

Phytochemistry of *Trattinnickia burserifolia*, *T. rhoifolia*, and *Dacryodes hopkinsii*: Chemosystematic Implications

M. da Paz Lima^a, Patrícia A. de Campos Braga^b, Mario Lopes Macedo^b, M. Fátima das G. F. da Silva^{*,b}, A. Gilberto Ferreira^b, João B. Fernandes^b and Paulo C. Vieira^b

^aInstituto Nacional de Pesquisa da Amazônia, Coordenação de Pesquisas em Produtos Naturais,
CP 478, 69011-970 Manaus - AM, Brazil

^bDepartamento de Química, Universidade Federal de São Carlos, CP 676, 13565-905 São Carlos - SP, Brazil

O estudo de *Trattinnickia burserifolia* levou ao isolamento dos triterpenos conhecidos ursanos α -amirenona, α -amirina, 3-*epi*- α -amirina, 3 α ,16 β -diidroxiolean-12-eno; oleananos β -amirenona, β -amirina, 3-*epi*- β -amirina, 3 α ,16 β -diidroxiolean-12-eno; tirucalanos ácidos 3 α -hidroxitirucal-8,24-dien-21-óico, 3 α -hidroxitirucal-7,24-dien-21-óico, e 3-oxotirucal-8,24-dien-21-óico; damaranos dammarenediol-II e 3 α ,20(S)-diidroxidamar-24-eno. Além desses foram ainda isolados o monoterpene novo 2(S*)-fenilacetoxi-4(R*)-p-menta-1(7),5-dieno, e os triterpenos novos 3 β -fenilacetoxiurs-12-eno, 3 β -fenilacetoxiolean-12-eno e 3 β ,16 β ,11 α -triidroxiolean-12-eno. Os triterpenos de *T. burserifolia*, *T. rhoifolia* e *Dacryodes* foram analisados em mistura. Os espectros de RMN ¹³C mostraram que os principais triterpenos eram α -amirina e β -amirina em *T. burserifolia*; α -amirina, β -amirina, 3-*epi*- α -amirina, 3-*epi*- β -amirina, lupenona, ácidos 3 α -hidroxitirucal-8,24-dien-21-óico e 3 α -hidroxitirucal-7,24-dien-21-óico em *T. rhoifolia*; α -amirina, β -amirina, lupeol, tirucalol, sitosterol e estigmasterol em *D. hopkinsii*. A quimiosistemática da tribo Protieae é discutida.

Trattinnickia burserifolia has yielded the known ursanes, α -amyrenone, α -amyrin, 3-*epi*- α -amyrin, 3 α ,16 β -dihydroxyurs-12-ene, the oleananes β -amyrenone, β -amyrin, 3-*epi*- β -amyrin, 3 α ,16 β -dihydroxyolean-12-ene, the tirucallane acids 3 α -hydroxytirucall-8,24-dien-21-oic, 3 α -hydroxytirucall-7,24-dien-21-oic and 3-oxotirucall-8,24-dien-21-oic, the dammaranes dammarenediol-II and 3 α ,20(S)-dihydroxydammar-24-ene. Besides it was isolated the new monoterpene 2(S*)-phenylacetoxi-4(R*)-p-mentha-1(7),5-diene and, the new triterpenes 3 β -phenylacetoxiurs-12-ene, 3 β -phenylacetoxiolean-12-ene and 3 β ,16 β ,11 α -trihydroxyurs-12-ene. The triterpenes from *T. burserifolia*, *T. rhoifolia* and *Dacryodes* were analyzed in mixture. Their ¹³C NMR spectra showed that the major triterpenes were in *T. burserifolia* α -amyrin and β -amyrin; in *T. rhoifolia* α -amyrin, β -amyrin, 3-*epi*- α -amyrin, 3-*epi*- β -amyrin, and lupenone; in *T. rhoifolia* α -amyrin, β -amyrin, 3-*epi*- α -amyrin, 3-*epi*- β -amyrin, 3 α -hydroxytirucall-8,24-dien-21-oic acid and 3 α -hydroxytirucall-7,24-dien-21-oic acid; in *D. hopkinsii* α -amyrin, β -amyrin, lupeol, tirucallol, sitosterol and stigmasterol. Aspects of chemosystematic of the tribe Protieae are discussed.

Keywords: Burseraceae, monoterpene, triterpenes, chemosystematic

Introduction

The Burseraceae has usually been considered to contain 21 genera and nearly 600 species. Engler (1931) classified these genera into three tribes.¹ The Protieae consist of four genera exhibiting many morphological characters regarded as primitives. Three Protieae genera occur in tropical America and one in Asia. The following group the Boswellieae contain eight genera centred in

Africa and Asia. In contrast, the Canarieae, represented by nine genera, appear more advanced in their morphology. This tribe is predominantly Paleotropical, therefore, two genera occur in South America. Later Lam (1932) recognised these tribes but replaced the name Boswellieae by Bursereae.²

Crepidospermum Hook. is a member of the Protieae and consists of five species distributed in the tropical South America. Swart in 1942 on morphological grounds described the genus *Hemicrepidospermum* to accommodate *C. rhoifolium*.³ However, other aspects of

* e-mail: dmfs@power.ufscar.br

their morphology have led Daly (1989) to consider *Hemicrepidosperrum* a section of *Crepidosperrum*; the two sections have three and two species, respectively.⁴ The following tropical S. American genera of the Proteaceae, *Tetragastris* and *Protium*, have long been considered closely related, in fact, many specimens of each genus have been mistakenly referred to the other.⁴ *Garuga* is the only representative of Asian Proteaceae and its morphology is easily recognisable.⁴

Trattinnickia was also a member of the Proteaceae, however, morphological and anatomical evidence have led Daly (1989) to transfer it into the Canarieae and to propose a taxonomic position close to *Dacryodes*.⁴

Within tribe Proteaceae phytochemical data were not available for *Crepidosperrum*, *Tetragastris*, *Trattinnickia* and *Dacryodes*. As part of our chemosystematic interest in the Brazilian Burseraceae, we recently reported the phytochemical investigation of *Crepidosperrum rhoifolium* Benth. and *Tetragastris altissima* (Aublet) Swart.⁵ Thus, we have now examined the resin, stem bark and branches of *Trattinnickia burserifolia* Engl., *T. rhoifolia* var. *willdenowii* Engl. and *Dacryodes hopkinsii* Daly.⁴

Results and Discussion

Chemical composition of the extracts

A chloroform-soluble fraction of the resin of *T. burserifolia* afforded one new monoterpene (**1**), three new triterpenes (**2-4**) and the known ursanes, α -amyrenone,⁶ α -amyrin (**5**), 3-*epi*- α -amyrin, 3 α ,16 β -dihydroxyurs-12-ene

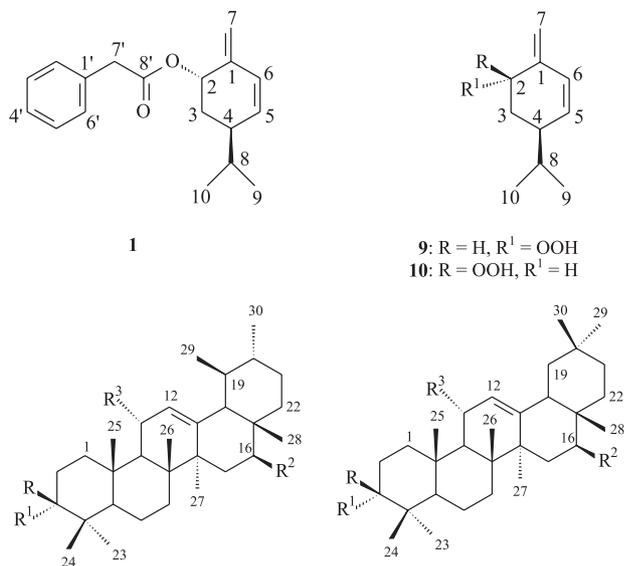
(**6**),⁷ the oleananes β -amyrenone,⁶ β -amyrin (**7**), 3-*epi*- β -amyrin, 3 α ,16 β -dihydroxyolean-12-ene (**8**),⁷ the tirucallane acids 3 α -hydroxytirucall-8,24-dien-21-oic,⁸ 3 α -hydroxytirucall-7,24-dien-21-oic,⁶ 3-oxotirucall-8,24-dien-21-oic,⁸ and the dammaranes dammarenediol-II and 3 α ,20(S)-dihydroxydammar-24-ene.⁵

