

Effects of Local Heat Input Conditions on the Thermophysical Properties of Super Duplex Stainless Steels (SDSS)

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Received: April 18, 2017; Revised: December 10, 2017; Accepted: December 10, 2017

The properties of the super duplex stainless steels (SDSS) are strongly affected by the thermal history imposed by welding procedures. The controlled dual phase microstructure (ferrite and austenite) guarantee excellent mechanical properties such as mechanical strength and corrosion resistance, in addition to small thermal expansion coefficient and high thermal conductivity. In this paper, we newly proposed a model able to predict the thermal history of the welding pieces coupled with local mechanical properties developed during welding procedure that combine the effects of temperature and phase changes during welding. We applied inverse method to fit the thermophysical parameters based on measured data. The model was verified by comparing measured and predicted temperature profiles using thermocouples located within the heat affected zone. Thus, an inverse method was implemented to obtain the parameters fitting for the grain growth evolution compatible with the final microstructure and grain size measured using SEM images and stereological techniques. We demonstrated that very small amount of non-equilibrium deleterious phases and nanosized precipitates are expected during the welding procedures depending upon the local conditions of temperature, compositions and alloying dilution evolutions.

Keywords: Super duplex stainless steel (SDSS), heat input, thermal conductivity, heat capacity, mathematical modeling, non equilibrium, nanosized precipitate.

1. Introduction

Enhanced properties of the off shore materials are the main subject to be addressed in order to attain safety and extend service life of the equipments. Two aspects are of primary concern: corrosion and residual stress. Microstructure plays a decisive role on both^{1,2}. The accurate determination of the residual stresses during the steels welding and corrosion resistance during service are, therefore, subjects that need better attention due to their technological importance³⁻⁶. The super duplex stainless steels (SDSS) are enhanced materials that could be used in several applications that need higher performance³. Nevertheless, the welding process can introduce deleterious and non-homogeneous properties, which are critical and have limited some noble application^{3,7}.

The SDSS are characterized by an excellent balance of the ferrite and austenite phases with size and distributions that confers special properties for these materials. Yet, the role of alloying elements can significantly modify their properties on the local basis. The phases balance during the welding procedure is a key feature for the final properties

developed in the joints of the pieces and equipments and hence, their performance. The balanced ferrite and austenitic phases on the SDSS can be strongly affected by the local heat input and strict control of the thermal distributions is the technological and practical parameters that can be handled in order to keep suitable temperature profile during the welding procedure. Therefore, these combinations of balanced phases and elements dilution into the phases and some rejection allowing the formation of fine precipitates can confer excellent properties such as corrosion resistance and good toughness^{3,7,8}. In welding process or working at elevated temperatures, secondary phases may appear as precipitates^{7,8}. These compounds can be deleterious for steel properties at heat affected zone, depending of the size and volume fractions distributions. The principal precipitate that appears in this case is the sigma phase, although other precipitates such as chi phase or nitrides could be encountered. It is well known that the chi phase can be a precursor of the sigma phase which is also deleterious for the properties of SDSS^{7,8}. The literature on characterization of the chi and sigma phases are abundant and indicated that these deleterious phases can be formed

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in large amount if the temperature is kept around 800°C for times longer than 30 min^{7,8}. This is not a typical time interval on the critical zone for common welding process of small pieces. However, for large equipments and welding pieces it is common to use multiple pass welding procedure and some regions can relay into the gap of temperature and time suitable to form the deleterious phases. Thus, the design of an adequate welding procedure is important to make or repair the equipment to work under safety conditions⁹⁻¹². In this work we propose an approach that could help on this task.

The heat input is a common measure and indication for the expected phases formed and distribution of the heat affected zone (HAZ). Nevertheless, depending on the local conditions of elements dilution and temperature distributions non-equilibrium phases and nanosized precipitates can be formed.

Models capable of predicting the local conditions and possibility of deleterious phases being formed are welcome and can drive good practice for avoiding the non-uniformity of properties¹²⁻²². This paper is focused on this challenge. Thus, our work focused on the development of a computational tool for virtually simulates the welding procedure in order to give guidance for the onsite execution. The computational code is based on the simultaneous solution of the energy equation coupled with the phase changes kinetics in order to predict the final phase balance. Additionally, an inverse method is newly applied using measured values of materials properties and model adjustments are carried out for the super duplex stainless steel (SDSS).

