.26</td>
<td>21.59</td>
</tr>
</tbody>
</table>

(1) Number of samples; (2) Standard Deviation; (3) Coefficient of Variation
Table 2 - Multiple regression equations developed from soil reflectance at ground level

<table>
<thead>
<tr>
<th>Attributes</th>
<th>Multiple Equations(1)</th>
<th>( R^2 )</th>
<th>RMSD(2)</th>
<th>EM(3)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sand (g kg(^{-1}))</td>
<td>( \text{Sand} = 492.670 + 9238.361<em>B1 – 17757.3</em>B7 + 13103.650<em>B8 – 16954.5</em>B13 + 9377.436<em>B16 + 9137.825</em>B19 + 35906.619<em>B5 – 2596.709</em>B8 – 2483.122*B12 )</td>
<td>0.71</td>
<td>106.31</td>
<td>21.36</td>
</tr>
<tr>
<td>Clay (g kg(^{-1}))</td>
<td>( \text{Clay} = 344.485 + 21560.041<em>B2 + 1973.068</em>B7 – 3895.359<em>B21 – 10231.5</em>B5 – 30019.3<em>B3 + 21789.5</em>B5 + 6386.753<em>B4 + 2877.608</em>B8 + 6253.900<em>B13 – 4353.122</em>B16 )</td>
<td>0.75</td>
<td>63.8</td>
<td>20.65</td>
</tr>
<tr>
<td>( \text{pH}_2 \text{O} )</td>
<td>( \text{pH}_2 \text{O} = 5.548 + 47.487<em>B8 – 105.410</em>B5 + 31.770<em>B7 + 15.732</em>B10 )</td>
<td>0.19</td>
<td>0.45</td>
<td>6.88</td>
</tr>
<tr>
<td>( \text{pH} \text{KCl} )</td>
<td>( \text{pH} \text{KCl} = 4.984 + 18.388<em>B10 + 48.163</em>B8 + 15.798*B9 )</td>
<td>0.24</td>
<td>0.46</td>
<td>7.99</td>
</tr>
<tr>
<td>OM(4) (g kg(^{-1}))</td>
<td>( \text{OM} = 24.887 – 160.911<em>B21 – 185.125</em>B3 )</td>
<td>0.32</td>
<td>4.16</td>
<td>37.88</td>
</tr>
<tr>
<td>P (mg kg(^{-1}))</td>
<td>( \text{P} = (0.84 + 2.016<em>B3 + 7.911</em>H5 – 1.408<em>H8 – 0.839</em>H11 + 0.994<em>H4 + 0.196</em>H12) )</td>
<td>0.31</td>
<td>11.29</td>
<td>57.54</td>
</tr>
<tr>
<td>K (mmol(^{+}) kg(^{-1}))</td>
<td>( \text{K} = 10(-0.225 + 11.788<em>H11 + 15.520</em>H3) )</td>
<td>0.1</td>
<td>1.08</td>
<td>106.42</td>
</tr>
<tr>
<td>Ca (mmol(^{+}) kg(^{-1}))</td>
<td>( \text{Ca} = 10(1.311 + 4.059<em>H10 + 9.541</em>H9 – 56.672<em>H5 + 17.047</em>H7) )</td>
<td>0.42</td>
<td>8.21</td>
<td>47.03</td>
</tr>
<tr>
<td>Mg (mmol(^{+}) kg(^{-1}))</td>
<td>( \text{Mg} = (0.810 + 4.797<em>H6 – 5.059</em>H8 + 2.575<em>B6 – 1.023</em>H12 – 2.695<em>H7 + 8.576</em>H5 + 0.751<em>B22 + 2.761</em>H1) )</td>
<td>0.41</td>
<td>0.15</td>
<td>9.04</td>
</tr>
<tr>
<td>H+Al(5) (mmol(^{+}) kg(^{-1}))</td>
<td>( \text{H+Al} = (5.797 + 26.891<em>H11 + 63.635</em>H5 + 27.232<em>H1 – 30.428</em>H3 + 23.031*H9) )</td>
<td>0.16</td>
<td>9.03</td>
<td>25.80</td>
</tr>
<tr>
<td>SiO(_2) (g kg(^{-1}))</td>
<td>( \text{SiO}_2 = 10(1.291 – 6.225<em>B22 + 1.396</em>B5 – 4.150<em>H11 – 2.955</em>H13 + 2.926*B14) )</td>
<td>0.44</td>
<td>10.67</td>
<td>44.57</td>
</tr>
<tr>
<td>Fe(_2)O(_3) (g kg(^{-1}))</td>
<td>( \text{Fe}_2\text{O}_3 = 10(1.242 + 30.182<em>H7 + 8.823</em>H11 – 9.005<em>H9 – 5.283</em>H4 – 5.957<em>B2 + 5.860</em>H3) )</td>
<td>0.82</td>
<td>3.53</td>
<td>24.13</td>
</tr>
<tr>
<td>V(9) (%)</td>
<td>( \text{V} = 46.148 + 465.445<em>H10 + 1733.207</em>H8 + 333.806<em>H12 – 892.560</em>H6 – 3230.406<em>H5 + 1174.916</em>H7 )</td>
<td>0.29</td>
<td>13.79</td>
<td>38.72</td>
</tr>
<tr>
<td>Fe(_2)O(_3) (g kg(^{-1}))</td>
<td>( \text{Fe}_2\text{O}_3 = 10(1.373 – 5.714<em>B20 – 2.883</em>B2 + 6.810<em>B13 – 4.254</em>B17) )</td>
<td>0.81</td>
<td>2.21</td>
<td>16.9</td>
</tr>
<tr>
<td>( \text{Ki} )</td>
<td>( \text{Ki} = 1.391 – 15.875<em>H11 + 10.696</em>H2 + 15.438<em>H3 – 25.592</em>H9 + 26.745<em>H7 + 2.890</em>B18 )</td>
<td>0.61</td>
<td>0.19</td>
<td>13.4</td>
</tr>
</tbody>
</table>

