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FOOD TECHNOLOGY

Oxidative stability during frozen storage of fillets from silver catfish (*Rhamdia quelen*) sedated with the essential oil of *Aloysia triphylla* during transport

Estabilidade oxidativa de filés congelados de jundiás (*Rhamdia quelen*) sedados com o óleo essencial de *Aloysia triphylla* durante o transporte

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ABSTRACT

This research aimed to evaluate whether the essential oil of Aloysia triphylla (EOAT) used in vivo as a sedative in the water for transporting fish could increase the oxidative stability of silver catfish (Rhamdia quelen) fillets during frozen storage. The chemical composition of EOAT and of fillets from fish exposed to $EOAT(0, 30 \text{ or } 40\mu L L^{-1})$ were assessed. The pH and lipid oxidation parameters (conjugated dienes, CD; thiobarbituric acid-reactivesubstances, TBARS) were evaluated in the fillets throughout the storage period (-18±2°C/17 months). The main compounds found in EOAT were α - and β -citral. Treatment with EOAT did not modify the proximate composition of the fillets, but 40µL L-1 EOAT reduced pH levels when compared to the control fillets (P < 0.05). Compared to the control fillets, the fillets from fish treated with 30 and $40\mu L$ L^{-1} EOAT had higher initial CD values (P<0.05), whereas fillets from fish treated with 40µL L⁻¹ EOAT had lower TBARS levels after 6, 9 and 17 months of storage (P<0.05). Results indicated that use of EOAT as a sedative in silver catfish transport water delays the degradation of primary oxidation products (CD) into secondary products (TBARS) in the frozen fillets. This delay in the lipid oxidation rate may increase the shelf life of frozen fillets.

Key words: lipid oxidation, transport, fish, TBARS, natural anesthetic.

RESUMO

O objetivo do trabalho foi avaliar se o uso do óleo essencial de Aloysia triphylla (OEAT) na água de transporte de peixes, in vivo como sedativo, poderia aumentar a estabilidade oxidativa de filés de jundiá (Rhamdia quelen) durante o armazenamento congelado. Avaliou-se a composição química do OEAT e dos filés dos peixes expostos ao OEAT (0, 30 ou 40 μ L L⁻¹), bem como o pH e indicadores de oxidação lipídica (dienos conjugados, DC; substâncias reativas ao ácido tiobarbitúrico, TBARS) dos filés ao longo do armazenamento (-18±2°C/17 meses). O α- e o β-citral foram os compostos majoritários do OEAT. O tratamento com OEAT não modificou a composição centesimal dos filés de jundiá, mas 40µL L⁻¹ de OEAT reduziu o pH dos filés, comparado ao controle (P<0,05). Foi observado maior teor inicial de DC nos filés dos tratamentos 30 e 40µL L⁻¹ de OEAT e menor valor de TBARS nos filés do tratamento 40µL L⁻¹ de OEAT após 6, 9 e 17 meses de congelamento, em comparação com os filés controle (P<0,05). Os resultados indicam que o uso do OEAT como sedativo na água de transporte de jundiás retarda a degradação de produtos primários da oxidação lipídica (DC) em produtos secundários (TBARS) nos filés congelados. Este retardo na velocidade de oxidação lipídica pode ampliar a vida útil dos filés congelados.

Palavras-chave: oxidação lipídica, transporte, pescado, TBARS, anestésico natural.

INTRODUCTION

Silver catfish (*Rhamdia quelen*) is a freshwater fish found from southern Mexico to central Argentina, whose importance in aquaculture has grown, especially in Brazil (BARCELLOS et al., 2006), and which is well accepted by consumers (BARCELLOS et al., 2013). In commercial aquaculture, fish endure stress during capture, handling and transportation (PARODI et al., 2012). Besides putting the fish at risk of physical damage, meat quality may be also affected by stress. The increased muscle activity that is associated to stress causes an early onset of *rigor mortis* and accelerates lipid and protein oxidation during storage (ASHELY,

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2007), which are critical in determining the quality and shelf life of fish (BOSWORTH et al., 2007).