Previous works have addressed the phase transformations and properties evaluation developed in the welding joints due to the heat input distributions⁹⁻²³. Furthermore, the process simulation using numerical methods have been a fundamental tool for new developments. This approach furnishes qualitative and quantitative analysis of the stresses and prediction of the resultant microstructure and mechanical properties of the welds¹⁷⁻²³. Thus, this study deals with the predictions of coupled phenomena of kinetic transformations under non-isothermal conditions for super duplex stainless steel. The model consists of coupled differential equations and takes into account grain growth effects, permitting the prediction of the progress of the phase balance and formation of deleterious phases simultaneously and correlates with local properties evolutions.

Thus, the present work is an attempt for numerically predicting the transformations coupled with local changes of properties occurring during the welding procedure of super duplex stainless steels. For this purpose, the kinetic model of Avrami-type with adjustable parameters is proposed^{17,8,13,16,24,25} and implemented using a computational code developed by the author. The code is based on the finite volume method (FVM)²⁶⁻²⁸ implemented in Visual Fortran programming language. It is important to mention that, to date, precise

properties such as thermal conductivity and heat capacity as functions of temperature and compositions are not available for the super duplex stainless steels. This confers to this study highly innovative features and technological importance.

Furthermore, the computer code used has been continuously updated by the authors and applied for different materials and welding conditions^{3,6,9-12}. In this paper we introduce new features to improve the model capability such as routines for kinetics of phase transformations for super duplex stainless steels^{7,8}.

2. Methods

The methodology adopted in this study considers the numerical implementation of the computational code using specific routines for local phase transformations and thermophysical evaluations. In order to obtain suitable thermophysical properties for solving the energy equation an inverse method is proposed with assumed boundary conditions. The selected measured transient temperatures are experimentally obtained in a controlled experiment and used for solving the minimization problem, where the thermal conductivities, k , and heat capacity, C_p , are unknowns. The kinetics rate equations for the chi, sigma phases are used from previously published work^{7,8}. The experimental procedure is carried out with the automatic measurements and monitoring system schematically shown in Figure 1. The system is composed of a controlled welding module and multiple acquisition data module for welding parameters of voltage, current and speed and a temperature acquisition module where simultaneous transient measurements are recorded. Figure 2 shows the positions and arrangements of the inner temperature monitoring for the dynamic thermophysical properties determination and model verification. A large number of transient temperature measurements at different positions, designed for the inverse method proposed in the study, are used for both dynamic estimation of the thermophysical properties and model verification.

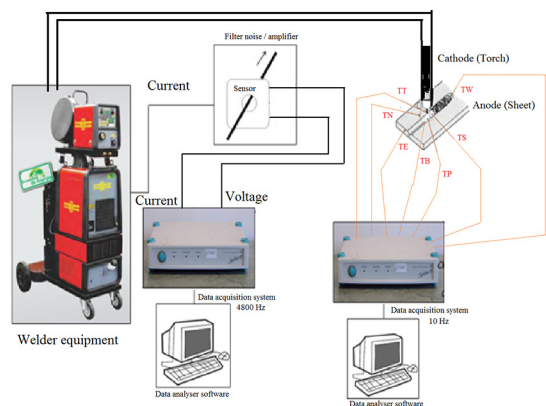


Figure 1. Welding experimental arrangement and data acquisition systems

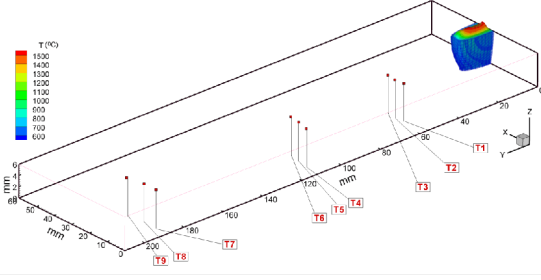


Figure 2. Temperature measurements arrangement using thermocouples and numerical predictions for first welding pass ($t=10s$ and heat input 0.7 kJ mm^{-1})- central thermocouples dynamic measurements for estimating thermophysical properties.

