



## Mathematical models for adjustments in the quantification of ammonia volatilization from urea fertilizer applied on tropical pastures

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**ABSTRACT:** In Brazil, urea is the most used nitrogen (N) fertilizer to improve forage production. However, their excessive use can cause environmental impacts through N losses, such as ammonia (NH<sub>3</sub>) volatilization. Therefore, the current study adjusted and estimated the NH<sub>3</sub> volatilization from urea applied on tropical pastures in three rainfall conditions using mathematical models. Data were collected from Marandu grass (*Brachiaria brizantha*) fertilized with 50 kg N ha<sup>-1</sup> during wet, intermediate, and dry conditions. Ammonia volatilization was measured in five semi-open chambers for 21 days. The linear, quadratic, exponential, Gompertz, Groot, and Richards models were tested for fitting and estimating the NH<sub>3</sub> volatilization. The Gompertz, Groot, and Richards models generated predictions similar to the observed data, with a high determination coefficient, indicating a better fit of these equations to data, with precision and accuracy. However, the Groot model was selected due to the lowest root mean square error of prediction (0.29 % total N lost as NH<sub>3</sub>). The greatest N loss as NH<sub>3</sub> volatilization occurred in the wet, followed by intermediate and dry conditions (20.2, 17.0, and 11.3 % total N lost as NH<sub>3</sub>, respectively). Therefore, nitrogen losses as NH<sub>3</sub> volatilization after application of 50 kg N ha<sup>-1</sup>, as urea source, are altered according to the weather conditions, reaching 20% of N added in the wet rainfall period. The Groot model is recommended for fitting and estimating the NH<sub>3</sub> volatilization from urea applied on Marandu grass pastures in the wet and dry rainfall conditions.

**Key words:** ammonia volatilization, marandu grass, mathematical model, synthetic fertilizer, tropical pasture, urea.

## Modelos matemáticos para ajustes na quantificação da volatilização de amônia do fertilizante ureia aplicado em pastagens tropicais

**RESUMO:** No Brasil, a ureia é o fertilizante nitrogenado mais utilizado para melhorar a produção de forragem. No entanto, seu uso excessivo pode causar impactos ambientais por meio de perdas de nitrogênio (N), como a volatilização da amônia (NH<sub>3</sub>). Portanto, o objetivo do presente estudo foi ajustar a volatilização de NH<sub>3</sub> da ureia aplicada em pastos tropicais em três condições de chuva utilizando modelos matemáticos. Dados foram coletados de pastos de capim-marandu (*Brachiaria brizantha*) adubado com 50 kg N ha<sup>-1</sup> em condições úmidas, intermediárias e secas. A volatilização da NH<sub>3</sub> foi medida em cinco câmaras semiabertas durante 21 dias. Os modelos, linear, quadrático, exponencial, Gompertz, Groot e Richards foram testados para ajuste e estimativa da volatilização do NH<sub>3</sub>. Os modelos de Gompertz, Groot e Richards geraram previsões semelhantes aos dados observados, com alto coeficiente de determinação, indicando um melhor ajuste dessas equações aos dados, com acurácia e precisão. No entanto, o modelo Groot foi selecionado devido ao menor erro quadrático médio das previsões (0,29% de N total perdido como NH<sub>3</sub>). A maior volatilização de NH<sub>3</sub> ocorreu em condições climáticas úmida, seguido por intermediária e seca (20,2; 17,0 e 11,3% de N total perdido como NH<sub>3</sub>, respectivamente). Portanto, as perdas de N como volatilização de NH<sub>3</sub> após a aplicação de 50 kg N ha<sup>-1</sup>, como fonte de ureia, são alteradas de acordo com as condições climáticas, atingindo a 20% do N adicionado nas condições úmidas. O modelo Groot é recomendado para ajuste e estimativa da volatilização de NH<sub>3</sub> da ureia aplicada em pastos de capim Marandu em condições úmidas e secas.

**Palavras-chave:** adubação sintética, capim-marandu, modelo matemático, pastagem tropical, ureia, volatilização de amônia.

## INTRODUCTION

In Brazil, around 163.1 million ha are used for grassland, supporting about 196.5 million cattle (ABIEC, 2022). Approximately 90 million ha of this area belongs to the genus *Brachiaria*, where

the cultivar Marandu occupies more than 50% (JANK et al., 2014). In grassland systems, nitrogen (N) fertilizer plays a vital role in increasing forage productivity (DELEVATTI et al., 2019). In Brazil, urea is the most used fertilizer because its cost per kilogram is lower than other N fertilizers (GURGEL

et al., 2020). SALES et al. (2019) reported that N fertilization doses between 50 to 75 kg N ha<sup>-1</sup> cycle<sup>-1</sup> in marandu grass pastures result in greater production and forage accumulation than doses of 25 and 100 kg N ha<sup>-1</sup> cycle<sup>-1</sup>.

