



Antioxidant activity of different extracts of lotus (*Nelumbo nucifera* Gaertn) rhizome by gradient solvent extraction

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Abstract

In vitro antioxidant activity of extracts of lotus rhizome with different solvent by gradient extraction was determined. The extraction yield and total phenolic content of extracts were determined. 1,1-diphenyl-2-picrylhydrazyl hydrate (DPPH) radical scavenging assay and β -carotene-linoleic acid assay were conducted to assess the antioxidant activity of the extracts. Results showed that the yield of butanol extract was the highest. The order of total phenolic content in different extracts was as follows: water extract < petroleum ether extract < chloroform extract < butanol extract < ethyl acetate extract, the same with that of the DPPH scavenging activity. The total phenolic content was significantly correlated with DPPH scavenging activity ($R^2 = 0.9363$). All extracts, except water extract with too poor antioxidant activity, showed obvious antioxidant activity against β -carotene-linoleic acid, and the antioxidant activity of butanol extract was significantly lower than others. There was no significant correlation between the antioxidant activity against β -carotene-linoleic acid and the total phenolic content ($R^2 = 0.1995$).

Keywords: lotus rhizome; extracts; antioxidant activity; solvent.

Practical Application: This study has provided a basis for further exploring the antioxidant value of lotus rhizome.

1 Introduction

Lotus (*Nelumbo nucifera* Gaertn) is both an ornamental plant and a dietary staple in Eastern Asia, particularly in China (Hu & Skibsted, 2002). All parts of lotus are used for various medicinal purposes in oriental medicine (Kashiwada et al., 2005). The seed of lotus is used in folk remedies as diuretic, cooling agent, antiemetic and antidote in the treatment of tissue inflammation, cancer, skin disease, leprosy and poisoning (Chopra et al., 1956; Liu et al., 2004). The leaf of lotus is considered best for “over-coming body heat”, and stopping bleeding (Bensky et al., 2004). It is used as a drug for hematemesis, epistaxis, hemoptysis, hematuria and metrorrhagia (Ono et al., 2006) in traditional Chinese medicine. Lotus rhizome can be cooked into different dishes or eaten raw as a common vegetable. Especially, it has been applied in Chinese herbal prescriptions to alleviate tissue inflammation, cancer, and liver cirrhosis for a long time (Mukherjee et al., 1997).

The extracts of lotus possess a wide variety of activities. The extracts of lotus talk shows anti-pyretic effect (Sinha et al., 2000), while extracts of lotus leaf and stamen show antioxidant effect and strong radical scavenging activity (Wu et al., 2003; Cho et al., 2003). It is reported that the extract of lotus rhizome exhibits high antioxidative capacity (Hu & Skibsted, 2002). However, few researchers study the antioxidant activities of the extract of lotus rhizome with different solvent by gradient extraction. In the current study, active substance of lotus rhizome was extracted with increased polarity organic solvent by gradient extraction. The extraction yield and total phenolics content of different extract were determined. 1,1-diphenyl-2-picrylhydrazyl hydrate (DPPH) radical scavenging assay and β -carotene-linoleic acid assay were conducted to assess the antioxidant activity of the extract.

2 Materials and methods

2.1 Materials and reagents

Lotus rhizome (*Dong he zao ou*, named in China) was purchased from Yiwu (Zhejiang, China) and washed by water. DPPH, β -carotene, Folin-Ciocalteu (FC) reagent, gallic acid, linoleic acid and polyoxyethylene sorbitan monopalmitate (Tween 40) were obtained from Sigma-Aldrich (Saint Louis, USA). All other chemicals used were of analytical grade.

2.2 Extraction method

One kg of randomly mixed lotus rhizome of different sizes and parts was homogenated, then extracted with 60% ethanol aqueous solution. After filtered, ethanol was evaporated at 35 °C, then the water solution of extract was extracted with petroleum ether, chloroform, ethyl acetate and butanol in sequence. The solvent of each extract liquid was evaporated at 35 °C and the remaining water solution was freeze-dried. The dried extracts were then weighed. The extraction process was showed in Figure 1.