2.1 Numerical procedure

The temperature field is dynamically calculated during the welding process evolution and the materials thermophysical properties are continuously updated. The basic equations to model the process are presented by Equations (1)-(7). The model takes into account the heat transfer by radiation, convection and conduction. The mass transfer due to consumable deposition and dilution within metal base melting and solidification are considered and the properties changes due to local composition and temperature are taken into account⁹⁻¹². The temperature distribution is obtained by solving the energy conservation, shown in Equation(1)⁹⁻¹². In Equation 1, ρ is the density, C_p is the specific heat capacity; k is thermal conductivity, U^* is velocity field, which accounts for buoyancy driven flows in the liquid pool or moving mesh to attain the geometry changes due to the consumable deposition, T is the temperature field and S are all source or sink heat due to moving arc torch, phase transformations, melting and solidification.

$$\frac{\partial}{\partial t}(\rho c_p T) + \text{div}[\rho c_p (\vec{U}) T] = \text{div}[k(\text{grad}(T))] + S \quad (1)$$

$$C_p(T) = A_0 + A_1 T + A_2 T^2 \quad (2)$$

$$k(T) = B_0 + B_1 T \quad (3)$$

The model equations are completed with the formulation of the heat source due to the torch and metal transfer mode. The heat input is supplied by the moving torch. We assume the well-known double-ellipsoid model^{9-13,22}. The heat source is a combination of two ellipses in the front and rear quadrants. Equations (4) and (5) present the volumetric heat flux distributions for the front and rear quadrants, respectively, and their constant parameters. The coordinates are time dependent and gives spatial transient distributions²².

$$q_f(x(t), y(t), z(t)) = \frac{6\sqrt{3}f_f Q}{ab_c \pi \sqrt{\pi}} e^{-3\left(\frac{x}{a}\right)^2} e^{-3\left(\frac{y}{b_c}\right)^2} e^{-3\left(\frac{z}{c}\right)^2} \quad (4)$$

$$q_r(x(t), y(t), z(t)) = \frac{6\sqrt{3}f_r Q}{ab_r \pi \sqrt{\pi}} e^{-3\left(\frac{x}{a}\right)^2} e^{-3\left(\frac{y}{b_r}\right)^2} e^{-3\left(\frac{z}{c}\right)^2} \quad (5)$$

The total heat input $Q = \eta VI$ is determined by the operational parameters current (I), voltage (V) and thermal efficiency η , respectively. The spatial distribution parameters f_f and f_r are the fraction of the total heat received in the front and rear quadrant, respectively. The parameters are chosen attaining the relationship^{9-12,22}, $f_f + f_r = 2$. The heat input constant parameters a, b_f, c and b_r , are selected to account for the torch heat distribution⁹⁻¹². These constant parameters selected in this model are assumed fixed values, as shown in Table 1. The factors f_f and f_r were assumed as 0.6 and 1.4 respectively while η was assumed 0.75 throughout the all cases analyzed^{3,9-12}. The model parameters assumed in this study are presented for the cases considered in Table 1. The experimental design was selected using the sinergic mode where the voltage aimed fixed and the current slightly adjusted for keeping constant welding power. The welding speed was changed by controlling the motion of the torch and thus, controlling the heat input. The constant parameters of the heat source model were kept constant, as shown in Table 1.

The initial condition is assumed with the whole workpiece at room temperature and the composition as received material. The geometry of the bead is actualized for each time step after metal deposition and motion of the torch arc. The effects of convective and radiative fluxes are considered using Equations 6 and 7, respectively, by applying the boundary conditions.