Nitrogen fertilizer over the recommended dose can lead to ammonia (NH<sub>3</sub>) losses (ZAMAN et al., 2009). The NH<sub>3</sub> is an important atmospheric pollutant responsible for cause negative environmental impacts (BEUSEN et al., 2008). Furthermore, N applied from urea fertilizer may be lost more than 50% as NH<sub>3</sub> volatilized to the environment (MORAIS et al., 2013; ROCHETTE et al., 2009). It is estimated that N global annual losses from synthetic N fertilizers are around 17 million tons (XU et al., 2019).

The Intergovernmental Panel on Climate Change (IPCC) suggests a default NH<sub>3</sub> emission factor of 15% of applied N (uncertainty range, 3–43%) for urea fertilizer for national greenhouse gas inventory methodology (IPCC, 2019). However, these large variations in the rate of NH<sub>3</sub> volatilization from urea are explained by several factors, like changes in weather conditions, such as air temperature and rainfall amount (BURCHILL et al., 2017; ENGEL et al., 2011; SANZ-COBENA et al., 2011; SIMAN et al., 2020), strong wind and moisture (NUNES et al., 2023), soil pH, application rate and placement depth (ROCHETTE et al., 2013). CORRÊA et al. (2021) using urea fertilization at a rate of 270 kg N ha<sup>-1</sup> in Marandu grass, reported 44.5% of N applied lost as NH<sub>3</sub> in a single dose; however, when it was divided into three applications, there was a reduction to 24.1% of N applied lost as NH<sub>3</sub>. For this reason, the NH<sub>3</sub> volatilization assessment from different weather conditions plays an important role in providing country-specific emissions data for agriculture inventory calculation.

Mathematical models have been used in several research areas. In the Agricultural Science, they have been used, for example, to adjust the kinetics of *in vitro* cumulative gas production (GURGEL et al., 2021a; ZORNITTA et al., 2021), in animal growth curve (GURGEL et al., 2021b; SOUZA et al., 2022), and in bacterial growth curve (ZWIETERING et al., 1990). According to HAN et al. (2022) there were several empirical models to estimate NH<sub>3</sub> volatilization. However, these models need to be able to describe this process with sufficient precision and accuracy. Therefore, this study compared mathematical models to adjust and estimate the NH<sub>3</sub> volatilization from urea fertilizer applied on tropical pastures in wet, intermediate, and dry rainfall conditions.

## MATERIALS AND METHODS

This study used a dataset from LONGHINI et al. (2020), ninety data were collected from pastures of *Brachiaria brizantha* cv. Marandu fertilized with 50 kg N ha<sup>-1</sup> during wet (3 May 2017), intermediate (4 April 2018), and dry (8 June 2018) rainfall conditions, using urea as a fertilizer. The study was carried out at Sao Paulo State University, Jaboticabal, Sao Paulo, Brazil (21°14'20" S, 48°17'27" W; 583 m a.s.l.). The total annual rainfall in this area is 1,424 mm and mean annual air temperature is 22.3°C. Daily rainfall, air temperature (maximum, average, and minimum), and relative humidity were obtained from the Agrometeorological Station, Department of Exact Sciences, UNESP, Jaboticabal Campus, located at 700 m from the experimental site (Figure 1 and 2). Soil samples (0–20-cm depth) were as follows: pH (CaCl<sub>2</sub>) 5.3; organic matter 32.4 g kg<sup>-1</sup>; cation exchange capacity 74.8 mmol<sub>c</sub> dm<sup>-3</sup>; P (ion-exchange resin extraction method) 10.9 mg dm<sup>-3</sup>; Mehlich-1 extractable Ca 28.3 mmol<sub>c</sub> dm<sup>-3</sup>; Mehlich-1 extractable Mg 9.7 mmol<sub>c</sub> dm<sup>-3</sup>; Mehlich-1 extractable K 4.2 mmol<sub>c</sub> dm<sup>-3</sup>; base saturation 561 g kg<sup>-1</sup>, respectively (LONGHINI et al., 2020). Soil texture was 340 g kg<sup>-1</sup> sand, 140 g kg<sup>-1</sup> silt, and 520 g kg<sup>-1</sup> clay (LONGHINI et al., 2020).