The ¹H NMR spectrum (Table 1) of compound **1** showed signals for a terminal methylene (δ 5.08, br s, and 5.00, br s), two olefinic protons which were coupled to each other (δ 5.80, dd, *J* 10.1 and 1.0 Hz; 6.14, dd, *J* 10.1 and 2.5 Hz), two methyl doublet (δ 0.88, d, *J* 6.8 Hz; 0.86, d, *J* 6.8 Hz), an oxymethine (δ 5.57, dd, *J* 5.0 and 2.8 Hz), an aryl substituted methylene (δ 3.61, 2H, s) and five protons as a multiplet between δ 7.30 and 7.25, clearly indicating the presence of a phenyl group. From HMBC experiments the observed correlations between the methylene protons at δ 5.08 and 5.00 and the ¹³C signals at δ 127.2 and 71.8, requiring the presence of a conjugated double bond and an allylic proton attached to a carbon adjacent to an oxygen atom. HSQC experiments showed correlations of terminal methylene and conjugated double bond protons (δ 5.80 and 6.14) with the ¹³C signals at δ 115.4, 133.7 and 127.2, respectively. HSQC also permitted the assignment of the signal at δ 41.8 to aryl substituted methylene at δ 3.61, which showed cross peaks with the ¹³C signals of the aromatic ring (C-1', C-2' and C-6') and carboxyl at δ 171.1, indicating a phenylacetoxyl substituent. This group must be connected allylic to terminal methylene, due to the observed downfield shifted proton signal at δ 5.57, which showed one-bond correlation with the ¹³C signal at δ 71.8. These correlations resulted in the construction of a CH₂=C[CH(R)OCOCH₂Ph]CH=CHR system.

Table 1. ¹³C and ¹H NMR spectral data for compounds **1** and the model compounds **9** and **10**

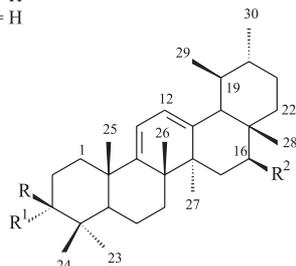
C	1	9	10	H	1 (2S)	9 (2S)	10 (2R)
1	140.5	139.4	141.7	OOH		7.73 s	7.87 s
2	71.8	82.6	82.5	2	5.57 dd (5.0, 2.8)	4.67 dd (3.5, 3.0)	4.68 dm (11.0)
				3a	1.89 dt (13.6, 5.0)	2.17 dtd (13.7, 4.9, 3.5, 1.4)	2.22 dt (11.0, 4.2)
3	30.4	29.0	30.1	3b	1.55 ddd (13.6, 9.8, 2.8)	1.57 td (13.7, 11.5, 3.0)	1.63 q (11.0)
4	37.8	36.9	42.0	4	2.20 m	2.26 m	2.25 m
5	133.7	134.1	133.3	5	5.80 dd (10.1, 1.0)	5.80 d (10.0)	5.71 d (9.7)
6	127.2	126.7	128.6	6	6.14 dd (10.1, 2.5)	6.13 dd (10.0, 2.5)	6.13 dd (9.7, 2.4)
7	115.4	116.7	109.9	7 _a	5.08 br s	5.15 s	5.17 s
8	31.4	31.4	31.9	7 _b	5.00 br s	5.13 s	4.98 s
9	19.5	19.2	19.6	8	1.60 m	1.69 sept.d (6.8, 5.5)	1.74 sept.d (6.8, 5.2)
10	19.4	19.2	19.2	9	0.88 d (6.8)	0.90 d (6.8)	0.94 d (6.8)
1'	134.3			10	0.86 d (6.8)	0.90 d (6.8)	0.92 d (6.8)
2'/6'	129.3			2'-6'	7.25-7.30 m		
3'/5'	128.5			7'	3.61 s		
4'	127.0						
7'	41.8						
8'	171.1						

Assignments based on HSQC and HMBC for **1**.



2: R = OCOCH₂Ph;
R¹, R², R³ = H
4: R, R², R³ = OH; R¹ = H
5: R = OH; R¹, R², R³ = H
6: R¹, R² = OH; R, R³ = H
6a: R¹, R² = OAc; R, R³ = H
11: R, R² = OH; R¹, R³ = H
12: R¹, R³ = OH; R, R² = H

3: R = OCOCH₂Ph,
R¹, R², R³ = H
7: R = OH, R¹, R², R³ = H
8: R¹, R² = OH; R, R³ = H
8a: R¹, R² = OAc; R, R³ = H



4a: R, R² = OAc; R¹ = H
13: R = OAc; R², R³ = H

The remaining unassigned ¹³C signals accounted for 2 CH₃, CH₂ and 2 CH, of which two methyls were coupled to a methine, suggesting the presence of a monoterpene *p*-menthane skeleton to be substituted with phenylacetoxyl. The identification of the nucleus as a *p*-menthane was supported by comparison of the ¹³C NMR spectrum (Table 1) with those of 2(*S*)-hydroperoxy-4(*R*)-*p*-mentha-1(7),5-diene (**9**) and 2(*R*)-hydroperoxy-4(*R*)-*p*-mentha-1(7),5-diene (**10**) obtained from photooxygenation of (-)-(*R*)- α -phellandrene.⁹ Based on the ¹H and ¹³C NMR data for **9** and **10**, in compound **1** the resonances for H-2 (δ 5.57, dd, *J* 5.0 and 2.8 Hz) and C-4 (δ 37.8) were characteristic of 2(*S*)-4(*R*)-*p*-menthane derivative, but this has not been confirmed. Compound **1** is thus 2(*S*^{*})-phenylacetoxyl-4(*R*^{*})-*p*-mentha-1(7),5-diene.

The two new triterpenes **2** and **3** showed a single spot on TLC in various solvent systems and attempts to separate this mixture into its constituents were not successful. They

also showed the spectral characteristics of a phenylacetoxyl substituent. The ¹H and ¹³C NMR spectra of this mixture in addition to signals described above for phenylacetoxyl, revealed resonances for C-1 and H-1 to C-30 and H-30 in close agreement with those for α -amyrin (**5**) and β -amyrin (**7**), respectively^{7, 10} (Table 2). The downfield shift of the signals for C-3 (δ 81.4) and H-3 (δ 4.48) in the ¹H and ¹³C NMR spectra, when compared with **5** and **7**, determined the position of the phenylacetoxyl at C-3 in both the compounds of the mixture. The phenylacetoxyl present at C-3 β was evident by resonance at δ 4.48 with a large coupling constant (*J* 11.0 and 5.1 Hz). The structure of the new natural products were thus established as 3 β -phenylacetoxyl-urs-12-ene (**2**) and 3 β -phenylacetoxyl-olean-12-ene (**3**).

