# Synthesis of New 1,2,4-Oxadiazoles Carrying (1'S,2'S)-t-Butyloxycarbonyl-1-amino-2-methyl-1-butyl and (1'S)-t-Butyloxycarbonyl-1'-amino-1'-ethyl Groups at C-5 

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#### Abstract

Uma síntese fácil e eficiente de 3-aril-5[(1S)-t-butiloxicarbonil-1-amino-(2S)-metil-1-butil)]-1,2,4-oxadiazóis 4a-f e 3-fenil-5-[(1S)-t-butiloxicarbonil-1-amino-1-etil]-1,2,4-oxadiazol $\mathbf{6}$ partindo de arilamidoximas, $N$ - $t$-Boc-L-isoleucina e $N$ - $t$-Boc-L-alanina é descrita. As estruturasedos intermediários e compostos finais foram determinadas a partir dos dados espectroscópicos.


A facile and efficient synthesis of 3-aryl-5-[(1S)-t-butyloxycarbonyl-1-amino-(2S)-methyl-1-butyl)]-1,2,4-oxadiazoles 4a-f and 3-phenyl-5-[(1S)-t-butyloxycarbonyl-1-amino-1-ethyl]-1,2,4oxadiazole 6 starting from arylamidoximes, $N$ - $t$-Boc-L-isoleucine and $N$ - $t$-Boc-L-alanine is decribed. The structures of the intermediates and final compounds have been deduced from spectroscopic data.

Keywords: chiral drugs, 1,2,4-oxadiazoles, arylamidoximes, isoleucine, alanine, antiinflammatory, antipyretic

## Introduction

It is known that stereoisomers of chiral drugs often exhibit pronounced differences in their pharmacokinetic and pharmacodynamic properties both in quantitative and qualitative terms. ${ }^{1}$ So, it is extremely necessary to study each stereoisomer separately. With the correct therapeutic information it is possible to maximize the properties of the drug by taking the most potent enantiomer, the eutomer, ${ }^{2}$ or eliminating the enantiomer that may cause side effects, the distomer. The latter can be linked to a secondary molecular target. ${ }^{3}$ The pharmaceutical market has converged strongly to this idea of single enantiomers as substitutes for their racemates. The change to chiral technologies offers many advantages to the pharmaceutical industry, including fewer toxicological tests, lower costs for liberation and less metabolites. These advantages are also beneficial for the patient's health. ${ }^{4}$ In the not too distant future, chiral drugs in single enantiomeric forms will probably substitute most of the existing racemates and will eventually touch every area of clinical medicine. ${ }^{7}$

[^0]Only in a few situations that are not common at all, it is more appropriate to use the racemate instead of the eutomer, for example with ibuprofen, a nonsteroidal antiinflammatory drug (NSAID) commercially available in the market. In this particular case, there is an enzymatic inversion of the $R(-)$ distomer to the eutomer $(S)$ responsible for the activity. ${ }^{5}$ In any event, it is a single enantiomer ( $S$ ) which is important. However, in other NSAIDs such as flurbiprofen ${ }^{6}$ and ketoprofen, ${ }^{7}$ this inversion is insignificant in humans, and so the application of the chiral technology for the use of the eutomer is more appropriate.

In view of the great importance of single enantiomers, our current research is focused on synthesizing new 1,2,4oxadiazoles containing asymmetric centers in their C-5 side chain. Our main goal was to synthesize compounds possessing an amino group in the $\mathrm{C}-5$ side chain. It is well known that a tertiary amine function attached to the C-5 alkyl chain of 1,2,4-oxadiazoles possess antiinflammatory as well as other interesting pharmacological activities. ${ }^{8,9}$ With this idea in mind, we synthesized seven Boc protected amines. In fact, two of these new compounds $\mathbf{4 d}$ and $\mathbf{4 e}$ showed antiinflammatory activity. These are very significant in the sense that they can be deprotected easily to remove the Boc group followed by their transformation
into $\mathrm{N}, \mathrm{N}$-dialkyl amino compounds. These compounds will be potential candidates for analgesic, antiinflammatory and antipyretic activity. Therefore, this communication describes some biologically active 1,2,4-oxadiazoles with one or two asymmetric centers in their side chain. Such compounds have not been cited in the literature before.

## Results and Discussion

The strategy used to achieve our goals explored known protected amino acids as starting materials. We have already published one paper relating to the synthesis of 3 -aryl-$1,2,4$-oxadiazoles carrying a protected L -alanine side chain, starting from arylamidoximes and protected aspartic acid, ${ }^{10}$ and it has been established that the products obtained from the condensation of arylamidoximes and an appropriately protected amino acid didn't lead to isomerization. Applying a similar strategy, but using Bocprotected L-isoleucine or L-alanine and arylamidoximes, we synthesized seven 1,2,4-oxadiazoles, six containing two asymmetric centers 4a-f and one oxadiazole having only one asymmetric center $\mathbf{6}$, as shown below (Scheme 1). Earlier, another group has observed that even in pyridine as reaction solvent, no significant racemization occurs in the case of 1,2,4-oxadiazoles having an asymmetric carbon attached to $\mathrm{C}-5 .{ }^{11}$

Since the reaction conditions in the present work are mild and similar to the one reported earlier, ${ }^{10}$ it is assumed that no racemization occurred during the synthesis of these $1,2,4$-oxadiazoles. This conclusion is supported by the ${ }^{1} \mathrm{H}$ NMR spectra, which showed the presence of only one H-3 absorption for 3a-f; had there been racemization, one would
expect two diastereomers hence more than one absorption. The $300 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectra of compounds $\mathbf{3 a - f}$ showed a triplet at $\delta 0.95 \mathrm{ppm}$ and a doublet at $\delta 1.02 \mathrm{ppm}$ for the terminal and C-3 methyl groups. In compounds 4a-f, the doublet due to methyl group shifted to higher field and generally overlapped with the triplet of the other methyl group. This shift is attributed to the ring current effect causing C-3 methyl group signals to move upfield.