Notes:
(1) bands and heights selected; (2) Root Mean Standard Deviation; (3) Error of the Mean; (4) organic matter; (5) hydrogen plus aluminum; (6) sum of bases (Ca + Mg + K); (7) Cationic Exchange Capacity (SB + H + Al); (8) saturation for bases [(SB/CEC*100)]; (9) Saturation for aluminum (Al/Al + SB) * 100
The other chemical attributes showed $R^2$ values lower than 0.5, while K showed the worst results, agreeing with the results of Sousa Junior, Demattê and Araújo (2011), which studied soils in the same region.

The determined equations were applied to the database of area 2 for all elements, and the estimated results were correlated with the real value, that is, determined by traditional methods (Figure 2). The best results were obtained from physical attributes with $R^2$ of 0.67 and 0.61 for clay and sand, respectively, and for the attributes determined by sulfuric attack with $R^2$ of 0.69, 0.81, 0.57 and 0.59 for $\text{Fe}_2\text{O}_3$, $\text{Al}_2\text{O}_3$, $\text{SiO}_2$ and K, respectively.

The other chemical elements (pH in KCl, OM, Al, CEC, v%, V%, pH in H$_2$O, P, K, Ca, Mg) showed $R^2$ values ranging from 0.02 to 0.3, which denotes that these attributes were not satisfactorily determined in the proposed method.

The statistical process used in the choice of the bands and heights used in the models to determine

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**Figure 2** - Comparison between the values of soil attributes determined in conventional laboratory (determined values) and estimated values by linear regression equations from soil reflectance data.
the studied attributes allowed to separate the most representative for each attribute, agreeing with studies of Huete and Escadafal (1991), Demattê and Garcia (1999), Nanni and Demattê (2006). However, in these studies, the authors found a number of different variables to determine the respective soil attributes, with variations in the number of variables, showing that the variations may change according to the dataset and study site. Demattê and Garcia (1999) indicated differences in estimates when the dataset was made for a region and from various regions together.

In regards to the $R^2$ values obtained for the equations (Table 2), similar studies for attributes obtained by sulphuric attack, Demattê and Garcia (1999) showed $R^2$ for $\text{Fe}_2\text{O}_3$ of 0.64, while Nanni and Demattê (2006) obtained $R^2$ of 0.95 for $\text{SiO}_2$, and 0.93 for $\text{Fe}_2\text{O}_3$. Although the values determined in this study for $\text{Al}_2\text{O}_3$, $\text{Fe}_2\text{O}_3$ and $\text{SiO}_2$ are less expressive, they are still considered satisfactory, as they remained within the range 0.6 to 0.9, which encompasses the limits of these attributes in research in this field, corroborating reports of Ben-Dor and Banin (1995a). In addition, the RMSD and EM values
showed good association between the estimated data and the real data used to generate the equations.