Table 2. ¹³C NMR spectrum data for compounds **2**, **3** and the model compounds **5** and **7**

C	2	3	5	7	C	2	3
1	38.5	38.4	38.7	38.7	1'	134.5	134.5
2	27.9	27.0	27.2	27.3	2'	129.3	129.3
3	81.4	81.4	78.3	79.0	3'	128.5	128.5
4	37.8	37.8	38.7	38.8	4'	126.9	126.9
5	55.3	55.3	55.2	55.3	5'	128.5	128.5
6	18.2	18.2	18.3	18.5	6'	129.3	129.3
7	32.5	32.5	32.9	32.8	7'	42.1	42.1
8	40.1	40.1	40.0	38.8	8'	171.3	171.3
9	47.7	47.6	47.7	47.7			
10	36.8	36.8	36.9	37.6			
11	23.4	23.7	23.3	23.6			
12	124.4	121.7	124.3	121.8			
13	139.7	145.0	139.3	145.1			
14	42.1	41.8	42.0	41.8			
15	27.9	26.2	28.7	26.2			
16	26.6	25.9	26.6	27.0			
17	32.9	32.8	33.7	32.5			
18	59.1	47.3	58.9	47.4			
19	39.7	46.8	39.6	46.9			
20	39.6	31.1	39.6	31.1			
21	31.3	34.8	31.2	34.8			
22	41.6	37.3	41.5	37.2			
23	28.1	28.1	28.1	28.2			
24	15.7	15.5	15.6	15.5			
25	16.8	16.6	15.6	15.6			
26	16.9	16.8	16.8	16.9			
27	23.3	25.9	23.3	26.0			
28	28.8	28.4	28.1	28.4			
29	17.5	33.8	17.4	33.3			
30	21.4	23.7	21.3	23.7			

The new triterpene **4** was identified on the basis of the following data. The ¹H NMR spectrum indicated the presence of three signals characteristics of protons attached to a carbon adjacent to an oxygen atom (δ 3.23, dd, *J* 10.4 and 5.8 Hz; 4.23, dd, *J* 11.1 and 5.2 Hz; 4.27, dd, *J* 8.7 and 3.2 Hz), one olefinic proton (δ 5.24, d, *J* 3.2 Hz), and eight

methyl groups, six of them on quaternary carbons and two of them on a methine group, suggesting a urs-12-ene skeleton. From the HMBC experiments (Table 3) the observed correlations between the two methyl protons at δ 0.77 and 0.97 and the ^{13}C signals at δ 78.7 (3J ; CH by DEPT), 55.4 (3J ; CH), 39.3 (2J ; quaternary carbon), 28.4 (3J for methyl at δ 0.97; CH_3) and 15.7 (3J for methyl at δ 0.77; CH_3) led to their assignments as C-3, C-5, C-4, C-23 and C-24, respectively. Based on the HSQC experiments the signal at δ 0.77, 0.97 and 3.23 were then assigned to Me-24, Me-23 and H-3, respectively. The methyl proton at δ 1.06 (δ_{C} 17.0) showed long-range correlation with the ^{13}C signal for C-5 (δ 55.4), permitting the assignment of these signals to H_3 -25 and C-25, respectively. The signal for H_3 -25 also showed correlations with the ^{13}C signals at 55.3 (CH), 40.9 (CH_2) and 38.1 (quaternary carbon), showing that these signals correspond to C-9, C-1 and C-10, respectively. The olefinic proton at δ 5.24 showed cross peaks with the C-9 signal (δ 55.3), and was coupled to the ^1H signal at δ 4.27, thus indicating a hydroxyl group to be located at C-11 and a double bond at C-12. The late oxymethine proton showed one-bond correlation with the ^{13}C signal at δ 68.2 and long-range correlation with the ^{13}C signals at δ 141.3 and 129.2, allowing the assignment of these to C-11, C-13 and C-12, respectively. Moreover, the existence of correlations between H-12 and the ^{13}C

signals at δ 59.9 (CH) and 44.3 (quaternary carbon) led to their assignments as C-18 and C-14, respectively. A fourth methyl proton at δ 1.20 (δ_{C} 24.3) was attributed to H_3 -27 by its correlations with the C-13 (δ 141.3) and C-14 (δ 44.3) signals. The H_3 -27 signal also showed a cross-peak with the signal at δ 43.6, confirming a methyl group at C-8. In the same way, the unsubstituted C-15 emerged from the correlation between the H_3 -27 signal and the ^{13}C signal at δ 36.0 (3J ; CH_2), which showed one-bond correlation with the ^1H signal at δ 1.36 (m). This signal was coupled to the ^1H signal at δ 4.23 (dd, J 11.1 and 5.2 Hz), requiring the presence of a hydroxyl function at C-16. The coupling constants indicated that the hydroxyl group was attached β (equatorial) to C-16 and was coupled only to H_2 -15, indicating C-17 fully substituted. This was supported by the relationship of the H-16 (δ 4.23) signal to the ^{13}C signal at δ 21.9, which showed one-bond correlation with the methyl proton at δ 0.73, and long-range correlation with the ^{13}C signals for C-18, C-16 and at δ 38.5 (quaternary carbon) and 35.2 (CH_2). The signals at δ_{H} 0.73, δ_{C} 21.9, 38.5 and 35.2 were then assigned to H_3 -28, C-28, C-17 and C-22, respectively. A sixth methyl proton at δ 1.05 (δ 18.0) was attributed to H_3 -26 by its correlation with the C-9 and C-14. H_3 -26 signal also showed cross peaks with the ^{13}C signal at δ 43.6 (quaternary carbon) and 33.7 (CH_2), which were attributed to C-8 and C-7, respectively. A seventh methyl proton at δ_{H} 0.82 (d, J 6.3 Hz; δ_{C} 17.8) was attributed to H_3 -29 by its correlation with the C-18 signal. H_3 -29 signal also showed cross peaks with the ^{13}C signal at δ 39.1 (CH) and (or) 39.5 (CH), suggesting a methine for C-20, indicating a methyl group to be located at C-20 and confirming a urs-12-ene skeleton. Thus, the eighth methyl proton at δ_{H} 0.91 (d, J 5.9 Hz; δ_{C} 21.5) was attributed to H_3 -30. The signal for C-21 was established as δ 30.4 (CH_2 ; δ_{H} 1.42 m, by HSQC) by the existence of a correlation between the H_3 -30 signal and this ^{13}C signal.

The stereochemistry suggested for **4** was based on the biosynthesis of urs-12-enes. However, for C-3, C-11 and C-16 the stereochemistry were assigned by coupling constants and NOESY experiments. A model shows that, in compound **4**, ring A is nearer to a chair conformation, in which H-3 and H-5 are on the α -side of the molecule. This was supported by NOESY experiments (Table 3), which showed correlation of the signal of H-3 α (δ 3.23; OH-3 β) with the signal of H-5 α (δ 0.72 m, by HSQC). Moreover, the existence of a correlation from H-3 to H_3 -23 (δ 0.97) confirmed that Me-23 is in the α -configuration. In addition, the signal of H-11 (δ 4.27) showed cross-peaks with the signals of H_3 -25 (δ 1.06) and H_3 -26 (δ 1.05), suggesting a spatial proximity of H-11 to Me-25 and Me-26, which requires 11-OH to be in the α -configuration. The

Table 3. HMBC assignments for **4** and **4a**, and G-NOESY for **4**

4 HMBC		4a HMBC	
H	C	H	C
3	24	3	Ac (170.0), 23, 24
9	5, 8, 10, 11, 26, 25		
11	12, 13	11	8, 10, 13
12	9, 14, 18	12	9, 14, 18
15	13, 16, 17		
16	28	16	Ac (169.8), 28
18	12, 13, 14, 16, 17, 19 (or 20)	18	12, 13, 14, 17
23	3, 4, 5, 24	23	3, 4, 5, 24
24	3, 4, 5, 23	24	3, 5, 23
25	1, 5, 9, 10	25	1, 5, 9, 10
26	7, 8, 9, 14	26	7, 8, 14
27	8, 13, 14, 15	27	8, 13, 14, 15
28	16, 17, 18, 22	28	16, 17, 18, 22
29	18, 19 (or 20)	29	18, 19, 20
30	19 (or 20), 21	30	19, 20, 21

4 G-NOESY	
H	H
2H-1	2H-2, Me-25
H-3	H-2, H-5, Me-23
2H-7	H-5, Me-26, Me-27
H-11	Me-25, Me-26
H-12	H-11, H-18, Me-29
H-16	Me-27

relationship of H-16 signal (δ 4.23) to the H₃-27 (α , δ 1.20) indicated that 16-OH is in the β -configuration.