The mass analyses agreed with the proposed structures. However the compounds $\mathbf{4 d}, \mathbf{e}$ and $\mathbf{f}$, show a sequence of fragmentations which occurs during the decomposition of $\mathrm{M}^{+}$. These include the loss of $t$-butyloxide radical (73 mass units), the extrusion of carbon monoxide ( 28 mass units), rearrangement and finally the elimination of butene2 molecule ( 56 mass units) to provide the major fragments a', at $m / z 204(\mathbf{4 d}), b^{\prime} m / z 208(4 \mathbf{e})$ and $c^{\prime}, m / z 219$ for the compound $\mathbf{4 f}$. These are depicted in Scheme 2.

The synthesis of compound $\mathbf{6}$ was carried out with the aim to shorten the side-chain containing the amino group and consequently reduce the lipophilicity.

## Experimental

Melting points were determined with a Thomas Hoover apparatus and are uncorrected. Elemental analyses of compounds $\mathbf{4 a} \mathbf{a} \mathbf{f}$ were performed in the Central Analítica do Departamento de Química Fundamental da Universidade Federal de Pernambuco. Infrared spectra were recorded on a Bruker spectrophotometer Model IFS66. $300 \mathrm{MHz}{ }^{1} \mathrm{H}-\mathrm{NMR}$ and $75 \mathrm{MHz}{ }^{13} \mathrm{C}$ NMR spectra were recorded on a Varian Unity plus instrument, using $\mathrm{CDCl}_{3}$ as solvent and TMS as internal reference. Thin-layer chromatography (tlc) was done


Scheme 1.

$\mathrm{a}^{\prime}, m / z=204$
$\mathrm{b}^{\prime}, m / z=208$
$c^{\prime}, m / z=219$
Scheme 2.
on plates coated with silica gel having a fluorescent indicator $\mathrm{P}_{254}$ (Merck) and the spots were detected under ultraviolet light. Arylamidoximes were obtained following the methodology reported in the literature. ${ }^{12}$

General Procedure for the Synthesis of $O-[(2 S, 3 S)-2-$ $t$-Butyloxycarbonylamino-3-methylpentanoyl] arylamidoximes ( $\mathbf{3 a - f}$ ). The appropriate arylamidoxime 1a-f ( 3.7 mmol ) was allowed to react with $N$-t-Boc-L-isoleucine 2 ( 3.7 mmol ) and $N, N^{\prime}$-dicyclohexylcrbodiimide ( 4.1 mmol ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{~mL}$ under $\mathrm{N}_{2}$ atmosphere at room temperature for periods varying from 3 to 5 hours. The completion of the reaction was monitored by TLC. Generally, the products were purified by column chromatography on silica gel using $n$-hexane-ethyl acetate (7.0:3.0) as eluent. The same solvent system was employed for developing the TLC plates. An ultraviolet lamp was used for the detection of the spots.

Data for $O-[(2 S, 3 S)-2-t-$ butyloxycarbonylamino-3methylpentanoyl]benzamidoxime (3a). White crystals (69\% yield), $\mathrm{mp} 110-111{ }^{\circ} \mathrm{C}$. IR (KBr) $v_{\max } / \mathrm{cm}^{-1}: 3483\left(\nu \mathrm{NH}_{2}\right.$, asym); $3347\left(\nu \mathrm{NH}_{2}\right.$, sym; $\left.\nu \mathrm{NH}\right) ; 2966(\nu \mathrm{C}-\mathrm{H}) ; 1740\left(\nu \mathrm{CO}_{2}\right)$; $1688\left(\nu \mathrm{NHCO}_{2}\right) ; 1614$ and $1519(\nu \mathrm{C}=\mathrm{C}$, ar); $1586(\nu \mathrm{C}=\mathrm{N})$; $1519\left(\nu \mathrm{C}=\mathrm{C}\right.$ band); $1172(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 0.95$ (3H, t, J 7.5 Hz ); 1.02 (3H, d, J 6.9 Hz ); 1.44 ( $9 \mathrm{H}, \mathrm{s}$ ); 1.14$1.68(2 \mathrm{H}, \mathrm{m}) ; 1.84-2.00(1 \mathrm{H}, \mathrm{m}) ; 4.31(1 \mathrm{H}, \mathrm{dd}, J 8.7 \mathrm{~Hz}, J 6.3$ $\mathrm{Hz}) ; 5.16(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 5.37(2 \mathrm{H}, \mathrm{bs}) ; 7.35-7.52(3 \mathrm{H}, \mathrm{m})$; 7.67-7.75 (2H, m). ${ }^{13} \mathrm{CNMR}\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-5) ; 15.3\left(\mathrm{CH}_{3}\right)$; 25.0 (C-4); $28.3\left(\mathrm{CH}_{3}\right)_{3} ; 37.8(\mathrm{C}-3) ; 57.3(\mathrm{C}-2) ; 80.1(\mathrm{C})$; 130.8 (C-1'); 128.1 (C-2' and C-6'); 128.7 (C-3' and C-5'); 131.1 (C-4'); 155.9 (C1"); 157.8 (CONH); 170.1 (C-1).