The field NH<sub>3</sub> volatilization was measured in Marandu grass pasture (1,200 m<sup>2</sup>), seeded in 2014. The area had not been grazed or treated with N (urea fertilizer or animal excreta) during the previous 2 yr. The experiment was a randomized complete block design, with five replicates. Treatments were urea fertilizer (50 kg N ha<sup>-1</sup>) and control without fertilizer (0 kg N ha<sup>-1</sup>). The evaluations were replicated three times during different natural rainfall conditions, which were classified as wet, intermediate, and dry (Table 1).

A semi-open chamber (0.008 m<sup>2</sup>) was used to quantify NH<sub>3</sub> volatilization from urea (ARAUJO et al., 2009). The methodological description and validation of these chambers using the <sup>15</sup>N technique were reported in the studies of ARAUJO et al. (2009), in Brazil and JANTALIA et al. (2012) in the United States. Urea fertilizer was applied by hand at a rate of 50 kg N ha<sup>-1</sup> (2.67 g of urea plot<sup>-1</sup>). Each plot measured 0.4 m × 0.6 m (0.24 m<sup>2</sup>), totaling 10 plots per replication (rainfall conditions). Before of the experimental period, Marandu grass was cut at a height of 10 cm. Ammonia volatilization was monitored for 21 d after urea application on the Marandu grass. After urea application, the foam strips were changed for new strips 1, 3, 5, 9, 14, and 21 d. Ammonia volatilization for treatment in each sampling interval was calculated following Equation described by LONGHINI et al. (2020):