The ESI-MSMS showed ions at m/z 457 [M - H]⁻, 439 [M - H - H₂O]⁻ and 421 [M - H - H₂O - H₂O]⁻, confirming the presence of hydroxyl groups and thus the molecular formula (C₃₀H₅₀O₃). Based in the above evidence the structure of this compound was thus established as 3 β ,16 β ,11 α -trihydroxyurs-12-ene (**4**). The structural assignment was also supported by comparison of the ¹³C NMR spectrum (Table 4) with those of 3 β ,16 β -dihydroxyurs-12-ene (**11**)⁷ and 3 α ,11 α -dihydroxyurs-12-ene (**12**).¹⁰ In order to confirm of the assignments for **6a** and **8a** discussed below, **4** was acetylated. This reaction involved dehydration of C-11 alcohol and acetylation of the C-3 and C-16 hydroxyl groups to give 3 β ,16 β -diacetoxys-9(11),12-diene (**4a**). The ¹H NMR spectrum

of **4a** revealed the downfield shift of the signals for H-3 (δ 4.51, dd, J 11.4 and 4.9 Hz) and H-16 (δ 5.46, dd, J 11.4 and 5.5 Hz). From the HMBC experiments (Table 3) the observed correlations between the two methyl protons at δ 0.88 and 0.90 and the ¹³C signals at δ 79.5, 50.0 (³ J ; CH), 36.8 (² J ; quaternary carbon), 27.1 (³ J ; CH₃) and 15.7 (³ J ; CH₃) led to their assignments as C-3, C-5, C-4, C-23 and C-24, respectively. The oxymethine proton at δ 4.51 showed long-range correlation with the ¹³C signal at δ 170.0 and with the C-24 (δ 15.7) and C-23 (δ 27.1) signals, confirming this proton signal to H-3 and allowing the assignment of the signal at δ 170.0 to C-3 acetoxy group. Moreover, the existence of correlations between H₃-28 (δ 0.91) and the ¹³C signals at δ 69.9 (CH), 57.8 (CH), 36.3 (quaternary carbon) and 34.4 (CH₂) led to their assignments as C-16, C-18, C-17 and C-22, respectively. Thus, the

Table 4. ¹³C NMR spectrum data for compounds **4**, **4a**, **6a**, **8a**, and the model compounds **6**, **8**, **11**, **12** and **13**^{10, 26-29}

C	4	11	12	4a ^b	13	6	6a	8	8a
1	40.9	38.7	33.5	35.9	37.0	33.3	32.6	33.1	32.4
2	27.4	27.2	25.4	23.1	24.3	25.2	22.7	25.2	22.7
3	78.7	78.9	76.0	79.5	80.6	76.1	78.1	76.1	78.1
4	39.3	38.7	37.5	36.8	38.6	37.4	36.3	37.3	36.3
5	55.4	55.2	48.8	50.0	51.2	48.9	49.2	48.9	49.2
6	18.3	18.3	18.2	17.3	18.2	18.3	18.1	18.3	18.1
7	33.7	32.8	35.2	30.8	32.0	32.8	32.3	32.6	31.9
8	43.6	40.0	43.5	42.2	40.7	40.2	40.2	40.1	40.0
9	55.3	46.1	55.8	153.1	154.2	46.8	46.6	46.6	46.6
10	38.1	36.8	38.2	37.5	37.9	36.9	36.8	36.9	36.8
11	68.2 ^a	23.3	68.4 ^a	114.6	115.5	23.3	23.2	23.5	23.4
12	129.2	125.1	128.7	122.2	123.0	125.2	125.3	122.4	122.6
13	141.3	137.9	142.9	137.7	141.4	137.9	137.5	143.5	142.9
14	44.3	44.0	42.2	41.8	43.1	44.2	43.9	43.9	43.7
15	36.0	35.9	27.9	30.7	28.2	35.9	33.8	35.6	33.6
16	66.7	67.0	27.7	69.9	26.1	67.0	70.8	66.1	69.8
17	38.5	38.5	33.6	36.3	33.7	38.6	37.6	37.4	36.5
18	59.9	60.7	58.1	57.8	57.3	60.8	60.8	49.1	49.9
19	39.5 ^c	39.5	39.4	38.2 ^c	39.0	39.6	39.5	46.6	46.5
20	39.1 ^c	39.4	39.3	37.5 ^c	39.5	39.5	39.4	30.9	30.8
21	30.4	30.5	31.1	29.5	31.2	30.5	30.5	34.2	34.3
22	35.2	35.2	41.3	34.4	41.4	35.2	35.4	30.6	30.8
23	28.4	28.1	28.7	27.1	28.2	28.3	27.8	28.3	27.8
24	15.7	15.6	22.4	15.7	17.4	22.3	22.6	22.3	22.6
25	17.0	15.7	16.6	17.0	17.6	15.5	15.5	15.3	15.3
26	18.0	16.8	18.0	21.7	22.2	16.9	16.8	16.9	16.8
27	24.3	24.5	23.3	24.4	25.5	24.7	24.4	27.3	27.0
28	21.9	21.9	28.6	22.2	28.7	21.9	21.4	21.5	21.9
29	17.8	17.6	17.5	16.5	16.8	17.6	17.6	33.3	33.0
30	21.5	21.3	21.3	21.2	21.5	21.3	21.3	23.9	23.8
MeCO				169.8	171.0		170.9		170.8
MeCO				170.0					
MeCO				20.3	21.3		21.2		21.2
MeCO				20.3					

Assignments based on HSQC, HMBC and G-NOESY for **4**, HMBC for **4a** and DEPT for **6a** and **8a**. In **4a**: C-3 acetoxy group δ 170.0; C-16 acetoxy group δ 169.8. The use of compound **13** as a model permitted to find out that Mahato and Kundu¹⁰ published the ¹³C NMR data for 3 β -hydroxyurs-9(11),12-diene, however, they were for 3 β -acetoxys-9(11),12-diene (see reference 26); ^aThe structural assignment was also supported by comparison of the ¹³C NMR spectrum with those of olean derivatives, 11 α -methoxyolean-12-ene and 11 β -hydroxyolean-12-ene derivatives.^{27, 28}; ^bDehydration of 11-hydroxyolean-12-ene and 11-hydroxyurs-12-ene see reference 29; ^cAssignment interchangeable.

proton signal at δ 5.46 (δ_c 69.9) was confirmed to H-16 and the ^{13}C signals at δ 169.8 and 22.2 were attributed to C-16 acetoxy group and C-28, respectively, due their correlations with the H-16 signal. The ^1H NMR also showed an olefinic proton to be coupled to H-12 (δ 5.49, d, J 5.6 Hz; 5.61 d, J 5.6 Hz), thus indicating the second double bond between C-9 and C-11. The olefinic proton signal at δ 5.49 was attributed to H-12 by its correlation with the C-18 (δ 57.8) signal. H-12 signal also showed cross peaks with the ^{13}C signal at δ 41.8 (quaternary carbon), which was attributed to C-14. In the same way, the olefinic proton signal at δ 5.61 to H-11 emerged from the correlations between the H_3 -25 and H-11 signals with the ^{13}C signals at δ 37.5, assigned to C-10. H-11 signal also showed cross peaks with the ^{13}C signal at δ 42.2, which was attributed to C-8. Moreover, H_3 -25 showed cross peaks with the C-5 signal (δ 50.0) and the ^{13}C signals at δ 153.1 and 35.9, leading to their assignments as C-9 and C-1, respectively. The signal for C-13 was established as 137.7 by the existence of a correlation between the H_3 -27 signal (δ 0.99) and this ^{13}C signal. H_3 -27 signal also showed cross peaks with the ^{13}C signals at δ 30.7, which was attributed to C-15. The relationship of ^1H signal at δ 1.69 (d, J 11.1 Hz) to the C-13, C-14, C-17 signals and the ^{13}C signal at δ 122.2 led to their assignments as H-18 and C-12. Thus, the second olefinic carbon at δ 114.6 was attributed to C-11. The ursa-9(11),12-diene system was also supported by the ^{13}C NMR spectrum which agreed closely with published data for 3β -acetoxyurs-9(11),12-diene (**13**).¹⁰ In the HMBC experiments several other long-range correlations were observed, which also confirmed the attribution of all the ^{13}C signals of the molecule (Table 3 and 4).

