Efficient Solid-phase Synthesis of 1,2,3-Benzotriazin-4-ones with SynPhase[™] Lanterns

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A síntese de 1,2,3-benzotriazin-4-onas em fase sólida através da ciclização de 2-aminobenzamidas é descrita. O desenvolvimento de um método de síntese eficiente é importante para o desenvolvimento de drogas em razão da atividade biológica variada desses compostos. Os precursores de 1,2,3-benzotriazin-4-onas, as 2-aminobenzamidas, foram preparadas a partir de aminas ligadas a suporte sólido e ácidos 2-nitrobenzóicos. Várias 1,2,3-benzotriazin-4-onas foram obtidas em alta pureza utilizando-se SynPhase[™] Lanterns."

We have developed a solid-phase synthesis of 1,2,3-benzotriazin-4-ones through cyclization of 2-aminobenzamides *via* diazotization. The development of an efficient synthesis is important from the viewpoint of drug discovery, considering the various bioactivities of these derivatives. The precursors of 1,2,3-benzotriazin-4-ones, 2-aminobenzamides, were prepared from solid-supported amines and 2-nitrobenzoic acids. Various 1,2,3-benzotriazin-4-ones were obtained with high purity using SynPhaseTM Lanterns.

Keywords: solid-phase synthesis, drug, quinazoline, 1,2,3-benzotriazin-4-one, SynPhaseTM Lantern, diazotization

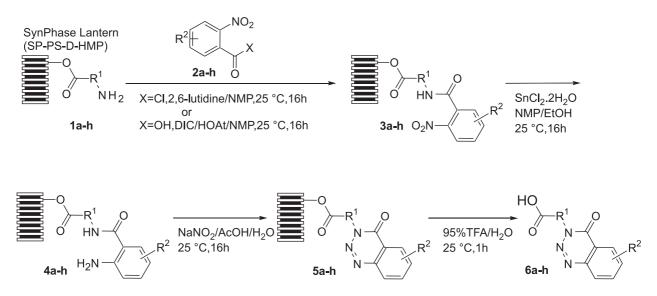
Introduction

Solid-phase synthesis of small-sized, non-peptidic molecules has emerged as an important drug discovery tool.¹ The synthesis of heterocyclic compounds on solidsupport, in particular, has been a focus of intensive study because of their applications toward a variety of drug targets.² Among various heterocycles, quinazolines are particularly attractive pharmacophores in view of their wide range of bioactivities.³ As a part of our project to develop an efficient synthetic protocol for quinazoline analogues on solid-support,⁴ we investigated the synthesis of 1,2,3-benzotriazin-4-ones. 1,2,3-Benzotriazin-4-ones have been known to possess various biological activities such as sedative,⁵ diuretic,⁶ anesthetic,⁷ antiarthritic⁸ and antitumor activity.9 In addition, numerous heterocycles such as quinazoline-2, 4-diones,^{4,10} 4-quinazolinones,^{4,11} 2-thioxoquinazolin-4-ones,4 3-(sulfanyl)-1,2,4benzothiadiazine 1,1-dioxide,⁴ 1,2,4-benzothiadiazin-3one 1,1-dioxides,4 2,1,3-benzothiadiazin-4-one 2-oxides,4 benzimidazole,¹² hydantoin¹³ and piperazinone¹⁴ have been prepared from the solid-supported amines. Therefore, the bioactivities of 1,2,3-benzotriazin-4-ones and these heterocycles can be efficiently compared if appropriate synthetic methods are developed. Here, we report the first solid-phase synthesis of 1,2,3-benzotriazin-4-ones.

Results and Discussion

Wang-type SynPhaseTM Lantern (1a)¹⁵ bearing a 3aminobenzoic acid ester⁴ was reacted with 2-nitrobenzoyl chloride (2a)/2,6-lutidine to give 3a (Scheme 1). The derivatized Lantern 3a was treated with SnCl₂ • 2H₂O/EtOH /NMP to give 2-aminobenzamide (4a) with high purity. Next, the cyclization of 4a through diazotization was attempted. Diazotization of arylamines has been often performed with NaNO, in highly acidic solvents such as TFA¹⁶ or concentrated HCl.⁵ However, compounds on Wang-type solid-supports are cleaved off when treated with these solvents. Therefore, numerous solvents were tested for the diazotization and AcOH/H₂O (1/1) was found as the best solvent to give 6a with high purity (Table 1). Other 1,2,3benzotriazin-4-ones (6b-d) were also synthesized using the solid-supported arylamines and 2-nitrobenzoyl chloride. This synthetic method was found to work with a solidsupported alkylamine and phenylalanine ester (6e). 1,2,3-Benzotriazin-4-ones with substitutions on the aromatic ring were also synthesized using substituted 2-nitrobenzoic