$$q_c = h(T - T_0) \quad (6)$$

$$q_r = \epsilon_{(T)} \sigma (T^4 - T_0^4) \quad (7)$$

In Equations (6) and (7), is the room temperature, $\epsilon_{(T)}$ is the emissivity and $\sigma = (5.67 \times 10^{-8} \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-4})$ is the Stefan-Boltzmann constant. The effective heat transfer coefficient accounting for the convective heat loss, h , was assumed $8 \text{ W} \cdot \text{m}^{-2} \cdot \text{K}^{-1}$. In this study is assumed constant emissivity (0.45) during welding simulations^{3,9-23}. The specific mass appeared in Equation 1 is assumed constant as $7700 \text{ kg} \cdot \text{m}^{-3}$. The constants used in Equations 2 and 3 are obtained by iterative procedure using the optimization problem for corrections of the constant parameters using transient measurements of the temperatures positioned within the working piece, as shown in Figure 2. The iterative process for determining the constants A_0, A_1, A_2 and B_0 and B_1 is shown in Figure 3 for one time increment, involving backward and forward time steps for solving the complete discrete equations based on the measured values. This iteration process is carried out for the whole transient calculations and averaged values are obtained and fitted for the equations 2 and 3 as temperature dependent functions. Therefore, simultaneous measurements of the temperatures for the same time step are required. The basic discrete equation used in this study is presented in Equations (8)-(12).

$$\frac{k(T^*)}{\rho(T^*)C_p(T^*)} = \frac{1}{\sqrt{\left(\frac{T_0^4 - 2T_1^4 + T_2^4}{\Delta x^2}\right) + \left(\frac{T_0^4 - 2T_1^4 + T_2^4}{\Delta y^2}\right) + \left(\frac{T_0^4 - 2T_1^4 + T_2^4}{\Delta z^2}\right)}} \quad (8)$$

$$\frac{k(T_i^{k+1})}{\rho(T_i^{k+1})C_p(T_i^{k+1})} = \frac{[T_i^{k+1} - T_i^k] \cdot s^k}{\left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta x^2} \right) + \left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta y^2} \right) + \left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta z^2} \right)} \quad (9)$$

$$\frac{k(T_i^k)}{\rho(T_i^k)C_p(T_i^k)} = \frac{[T_i^k - T_i^{k-1}] \cdot s^k}{\left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta x^2} \right) + \left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta y^2} \right) + \left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta z^2} \right)} \quad (10)$$

$$\frac{k(T_i^{k+1})}{\rho(T_i^{k+1})C_p(T_i^{k+1})} = \frac{[T_i^{k+1} - T_i^k] \cdot s^k}{\left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta x^2} \right) + \left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta y^2} \right) + \left(\frac{T_i^{k+1} - 2T_i^k + T_i^{k-1}}{\Delta z^2} \right)} \quad (11)$$

$$\frac{k(T_i^k)}{\rho(T_i^k)C_p(T_i^k)} = \frac{[T_i^k - T_i^{k-1}] \cdot s^k}{\left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta x^2} \right) + \left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta y^2} \right) + \left(\frac{T_i^k - 2T_i^{k-1} + T_i^{k-2}}{\Delta z^2} \right)} \quad (12)$$

The values used in the right side of the Equations (8)-(12) are measured and the left side after substitutions using equations (2) and (3). The superscript k indicates the time step measured of the temperature values. The resulting system of algebraic equations are solved for the unknown constants and using the procedure schematically shown in Figure 3 for all transient measured values for the welding conditions shown in Table 1. The constants are then used on the direct method solution of the transient temperature distributions of the whole sheet and can be universally used for the SDSS.

2.2 Experimental procedure

We prepared samples of super duplex stainless steel (SDSS) sheets with dimensions of 60 x 210 x 8.0 mm to use in the experimental runs. The chemical composition of the main elements was determined by the technique inductive coupled

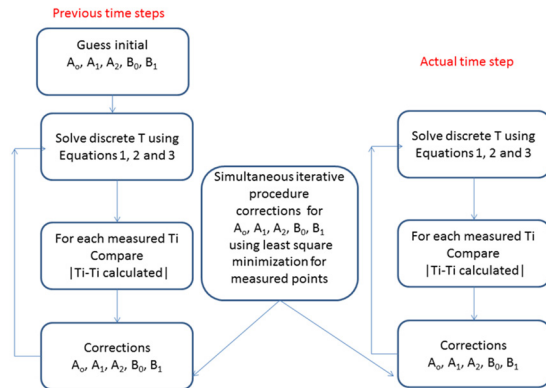


Figure 3. Flowchart of the iterative procedure to estimate the constant parameters for equations 2 and 3 based on discrete measurements of transient temperatures