One strategy to reduce pre-slaughter stress is the use of anesthetics (PARODI et al., 2012; ZEPPENFELD et al., 2013). However, the only anesthetic approved by the Food and Drug Administration for use in aquaculture, Tricaine Methanesulfonate (MS-222), cannot be applied within 21 days of slaughter if the fish is destined for human consumption in the U.S. or within 5 days in Canada (MEINERTZ et al., 2014). Accordingly, there has been increased interest in natural anesthetics and an eugenol-based product (AQUI-S® 20E) was approved for sedating fish in New Zealand, Australia, Chile and Korea (ASHLEY, 2007). In Brazil, there are not yet laws to regulate the use of anesthetics for fish and the most commonly used anesthetic is benzocaine (FAÇANHA & GOMES, 2005). The essential oil of Aloysia triphylla was recently patented as a sedative and anesthetic for aquatic animals (PI0904839-1A2).

A. triphylla(L'Herit)Britton(Verbenaceae) is an herb species that grows naturally in South America and has been used in popular medicine as an infusion to treat insomnia and anxiety and as an analgesic and sedative (CARNAT et al., 1999). A. triphylla leaves are used to give a lemon flavor to fish, chicken, vegetable and salad dishes, among others (FUNES et al., 2009). Anesthetic effects of its essential oil were attributed to its main components, α-citral, β-citral and limonene (CARNAT et al., 1999; PARODI et al., 2012). In addition, in the in vitro DPPH (1,1-diphenyl-2-picrylhydrazyl) radical scavenging assay, the essential oil of A. triphylla presented moderate antioxidant activity in comparison to the synthetic antioxidant butylhydroxyanisole (BHA), which is utilized commercially (ALI et al., 2011).

Among animal source foods, fish is more susceptible to deterioration because its pH is close to neutral and it contains a high content of nutrients that are easily utilized by micro-organisms (HOSSEINI et al., 2010). In addition, its high content of unsaturated fatty acids also leads to a reduced shelf life due to hydrolysis and especially lipid oxidation reactions, which results in undesirable alterations of color, flavor, odor and nutrient value (SAMPELS, 2013).

Lipid oxidation is a free radical chain reaction that comprises three main stages namely initiation (auto-oxidation), propagation and termination (SAMPELS, 2013). The conjugated dienes (CD) and hydroperoxides, which are formed during the initial stages, are considered primary products, while secondary products of lipid oxidation are formed during the termination stage. These secondary products can be evaluated in fish by measuring thiobarbituric acid-reactive-substances (TBARS), which show the malondialdehyde (MDA) concentration (SAMPELS, 2013; VEECK et al., 2013; 2015). Natural antioxidants, such as the essential oils of plants may delay lipid oxidation due to the presence of terpenoid compounds, especially phenolic monoterpenes (VANDAR-ÜNLÜ et al., 2003). The π bonds are responsible for the chain-breaking antioxidant activity of terpenes (WOJTUNIK et al., 2005).

The aim of this research was to evaluate the use of essential oil of *A. triphylla* (EOAT) *in vivo*, as a sedative in the water for transporting fish to increase the oxidative stability of silver catfish (*R. quelen*) fillets during frozen storage.

MATERIAL AND METHODS

A. triphylla (L'Her.) Britton Essential Oil (EOAT)

The plant was collected in Frederico Westphalen (27°22"S; 53°25"W), Rio Grande do Sul, Brazil, identified by the botanist Dr. Gilberto Dolejal Zanetti (Dept. of Industrial Pharmacy, UFSM) and deposited in the herbarium at the Dept. of Biology, UFSM (SMDB No. 11169). *A. triphylla* was obtained from fresh leaves by hydrodistillation in a Clevenger during 3h and stored in amber glass vials (-20°C).