Data for $O-[(2 S, 3 S)-2-t-$ butyloxycarbonylamino-3-methylpentanoyl]m-tolylamidoxime (3b). White crystals ( $73 \%$ yield, mp $118-119{ }^{\circ} \mathrm{C}$. IR (KBr) $v_{\max } / \mathrm{cm}^{-1}: 3494$ $\left(\nu \mathrm{NH}_{2}\right.$, asym $) ; 3343\left(\nu \mathrm{NH}_{2}\right.$, sym; $\left.\nu \mathrm{NH}\right) ; 2967(\nu \mathrm{C}-\mathrm{H}) ; 1742$
$\left(\nu \mathrm{CO}_{2}\right) ; 1691\left(\nu \mathrm{NHCO}_{2}\right) ; 1621$ and $1522(\nu \mathrm{C}=\mathrm{C}$, ar) $; 1587$ $(\nu \mathrm{C}=\mathrm{N}) ; 1168(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{HNMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.93(3 \mathrm{H}, \mathrm{t}, J 7.4$ $\mathrm{Hz}) ; 1.00(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6.9 \mathrm{~Hz}) ; 1.43(9 \mathrm{H}, \mathrm{s}) ; 1.10-1.70(2 \mathrm{H}, \mathrm{m})$; $1.82-2.00(1 \mathrm{H}, \mathrm{m}) ; 2.36(3 \mathrm{H}, \mathrm{s}) ; 4.30(1 \mathrm{H}, \mathrm{dd}, J 6.0 \mathrm{~Hz}, J 8.7$ $\mathrm{Hz}) ; 5.19(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 5.29(2 \mathrm{H}, \mathrm{bs}) ; 7.62(2 \mathrm{H}, \mathrm{d}, J 9.3$ $\mathrm{Hz}) ; 8.00(2 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-5) ; 15.5$ $\left(\mathrm{CH}_{3}\right) ; 21.4\left(\mathrm{Ar}^{2} \mathrm{CH}_{3}\right) ; 25.0(\mathrm{C}-4) ; 28.3\left(\mathrm{CH}_{3}\right)_{3} ; 37.6(\mathrm{C}-3)$; 57.5 (C-2); 80.0 (C); 123.8(C-6’); 127.6 (C-1'); 128.6 (C2'); 130.2 (C-5'); 132.1 (C-4'); 138.6 (C-3'); 156.0 (C1"); 158.3 (CONH); 170 (C-1).

Data for O-[(2S, 3S)-2-t-butyloxycarbonylamino-3-methylpentanoyl]p-tolylamidoxime (3c). White crystals ( $65 \%$ yield), $\mathrm{mp} 105-106{ }^{\circ} \mathrm{C}$. IR (KBr) $v_{\max } / \mathrm{cm}^{-1}: 3500$ $\left(\nu \mathrm{NH}_{2}\right.$, asym) $3358\left(\nu \mathrm{NH}_{2}\right.$, sym $) ; 3333(v \mathrm{NH}) ; 2965$ $(\nu \mathrm{C}-\mathrm{H}) ; 1746\left(\nu \mathrm{CO}_{2}\right) ; 1693\left(\nu \mathrm{NHCO}_{2}\right) ; 1614$ and 1521( $\nu \mathrm{C}=\mathrm{C}$, ar); $1587(\nu \mathrm{C}=\mathrm{N}) ; 1175(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 0.94(3 \mathrm{H}, \mathrm{t}, J 7.4 \mathrm{~Hz}) ; 1.01(3 \mathrm{H}, \mathrm{d}, J 6.6 \mathrm{~Hz}) ;$ 1.44 (9H, s); 1.14-1.68 ( $2 \mathrm{H}, \mathrm{m}$ ); 1.82-2.00 ( $1 \mathrm{H}, \mathrm{m}$ ); 2.38 $(3 \mathrm{H}, \mathrm{s}) ; 4.30(1 \mathrm{H}, \mathrm{dd}, J 6.3 \mathrm{~Hz}, J 8.1 \mathrm{~Hz}) ; 5.21(1 \mathrm{H}, \mathrm{d}, J 8.4$ $\mathrm{Hz}) ; 5.44(2 \mathrm{H}, \mathrm{bs}) ; 7.35(2 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 7.61(2 \mathrm{H}, \mathrm{d}, J 8.4$ $\mathrm{Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-5) ; 15.5\left(\mathrm{CH}_{3}\right) ; 21.2(\mathrm{Ar}-$ $\left.\mathrm{CH}_{3}\right) ; 25.1(\mathrm{C}-4) ; 28.3\left(\mathrm{CH}_{3}\right)_{3} ; 37.5(\mathrm{C}-3) ; 57.4(\mathrm{C}-2) ; 80.1$ (C); 123.8 (C-1'); 127.4 (C-2' and C-6'); 129.5 (C-3' and C-5'); 141.5 (C-4'); 155.9 (C-1"); 158.3 (CONH); 170.0 (C-1).