plasma optical emission spectrometry (ICP-OES) for the as received material, as shown in Table 2. The experimental procedure was carried out with thermocouples inserted within the specimen and the temperature recorded and used to apply the inverse method. The recorded data were used to obtain the basic effective properties used to calculate the temperature field, thermal conductivity and heat capacity, k and C_p , respectively, applying Equations (2) and (3). Three runs of experimental conditions were conducted for different heat inputs, 0.7, 1.2 and 1.7 $\text{kJ}\cdot\text{mm}^{-1}$, respectively. The transient average temperature values were used and for each heat input the same position of the thermocouples undergoes different thermal history, which was used to fit the constants of equations 2 and 3. Therefore a large number of temperature values are measured and can be used for solving the minimization problem obtained by applying the inverse method with the unknown constants (A_0, A_1, A_2, B_0, B_1) of equations 2 and 3, respectively. The measured transient temperatures obtained from the thermocouples 7 central positions (TS, TN, TW, TE, TB, TT and TP) are used to generate 5 discretized equations for 5 sequential time steps, which are solved for the unknowns A_0, A_1, A_2, B_0, B_1 , as shown in Equations (8)-(12). This procedure is repeated for a large number of time steps and the average values are computed. These computed values are now used for the direct numerical method calculations of the temperature field. The temperature history are compared for the whole transient calculation on the positions (T1,T2,T3,T4,T5 and T6) and small adjustment are iteratively allowed for minimizing the average absolute difference of the measured and calculated temperatures (see Figure 3).

The austenitic grain size plays important role on the mechanical properties and during the phase decomposition process. We applied a temperature-dependent rate equation, as shown in Equation (13), with the parameters obtained by measuring the final grain size of the HAZ previously determined³.

$$\frac{dD}{dt} = \frac{K_A(T)}{n_A D^{n_A-1}} \quad (13)$$

Table 1. Selected double ellipsoid model parameters and welding data

Voltage (V)	Current (A)	Speed ($\text{mm}\cdot\text{s}^{-1}$)	Thickness (mm)	Heat input ($\text{kJ}\cdot\text{mm}^{-1}$)	Double-ellipsoid model constants (mm)
30	179	7.41	6.35	0.7	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
30	178	4.33	6.35	1.2	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
30	180	3.05	6.35	1.7	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
25.7	170	6.76	8.00	0.5	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
25.7	166	2.24	8.00	1.5	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
25.7	169	1.59	8.00	2.2	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
30	178	4.33	10.50	1.2	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
30	179	4.33	6.35	1.2	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$
30	179	4.33	2.00	1.2	$a = 5.35; b_f = 2.3; c = 7.8; b_r = 14$

The deleterious phase formation kinetic was modeled as Avrami-type equation with the parameters previously determined by Fonseca et al⁷. We applied the constants obtained using the published data for the mechanism 1, presented by Fonseca et al⁷, which is compatible with the welding typical times carried out on this study.

$$V_v = 1 - \text{EXP}\left(-k_0 t^n \text{EXP}\left(\frac{-Q}{RT}\right)\right) \quad (14)$$

In Equation (14) V_v is volume fractions of the phases, R is apparent activation energy, R is universal constant, t is residence time and T is temperature. The values obtained from the data fitting for the mechanism 1, $K_0 = 1.54 \times 10^{20}$, $Q = 421 \text{ KJ} \cdot \text{mol}^{-1}$, and $n = 2.68$ were used for the computations of the dynamic calculation of the transformed phases. In the present model, the initial phase composition of the material, chemical composition and average grain size are input parameters for the model. Thus, when local transformations temperature is achieved the transformations are likely to occur following the local kinetic conditions.