The composition of EOAT was evaluated in a Varian gas chromatographer (GC; Model CP - 3800) equipped with a Flame Ionizer Detector (FID) and a Saturno Mass Spectrometer (MS) with a 30m x 0.25mm x 0.25µm fused silica capillary column, VF-5 EM (Varian). The sample (1µL) was injected in split mode (1:20) for GC-MS and in splitless mode for GC-FID. The injector temperature was 250°C, the carrier gas was He (1mL.min⁻¹), and the column oven temperature was maintained at 50°C for 4min and then heated to 280°C at 4°C min⁻¹. The detector temperature was 310°C. The GC-MS analysis was performed using electronionization (70eV). The AT components were identified by comparing the retention indices (RI) of the sample with those determined using a calibration curve of a homologous series of n-alkanes (C_8 - C_{32}) injected under the same chromatographic conditions as the samples and using fragmentation models of the mass spectra (ADAMS, 2007; NIST, 1998). Quantitative data were obtained from the peak areas in the GC-FID analysis.

Treatment of fish and freezing of fillets

Forty-five silver catfish (*R. quelen*; 247.2 \pm 45.1g; 32.8 \pm 1.4cm) obtained from a fish farm (fed with commercial feed and raised in ponds) were

placed in plastic bags at a load density of 260g L⁻¹ water (5 fish per bag), and transported on a paved road for 6h. Three independent replicates were performed for each of three treatments (3 bags by treatment, resulting in 15 fish by treatment) applied to the water used for transportation: control (water alone), 30 or 40µL L⁻¹ of EOAT (equivalent to 0, 27 or 36mg L⁻¹, respectively, as the density of the essential oil is 0.90). The EOAT was diluted in ethanol (1:10, v/v) and added to the transport water. Transport time, loading density and EOAT concentrations were chosen based on previous studies (GOLOMBIESKI et al., 2003; PARODI et al., 2012; PARODI et al., 2013). The amount of dissolved carbon dioxide, pH, temperature, and total and unionized ammonia were verified in the water before and after transport and presented appropriate levels for the species (data published in DANIEL et al., 2014).

After transport, the fish were slaughtered by hypothermia, washed in potable water and cut into fillets by hand. One fillet from each repetition of each treatment was utilized immediately after slaughter to evaluate the chemical composition and lipid oxidation and the remaining fillets were stored at -18±2°C, in individual polystyrene trays covered by PVC film. Lipid stability of the fillets was evaluated by determining CD and TBARS levels throughout the 17-month period of frozen storage.

Analysis of the fillets

The proximate composition of the fillets was assessed immediately after slaughter by the AOAC method (1996), except for fat, which was evaluated according to BLIGH & DYER (1959). Determination of pH was performed at a ratio of 1g per 10mL distilled water (PASTORIZA & SAMPEDRO, 1994).

The CDs were determined in fat extracted by the method of BLIGH & DYER (1959), where lipids were extracted and dissolved in 3mL of cyclohexane and measured at 233nm (RECKNAGEL & GLENDE, 1984). To determine TBARS, the fillet samples were homogenized with 1.5% KCl (1:5, m/v) using an Ultra Turrax (IKA T18 Basic) and then centrifuged at 3.000 x g for 10 min. The supernatant was incubated at 100°C for 15min in a medium containing trichloroacetic acid and thiobarbituric acid. The reaction product was extracted with n-butanol and measured at 535nm (BUEGE & AUST, 1978).

Statistical analysis

The Statistica 9.1 software was used to perform analyses. Data from the composition of the fillets were analyzed using one-way analysis of variance (ANOVA) and the remaining results were analyzed using two-way factorial ANOVA (3 treatments x 6 time points). Differences between means were *post hoc* evaluated using Tukey's test. Differences were considered significant at P<0.05.