Compounds $3\alpha,16\beta$ -dihydroxyurs-12-ene (**6**) and $3\alpha,16\beta$ -dihydroxyolean-12-ene (**8**) have previously been isolated from *Canarium album*.⁷ In the present investigation the physical separation of these compounds was not achieved, even after acetylation with anhydride in pyridine, but the ^{13}C NMR data (Table 4) left no doubt that they were $3\alpha,16\beta$ -diacetoxy derivatives **6a** and **8a**. The presence of urs- and olean-12-ene systems were indicated by the characteristic olefinic ^{13}C resonances at δ 125.3 and 137.5, 122.6 and 142.9, respectively. The latter ^{13}C signals were more intense than the former, indicating the olean-12-ene derivative as the major compound. In the same way, the intense olefinic ^1H signal at δ 5.26 (t, J 3.6 Hz) was then assigned to H-12 of olean-12-ene, whereas δ 5.20 (t, J 3.6 Hz) to H-12 of urs-12-ene derivative. The ^1H and ^{13}C NMR spectra revealed the presence of two acetoxy groups. The identification of the A-ring containing only 3α -acetoxy group as substituent was supported by the coupling constants for H-3 (δ 4.63, t, J 2.7 Hz, **8a**; 4.50, t,

J 2.7 Hz, **6a**) and ^{13}C NMR spectrum, which showed that the acetylation accentuated the α -effect (C-3, δ 78.1) and diminished the β -effect (C-2, δ 22.7; C-4, δ 36.3) when compared with published data for C-2 to C-4 in **6** and **8**. Inspection of the ^{13}C NMR data of various hydroxy urs- and olean-12-enes, revealed that introduction of a hydroxyl group only on D- and E-ring causes a significant alteration of the chemical shift of C-18 (δ 58.9, **5**; 60.8, **6**; 47.4, **7**; 49.1, **8**).¹⁰ Although, in urs- and olean-12-enes containing 21β -, 21α -, 22α - and 22β -hydroxyl groups, the C-21 and C-22 resonate at δ 70.0 to 76.7.¹⁰ In olean-12-enes containing 15α - or 16β -hydroxyl groups, these carbonyl carbons resonate at δ 65.9 to 68.2.¹⁰ The coupling constants of ^1H signals of an oxymethine for **6a** and **8a** (δ 5.46, dd, J 11.9 and 5.0 Hz, **8a**; 5.48, dd, J 11.9 and 5.0 Hz, **6a**) indicated that the acetoxy group would be attached equatorial to C-15 (α) or C-16 (β), thus the above compounds are excellent models. The presence of a hydroxyl group at C-15 α has a pronounced effect on the olefinic carbon resonances. The chemical shift of C-12 and C-13 in $3\beta,15\alpha$ -dihydroxyolean-12-ene appear at δ 123.0 and 146.1, respectively. The corresponding resonances for $3\beta,16\beta$ -dihydroxyolean-12-ene appear at δ 122.2 and 143.4, i.e. C-13 is shielded by $\Delta\delta$ 2.7.¹⁰ The shielded resonance observed for **8a** (δ 122.6 and 142.9) is typical of 16β -hydroxyolean-12-ene, determining the position of the second acetoxy group at C-16 β . Although, no urs-12-ene containing a 15α -hydroxyl group appears to have been isolated so far. However, for the analogous structural situation in 3β -hydroxyolean-12-en-27,28-dioic (C-13, δ 138.1) and 3β -hydroxyurs-en-27,28-dioic (C-13, δ 134.2) acids a similar shield is observed for C-13 in both systems [C-13, δ 145.1 (**7**) – 138.1 = $\Delta\delta$ 7.0; δ 139.3 (**5**) – 134.2 = $\Delta\delta$ 5.1].¹⁰ Placement of the acetoxy group at C-16 received further support from acetylation of **4**, which yielded **4a**. In the ^{13}C NMR spectrum of **4a** (Table 4) the signal for C-16 was observed at δ 69.9, in close agreement with the resonance for the corresponding carbon in **6a** (δ 70.8) and **8a** (δ 69.8). All the above data suggested that **6a** was $3\alpha,16\beta$ -diacetoxyurs-12-ene and **8a** was $3\alpha,16\beta$ -diacetoxyolean-12-ene.

The ^1H NMR spectra of concentrated MeOH extracts of stem bark of *T. burserifolia*, branch and resin of *T. rhoifolia*, branch and resin of *Dacryodes hopkinsii* showed signals corresponding to triterpenes isolated from resin of *T. burserifolia*. Thus, only CH_2Cl_2 -soluble fractions of these extracts were further examined by their ^{13}C NMR spectra which showed that the major triterpenes were in stem bark of *T. burserifolia* α -amyirin (**5**) and β -amyirin (**7**); in branch of *T. rhoifolia* α -amyirin (**5**), β -amyirin (**7**), 3-*epi*- α -amyirin, 3-*epi*- β -amyirin, lupenone¹¹ and sitosterol; in

resin of *T. rhoifolia* α -amyirin (**5**), β -amyirin (**7**), 3-*epi*- α -amyirin, 3-*epi*- β -amyirin, 3 α -hydroxytirucall-8,24-dien-21-oic acid⁸ and 3 α -hydroxytirucall-7,24-dien-21-oic acid;⁶ in branch of *D. hopkinsii* α -amyirin (**5**), β -amyirin (**7**), lupeol,¹⁰ tirucallol,¹² sitosterol and stigmasterol; and in resin of *D. hopkinsii* α -amyirin (**5**) and β -amyirin (**7**).

Chemosystematic implications

Burseraceous genera are characterised by the production of tetracyclic tirucallane, dammarane, cycloartane, lanostane, and pentacyclic lupane, ursane and oleanane triterpenes.¹³ Ursanes and/or oleananes are omnipresent. All genera produce volatile oils which are often represented by many different sesquiterpenes, aromadendranes, humulanes, germacrane, eudesmanes, elemanes, guaianes, pseudo-guaianes, bourbonanes, caryophyllanes, cubebanes, cadinanes, copaanes and bisabolanes.¹³ Furosesquiterpenes have been isolated from *Commiphora* species.¹³ Lignans are less common and were for long known only from *Bursera*, but have now been recorded in *Commiphora* and *Protium*.^{14, 15}

Until recently the phytochemical knowledge of Protieae were the records of three dammaranes (dammaradienol in *Garuga pinnata*,¹⁶ 3 α ,20(S)-dihydroxydammar-24-ene in *Crepidosperrum rhoifolium*,⁵ and cabraleadiol in *Protium apiculatum*⁵), one tirucallane (butirospermol in *G. pinnata*¹⁶), two ursanes (α -amirin and 3-*epi*- α -amirin in *G. pinnata*,¹⁶ and only the former in *P. paniculatum* Engl¹⁷ and *P. icariba*⁵), one oleanane (β -amirin in *P. paniculatum*¹⁷ and *P. icariba*⁵), one cycloartane (3 β ,24-dihydroxycycloart-25-ene in *C. rhoifolium*⁵), one multiflorane (secoisobryononic acid in *Tetragastris altissima*⁵), one friedelane (friedelin in *T. altissima*⁵), one taraxerane (taraxerol in *T. altissima*⁵), one lupane (lupeol in *P. icariba*⁵ and *P. apiculatum*⁵), four sterols (sitosterol, stigmasterol, campesterol and 3 β -O- β -D-glucopyranosylsitosterol in *C. rhoifolium*⁵, only the two first in *P. paniculatum*;¹⁷ and only the former in *P. opacum* Swart,¹⁸ *P. apiculatum*⁵ and *T. altissima*⁵), four lignans [(+)-(2S,3S)-2-(3',4'-methylenedioxy-acetophenone)-butyrolactone, (-)-cubebin epimers in *P. tenuifolium* Engl,¹⁵ parabenzolactone in *C. rhoifolium*⁵ and (-)-savinin in *T. altissima*⁵], one coumarin (propacin in *P. opacum*¹⁸), one biflavonoid (amentoflavone in *G. pinnata*¹⁹) and two macrocyclic biphenyl ether (garuganin I and III in *G. pinnata*¹⁶). Protieae genera have been shown to have a chemical profile that is comparable to those of all other burseraceous tribes.^{5, 13}