2.3 Determination of total phenolics content

Determination of total phenolics content was carried out using Folin-Ciocalteu (FC) method with gallic acid as standard according to Slinkard & Singleton (1997) with some modifications. 1.0 mL of FC reagent and proper amount of sample solution were added in a flask. The flask was shaken up to mix the solutions thoroughly. After 3 min, 3.0 mL of

Received 24 Jan., 2021

Accepted 23 Feb., 2021

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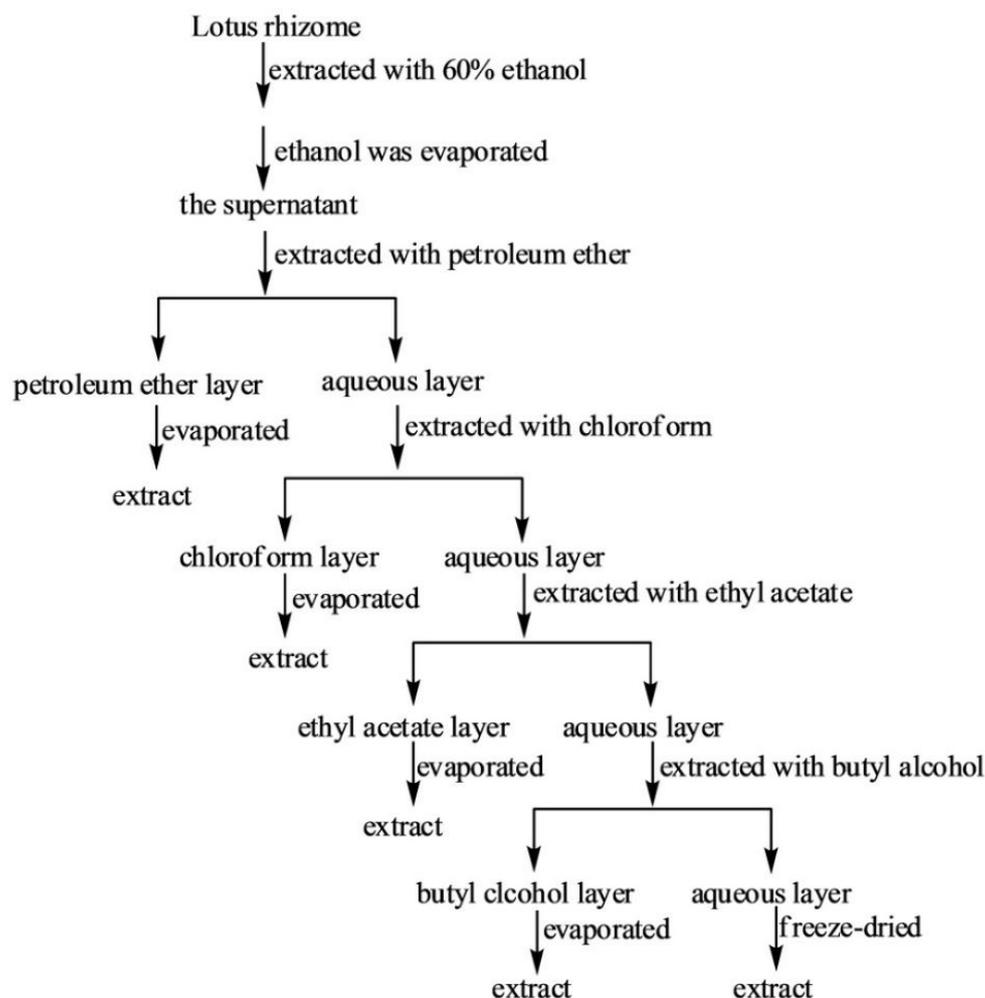


Figure 1. Extraction process of antioxidant constituents from lotus rhizome.

Na_2CO_3 solution (2%) was added. The solutions were mixed again and left at room temperature for 2 h. The absorbance was measured at 765 nm using a UV-Vis spectrophotometer. Total phenolics content of the extracts was calculated with a linear regression equation obtained from gallic acid standard graph: Absorbance = $8.3886 \times \text{gallic acid } (\mu\text{g}) - 0.0114$. Results were expressed as milligrams of gallic acid equivalents (GAE) per 100 g of extracts.