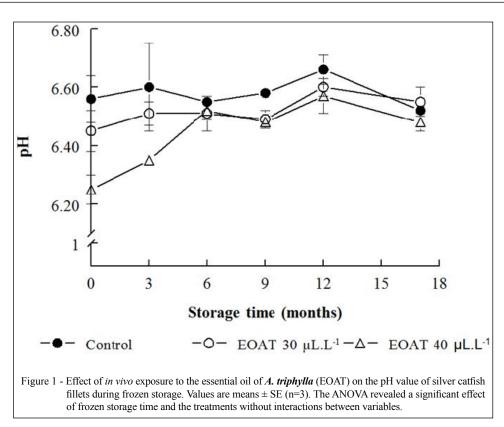
RESULTS AND DISCUSSION

Forty-one compounds were identified in the EOAT, most of which were mono- and sesquiterpenoids. Compounds found in the highest concentrations were: (in g $100g^{-1}$ oil) 29.4 α -citral; 20.8 β-citral; 11.9 limonene; 5.6 caryophyllene, 3.3 cubenol, 3.0 geranyl acetate and 2.3 caryophyllene oxide; the remaining compounds were found in concentrations under 2g 100g-1. Results are in corroboration with other studies that found citral to be the main component of EOAT (ALI et al., 2011; PARODI et al., 2012). Generally, α-citral is the main component, followed by β -citral and limonene (PARODI et al., 2012), all of which are considered to be safe food additives for human consumption by regulatory agencies in the U.S. (Code of Federal Regulation 21 CFR part 182.20; U.S. Food and Drug Administration) and Europe (Commission Implementing Regulation No. 872/2012; European Union).

Treatment with EOAT during transport did not modify the proximate composition of the fish fillets, which presented mean composition $(3.0\pm0.6\%$ fat, $13.4\pm1.0\%$ protein, $1.1\pm0.0\%$ ash and $80.3\pm1.4\%$ moisture), for all of the treatments (n=9), similar to those reported by VEECK et al. (2013) for the species.

Throughout the storage period, the frozen fillets presented pH levels under 6.8 (Figure 1), which is the maximum allowed limit for fish destined for human consumption, according to Brazilian legislation (6.5 and 6.8 for internal and external fish muscles, respectively, BRASIL, 1997). There was an effect of storage time and treatment on pH levels (P<0.05), with no interaction between these variables. Levels of pH increased after 12 months frozen storage when compared to initial levels (P<0.05) for all treatments. Fillets from fish treated with 40µL L⁻¹ EOAT presented pH levels lower than those of the controls (P<0.05) throughout the storage period (Figure 1). The lower pH levels found for fillets from fish treated with EOAT may be the result of lower stress levels, which would lead to reduced glycogen use during transport. Following slaughter, muscle glycogen is degraded and releases glucose for energy production by the anaerobic glycolytic pathway, resulting in an accumulation of lactic acid and a reduction in pH (BAGNI et al., 2007).

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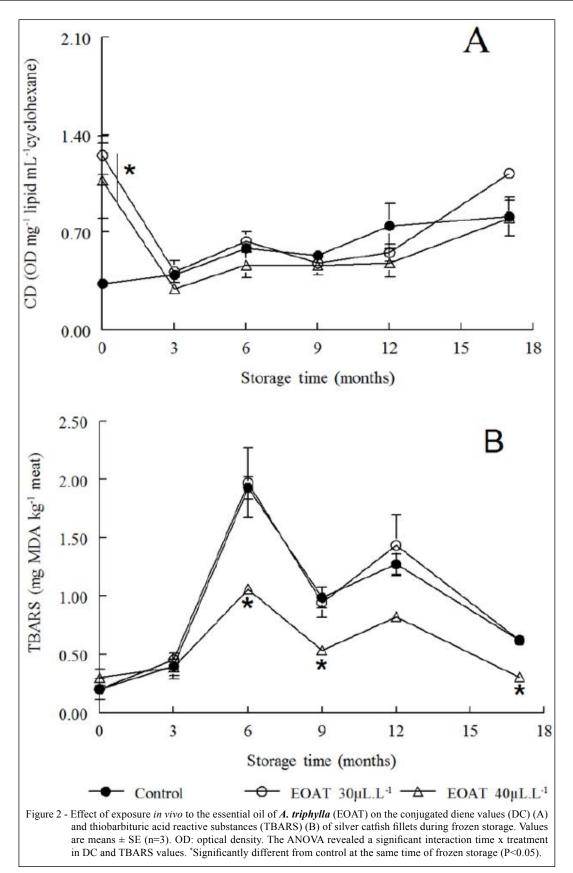


Therefore, the greater reserve of muscle glycogen at the time of slaughter resulted in greater accumulation of lactic acid and lower pH in the fish treated with EOAT. This hypothesis is in agreement with data of ZEPPENFELD et al. (2014), who reported higher lactate content in the muscle of silver catfish sedated with EOAT during transport in comparison to controls.