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Scheme 1. Solid-phase synthesis of 1,2,3-benzotriazin-4-ones via diazotization of 2-aminobenzamides.

acids. 2-Nitrobenzoic acids were coupled with 3aminobenzoic acid ester using N,N'-diisopropylcarbodiimide (DIC)/1-hydroxy-7-azabenzotriazole (HOAt) to give **3f-h**. The usage of HOAt was important as the acylation was not completed with HOBt. Following a similar reaction sequence, substituted 1,2,3-benzotriazin-4-ones (**6fh**) were successfully synthesized with high purity. Yields of products ranged from 46 - 83% based on the theoretical loading weights of the target molecules.¹⁷ The structures of all the products in this manuscript were confirmed by ¹H NMR and LC-MAS (ESI mass spectrometry).

In conclusion, the solid-phase synthesis of 1,2,3benzotriazin-4-ones has been accomplished by cyclization of 2-aminobenzamides via diazotization from various solid-supported amines and 2-nitrobenzoic acids. Although the size of the library is not large due to the limited diversity points (two diversity points), the same solidsupported amines can be derivatized into various heterocycles as described above, making this library an addition to the larger heterocycle library.

Experimental

General

Commercial reagents were used without further purification. ¹H NMR spectra were recorded on Varian VXR-300S (300 MHz) spectrometers using tetramethylsilane as an internal standard. Liquid chromatography was performed using symmetry C_{18} column with ESI/PDA detection on Micromass platform.

Representative procedure: synthesis of 3-(6-chloro-4-oxo-4H-benzo[d][1,2,3]triazin-3-yl)-benzoic acid (6f)

SynPhaseTM Lanterns bearing a 3-aminobenzoic acid ester (1f) were prepared according to the previous report.⁴ The Lantern was treated with 5-chloro-2-nitrobenzoic acid /HOAt/DIC/NMP (1.0 mg/1.0mg/25 µL/2 mL) for 16 h at 60 °C to give 3f. The Lanterns were washed with DMF (2 mL x 3) and CH₂Cl₂ (2 mL x 3) and dried under vacuum. After reduction of the nitro group with SnCl₂ • 2H₂O/NMP/ EtOH (1.0 g/2.0 mL/0.1 mL) for 16 h, the lantern was washed with DMF (2 mL x 3) and CH₂Cl₂ (2 mL x 3), and dried under vacuum for 1h to give 4f. Acetic acid (1.0 mL) and H₂O (1.0 mL) were added to the Lantern in a 2.5 mL syringe with a cap¹⁷, and immediately after adding solid NaNO₂ (100 mg), the syringe was sealed and put into a 50 mL Falcon tube¹⁸ to prevent the syringe from opening. After shaking the tube for 16 h, the tube and the syringe cap were carefully removed and the lantern was washed with DMF (2 mL x 3) and CH₂Cl₂ (2 mL x 3) and dried under vacuum to give 5f. The lantern was treated with 95% TFA/H₂O for 1 h and the solution was concentrated with a Genevac evaporater.¹⁹ The residue was dissolved with 50% CH₂CN/H₂O and lyophilized to give the product 6f (Entry f in Table 1) in 79% yield based on the theoretical loading weight of the target molecule.

3-(4-Oxo-4H-benzo[d][1,2,3]triazin-3-yl)-benzoic acid (6a). ¹H NMR (DMSO- d_6) δ 7.73 (dd, 1H, J 7.8, 7.8 Hz), 7.94 (ddd, 1H, J 1.2, 2.4, 8.1 Hz), 7.7-8.03 (m, 1H), 8.08-8.18 (m, 2H), 8.24 (dd, 1H, J 1.7, 1.8 Hz), 8.28-8.36 (m, 2H). MS m/z 268 (M + 1)⁺.

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Table 1. Various 1,2,3-benzotriazin-4-ones (**6a-h**) synthesized according to Scheme 1. ^a Reverse-phase HPLC was carried out using rapid water (0.05% TFA)/acetonitrile (0.04% TFA) linear gradients from 5% organic to 98% organic component over 5 min. Flow: 2 mL min⁻¹. Column: Waters Symmetry C₁₈ (3.5 μ m) 4.6 x 50 mm. HPLC purities were determined by summation of integrated HPLC peak areas at (210 + 3N) nm, N = 0 - 30. ^b Crude yields based on the theoretical loading weight of the target molecules. ^c Diazotization of 2-aminobenzamides was repeated 3 times to complete the reaction

Entry	R ² K NO ₂ NO ₂ X O	6	
		purity (%) ^a	yield (%) ^b
a	C1 O	> 95	76
b	C1 O	> 95	66
с	Cl O	> 95	64
d	Cl O	> 95	46
e	Cl O	> 95	83
f	C1 OH	92	79
g	CF ₃ (c) OH	> 95	61
h	NO2 OH	> 95	55

3-[2-(4-Oxo-4H-benzo[d][1,2,3]triazin-3-yl)-phenyl]acrylic acid (**6b**). ¹H NMR (DMSO- d_6) δ 6.61 (d, 1H, J 16.0 Hz), 7.67 (d, 1H, J 16.0 Hz), 7.62-7.73 (m, 2H), 7.84-7.89 (m, 1H), 7.98-8.04 (m, 2H), 8.13-8.19 (m, 1H), 8.29-8.36 (m, 2H). MS m/z 294 (M + 1)⁺.