Data for O-[(2S, 3S)-2-t-butyloxycarbonylamino-3-methylpentanoyl]p-anisylamidoxime (3d). Yellow crystals ( $80 \%$ yield), $\mathrm{mp}=112-113{ }^{\circ} \mathrm{C}$. IR (KBr) $v_{\max } / \mathrm{cm}^{-1}: 3493$ $\left(v-\mathrm{NH}_{2}\right.$, asym $) ; 3344\left(v-\mathrm{NH}_{2}\right.$, sym; $\left.v \mathrm{NH}\right) ; 2970(v \mathrm{C}-\mathrm{H}) ; 1744$ $\left(v-\mathrm{CO}_{2}\right) ; 1692\left(v-\mathrm{NHCO}_{2}\right) ; 1615$ and $1520(v \mathrm{C}=\mathrm{C}$, ar $) ; 1565$ $(\nu \mathrm{C}=\mathrm{N}) ; 1165(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 0.96(3 \mathrm{H}, \mathrm{t}, J 7.4$ Hz ); 1.04 ( $3 \mathrm{H}, \mathrm{d}, J 6.9 \mathrm{~Hz}$ ); $1.45(9 \mathrm{H}, \mathrm{s}) ; 1.16-1.70(2 \mathrm{H}, \mathrm{m})$; $1.86-2.00(1 \mathrm{H}, \mathrm{m}) ; 3.81(3 \mathrm{H}, \mathrm{s}) ; 4.30(1 \mathrm{H}, \mathrm{dd}, J 6.3 \mathrm{~Hz}$, $J 8.1 \mathrm{~Hz}) ; 5.13(1 \mathrm{H}, \mathrm{d}, J 8.4 \mathrm{~Hz}) ; 5.49(2 \mathrm{H}, \mathrm{bs}) ; 7.91(2 \mathrm{H}, \mathrm{d}$,
$J 9.0 \mathrm{~Hz}) ; 8.27(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 11.4$ $(\mathrm{C}-5) ; 15.4\left(\mathrm{CH}_{3}\right) ; 25.1(\mathrm{C}-4) ; 28.3\left(\mathrm{CH}_{3}\right)_{3} ; 37.5(\mathrm{C}-3) ; 57.4$ (C-2); $55.3\left(\mathrm{O}_{\left.-\mathrm{CH}_{3}\right)}\right) ; 80.1(\mathrm{C}) ; 113.9\left(\mathrm{C}-3 \prime\right.$ and $\left.\mathrm{C}-5{ }^{\prime}\right) ; 122.5$ (C-1'); 128.4 (C-2' and C-6'); 155.9 (C-4'); 157.8 (C-1"); 161.9 (CONH); 170.1 (C-1).

Data for $O-[(2 S, 3 S)-2-t-b u t y l o x y c a r b o n y l a m i n o-3-$ methylpentanoyl]p-chlorobenzamidoxime (3e). White crystals ( $80 \%$ yield), mp 114-115 ${ }^{\circ} \mathrm{C}$. IR (KBr) $v_{\max } / \mathrm{cm}^{-1}$ : $3500\left(v-\mathrm{NH}_{2}\right.$, asym $) ; 3358\left(\nu \mathrm{NH}_{2}\right.$, sym $) ; 3315(v-\mathrm{NH}) ; 2963$ $(\nu \mathrm{C}-\mathrm{H}) ; 1751\left(\nu \mathrm{CO}_{2}\right), 1692\left(\nu \mathrm{NHCO}_{2}\right) ; 1615$ and 1522 $(\nu \mathrm{C}=\mathrm{C}, \operatorname{ar}) ; 1587(\nu \mathrm{C}=\mathrm{N}) ; 1172(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right)$ $\delta: 0.95(3 \mathrm{H}, \mathrm{t}, J 7.5 \mathrm{~Hz}) ; 1.02(3 \mathrm{H}, \mathrm{d}, J 6.9 \mathrm{~Hz}) ; 1.45(9 \mathrm{H}, \mathrm{s}) ;$ 1.16-1.70 (2H, m); 1.80-2.00 (1H, m); $4.32(1 \mathrm{H}, \mathrm{dd}, J 6.0$ $\mathrm{Hz}, J 8.9 \mathrm{~Hz}) ; 5.13(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 5.19(2 \mathrm{H}, \mathrm{bs}) ; 7.25$ $(2 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}) ; 7.60(2 \mathrm{H}, \mathrm{d}, J 8.1 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : $11.4(\mathrm{C}-5) ; 15.6\left(\mathrm{CH}_{3}\right) ; 25.1(\mathrm{C}-4) ; 28.4\left(\mathrm{CH}_{3}\right)_{3} ; 37.6(\mathrm{C}-3)$; 57.5 (C-2); 80.2 (C); 128.3 (C-2' and C-6’); 128.9 (C-3' and C-5'); 129.2 (C-1'); 137.2 (C-4'); 155.1 (C-1"); 157.1 (CONH); 170.3 (C-1).