All the burseraceous tribes yielded tirucallanes, ursanes, oleananes and lupanes, while dammaranes appear to have been recorded only from Protieae and Boswellieae

genera.⁵ *Trattinnickia* has in common tirucallane, ursane and oleanane types with *Dacryodes*. Lupanes have been found only in *Dacryodes*. Furthermore, *Dacryodes* contains peculiar 3,4-*secolupanes*¹³ which could be taken as indicative of an affinity to the Boswellieae where similar lupanes occur.¹³ It is also chemical evidence favouring its classification in the Canarieae, notably by the co-occurrence of 3,4-*secolupanes* in *Canarium muelleri* and *C. zeylanicum*.¹³

The co-occurrence of dammaranes in *Trattinnickia*, *Garuga*,¹⁷ *Crepidosperrum*,⁵ *Protium*⁵ and *Commiphora*.^{20, 21} suggests some affinity between Protieae and Boswellieae (Bursereae). The simplest dammaranes, in which the side chain is undegraded (as in dammarendiol-II and 3 α ,20(S)-dihydroxydammar-24-ene from *T. burserifolia*), are typical of the Protieae and that increasing ability to lose the entire C-17 side chain (as in mansumbinanes²¹) occurs through the Boswellieae (in *Commiphora*). This can be seen as a further advance in oxidative mechanisms and appears to agree closely with the suggested phylogenetic sequence within the Burseraceae; Protieae considered the most primitive, Boswellieae intermediate. Thus, the isolation of two undegraded dammaranes from *T. burserifolia* suggest that *Trattinnickia* appears to have a less pronounced relationship to the Canarieae than to the Protieae, since they do not occur, at present, in the former. Thus, the presence of dammaranes in *T. burserifolia* does not support Daly's taxonomic conclusions.⁵

The most common tetracyclic triterpenes in the Rutales families are tirucall-7-en derivatives which are the precursors of the limonoids and quassinoids that are major chemotaxonomic characters of the order.^{22, 23} The Burseraceae is an exception, no quassinoids or limonoids have so far been isolated. Burseraceous genera have heretofore yielded relatively few variety of compounds. This fact can be rationalised by the presence in their species of massive quantities of tannins (possibly also essential oils), general defences which make the presence of specific alleochemics superfluous.²⁴ Essential oils represented by many different sesquiterpenes, appear to inhibit the formation of squalene, potential precursor of limonoids.²⁵

Experimental

General

NMR on a Bruker DRX 400, with TMS as internal standard; ESI-MSMS: low resolution on a triple quadrupole Micromass Quattro LC instrument, equipped with a "Z-spray" ion source; GC-MS: Shimadzu GC-17A gas chromatograph fitted with a fused silica DB-5 (30 m x

0.25 mm ID, 0.25 μm film thickness) capillary column with helium as the carrier gas at a flow rate of 1.6 mL min^{-1} . The temperature was programmed initially at 60 $^{\circ}\text{C}$ for 2 min, then increased with a rate of 3 $^{\circ}\text{C min}^{-1}$ to 240 $^{\circ}\text{C}$. The injection was split and its temperature was 225 $^{\circ}\text{C}$. The interface temperature was 250 $^{\circ}\text{C}$. The chromatograph was coupled to a Shimadzu QP5000 mass selective detector at 70 eV; IR (BOMEN - Ft/IR). $[\alpha]_D$: Perkin Elmer 241 instrument; IR (KBr, BOMEN - Ft/IR); R-HPLC: Recycling High-Performance Liquid Chromatography on a model Shimadzu LC-6AD; the column used was a Shim-pack Prep-Sil (H), 250 mm X 20 mm, 5 mm particle size, 100 \AA pore diameter; eluant: CHCl_3 ; flow rate: 8.0 mL min^{-1} and 5.0 mL min^{-1} ; detection (Shimadzu SPD-6AV): UV λ 254 nm.

Plant material

Trattinnickia burserifolia, *T. rhoifolia* and *Dacryodes hopkinsii* were collected from Forest Reserve Adolpho Ducke, Amazonas, Brazil; vouchers (184.962, 178.219, 178.240, respectively) were deposited in the Herbarium of Instituto Nacional de Pesquisa da Amazônia (INPA), Manaus, AM.

Extraction and isolation from resin of *T. burserifolia*

The resin was dissolved in CHCl_3 , filtered and concentrated under vacuum. The concentrated (60 g) was partitioned into CHCl_3 , MeOH and H_2O soluble fractions. The concentrated CHCl_3 -soluble fraction was subjected to column chromatography over silica gel. Elution with a hexane- CH_2Cl_2 -MeOH gradient afforded 6 fractions (3 hexane-fractions, 1 hexane/ CH_2Cl_2 fraction, 1 CH_2Cl_2 fraction and 1 MeOH fraction). The hexane fractions were combined in 2 groups on the basis of analytical TLC. The hexane-fraction 2-3 gave a mixture (12.6 g) of α -amyrin (**5**) and β -amyrin (**7**). The hexane-fraction 1 was subjected to column chromatography over silica gel eluting with a hexane-EtOAc gradient to afford a mixture of **5** and **7** (3 g) and fractions A, B and C. Fraction A was twice flash chromatographed on silica gel, eluting with hexane-EtOAc gradient and finally with hexane-EtOAc (98:2) affording impure **1**. Compound **1** was purified by R-HPLC (CHCl_3 ; detection UV λ 254 nm, flow rate: 8.0 mL min^{-1} ; see above) affording pure **1** at 1 first peak (2.4 mg). Fraction B was flash chromatographed on silica gel, eluting with hexane-EtOAc gradient yielding 32 fractions. These fractions were combined in 5 groups on the basis of analytical TLC. The 5 groups were monitored by ^1H NMR (200 MHz) and were examined only those which showed features of taxonomic

interest. Group 2 (fractions B17-24) was twice flash chromatographed on silica gel, eluting with benzene- CH_2Cl_2 (9:1) and finally with hexane-EtOAc 95:5 affording a mixture (5 mg) of **2** and **3**. Group 3 (fractions B25-27) was flash rechromatographed on silica gel eluting with hexane-EtOAc 95:5 yielding an amorphous solid which was purified by preparative TLC (silica gel; hexane- CH_2Cl_2 -THF, 10:1.0:0.25) to yield a mixture (7.9 mg) of α - and β -amyrenone. Group 4 (fraction B28) was purified three times by preparative TLC (silica gel; benzene) to yield 3-*epi*- α -amyrin (8.4 mg). Group 5 (fractions B29-32) was flash rechromatographed on silica gel eluting with hexane-EtOAc 9:1 yielding 3-*epi*- β -amyrin (13 mg) after crystallization in hexane. Fraction C yielded a precipitate (458 mg) from which 80 mg were methylated with CH_2N_2 yielding the corresponding methyl ester of 3 α -hydroxytirucall-8,24-dien-21-oic acid.