3. Results and Discussion

3.1. Estimation of the thermophysical properties for SDSS using inverse method

The welding predictions are strongly dependent on the themophysical properties, as shown in equation (1). This section discuss the results obtained for the constants of the Equations (2) and (3) and the comparisons of the model predictions for the recorded temperatures of the selected points where the thermocouples are positioned. Three different heat input were selected and the proposed method was applied. Figure 4, 5 and 6 show the comparison of the calculated and measured

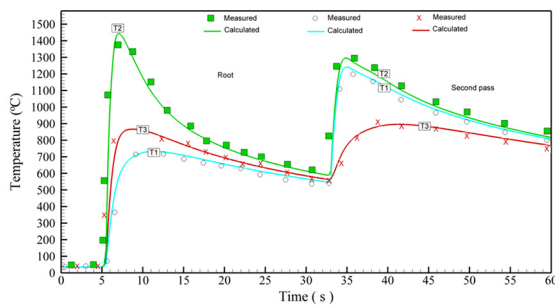


Figure 4. Transient temperature measurements and comparison with numerical predictions after thermophysical properties determination using inverse method (heat input = 0.7 kJ mm^{-1}).

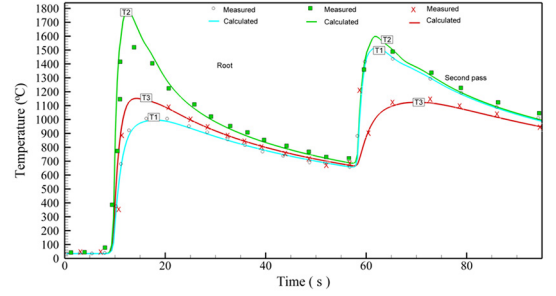


Figure 5. Transient temperature measurements and comparison with numerical predictions using inverse method (heat input = 1.2 kJ mm^{-1}).

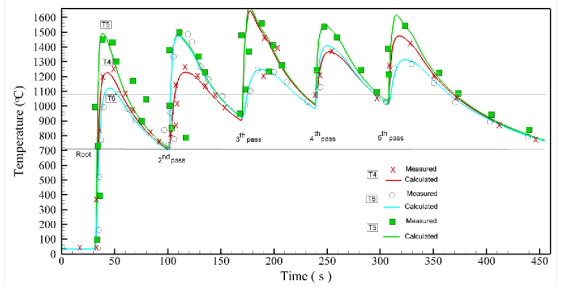


Figure 6. Transient temperature measurements and comparison with numerical predictions using inverse method (heat input = 1.7 kJ mm^{-1}).

transient temperatures for the heat input of 0.7 , 1.2 and $1.7 \text{ kJ} \cdot \text{mm}^{-1}$, respectively, used to estimate the constants presented in Table 3. As can be shown in Figures 4, 5 and 6, the measured and calculated transient results are in close agreement for all compared results, which validates the proposed inverse method procedure used in this study. Therefore, the constants presented in Table 3 adequately represent the behavior of the SDSS during the whole thermal cycle.

The evolution of the heat capacity and conductivity are presented in Figures 7 and 8, respectively, as functions of time and temperature for a measured position (TP). The calculated values of the thermal conductivity, k and capacity, C_p , were obtained after substitution of the constants presented in Table 3 in the Equations (2) and (3), respectively. As observed, the measured temperature fitted well with the local predicted with the new constants obtained in this study. The spatial variations of the thermal conductivity and heat capacity are presented in Figure 9 and 10, respectively, for the first pass showing the strong dependence with the temperature variations. Similar results are obtained for the analyzed cases, reflecting the effect of temperature.

Table 2. Super duplex stainless steel (SDSS) chemical composition used.

C (%)	Mn (%)	Si (%)	P (%)	S (%)	Cr (%)	Ni (%)	Mo (%)	Cu (%)	N (%)
0.020	0.850	0.328	0.027	0.0009	24.890	6.820	3.720	0.156	0.278

$PRE_N = \%Cr + 3.3 \times (\%Mo) + 16 \times (\%N) = 41.62$ - Note: chemical analysis carried out using ICP-OES technique.

Table 3. Constants for the heat capacity (C_p) and thermal conductivity (k) of super duplex stainless steel (SDSS) determined using the proposed method (Equations 2 and 3).