Data for $O-[(2 S, 3 S)-2-t-b u t y l o x y c a r b o n y l a m i n o-3-$ methylpentanoyl]p-nitrobenzamidoxime (3f). White crystals ( $75 \%$ yield), $\mathrm{mp} 110-112{ }^{\circ} \mathrm{C}$. IR (KBr) $\nu_{\max } / \mathrm{cm}^{-1}: 3477$ $\left(\nu \mathrm{NH}_{2}, \operatorname{asym}\right) ; 3341\left(\nu-\mathrm{NH}_{2}, \mathrm{sym} ; ~ v \mathrm{NH}\right) ; 2967(\nu \mathrm{C}-\mathrm{H}) ; 1755$ $(\nu \mathrm{CO} 2) ; 1691(\nu \mathrm{NHCO} 2) ; 1639$ and $1522(\nu \mathrm{C}=\mathrm{C}$, ar $) ; 1585$ $(\nu \mathrm{C}=\mathrm{N}) ; 1168(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.94(3 \mathrm{H}, \mathrm{t}, J 7.5$ $\mathrm{Hz}) ; 1.02(3 \mathrm{H}, \mathrm{d}, J 6.9 \mathrm{~Hz}) ; 1.44(9 \mathrm{H}, \mathrm{s}) ; 1.10-1.76(2 \mathrm{H}, \mathrm{m}) ;$ $1.80-2.00(1 \mathrm{H}, \mathrm{m}) ; 4.31(1 \mathrm{H}, \mathrm{dd}, J 6.0 \mathrm{~Hz}, J 8.7 \mathrm{~Hz}) ; 5.15$ $(1 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) ; 5.26(2 \mathrm{H}, \mathrm{bs}) ; 7.25-7.30(2 \mathrm{H}, \mathrm{m}) ; 7.40-7.50$ $(2 \mathrm{H}, \mathrm{m}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 11.3(\mathrm{C}-5) ; 15.5\left(\mathrm{CH}_{3}\right) ; 25.0(\mathrm{C}-$ 4); $28.2\left(\mathrm{CH}_{3}\right)_{3} ; 37.4(\mathrm{C}-3) ; 57.4(\mathrm{C}-2) ; 80.3(\mathrm{C}) ; 123.8(\mathrm{C}-3$, and C-5'); 127.9 (C-2' and C-6'); 136.9 (C-1'); 149.3 (C-4'); 155.9 (C-1"); 156.1 (CONH);169.9 (C-1).

General Procedure for the Synthesis of 5-[(1'S, 2'S)-1-t-Butyloxycarbonylamino-2-methylbutyl)]-3-aryl-1,2,4oxadiazoles 4a-f and 5-[(2S)-1-t-Butyloxycarbonyl-1-aminoeth-1-yl ]-3-phenyl-1,2,4-oxadiazole (6)

The intermediates $\mathbf{3}$ were heated at temperatures that varied from 100 to $120^{\circ} \mathrm{C}$. The reactions were acompanied by TLC and discontinued when the TLC plate showed the disappearance of the starting compound, in three to five hours. In the case of the compound starting from $N-t-B o c-L-$ alanine 5, it was not possible to isolate the intermediate. The compounds obtained 4a-f initially had a pasty appearance, but after some days in a dry atmosphere under vacum the compounds started to crystallize. Conversion of intermediates $\mathbf{3}$ a-f to $\mathbf{4}$ a-f were quantitative; however, the yields of the pure and crystalline products are given below.

Data for 5-[(1'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-phenyl-1,2,4-oxadiazole (4a). Colorless
crystals ( $69 \%$ yield), $\mathrm{mp} 73.3-74.4^{\circ} \mathrm{C}$ (ethanol-water). $[\alpha]_{\mathrm{D}}^{22}-44^{\circ} \pm 6\left(\mathrm{c}=0.005, \mathrm{CHCl}_{3}\right) . \mathrm{IR}(\mathrm{KBr}) v_{\max } / \mathrm{cm}^{-1}: 3356$ $(\nu \mathrm{NH}) ; 2976(\nu \mathrm{C}-\mathrm{H}) ; 1716(\nu \mathrm{C}=\mathrm{O}) ; 1569(\nu \mathrm{C}=\mathrm{N}) ; 1523$ $(\nu \mathrm{C}=\mathrm{C}, \operatorname{ar}) ; 1160(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.95(3 \mathrm{H}, \mathrm{t}, J$ $7.2 \mathrm{~Hz}) ; 0.94(3 \mathrm{H}, \mathrm{d}, J 6.3 \mathrm{~Hz}) ; 1.46(9 \mathrm{H}, \mathrm{s}) ; 1.16-1.70(2 \mathrm{H}$, m); 1.94-2.12 (1H, m); 5.06 (1H, dd, J 5.7 Hz, J 8.7 Hz); $5.26(1 \mathrm{H}, \mathrm{d}, J 9.3 \mathrm{~Hz}) ; 7.40-7.56(3 \mathrm{H}, \mathrm{m}) ; 8.04-8.14(2 \mathrm{H}$, m). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-4 ') ; 15.1\left(\mathrm{CH}_{3}\right) ; 25.0(\mathrm{C}-$ $\left.3^{\prime}\right) ; 28.3\left(\mathrm{CH}_{3}\right)_{3} ; 39.2\left(\mathrm{C}-2^{\prime}\right) ; 52.6(\mathrm{C}-1 ’) ; 80.36(\mathrm{C}) ; 126.6$ (C-1"); 127.5 (C-2" and C-6"); 128.8 (C-3" and C-5"); 131.2 (C-4"); 155.1 (C-3); 168.1 (C-5); 178.9 (CO). MS, $\mathrm{m} / \mathrm{z}(\%): 332\left[\mathrm{MH}^{+}, 12 \%\right], 174$ (69\%), 146 (81\%), 119 ( $100 \%$ ), 104 (48\%). Anal. Calc. for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{3}: \mathrm{C}, 65.17$; H, 7.54; N, 12.67. Found: C, 65.52; H, 7.53; N, 12.98.