Hexane/ CH_2Cl_2 fraction from the concentrated CHCl_3 -soluble fraction of resin was subjected to column chromatography over silica gel eluting with a hexane-EtOAc gradient to afford 42 fractions. These fractions were combined in 7 groups on the basis of analytical TLC. The 7 groups were monitored by ^1H NMR (200 MHz) and were examined only those which showed features of taxonomic interest. Group 3 (fractions G3D17-19) yielded a precipitate which was dissolved in Me_2CO and kept in the refrigerator overnight. The residue was flash rechromatographed on silica gel eluting with hexane- CH_2Cl_2 -MeOH 20:5:1 yielding a mixture of 3 α -hydroxytirucall-8,24-dien-21-oic acid and 3 α -hydroxytirucall-7,24-dien-21-oic acid. The filtrate was evaporated and the residue was rechromatographed as above affording a mixture (160 mg) of α -amyrin (**5**) and β -amyrin (**7**) and 3 α -hydroxytirucall-8,24-dien-21-oic acid (6.7 mg). Group 5 (fractions G5E28-37) was twice flash chromatographed on silica gel, eluting with hexane- CH_2Cl_2 -MeOH 10:1:1 and finally with hexane-EtOAc (8:2) affording fraction G5E-X and 3-oxotirucall-8,24-dien-21-oic acid (16 mg) after crystallisation in hexane- Me_2CO . Fraction G5E-X was acetylated with Ac_2O -pyridine to give acetate derivatives which were subsequently purified by flash chromatography eluting with hexane- Me_2CO 8:2 affording fraction G5E-Xa and fraction G5E-Xb. Fraction G5E-Xa was flash rechromatographed on silica gel eluting with benzene-EtOAc 95:5 to yield a mixture (102 mg) of **6a** and **8a**. Fraction G5E-Xb was rechromatographed as above to afford a mixture (4.4 mg) of C3-epimers 3-acetoxylammarenediol-II and 3 α -acetoxo-20(S)-hydroxydammar-24-ene.

The MeOH fraction from the concentrated CHCl_3 -soluble fraction of resin was subjected to column chromatography over silica gel eluting with a hexane-EtOAc-MeOH gradient to afford 15 fractions. These

fractions were combined in 7 groups on the basis of analytical TLC. The 7 groups were monitored by ^1H NMR (200 MHz) and were examined only those which showed features of taxonomic interest. Group 4 (fractions G4F5-8) was three times flash rechromatographed on silica gel eluting with CH_2Cl_2 -EtOAc 7:3, then CH_2Cl_2 -EtOAc 7:3 and finally CH_2Cl_2 -MeOH 95:5, affording **4** (2.6 mg). Compound **4** was allowed to react overnight with an excess of Ac_2O in pyridine. Work-up as usual yielded $3\beta,16\beta$ -diacetoxyurs-9(11),12-diene (**4a**).

Extractions from stem bark of Trattinnickia burserifolia, branch of T. rhoifolia, resin of T. rhoifolia, branch of Dacryodes hopkinsii and resin of D. hopkinsii

These organs ground were extracted with MeOH. The ^1H NMR spectra of concentrated MeOH extracts showed signals corresponding to triterpenes isolated from resin above. They were partitioned into hexane, CH_2Cl_2 and MeOH soluble fractions. The ^1H NMR spectra of these fractions showed that the major triterpenes were in concentrated CH_2Cl_2 -soluble fraction. The ^{13}C NMR spectra of these fractions showed that the major triterpenes were in: a) stem bark of *T. burserifolia*, α -amyrin (**5**) and β -amyrin (**7**); b) branch of *T. rhoifolia*, α -amyrin (**5**), β -amyrin (**7**), 3-*epi*- α -amyrin, 3-*epi*- β -amyrin, lupenone and sitosterol; c) resin of *T. rhoifolia*, α -amyrin (**5**), β -amyrin (**7**), 3-*epi*- α -amyrin, 3-*epi*- β -amyrin, 3 α -hydroxytirucall-8,24-dien-21-oic acid and 3 α -hydroxytirucall-7,24-dien-21-oic acid; d) branch of *Dacryodes hopkinsii*, α -amyrin (**5**), β -amyrin (**7**), lupeol, tirucallol, sitosterol and stigmasterol; e) resin of *D. hopkinsii*, α -amyrin (**5**) and β -amyrin (**7**).

2(S)-Phenylacetoxy-4(R*)-p-mentha-1(7),5-dien (1)*

Amorphous solid; $[\alpha]_{\text{D}}^{26} + 2.5^\circ$ (CHCl_3 ; c 0.0024); IR $\nu_{\text{max}}/\text{cm}^{-1}$: 2956, 2923, 2851 (aliphatic CH), 1733 (ester), 1457 (aromatic C=C), (liq. film); ^1H NMR (400 MHz, CDCl_3): see Table 1; ^{13}C NMR (100 MHz, CDCl_3 , multiplicities assigned from DEPT 135 experiment): see Table 1; HMBC and HSQC (400/100 MHz, CDCl_3): see discussion. GC-MS: R_f 22.27, EIMS: m/z (rel. int.): **1** failed to give an $[\text{M}]^{+}$, the main fragments observed being 150 for $[\text{M} - \text{C}_6\text{H}_5\text{CH}_2\text{CO} - \text{H}]^{+}$ (10), 149 for $[\text{150} - \text{H}]^{+}$ (100), 119 for $[\text{C}_6\text{H}_5\text{CH}_2\text{CO}]^{+}$ (10), 91 for $[\text{C}_6\text{H}_5\text{CH}_2]^{+}$ (20).

Mixture of 3 β -phenylacetoxyurs-12-en (2) and 3 β -phenylacetoxyolean-12-en (3)

Amorphous solid; IR $\nu_{\text{max}}/\text{cm}^{-1}$: 2937, 2857 (aliphatic CH), 1733 (ester), 1455 (aromatic C=C), (liq. film); ^1H NMR

(400 MHz, CDCl_3): δ 5.16 (1H, t, J 3.6 Hz, H-12, **3**); 5.11 (1H, t, J 3.6 Hz, H-12, **2**), 4.48 (1H, dd, J 11.0, 5.1 Hz, H-3); 3.60 (2H, s, H-7'); 1.11-0.76 (16 Me, s). ^{13}C NMR (100 MHz, CDCl_3 , multiplicities assigned from DEPT 135 experiment): see Table 2. GC-MS: R_f : 54.6 min. and 54.13 min. EIMS: m/z (rel. int.): 544 $[\text{M}]^{+}$ (5), 218 (100): associated with retro-Diels-Alder cleavage of C-ring, 119 for $[\text{C}_6\text{H}_5\text{CH}_2\text{CO}]^{+}$ (20), 91 for $[\text{C}_6\text{H}_5\text{CH}_2]^{+}$ (60).

3 $\beta,16\beta,11\alpha$ -Trihydroxyurs-12-en (4)

Amorphous solid; $[\alpha]_{\text{D}}^{24} + 13.5^\circ$ (CHCl_3 ; c 0.002); IR $\nu_{\text{max}}/\text{cm}^{-1}$: 3424 (OH), 2928, 2867 (aliphatic CH), 1710 (ester), (liq. film); ^1H NMR (400 MHz, CDCl_3 ; resonances were confirmed by HSQC and HMBC experiments): δ 5.24 (1H, d, J 3.2 Hz, H-12), 4.27 (1H, dd, J 8.7, 3.2 Hz, H-11), 4.23 (1H, dd, J 11.1, 5.2 Hz, H-16), 3.23 (1H, dd, J 10.4, 5.8 Hz, H-3), 2.20 (2H, dt, J 13.5, 3.4 Hz, H_2 -1), 2.03 (2H, dt, J 13.9, 3.1 Hz, H_2 -7), 1.65 (2H, m, H_2 -2), 1.62 (2H, m, H_2 -22), 1.59 (1H, m, H-6a), 1.57 (1H, d, J 11.3, H-18), 1.48 (1H, d, J 8.7, H-9), 1.42 (2H, m, H_2 -21), 1.38 (1H, m, H-6b), 1.36 (2H, m, H_2 -15), 1.20 (3H, s, Me-27), 1.06 (3H, s, Me-25), 1.05 (3H, s, Me-26), 0.97 (3H, s, Me-23), 0.91 (3H, d, J 5.9 Hz, Me-30), 0.82 (3H, d, J 6.3 Hz, Me-29), 0.77 (3H, s, Me-24), 0.73 (3H, s, Me-28), 0.72 (1H, m, H-5). ^{13}C NMR (100 MHz, CDCl_3 , multiplicities assigned from DEPT 135 experiment): see Table 4. ESI-MSMS (probe) 25 eV, m/z (rel. int.): 457 $[\text{M} - \text{H}]^{-}$ (100), 439 $[\text{M} - \text{H} - \text{H}_2\text{O}]^{-}$ (40), 421 $[\text{M} - \text{H} - \text{H}_2\text{O} - \text{H}_2\text{O}]^{-}$ (10), 247 (30).