2.4 Determination of DPPH radical scavenging capacity

The scavenging activity on DPPH radicals was measured according to the method of Sánchez-Moreno et al. (1998) with some modifications. 0.2 mL of methanol solution of extracts was added to 2.8 mL of methanol solution of DPPH. The solutions were mixed thoroughly. After 3 min, absorbance of the solutions was measured at 517 nm using a UV-Vis spectrophotometer. Methanol was used as a control. The DPPH radical scavenging activity of each sample was calculated by the DPPH inhibition (absorbance decrease) according to the following equation: DPPH inhibition (%) = $100 \times (A_{\text{control}} - A_{\text{sample}}) / A_{\text{control}}$ (with A: absorbance).

2.5 Determination of antioxidant activity by β -carotene bleaching method

In β -carotene-linoleic acid assay, antioxidant capacity is determined by measuring the inhibition of the volatile organic compounds and the conjugated diene hydroperoxides arising from linoleic acid oxidation (Kartal et al., 2007). Total antioxidant activity of lotus rhizome extracts was measured according to the method of Moure et al. (2000). 2.0 mg of β -carotene were dissolved in 10 mL chloroform and 1.0 mL of the β -carotene solution was mixed with 20 mg of purified linoleic acid and 200 mg of Tween40 emulsifier in a round-bottom flask, chloroform was then removed in a rotary vacuum evaporator. After evaporation, the mixture was immediately diluted with 50 mL of distilled water and the mixture was stirred in a sonicator. 100 μL of methanol solution of extracts was added to 9.9 mL of the β -carotene/linoleic acid emulsion. Methanol was used as a control. Absorbance at 470 nm was immediately recorded after addition of sample. The vials were then gently capped and placed at 50 °C in a water bath. The absorbance was determined every 20 min until 120 min. Antioxidant activity coefficient (AAC) was measured in terms of successful bleaching of β -carotene by using a slightly modified version of the formula: $\text{AAC} = 1000 \times (A_{S(120)} - A_{C(120)}) / (A_{C(0)} - A_{C(120)})$, where $A_{C(0)}$ is the absorbance values measured at initial time of

the incubation for control at 0 min, while $A_{S(120)}$ and $A_{C(120)}$ are the absorbance values measured in the samples or control at 120 min, respectively.

2.6 Statistical analysis

All analyses were run in triplicates and results averaged. Statistical analyses were performed with the Excel and SPSS software package.

3 Results and discussion

3.1 Yield of lotus rhizome extracts

As shown in Figure 2, the yields of lotus rhizome extract from wet weight by gradient extraction with different solvents were varied from 0.027 to 0.229 g extracts/100 g lotus rhizome, ranging from low to high in the following order: ethyl acetate extract < chloroform extract < petroleum ether extract < water extract < butanol extract. The yield of butanol extracts was 8.48 times of that of ethyl acetate extract. This indicated that substances in lotus rhizome were quite polar and lots of polar compounds were mainly dissolved in water and alcohol.

3.2 Total phenolic content of lotus rhizome extracts

The total phenolic content of the extract, affected by the extracting solvents, was showed in Figure 3. The total phenolic content (mg GAE/100g extract) ranked from low to high in the following order: water extract (0.7) < petroleum ether extract (1.5) < chloroform extract (2.1) < butanol extract (4.1) < ethyl acetate extract (5.7), and there was a significant difference between any two of five extracts. This was in concordance with the research results of Wangenstein et al. (2004).

3.3 DPPH radical scavenging activity of lotus rhizome extracts

It is well accepted that the DPPH radical scavenging by antioxidant is attributable to their hydrogen donating activity (Krings & Berger, 2001). This test system can be used for

the primary characterization of the scavenging potential of compounds, and is thus used in the present work for evaluation of the effectiveness of extracts of lotus rhizome. In experiment, the proper concentration of extract of petroleum ether, chloroform, ethyl acetate, butanol, and BHA was 41-400, 20-300, 11-55, 15.43-66.86, and 12.5-100 $\mu\text{g}/3\text{ mL}$, respectively. The DPPH radical scavenging activity of water extract was too poor to consider in this study. Results were shown in Figure 4 and 5. The DPPH inhibition rate of all extracts and BHA increased when their concentrations increased. The order of DPPH inhibition rate of different extracts and BHA was as follows: petroleum ether extract < chloroform extract < butanol extract < BHA < ethyl acetate extract. IC₅₀, the half maximal inhibitory concentration, represents the concentration of an inhibitor that is required for 50% inhibition. The IC₅₀ value of ethyl acetate extract, BHA extract, butanol extract, chloroform extract and petroleum