Lipid stability of the frozen fish fillets was assessed by measuring CD and TBARS levels (Figure 2). There was a time x treatment interaction (P<0.05) for CD levels (primary products of lipid oxidation) and for TBARS (secondary products of lipid oxidation). Fillets from fish treated with EOAT (30 and $40\mu L L^{-1}$) had higher CD levels than those found in the controls at time point 0 (Figure 2A). However, fillets from fish treated with $40\mu L L^{-1}$ EOAT had lower TBARS levels than those found in the controls after 6, 9 and 17 months of frozen storage (P<0.05; Figure 2B), which did not occur for the treatment with 30µL L-1 of EOAT. These results indicated that exposure to 40µL L⁻¹ EOAT during transport led to an antioxidant effect on the frozen fillets, by delaying the degradation of primary lipid oxidation products (CD) into secondary products (TBARS). Similar results were found in fillets sprinkled with Lippia alba extracts and in fillets from fish exposed in vivo to the essential oil of L. alba, which showed delayed degradation of hydroperoxides into secondary oxidation products (TBARS), indicating an antioxidant activity (VEECK et al., 2013; 2015). In fact, some antioxidants (e.g., tocopherols) can increase primary lipid oxidation products by donating hydrogen to a peroxyl radical to form a lipid hydroperoxide while simultaneously decreasing the formation of low molecular weight volatile secondary oxidation products (DECKER et al., 2005). The reduction in the lipid oxidation rate caused by EOAT, especially with regard to the lower TBARS levels, may be important to increase the shelf life of frozen fish fillets, since the aldehydes resulting from lipid oxidation (assessed by TBARS levels) are responsible for a rancid odor and consequent rejection of oxidized food (DAMODARAN et al., 2008).

The proposed maximum acceptable limit for MDA levels were 1-2mg kg⁻¹ (GILL, 1990). The fish fillets from control group and those treated with $30\mu L L^{-1}$ EOAT reached this limit at the sixth month of frozen storage, while the fillets from fish treated with $40\mu L L^{-1}$ EOAT did not reach this limit even after 17 months of storage (Figure 2B).

The delay in lipid oxidation of fillets from fish sedated with EOAT during transport may be



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related to the antioxidant capacity and inhibition of lipid peroxidation shown by EOAT when used as an anesthetic in vivo in silver catfish (GRESSLER et al., 2012). Exposure in vivo to EOAT immediately before slaughter reduced stress in the catfish by suppressing cortisol release and increasing the activity of antioxidant enzymes, catalase and glutathione S-transferase (GRESSLER et al., 2012). In addition, the antioxidant effect of EOAT may also be related to the removal of free radicals by antioxidant compounds present in the oil, since, in vitro, the oil presented moderate capacity to remove the DPPH radical when compared to BHA (ALI et al., 2011). Despite presenting a promising antioxidant effect on the frozen fillets and even though EOAT is a natural compound extracted from a plant that is widely used in tea and for spice in the human diet, it is still necessary to evaluate its toxicological safety before it can be used in foods.

CONCLUSION

The use of *A. triphylla* essential oil $(40\mu L L^{-1})$ in fish transport water reduced muscle pH after slaughter and delayed lipid oxidation of frozen silver catfish fillets, especially the degradation of primary products of lipid oxidation (CD) into secondary products (TBARS). The use of EOAT as a sedative in fish transport water contributes to extend the shelf life of frozen fillets.

BIOETHICS AND BIOSSECURITY COMMITTEE APPROVAL

This study was approved by the Ethics Committee on Animal Use (092/2011).

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