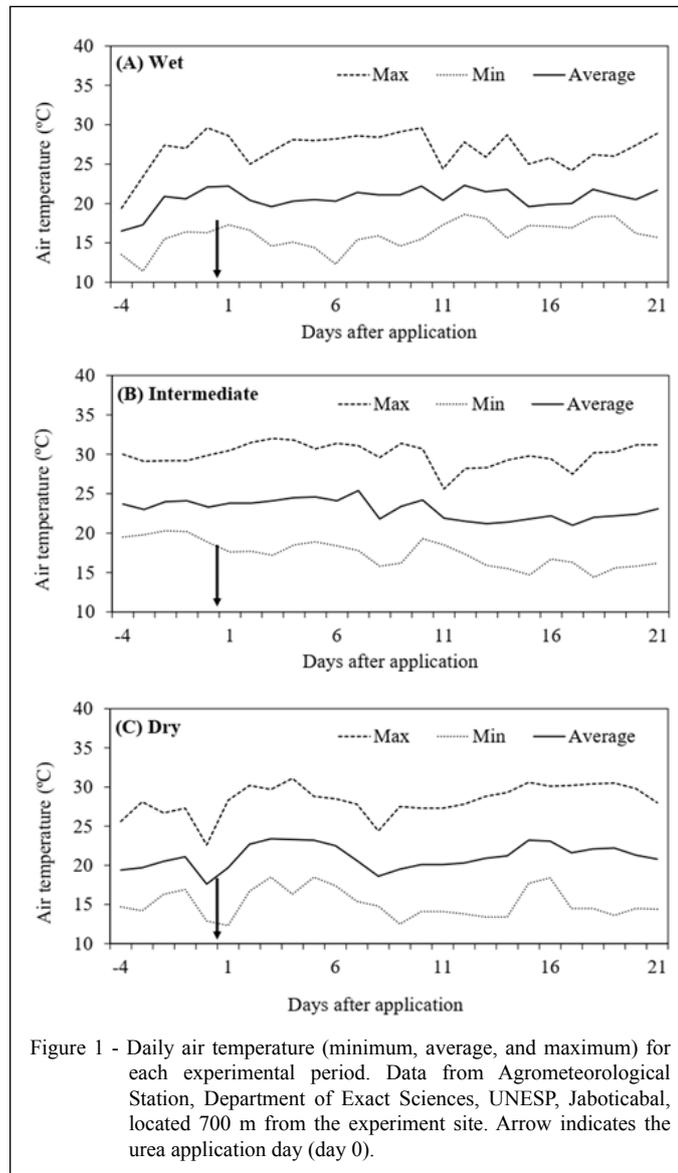


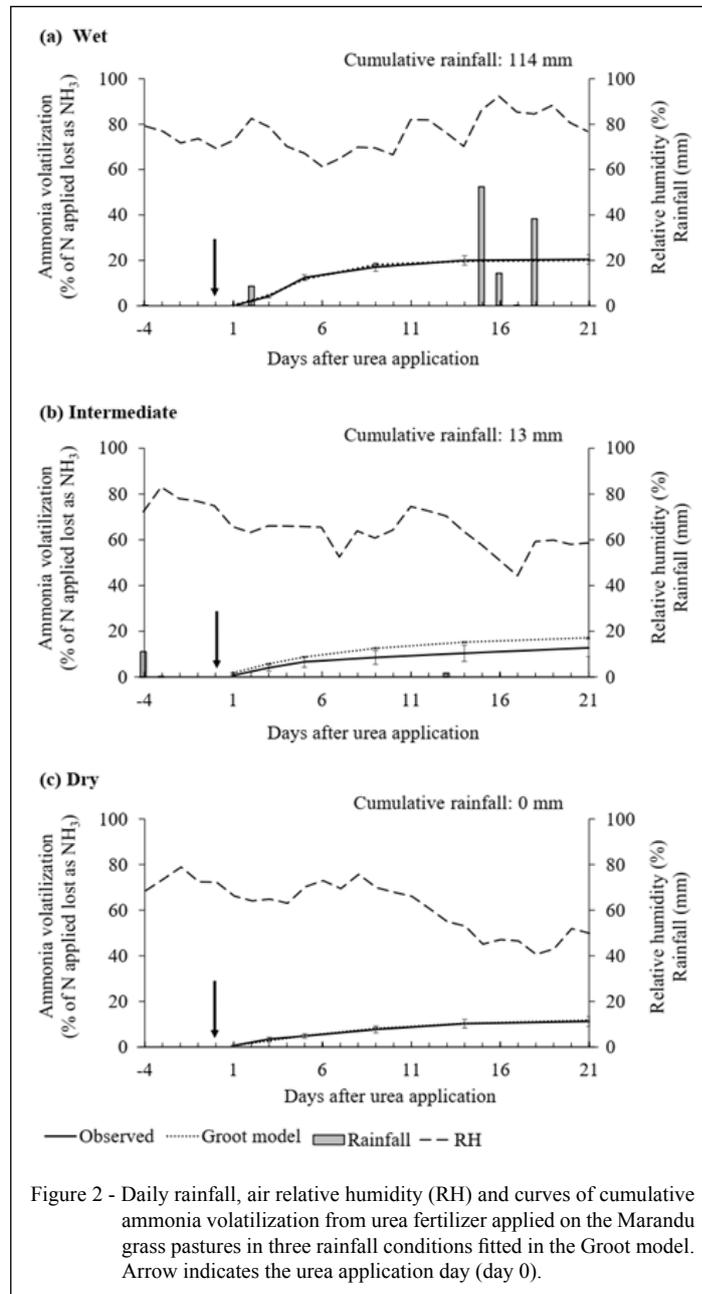
Figure 1 - Daily air temperature (minimum, average, and maximum) for each experimental period. Data from Agrometeorological Station, Department of Exact Sciences, UNESP, Jaboticabal, located 700 m from the experiment site. Arrow indicates the urea application day (day 0).

$$\text{Ammonia volatilization (\%)} = \frac{[NH_{3(\text{urea})} - NH_{3(\text{control})}]}{N_{(\text{applied})}}$$

Where:  $NH_{3(\text{urea})}$  is the amount of N applied lost as  $NH_3$  for the urea fertilizer treatment;  $NH_{3(\text{control})}$  is the amount of N from air + soil + Marandu grass without N addition lost as  $NH_3$  for the control treatment; and  $N_{(\text{applied})}$  is the amount of N applied in the area covered by the chamber ( $kg\ N\ ha^{-1}$ ). Cumulative  $NH_3$  volatilization (% total N lost as  $NH_3$ ) was calculated for the rainfall condition by summing the amounts of  $NH_3$  volatilized in each sampling interval (0–1, 1–3, 3–5, 5–9, 9–14, and 14–21 d). More details are described in LONGHINI et al. (2020).