3 $\beta,16\beta$ -diacetoxyurs-9(11),12-diene (4a)

Amorphous solid; ^1H NMR (400 MHz, CDCl_3 ; resonances were confirmed by HMBC experiments): δ 5.61 (1H, d, J 5.6 Hz, H-11), 5.49 (1H, d, J 5.6 Hz, H-12), 5.46 (1H, dd, J 11.4 and 5.5 Hz, H-16), 4.51 (1H, dd, J 11.4 and 4.9 Hz, H-3), 2.06 (3H, s, MeCO), 2.05 (3H, s, MeCO), 1.69 (1H, d, J 11.1 Hz, H-18), 1.23 (3H, s, Me-25), 1.20 (3H, s, Me-26), 0.99 (3H, s, Me-27), 0.95 (3H, d, J 4.6 Hz, Me-30), 0.91 (3H, s, Me-28), 0.90 (3H, s, Me-23), 0.88 (3H, s, Me-24), 0.81 (3H, d, J 6.5 Hz, Me-29). ^{13}C NMR (100 MHz, CDCl_3): see Table 4.

Mixture of compounds 3 $\alpha,16\beta$ -diacetoxyurs-12-ene (6a) and 3 $\alpha,16\beta$ -diacetoxyolean-12-ene (8a)

^1H NMR (400 MHz, CDCl_3): δ 5.20 (1H, t, J 3.6 Hz, H-12, **6a**), 5.26 (1H, t, J 3.6 Hz, H-12, **8a**), 5.46 (1H, dd, J 11.9 and 5.0, H-16, **8a**), 5.48 (1H, dd, J 11.9 and 5.0, H-16, **6a**), 4.63 (1H, t, J 2.7 Hz, H-3, **8a**), 4.50 (1H, t, J 2.7 Hz, H-3, **6a**), 1.30-0.83 (16 Me). ^{13}C NMR (100 MHz, CDCl_3): see Table 4.

Acknowledgements

The authors thank Conselho Nacional de Desenvolvimento Científico e Tecnológico (CNPq), Fundação de Amparo à Pesquisa do Estado de São Paulo (FAPESP), Coordenação de Aperfeiçoamento de Pessoal de Ensino Superior (CAPES), Financiadora de Estudos e Projetos (FINEP) for financial support.

References

1. Engler, A. von In *Die Natürlichen Pflanzenfamilien*; Engler, A.; von ; Prantl, K., eds., 2nd ed., Engelmann: Leipzig, 1931, vol. 10, p.187.
2. Lam, H. J.; *Bull. Jard. Bot. Buitenzorg III* **1932**, *12*, 281.
3. Swart, J. J.; *A Monograph of the Genus Protium and some Allied Genera (Burseraceae)*; Dukkerij Koch en Knuttel: Gouda, 1942.
4. Daly, D. C.; *Britonia* **1989**, *41*, 17.
5. Lima, M. da P.; Castro, F. B. G. de; Ferreira, A. G.; Rodrigues Fo, E.; Silva, M. F. das G. F. da; Fernandes, J. B.; Vieira, P. C.; *Rev. Latinoamer. Quim.* **2001**, *29*, 135.
6. Guang, L.; Gray, A. I.; Waterman, P. G.; *Phytochemistry* **1988**, *27*, 2283.
7. Tamai, M.; Watanabe, N.; Someya, M.; Kondoh, H.; Omura, S.; Ling, Z. P.; Chang, R.; Ming, C. W.; *Planta Med.* **1989**, *55*, 44.
8. Sawadogo, M.; Tessier, A. M. V.; Delaveu, P.; *Ann. Pharm. Fr.* **1985**, *43*, 89.
9. Matusch, R.; Schmidt, G.; *Helv. Chim. Acta* **1989**, *72*, 51.
10. Mahato, S. B.; Kundu, A. P.; *Phytochemistry* **1994**, *37*, 1517.
11. Wenkert, E.; Baddeley, G. V.; Burfitt, I. R.; Moreno L. N.; *Org. Magn. Reson.* **1978**, *11*, 337.
12. Polonsky, J.; Varon, Z.; Rabanal, R. M.; Jacquemin, H.; *Isr. J. Chem.* **1977**, *16*, 16.
13. Khalid, S. A. In *Chemistry and Chemical Taxonomy of the Rutales*; Waterman, P. G.; Grundon, M. F.; eds., Academic Press: New York, 1983, p. 281.
14. O'Sullivan, J. In *Chemistry and Chemical Taxonomy of the Rutales*; Waterman, P. G.; Grundon, M. F.; eds., Academic Press: New York, 1983, p. 267.
15. Siqueira, J. B. G.; Zoghbi, M. das G. B.; Cabral, J. A.; Wolter-Filho, W.; *J. Nat. Prod.* **1995**, *58*, 730.
16. Mishra, A. K.; Haribal, M. M.; Sabata, B. K.; *Phytochemistry* **1985**, *24*, 2463.
17. Zoghbi, M. das G. B.; Siqueira, J. B. G.; Wolter, E. L. A.; Júnior, O. L. P.; *Acta Amazonica* **1994**, *24*, 59.
18. Zoghbi, M. das G. B.; Roque, N. F.; Gottlieb, O. R.; *Phytochemistry* **1981**, *20*, 180.
19. Ansari, F. R.; Ansari, W. H.; Rahman, W.; *Indian J. Chem., Sect. B* **1978**, *16B*, 846.
20. Ampofo, S.; Waterman, P. G.; *Phytochemistry* **1985**, *24*, 2925.
21. Provan, G. J.; Waterman, P. G.; *Phytochemistry* **1986**, *25*, 917.
22. Silva, M.F. das G.F. da; Gottlieb, O.R.; Dreyer, D.L.; *Biochem. Syst. Ecol.* **1984**, *12*, 299.
23. Silva, M.F. das G.F. da; Gottlieb, O.R.; *Biochem. Syst. Ecol.* **1987**, *15*, 85.
24. Gottlieb, O. R.; *J. Ethnopharmacol.* **1982**, *6*, 227.
25. Kaplan, M. A. C.; Gottlieb, O. R.; *Interciencia* **1990**, *15*, 26.
26. Matsunaga, S.; Tanaka, R.; Akagi, M.; *Phytochemistry* **1988**, *27*, 535.
27. Mathiasa, L.; Vieira, I. J. C.; Braz-Filho, R.; Rodrigues-Filho, E.; *J. Braz. Chem. Soc.* **2000**, *11*, 195.
28. Barreiros, M. L.; David, J. M.; Pereira, P. A. P.; Guedes, M. L. S.; David, J. P.; *J. Braz. Chem. Soc.* **2002**, *13*, 669.
29. Barnes, R. A.; Pereira, A. L.; Scofield, C. V.; Braz-Filho, R.; Pinto, A. C. *Chem. Pharm. Bull.* **1984**, *32*, 3674.

Received: July 17, 2003

Published on the web: May 10, 2004

FAPESP helped in meeting the publication costs of this article.