3-[3-(4-Oxo-4H-benzo[d][1,2,3]triazin-3-yl)-phenyl]acrylic acid (**6c**). ¹H NMR (DMSO- d_6) δ 6.59 (d, 1H, J 15.9 Hz), 7.22 (d, 1H, J 15.9 Hz), 7.64-7.69 (m, 3H), 8.01-8.10 (m, 2H), 8.18-8.23 (m, 1H), 8.34 (d, 1H, J 1.5 Hz), 8.37 (d, 1H, J 1.2 Hz). MS m/z 294 (M + 1)⁺.

3-[4-(4-Oxo-4H-benzo[d][1,2,3]triazin-3-yl)-phenyl]acrylic acid (**6d**). ¹H NMR (DMSO-*d*₆) δ 6.65 (d, 1H, J 1.6 Hz), 7.69 (d, 1H, J 1.6 Hz), 7.72 (d, 2H, J 8.7 Hz), 7.92 (d, 2H, J 8.7 Hz), 7.98-8.03 (m, 1H), 8.13-8.19 (m, 1H), 8.28-8.29 (m, 1H), 8.31-8.36 (m, 1H). MS m/z 294 (M + 1)⁺.

2-(4-Oxo-4H-benzo[d][1,2,3]triazin-3-yl)-3-phenylpropionic acid (**6e**). ¹H NMR (DMSO-*d*₆) δ 3.53 (dd, 1H, J 11.0, 14.2 Hz), 3.65 (dd, 1H, J 5.1, 14.2 Hz), 7.08-7.16 (m, 5H), 7.91-7.96 (m, 1H), 8.07-8.13 (m, 1H), 8.17-8.23 (m, 2H). MS m/z 296 (M + 1)⁺.

3-(6-Chloro-4-oxo-4H-benzo[d][1,2,3]triazin-3-yl)benzoic acid (**6f**). ¹H NMR (DMSO-*d*₆) δ 7.72-7.77 (m, 1H), 7.91-7.95 (m, 1H), 8.09-8.12 (m, 1H), 8.18-8.25 (m, 2H), 8.31-8.36 (m, 2H). MS m/z 302, 603 (M + 1)⁺.

3-(4-Oxo-7-trifluoromethyl-4H-benzo[d][1,2,3]triazin-3-yl)-benzoic acid (**6g**). ¹H NMR (DMSO-*d*₆) δ 7.76 (dd, 1H, *J* 7.8, 7.8 Hz), 7.95 (ddd, 1H, *J* 1.3, 2.2, 7.8 Hz), 8.12 (ddd, 1H, *J* 1.3, 1.4, 7.8 Hz), 8.26 (dd, 1H, *J* 2.2, 2.2 Hz), 8.30 (dd, 1H, *J* 1.4, 8.2 Hz), 8.53 (d, 1H, *J* 8.2 Hz), 8.73 (s, 1H). MS m/z 336 (M + 1)⁺.

3-(8-Methyl-4-oxo-4H-benzo[d][1,2,3]triazin-3-yl)benzoic acid (**6h**). ¹H NMR (DMSO- d_6) δ 7.73 (dd, 1H, J 7.8, 7.8 Hz), 7.87 (dd, 1H, J 7.8, 7.8 Hz), 7.92-7.95 (m, 1H), 7.96-8.01 (m, 1H), 8.07-8.11 (m, 1H), 8.17 (dd, 1H, J 0.9, 7.5 Hz), 8.23 (dd, 1H, J 1.8 Hz). MS m/z 282 (M + 1)⁺.