Data for 5-[(1'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-m-tolyl-1,2,4-oxadiazole (4b). Colorless crystals ( $80 \%$ yield), mp 59.6-60.2 ${ }^{\circ} \mathrm{C}$ (ethanol-water). IR $(\mathrm{KBr}) \nu_{\max } / \mathrm{cm}^{-1}: 3360(\nu \mathrm{NH}) ; 2982(\nu \mathrm{C}-\mathrm{H}) ; 1685(\nu \mathrm{C}=\mathrm{O})$; $1564(\nu \mathrm{C}=\mathrm{N}) ; 1527(\nu \mathrm{C}=\mathrm{C}, \mathrm{ar}) ; 1176(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right)$ $\delta: 0.94(3 \mathrm{H}, \mathrm{t}, J 7.4 \mathrm{~Hz}) ; 0.93(3 \mathrm{H}, \mathrm{d}, J 6.3 \mathrm{~Hz}) ; 1.45(9 \mathrm{H}, \mathrm{s}) ;$ 1.14-1.82 (2H, m); 1.94-2.12 (1H, m); $2.40(3 \mathrm{H}$, s) $5.04(1 \mathrm{H}$, dd, $J 6.0 \mathrm{~Hz}, J 8.9 \mathrm{~Hz}) ; 5.28(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 7.27(2 \mathrm{H}, \mathrm{d}, J$ $8.1 \mathrm{~Hz}) ; 7.96(2 \mathrm{H}, \mathrm{d}, J 8.4 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-$
 39.3 (C-2'); 52.7 (C-1'); 80.3 (C); 123.8 (C-6"); 127.6 (C1"); 128.6 (C-2"); 130.2 (C-5"); 132.0 (C-4"); 138.6 (C-3"); 155.2 (C-3); 168.2 (C-5); 178.9 (CO). MS, m/z (\%): 345 [ $\left.\mathrm{M}^{+}, 8 \%\right], 289$ (53\%), 233 ( $71 \%$ ), 188 ( $100 \%$ ), 160 ( $52 \%$ ), 133 (86\%), 118 (37\%). Anal. Calc. for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3}: \mathrm{C}, 66.00$; H, 7.82; N, 12.16. Found: C, 66.52; H, 7.63; N, 12.28.

Data for 5-[(1'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-p-tolyl-1,2,4-oxadiazole (4c). Colorless crystals ( $75 \%$ yield), $\mathrm{mp} 65.4-66.8^{\circ} \mathrm{C}$ (ethanol-water). $[\mathrm{D}]_{\mathrm{D}}{ }^{22}-58^{\circ} \pm 3\left(\mathrm{c}=0.01, \mathrm{CHCl}_{3}\right) . \mathrm{IR}(\mathrm{KBr}) v_{\max } / \mathrm{cm}^{-1}: 3357$ $(\nu-\mathrm{NH}) ; 2970(\nu \mathrm{C}-\mathrm{H}) ; 1687(\nu \mathrm{C}=\mathrm{O}) ; 1569(\nu \mathrm{C}=\mathrm{N}) ; 1510$ $(\nu \mathrm{C}=\mathrm{C}, \operatorname{ar}), 1159(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta: 0.93(3 \mathrm{H}, \mathrm{d}$, $J 6.3 \mathrm{~Hz}), 0.94(3 \mathrm{H}, \mathrm{t}, J 7.2 \mathrm{~Hz}),, 1.45(9 \mathrm{H}, \mathrm{s}) ; 1.15-1.90$ (2H, m); 1.94-2.12 (1H, m); $2.41(3 \mathrm{H}, \mathrm{s}) ; 5.05(1 \mathrm{H}, \mathrm{dd}, J 5.4$ $\mathrm{Hz}, J 8.9 \mathrm{~Hz}) ; 5.29(1 \mathrm{H}, \mathrm{d}, J 9.3 \mathrm{~Hz}) ; 7.27-7.40(2 \mathrm{H}, \mathrm{m})$; 7.84-7.92 (2H, m). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4$ (C-4'); 15.1
 52.7 (C-1'); 80.3 (C); 123.8 (C-1"); 127.4 (C-2" and C$6 ") ; 129.5$ (C-3" and C-5"); 141.5 (C-4"); 155.1 (C-3); 168.1 (C-5); 178.7 (CO). MS, $m / z$ (\%): 345 [M+, 6\%], 289 (71\%), 188 (98\%), 160 (48\%), 133 (100\%), 118 (44\%). Anal. Calc. for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{3}: \mathrm{C}, 66.00 ; \mathrm{H}, 7.82 ; \mathrm{N}, 12.16$. Found: C, 65.93; H, 8.11; N, 12.73.