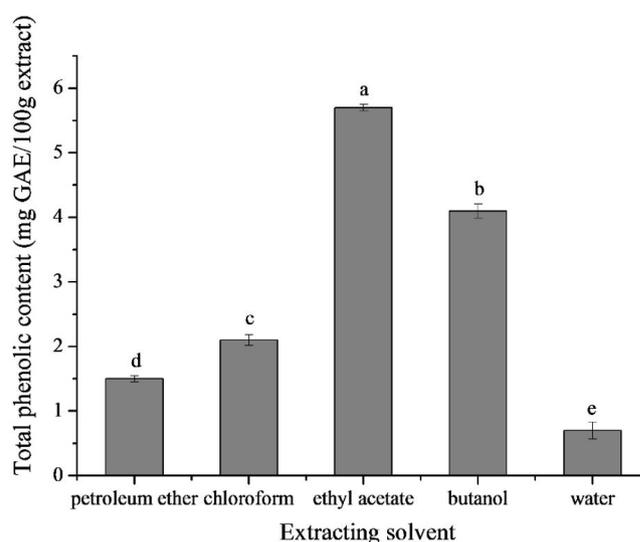


Figure 3. Total phenolic content of lotus rhizome extracts. Values in a column with different letters were significantly different ($P < 0.05$).

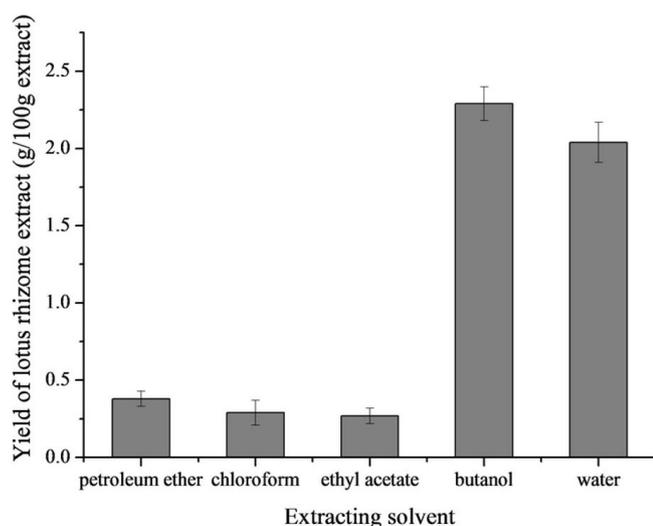


Figure 2. Yield of lotus rhizome extracts.

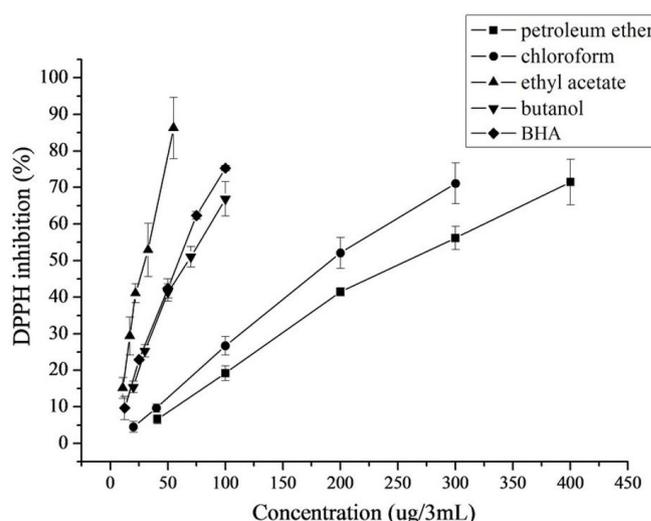


Figure 4. DPPH radical scavenging activity of lotus rhizome extracts.

extract were 0.41, 0.83, 0.93, 2.71, and 3.59 mg/mL, respectively. It was also found that the DPPH radical scavenging activity of different extracts was significantly correlated with the total phenolic content ($R^2 = 0.9363$). This indicated that the DPPH radical scavenging activity mainly attributed to the phenols in lotus rhizome.