The linear, quadratic, exponential, Gompertz, Groot, and Richards models were tested

to fitting the cumulative  $NH_3$  volatilization in 21 d (Table 2). The sigmoidal model Gompertz was described by SCHOFIELD et al. (1994). GROOT et al. (1996) and RICHARDS (1959) described the sigmoidal model's equations. The parameters of the equations are defined as:  $V(t)$  is the cumulative  $NH_3$  volatilization in time  $t$  (% total N lost as  $NH_3$ ); The parameter  $A$  is the volume of gases derived from the volatilization of  $NH_3$  when  $t \rightarrow \infty$ ; The parameter  $t$  is the time (days), and  $e$  is exponential; In the linear, quadratic, exponential, Gompertz, and Richards models, the parameter  $b$  represents interaction constant; in the Groot model, it is the time after urea application at which half of the asymptotic level was reached (days); In



the Gompertz and Richards models, parameter  $k$  represents the fractional rate of gas production ( $\% \text{ h}^{-1}$ ); in the Groot model, it is an integration constant that determines the sharpness of the curve. In the Richards model, the parameter  $M$  is a shape parameter. The variables obtained from the chosen model, time of curve inflection ( $T_i$ ), time at which volatilization rate is maximum ( $T_{\text{rmax}}$ ), and maximum fractional rate of volatilization in the  $T_{\text{rmax}}$  ( $R_{\text{max}}$ ) were calculated as described in GROOT et al. (1996).

A descriptive statistical analysis was performed using the PROC SUMMARY procedure in SAS (SAS University Edition, SAS Institute Inc. Cary, CA, USA). Pearson correlation coefficients between variables were estimated using the PROC CORR procedure in SAS. Model adjustments and variable selection were performed using PROC REG in SAS. The STEPWISE option and Mallow's  $C_p$  were used to select the variables included in the equations. Outliers were tested by evaluating the studentized residuals

Table 1 - Summary of weather conditions during each experimental period.

	-----Rainfall condition-----		
	Wet	Intermediate	Dry
Mean air temperature (°C)	21	23	21
Mean relative humidity (%)	76	62	60
Cumulative rainfall 4 d before the experimental period	0.2	11.3	0.0
Cumulative rainfall (mm)	113.4	1.7	0.0
Rain days	5	1	0
Initial volumetric soil moisture (%)	30	29	21

in relation to the values predicted by the equations. Residues that fell outside the range of -2.5 to 2.5 were removed. The goodness of fit of the developed equations was evaluated by the coefficients of determination ( $R^2$ ) and root mean square error (RMSE).

The data estimated by the equations that obtained the best adjustments were compared with the real values, using the regression model:  $Y = \beta_0 + \beta_1 \times X$ , where  $Y$  was the observed value;  $\beta_0$  and  $\beta_1$  represent the intercept and slope of the regression equation, respectively; and  $X$  was the value predicted by the equations. The criteria for assessing the adequacy of the equations were: the coefficient of determination ( $R^2$ ); F test, for the identity of the parameters ( $\beta_0 = 0$  and  $\beta_1 = 1$ ) of the regression of the predicted data by the observed ones. In addition, the Model Evaluation System version 3.2.2 program was used to estimate the coefficient of correlation and concordance (CCC); the square root of the mean square of the prediction error (RMSPE); and the decomposition of the mean square of the prediction error (MSPE) into mean error, systematic bias, and random error (TEDESCHI,

2006). The significance level was 5% probability in all statistical analyses.

The variables  $T_i$ ,  $T_{max}$ , and  $R_{max}$  were subjected to analysis of variance by the PROC GLM of the SAS statistical package. The means were compared using Tukey's test. Differences were considered significant at  $P \leq 0.05$ .

## RESULTS

The Gompertz, Groot, and Richards models showed average cumulative  $NH_3$  volatilization estimates and standard deviation close to the observed data as well as high determination coefficients (above 98%) of the regression of predicted on observed data (Table 3). The  $NH_3$  volatilization average observed was 8.63% of N applied lost as  $NH_3$ . Overall, all models presented predictions similar to the observed data ( $\beta_0 = 0$  and  $\beta_1 = 1$ ), except for the quadratic model, which was different from the data ( $P \leq 0.05$ ).

The cumulative  $NH_3$  volatilization curves from urea applied on the Marandu grass pasture in

Table 2 - Nonlinear models and equations considered in this study to describe the ammonia volatilization from urea fertilizer applied on the Marandu grass pastures in three rainfall conditions.

Models	Equation <sup>1</sup>	Parameters <sup>2</sup>
Linear	$V(t) = A + b.t$ (2)	2 ( $A, b$ )
Quadratic	$V(t) = A + b.t + k.t^2$ (3)	3 ( $A, b, k$ )
Exponential	$V(t) = A.e^{(b.t)}$ (4)	2 ( $A, b$ )
Gompertz	$V(t) = A.e(-b.e^{-k.t})$ (5)	3 ( $A, b, k$ )
Groot	$V(t) = A/(1 + (b^k/t^k))$ (6)	3 ( $A, b, k$ )
Richards	$V(t) = A.(1-b.e^{(-k.t)})^M$ (7)	4 ( $A, b, k, M$ )