Data for 5-[(1'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-p-anisyl-1,2,4-oxadiazole (4d). Colorless crystals ( $73 \%$ yield), mp 70.3-71.2 ${ }^{\circ} \mathrm{C}$ (ethanol-water). IR $(\mathrm{KBr}) \nu_{\max } / \mathrm{cm}^{-1}: 3349(v-\mathrm{NH}) ; 2968(\nu \mathrm{C}-\mathrm{H}) ; 1689(\nu \mathrm{C}=\mathrm{O})$;
$1568(\nu \mathrm{C}=\mathrm{N})$; 1529 ( $\nu \mathrm{C}=\mathrm{C}$, ar); 1173 ( $\nu \mathrm{C}-\mathrm{O}$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 0.93(3 \mathrm{H}, \mathrm{d}, J 6.3 \mathrm{~Hz}) ; 0.94(3 \mathrm{H}, \mathrm{t}, J 6.6 \mathrm{~Hz})$; 1.16-1.70 ( $2 \mathrm{H}, \mathrm{m}$ ); $1.45(9 \mathrm{H}, \mathrm{s}) ; 1.90-2.10(1 \mathrm{H}, \mathrm{m}) ; 3.85$ ( $3 \mathrm{H}, \mathrm{s}$ ); 5.03 ( 1 H, dd, $J 5.7 \mathrm{~Hz}, J 9.0 \mathrm{~Hz}$ ); 5.28 ( $1 \mathrm{H}, \mathrm{d}, J 9.3$ $\mathrm{Hz}) ; 6.97(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) ; 8.01(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4\left(\mathrm{C}-4{ }^{\prime}\right) ; 15.1\left(\mathrm{CH}_{3}\right) ; 25.1\left(\mathrm{C}-3{ }^{\prime}\right) ; 28.3\left(\mathrm{CH}_{3}\right)_{3}$; $34.9\left(\mathrm{O}^{3}-\mathrm{CH}_{3}\right) ; 39.3$ (C-2'); 52.7 (C-1'); 80.4 (C); 114.2 (C3 " and C-5");119.1 (C-1"); 129.1 (C-2" and C-6"); 155.1 (C-3);161.9 (C-4"); 167.8 (C-5); 178.7 (CO). MS, $m / z(\%):$ 204 (100\%), 176 ( $47 \%$ ), 149 (70\%), 133 (11\%). Anal. Calc. for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{4}: \mathrm{C}, 63.15 ; \mathrm{H}, 7.47$; $\mathrm{N}, 11.63$. Found: C, 63.35; H, 7.27; N, 12.22.

Data for 5-[(1'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-p-chlorophenyl-1,2,4-oxadiazole (4e). Colorless crystals ( $65 \%$ yield), mp 68.2-68.8 ${ }^{\circ} \mathrm{C}$ (ethanolwater). $[\alpha]_{\mathrm{D}}^{22}=-60^{0} \pm 4\left(\mathrm{c}=0.01, \mathrm{CHCl}_{3}\right)$. IR $(\mathrm{KBr}) v_{\max } / \mathrm{cm}^{-1}$ : 3364 ( $\nu-\mathrm{NH}) ; 2968(\nu \mathrm{C}-\mathrm{H}) ; 1685(\nu \mathrm{C}=\mathrm{O})$; $1562(\nu \mathrm{C}=\mathrm{N})$; 1518 ( $\nu \mathrm{C}=\mathrm{C}$, ar); $1174(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 0.93$ (3H, d, J 6.0 Hz ); 0.95 ( $3 \mathrm{H}, \mathrm{t}, J 7.5 \mathrm{~Hz}$ ); 1.14-1.80 ( $2 \mathrm{H}, \mathrm{m}$ ); 1.45 ( $9 \mathrm{H}, \mathrm{s}$ ); 1.92-2.12 ( $1 \mathrm{H}, \mathrm{m}$ ); 5.05 ( 1 H , dd, $J 5.7 \mathrm{~Hz}, J$ $8.7 \mathrm{~Hz}) 5.25(1 \mathrm{H}, \mathrm{d}, J 8.7 \mathrm{~Hz}) ; 7.45(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) ; 8.02$ $(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz})$. ). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.4(\mathrm{C}-4$ '); 15.1 $\left(\mathrm{CH}_{3}\right) ; 25.0\left(\mathrm{C}-3^{\prime}\right) ; 28.2\left(\mathrm{CH}_{3}\right)_{3} ; 39.1\left(\mathrm{C}-2^{\prime}\right) ; 52.6\left(\mathrm{C}-1{ }^{\prime}\right)$; 80.4 (C); 125.1 (C-1"); 126.8 (C-2" and C-6"); 129.1 (C3 " and C-5"); 137.4 (C-4"); 155.1 (C-3); 167.3 (C-5); 179.0 (CO). MS, $m / z$ (\%): 208 (100\%), 180 ( $47 \%$ ), 153 (89\%), 138, 20\%). Anal. Calc. for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{Cl}$ : C, 59.10; $\mathrm{H}, 6.57$; N, 11.49. Found: C, 59.39; H, 6.77; N, 11.14.