For the Ki attribute, Demattê and Garcia (1999) used basalt soils and observed $R^2$ of 0.69 for this attribute, inferring that the source material and, consequently, the study site interferes with the relationship between the weathering rates and their spectral response, which is no longer the case with soil physical attributes that show a closer relationship, as observed by several authors (CHANG et al., 2001; SHEPHERD; WALSH, 2002; SOUSA JUNIOR; DEMATTÊ; ARAÚJO, 2011), proving that estimating these attributes by remote sensing is efficient.

Variations of $R^2$ in several studies on the use of spectral reflectance to determine soil attributes are attributed to a number of factors, including observed soil variability, samples representativeness, number and specificity of the bands selected and equipment used as acquisition geometry. Regarding OM, Ben-Dor and Banin (1995b) reported that the degradation stage (type of OM) should be taken into account when predicting this attribute, justifying the $R^2$ 0.5 obtained in their studies.

For all other elements, the results were contrary to those observed by Dunn et al. (2002), who obtained $R^2$ of 0.72 for K in soils of southern Australia. Similar satisfactory results, but in different locations, were found by Chang et al. (2001) who obtained CEC (0.81), Ca (0.75), K (0.55), and Mg (0.68). Shepherd and Walsh (2002) obtained 0.88 for CEC, 0.88 for Ca, and 0.81 for Mg, similar to Islam, Singh and Mcbratney (2003), who obtained CEC (0.64), Ca (0.67), and Mg (0.63). However, these results are related to $R^2$ values obtained from the relationship between observed and determined data that were used to generate the equations. At this stage of this study, the $R^2$ values were obtained also from the relationship between the estimated and observed data, but without any dependency between these two datasets. Thus, the fact that the $R^2$ values are low in certain cases is a result of the equations generated, which were not satisfactory.

In turn, the physical attributes and those determined conventionally by sulfuric attack were best estimated by the equations. In the case of physical attributes, these findings reinforce several works, such as Ben-Dor and Banin (1995a), Shepherd and Walsh (2002), Islam, Singh, and Mcbratney (2003), Nanni and Demattê (2006), who reported the potential of this methodology to predict these attributes.

With regard to the attribute obtained by sulphuric attack, Nanni and Demattê (2006) had already found high correlation values between these attributes and those obtained in the spectral method, the values of $R^2$ of 0.88 ($Fe_2O_3$), 0.87 ($Al_2O_3$), and 0.72 ($SiO_2$). However, in these works, the equations were not applied to a database from a different area. The discussion about the attribute $Fe_2O_3$ is relevant as it is directly related to a particular soil class (EMPRESA BRASILEIRA DE PESQUISA AGROPECUÁRIA, 2006) inferring mainly in its classification. However, in most pedological works, analyses of sulphuric acid digestion determined are a small percentage of the total number of samples. This happens because of the high cost of these analyses and the slowness to obtain the results. Thus, determining the efficiency of these equations within an area allows to determine quantitative values of $Fe_2O_3$, $Al_2O_3$, $SiO_2$, $Ki$, among others, for all sampled points, along with the low cost of analyses and rapid results.

CONCLUSIONS

1. The proposed method was effective as an alternative to determine the attributes $Fe_2O_3$, $Al_2O_3$, $SiO_2$, sand and clay, with $R^2$ values reaching 0.8, providing faster analyses and less environmental impact since chemical reagents are not required;

2. The pH elements in $H_2O$, pH in KCl, OM, P, K, Ca, Mg, H+Al, Al, SB, CEC, m% and V% were not properly estimated, obtaining low $R^2$ values. Elements related to soil chemistry showed good and weak values, which denotes the need for further studies;

3. These results can be highly beneficial in various applications such as soil mapping and surveying, determining analyses and in precision agriculture. The method does not exclude the importance of traditional determinations, but it can aggregate information and optimize the conventional work.

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