3.4 Antioxidant activity determined by β -carotene bleaching method

The antioxidant activity of different extracts of lotus rhizome determined by β -carotene bleaching method was shown in Figure 6. All extracts, except water extract with too poor antioxidant activity, showed antioxidant activity against β -carotene-linoleic acid. The AAC values of the extracts were ranked in the order: butanol extract < ethyl acetate extract < chloroform extract < petroleum ether extract. The antioxidant activity of butanol extract was significantly lower than others.

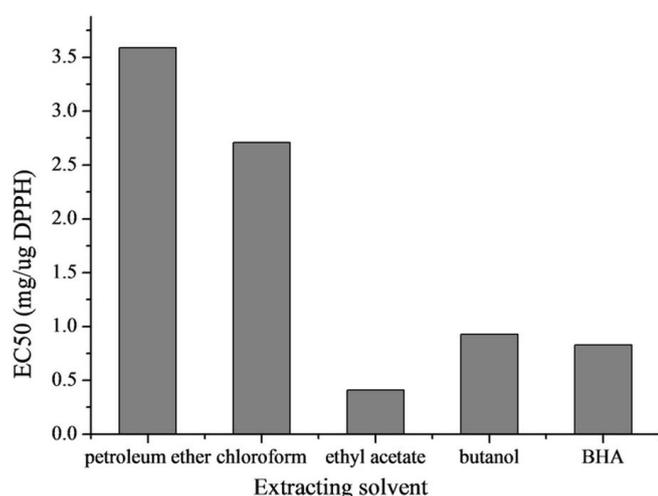


Figure 5. IC50 of lotus rhizome extracts.

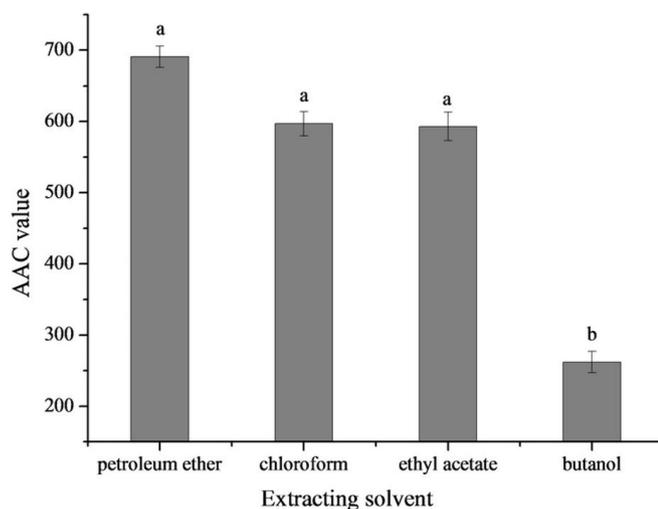


Figure 6. Antioxidant activity of lotus rhizome extracts determined by β -carotene bleaching method. Values in a column with different letters were significantly different ($P < 0.05$).

There was no significant correlation between the antioxidant activity against β -carotene-linoleic acid and the total phenolic content ($R^2 = 0.1995$). It could be concluded that not all the antioxidant activities against β -carotene-linoleic acid were caused by the phenols in lotus rhizome.

4 Conclusion

The antioxidant activity of the extract of lotus rhizome with different solvent by gradient extraction was determined in this study. The yield of butanol extract was the highest. The order of total phenolic content in different extracts was as follows: water extract < petroleum ether extract < chloroform extract < butanol extract < ethyl acetate extract, the same with that of the DPPH scavenging activity. The total phenolic content was significantly correlated with DPPH scavenging activity ($R^2 = 0.9363$). All extracts, except water extract with too poor antioxidant activity, showed obvious antioxidant activity against β -carotene-linoleic acid, and the antioxidant activity of butanol extract was significantly lower than others. There was no significant correlation between the antioxidant activity against β -carotene-linoleic acid and the total phenolic content ($R^2 = 0.1995$).

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