Data for 5-[(l'S, 2'S)-1'-t-Butyloxycarbonylamino-2'-methylbutyl)]-3-p-nitrophenyl-1,2,4-oxadiazole (4f). Yellow crystals ( $80 \%$ yield), mp 76.1-77.2 ${ }^{\circ} \mathrm{C}$ (ethanolwater). IR (KBr) $v_{\max } / \mathrm{cm}^{-1}: 3337(\nu \mathrm{NH}) ; 2969(\nu \mathrm{C}-\mathrm{H}) ; 1690$ $(\nu \mathrm{C}=\mathrm{O}) ; 1575(\nu \mathrm{C}=\mathrm{N}) ; 1528$ ( $\nu \mathrm{C}=\mathrm{C}$, ar);1172 ( $\nu \mathrm{C}-\mathrm{O})$. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 0.97(3 \mathrm{H}, \mathrm{t}, J 7.4 \mathrm{~Hz}) ; 0.97(3 \mathrm{H}, \mathrm{d}, J 6.6$ $\mathrm{Hz}) ; 1.47$ (9H, s); 1.16-1.80 (2H, m); 1.96-2.12 (1H, m); $5.08(1 \mathrm{H}, \mathrm{dd}, J 6.0 \mathrm{~Hz}, J 9.0 \mathrm{~Hz}) ; 5.25(1 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz})$, $8.29(2 \mathrm{H}, \mathrm{d}, J 9.0 \mathrm{~Hz}) ; 8.35(2 \mathrm{H}, \mathrm{d}, J 9.3 \mathrm{~Hz})$. ). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 11.22\left(\mathrm{C}-4\right.$ '); $15.18\left(\mathrm{CH}_{3}\right) ; 25.08\left(\mathrm{C}-3{ }^{\prime}\right)$ ); 28.25 $\left(\mathrm{CH}_{3}\right)_{3} ; 39.07\left(\mathrm{C}-2^{\prime}\right) ; 52.81\left(\mathrm{C}-1^{\prime}\right) ; 80.63(\mathrm{C}) ; 124.0\left(\mathrm{C}-3^{\prime \prime}\right.$ and C-5"); 126.5 (C-2" and C-6"); 132.5 (C-1"); 149.6 (C-4"); 155.1 (C-3); 166.6 (C-5); 179.9 (CO). MS, $m / z(\%):$ 219 ( $100 \%$ ), 191 ( $35 \%$ ), 164 ( $52 \%$ ). Anal. Calc. for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{~N}_{4} \mathrm{O}_{5}: \mathrm{C}, 57.44 ; \mathrm{H}, 6.43$. Found: C, $57.51 ; \mathrm{H}, 6.72$.

Data for 5-[(1'S)-t-Butyloxycarbonylaminoeth-1'-yl]-3-phenyl-1,2,4-oxadiazole (6) White crystals (70\% yield), $\mathrm{mp} 89.3-90.3^{\circ} \mathrm{C}$ (ethanol-water). $[\alpha]_{\mathrm{D}}{ }^{22}-70^{\circ} \pm 2$ ( $\mathrm{c}=0.02$, $\left.\mathrm{CHCl}_{3}\right)$. IR (KBr) $v_{\text {max }} / \mathrm{cm}^{-1}: 3370(\nu \mathrm{NH}) ; 2983(\nu \mathrm{C}-\mathrm{H}) ; 1687$ $(\nu \mathrm{C}=\mathrm{O}) ; 1571(\nu \mathrm{C}=\mathrm{N}) ; 1518\left(\nu \mathrm{C}=\mathrm{C}\right.$, ar); $1164(\nu \mathrm{C}-\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta: 1.43(9 \mathrm{H}, \mathrm{s}) ; 1.60(3 \mathrm{H}, \mathrm{d}, J 6.9 \mathrm{~Hz}) ; 5.06-$ $5.24(1 \mathrm{H}, \mathrm{m}) ; 5.44(1 \mathrm{H}, \mathrm{d}, J 7.8 \mathrm{~Hz}) ; 7.38-7.52(3 \mathrm{H}, \mathrm{m})$; 8.00-8.10 ( $2 \mathrm{H}, \mathrm{m}$ ). ). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): 20.1(\mathrm{C}-2$ '); 28.3
$\left(\mathrm{CH}_{3}\right)_{3} ; 44.8$ (C-1'); $80.0(\mathrm{C}) ; 126.5(\mathrm{C}-1 ") ; 127.5$ (C-2" and C-6"); 128.8 (C-3" and C-5"); 131.2 (C-4"); 168.3 (C3); 173.7 (C-5); 180.1 (CO). MS, $m / z(\%): 289$ [M $\left.{ }^{+}, 8 \%\right]$, 234 ( $51 \%$ ), 174 ( $72 \%$ ), 160 ( $100 \%$ ), 119 ( $98 \%$ ), 104 ( $80 \%$ ). Anal. Calc. for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{3}: \mathrm{C}, 62.49 ; \mathrm{H}, 6.30 ; \mathrm{N}, 14.57$. Found: C, 62.58; H, 6.27; N, 14.87.

## Conclusion

In conclusion, we have been able to synthesize and prove the structures of seven new 1,2,4-oxadiazoles as well as six of their intermediates 3a-f derived from L-(+)isoleucine.

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