A_0	A_1	A_2	B_0	B_1
558.14	0.3253	0.000525	10.50	0.0108

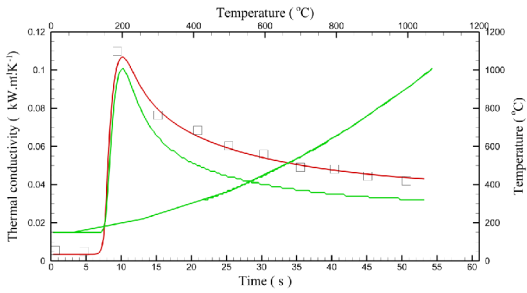


Figure 7. Transient thermal conductivity predictions for position TP using equation 2 and table 3 (heat input = 1.2 kJ mm^{-1}).

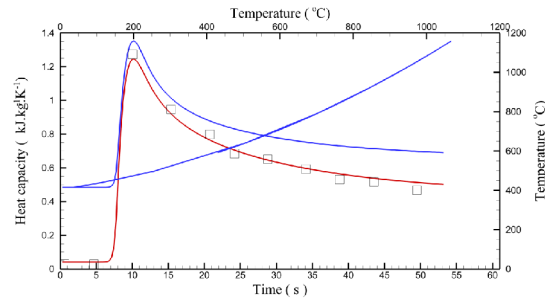


Figure 8. Transient heat capacity calculations at position TP using equation 3 and table 3 (heat input = 1.2 kJ mm^{-1}).

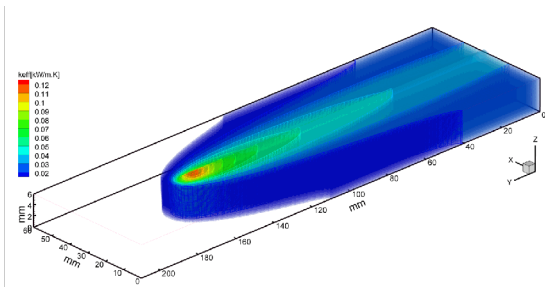


Figure 9. Transient thermal conductivity (k) distributions during the first welding pass (time = 32 s and heat input = 1.2 kJ mm^{-1}).

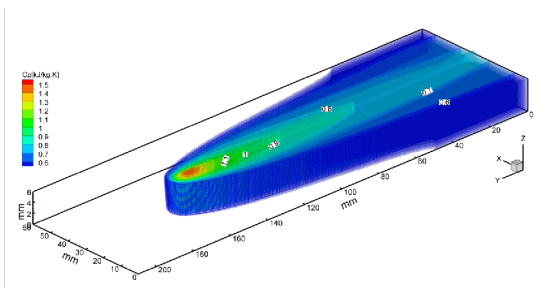


Figure 10. Transient heat capacity (C_p) distributions during the finishing welding pass (time = 32 s and heat input = 1.2 kJ mm^{-1}).

3.2. Effects of welding parameters on deleterious phases formation

The sigma phase formation in SDSS usually requires larger times at during the cooling step at the range of $750\text{--}850 \text{ }^\circ\text{C}$. We applied model to the conditions and times where are expected significant amount of this phase formation. Figure 11 shows the comparison of sigma phase volume formation and those predicted by the model under such conditions. As expected, the model fitted well the experimental data in those conditions, therefore we can have confidence on the predictions under the welding procedures carried out in this study. The effects of welding parameters on the maximum volume fractions of phases on the HAZ are presented in Table 4. The increase of the heat input promotes the increase of the maximum grain size, austenite reformed and sigma phases. The sigma and reformed austenite phases distributions for the intermediate heat input are shown in Figure 12 and 13, respectively. Table 4 shows the maximum values for all cases analyzed and their respective welding parameters.

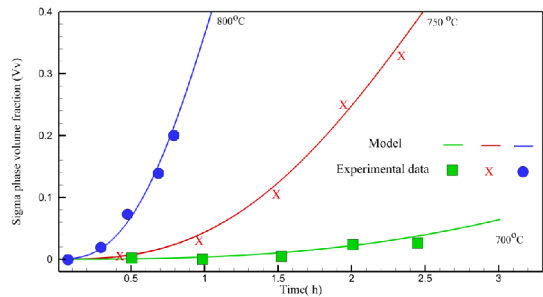


Figure 11. Model predictions confronted with experimental predictions of sigma phase under favorable formation conditions.