<sup>1</sup> $V(t)$  is the cumulative  $NH_3$  volatilization in time  $t$  (% total N lost as  $NH_3$ ); <sup>2</sup> The parameter  $A$  is the volume of gases derived from the  $NH_3$  volatilization when  $t \rightarrow \infty$ ;  $t$  is the time (days);  $e$  is exponential; In the linear, quadratic, exponential, Gompertz, and Richards models, the parameter  $b$  represents interaction constant; in the Groot model,  $b$  it is the time after urea application at which half of the asymptotic level was reached (days); In the Gompertz and Richards models, parameter  $k$  represents the fractional rate of gas production (%  $h^{-1}$ ); in the Groot model, it is an integration constant that determines the sharpness of the curve. In the Richards model, the parameter  $M$  is a shape parameter.

Table 3 - Evaluation of the models fitting to estimate the ammonia volatilization from urea fertilizer applied on the Marandu grass pastures in three rainfall conditions.

Model	Mean	SD	R <sup>2</sup>	P-value	CCC	RMSEP	-----Decomposition of MSEP (%)-----		
							ME	SB	RE
Observed data	8.63	5.63	-	-	-	-	-	-	-
Linear	8.60	5.21	0.92	0.99	0.92	2.02	0.02	0.02	99.96
Quadratic	9.76	6.89	0.98	0.05	0.95	1.91	35.11	41.96	22.96
Exponential	8.86	4.43	0.85	0.96	0.82	2.75	0.73	1.21	98.01
Gompertz	8.72	5.47	0.99	0.87	0.99	0.58	2.76	3.84	93.36
Groot	8.66	5.54	0.99	0.86	0.99	0.29	1.48	5.83	92.70
Richards	8.53	5.60	0.99	0.82	0.99	0.31	8.94	0.23	90.82

SD: standard deviation; R<sup>2</sup>: coefficient of determination; P-value: probability value associated with the simultaneous F-test for the identity of parameters ( $\beta_0 = 0$  and  $\beta_1 = 1$ ) of the regression of observed vs. predicted data; CCC: concordance correlation coefficient; RMSEP: root mean square error of prediction; MSEP: mean square error of prediction. ME: mean error; SB: systematic bias; RE: random error.

three rainfall conditions (wet, intermediate, and dry), projected from the parameters estimated by each model, are shown in figure 3. The evaluation of models fitting the criteria presented means close to the observed data; although, standard deviations were lower when fitted into models (Table 3). The Gompertz, Groot, and Richards models presented the higher R<sup>2</sup> indicating a better fit of these equations to the NH<sub>3</sub> volatilization data. In addition, the CCC presented the same pattern as R<sup>2</sup>, with the Gompertz, Groot, and Richards models closer to the ideal coefficient than other models, reflecting precision and accuracy (Tables 3, 4 and Figure 3). Finally,

the Groot model ( $V(t) = 15.79/(1 + (5.29^{1.85}/t^{1.85}))$ ) was considering the best to fitting NH<sub>3</sub> volatilization from urea fertilizer applied on tropical pastures in the wet, intermediate, and dry rainfall conditions, because showed the smaller root mean square error of prediction (RMSEP) (Table 3).

There was an adjustment between predicted and observed curves of cumulative NH<sub>3</sub> volatilization for the Groot model (Figure 2), except for the intermediate rainfall conditions, in which the Groot model overestimated the NH<sub>3</sub> volatilization of the intermediate rainfall conditions (Figure 2B). Across rainfall conditions, the urea fertilizer applied in the wet

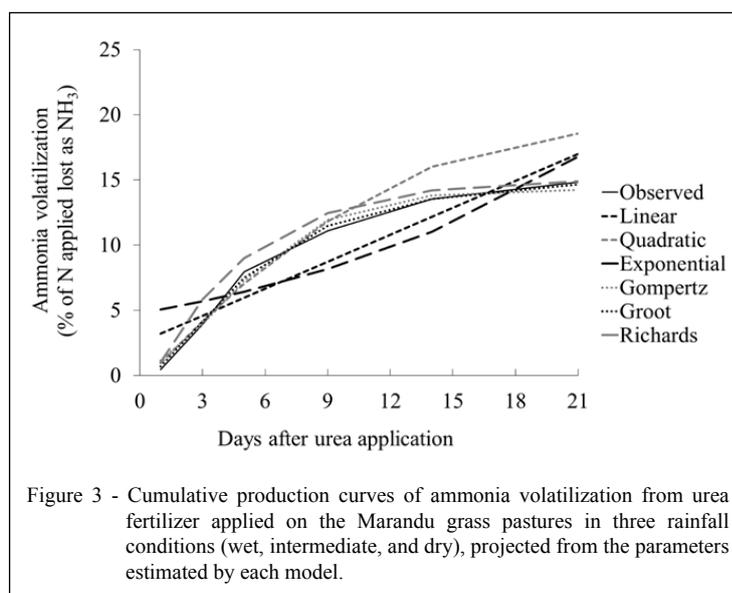


Table 4 - Models parameters to estimate the ammonia volatilization from urea fertilizer applied on the Marandu grass pastures in three rainfall conditions.