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References

- Thompson, L. A.; Ellman, J.; A. *Chem. Rev.* **1996**, *96*, 555; Booth, S.; Hermkens, P. H. H.; Ottenheijm, H. C. J.; Rees, D. C.; *Tetrahedron* **1998**, *54*, 15385.
- Dressman, B. A.; Spangle, L. A.; Kaldor, S. W.; *Tetrahedron Lett.* **1996**, *37*, 937; Hanessian, S.; Yang, R. Y.; *Tetrahedron Lett.* **1996**, *37*, 5835; Bunin, B. A.; Ellman, J. A.; *J. Am. Chem. Soc.* **1992**, *114*, 1997; Buckman, B. O.; Mohan, R.; *Tetrahedron Lett.* **1996**, *37*, 4439; Gouilleux, L.; Fehrentz, J. -A.; Winternitz, F.; Martinez, J.; *Tetrahedron Lett.* **1996**, *37*, 7031; Shao, H.; Colucci, M.; Tong, S.; Zhang, H.; Castelhano, A. L.; *Tetrahedron Lett.* **1998**, *39*, 7235.
- de Laszlo S. E.; Quagliato, C. S.; Greenlee, W. J.; Patchett, A. A.; Chang, R. S. L.; Lotti, V. J.; Chen, T. -B.; Scheck, S. A.; Faust, K. A.; Kivlighn, S. S.; Schorn, T. S.; Zingaro, G. J.; Siegl, P. K. S.; *J. Med. Chem.* 1993, *36*, 3207; Hutchinson, J. H.; Cook, J. J.; Brashear, K. M.; Breslin, M. J.; Glass, J. D.; Gould, R. J.; Halczenko, W.; Holahan, M. A.; Lynch, R. J.; Sitko, G. R.; Stranieri, M. T.; Hartman, G. D.; *J. Med. Chem.* 1996, *39*, 4583.

- Makino, S.; Suzuki, N.; Nakanishi, E.; Tsuji, T.; *Tetrahderon Lett.* 2000, *41*, 8333; Makino, S.; Suzuki, N.; Nakanishi, E.; Tsuji, T.; *Synlett* 2000, 1670; Makino, S.; Nakanishi, E.; Tsuji, T.; *Tetrahderon Lett.* 2001, *42*, 1749; Makino, S.; Suzuki, N.; Nakanishi, E.; Tsuji, T.; *Synlett* 2001, 333; Makino, S.; Okuzumi, T.; Nakanishi, E.; Tsuji, T.; *Tetrahderon Lett.* 2002, *43*, 8401; Makino, S.; Nakanishi, E.; Tsuji, T.; *J. Combi. Chem.* 2003, *5*, 73; Makino, S.; Nakanishi, E.; Tsuji, T.; *Bull. Korean Chem. Soc.* 2003 in press.
- 5. Gadekar, S. M.; Ross, E.; J. Org. Chem., 1961, 26, 613.
- Gadekar, S. M.; Frederick, J. L.; J. Org. Chem., 1962, 27, 1383.
- Caliendo, G.; Fiorino, F.; Grieco, P.; Perissutti, E.; Santagada,
 V.; Meli, R.; Raso, G. M.; Zanesco, A.; Nucci, G. D.; *Eur. J. Med. Chem.* **1999**, *34*, 1043.
- 8. Zandt, V.; Michael, C.; PCT patent, WO9743239, 19xx.
- 9. Rosowsky, A.; PCT patent, WO9304051, 19xx.
- Gordeev, M. F.; Hui, H. C.; Gordon, E. M.; Patel, D. V.; *Tetrahedron Lett.* **1997**, *38*, 1729; Gordeev, M. F.; Luehr, G. W. Hui, H. C.; Gordon, E. M. Patel, D. V.; *Tetrahedron* **1998**, *54*, 15879.
- Mayer, J. P.; Lewis, G. S.; Curtis, M. J. Zhang, J. W.; *Tetrahe*dron Lett. **1997**, 38, 8445.
- 12. Tumelty, D.; Schwarz, M. K.; Cao, K.; Needels, M. C.; *Tetrahedron Lett.* **1999**, *40*, 6185.
- Nefzi, A.; Ostresh, J. M.; Giulianotti, M. Houghten, R. A.; *Tetrahedron Lett.* **1998**, *39*, 8199; Xiao, X. Y.; Ngu, K.; Chao, C.; Patel, D. V.; *J. Org. Chem.* **1997**, *62*, 6968.
- Mohamed, N.; Bhatt, U.; Just, G.; *Tetrahedron Lett.* 1998, *39*, 8213.
- 15. SynPhase[™] Lanterns are available from Mimotopes (Clayton, Victoria, Australia). The type of Lantern used in this report was SP-PS-D-HMP (long chain hydroxymethyl phenoxy linker), loading 35 mmol / lantern.
- Pinney, K. G.; Katzenellenbogen, J. A.; J. Org. Chem. 1991, 56, 3125.
- 17. Disposable polypropylene / polyethylene syringes are available from Aldrich (Milwaukee, WI).
- FALCON, BLUE MAXTM 50-ml polypropylene conical tube available from Becton Dickinson Labware (Franklin Lakes, NJ. USA 07417-1886).
- Genevac HT-8 available from Genevac Limited (Farthing Road, Ipswich, IP1 5AP, UK).

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