Model	A ± SD	b ± SD	k ± SD	M ± SD	N	R <sup>2</sup>	RMSE	P-value
Linear	2.50 ± 0.96	0.69 ± 0.08	-	-	90.0	0.42	5.62	<0.0001
Quadratic	- 0.77 ± 1.33	1.76 ± 0.33	- 0.04 ± 0.01	-	90.0	0.49	5.30	<0.0001
Exponential	4.76 ± 0.74	0.06 ± 0.00	-	-	90.0	0.73	5.93	<0.0001
Gompertz	14.29 ± 1.34	3.50 ± 1.40	0.33 ± 0.11	-	90.0	0.79	5.30	<0.0001
Groot	15.79 ± 2.53	5.29 ± 1.40	1.85 ± 0.71	-	90.0	0.79	5.28	<0.0001
Richards	15.12 ± 2.04	1.15 ± 0.20	0.18 ± 0.14	1.16 ± 1.03	90.0	0.78	5.31	<0.0001

A is the volume of gases derived from the NH<sub>3</sub> volatilization when t→∞; In the linear, quadratic, exponential, Gompertz, and Richards models, the parameter b represents interaction constant; in the Groot model, b it is the time after urea application at which half of the asymptotic level was reached (days); In the Gompertz and Richards models, parameter k represents the fractional rate of gas production (% h<sup>-1</sup>); in the Groot model, it is an integration constant that determines the sharpness of the curve. In the Richards model, the parameter M is a shape parameter; SD is the standard deviation; N is the number of observations; R<sup>2</sup> is the coefficient of determination; RMSE: root mean squared error.

presented greater NH<sub>3</sub> volatilization (20.2% of N lost as NH<sub>3</sub>), followed by intermediate (17.0% of N lost as NH<sub>3</sub>) and the smallest in the dry (11.3% of N lost as NH<sub>3</sub>) (Table 5; P ≤ 0.05). In wet rainfall condition, the NH<sub>3</sub> volatilization maximum fractional rate in tropical pastures fertilized with 50 kg of N ha<sup>-1</sup> using urea was greater, reaching up to 36% of N added in the pasture area, on the day where the loss was maximum. Overall, the NH<sub>3</sub> volatilization maximum peak occurred 4.58 d after the urea application on the Marandu grass.

**DISCUSSION**

The linear models did not adjust the parameters appropriately for the evaluation periods. Because there were increases in the NH<sub>3</sub> volatilization

in the first days after the urea application, which was followed by an inflection curve due to the reduction of the availability of N in the soil until it reached a plateau (Figure 3). For this reason, nonlinear models, which has a sigmoidal function, were the best for fitting NH<sub>3</sub> volatilization from urea fertilizer applied on tropical pastures, and they showed satisfactory precision and accuracy. Previous studies have recommended the Groot model to adjust the pattern of the phenomena biological as a time function (GURGEL et al., 2021a; ZORNITTA et al., 2021). This model does not assume a constant fractional rate, as occurs in the Richards and Gompertz model, simulating a real pattern of the NH<sub>3</sub> volatilization, which provided a more accurate adjustment to the phenomena that exist in the environmental conditions

Table 5 - Parameters and variables obtained by the Groot model to estimate the ammonia volatilization from urea fertilizer applied on the Marandu grass pastures in three rainfall conditions.

	-----Rainfall conditions <sup>1</sup> -----			SEM	P-value
	Wet	Intermediate	Dry		
A	20.4 a	21.2 a	13.7 b	0.240	0.0410
b	4.49	6.64	7.04	0.013	0.0569
k	3.02 a	1.25 b	1.66 b	0.067	0.0003
Cumulative NH <sub>3</sub> volatilization (% total N loss as NH <sub>3</sub> )	20.2 a	17.0 ab	11.3 b	0.243	0.0311
Ti (days)	3.57	1.78	2.72	0.175	0.0960
TR max (days)	5.66	3.28	4.79	0.218	0.1666
R max (%)	36 a	13 b	15 b	0.025	0.0001

<sup>1</sup>Means followed by different letters differ statistically by Tukey’s test at 5% significance. A is the volume of gases derived from the NH<sub>3</sub> volatilization when t→∞; b is the time after urea application at which half of the asymptotic level was reached (days); k is an integration constant that determines the sharpness of the curve. Variables obtained from the parameters: Ti (d): time of curve inflection; TRmax (d): time at which volatilization rate is maximum; Rmax: maximum fractional rate of volatilization in the TRmax (%).

(GURGEL et al., 2021a). The authors also explained that the Groot model uses three parameters in the equation, resulting in greater degrees of freedom.

Changes in weather conditions are the most responsible for the variations in  $\text{NH}_3$  volatilization (BURCHILL et al., 2017; SIMAN et al., 2020). Previous studies have shown that the peak of  $\text{NH}_3$  volatilization occurs 1-3 d after urea application on wet soils (NUNES et al., 2023; RECH et al., 2017; TURNER et al., 2012). Ideal weather conditions such as strong wind and high soil temperature and moisture are responsible for fast urea hydrolysis (NUNES et al., 2023). Conversely, in drier soil, TURNER et al. (2012) reported that the  $\text{NH}_3$  volatilization peak did not occur until the 6-7 d after urea was applied, while in another experiment, the peak occurred at 4 d affected by a rainfall that occurred at 2 d. On this occasion, there is a lack of moisture or water to dissolve the urea granules, delaying the occurrence of the  $\text{NH}_3$  volatilization peak. In our study, the maximum  $\text{NH}_3$  volatilization rate was attended at 4.58 d (ranging from 3.28 – 5.66 d) after the urea application, regardless of the rainfall conditions.

Although, the peak of  $\text{NH}_3$  volatilization from urea was the same between rainfall conditions, the maximum N fractional rate of  $\text{NH}_3$  volatilization in the wet conditions was almost three times that in the intermediate and dry rainfall conditions. This pattern resulted in different intensities in the total N lost as  $\text{NH}_3$  volatilization. The greatest cumulative  $\text{NH}_3$  volatilization occurred in the wet condition, favored for the rain that fell at 2 d after the urea application (Figure 2A). The amount of rain was low (8.6 mm); however, it was enough to elevate the soil moisture and relative air humidity (above 76%). Urea fertilizer is hygroscopic and absorbs the moisture, resulting in urea hydrolysis and high N losses as  $\text{NH}_3$  volatilization (KROL et al., 2020). Conversely, if the high intensities of rainfall at the end of the evaluation had occurred after urea application, the  $\text{NH}_3$  volatilization could reduce due to the incorporation of the urea (LIU et al., 2020). For this reason, to avoid  $\text{NH}_3$  volatilization during the rainy season, the urea fertilizer should be applied in dry soil followed by rain or irrigation to allow the N infiltration into the soil. Nonetheless, high rainfall intensity in a short time can increase the  $\text{NH}_3$  volatilization due to the soil flood and urea exposition to the environment; this high amount of water can saturate the soil porosity and difficult the urea infiltration (JIANG et al., 2023).

This study showed that the  $\text{NH}_3$  volatilization had a different pattern according to the soil moisture and amount of rainfall during the urea application on the Marandu grass. The best Groot

model adjustment for evaluating the urea fertilizer means that there was an understanding of how the pattern of  $\text{NH}_3$  volatilization occurs. However, despite the good fit, none of the models studied adjusted to data in the intermediate rainfall conditions, which overestimated the  $\text{NH}_3$  volatilization. For this reason, other models should be applied better to understand the  $\text{NH}_3$  volatilization pattern in these rainfall conditions.

## CONCLUSION

Nitrogen losses as  $\text{NH}_3$  volatilization can reach 20% using urea at the dose of 50 kg N  $\text{ha}^{-1}$  in wet rainfall conditions, which was greater than the same rate applied in intermediate and dry rainfall conditions. The Groot model showed satisfactory precision and accuracy to adjust and estimate the  $\text{NH}_3$  volatilization from urea fertilizer applied on tropical pastures in wet and dry rainfall conditions.

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## DECLARATION OF CONFLICT OF INTEREST

The authors declare no conflict of interest. The founding sponsors had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, and in the decision to publish the results.

## AUTHORS' CONTRIBUTIONS

All authors contributed equally for the conception and writing of the manuscript. All authors critically revised the manuscript and approved of the final version.

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