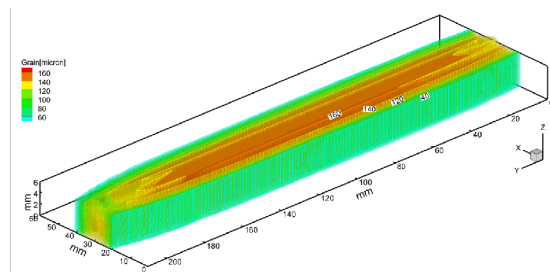


Figure 12. Grain size evolution resulted after the finishing welding pass (heat input = 1.2 kJ mm^{-1}).

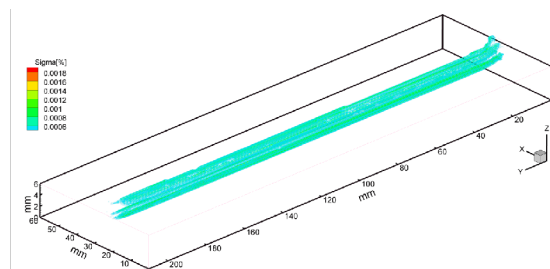


Figure 13. Predicted sigma phase evolution resulted after the finishing welding pass (heat input = 1.2 kJ mm^{-1}).

Table 4. Maximum sigma and reformed austenite phases volumes fractions and grain size predicted for the welding conditions

Voltage (V)	Current (A)	Speed (mm.s ⁻¹)	Thickness (mm)	Heat input (kJ.mm ⁻¹)	Ferritic Grain size (μm)	Austenite reformed (%)	Maximum Sigma Vv (%)
30	179	7.41	6.35	0.7	159	20.7	0.0006
30	178	4.33	6.35	1.2	175	27.5	0.0019
30	180	3.05	6.35	1.7	210	57.3	0.0269
25.7	170	6.76	8.00	0.5	73 (75*)	36.2 (36.9*)	0.0011
25.7	166	2.24	8.00	1.5	217 (232*)	42.5 (41.6*)	0.0026
25.7	169	1.59	8.00	2.2	344 (347*)	49.7 (48.2*)	0.0345
30	178	4.33	10.50	1.2	154	21.9	0.0013
30	179	4.33	6.35	1.2	174	27.2	0.0018
30	179	4.33	2.00	1.2	174	27.5	0.0055

*Averaged measured values using 10 randomly samples (about 5% of dispersion for grain size and 2% for austenite volume fractions).

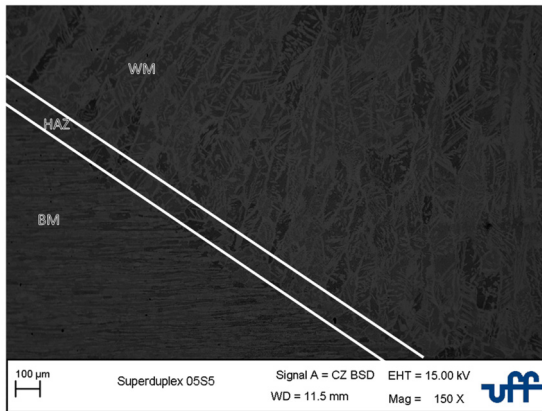


Figure 14. SEM image of the welding regions for heat input of 0.5 kJ mm⁻¹.

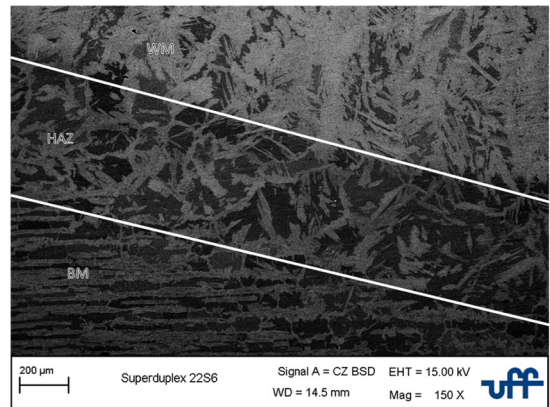


Figure 16. SEM image of the welding zones on the SDSS for heat input of 2.2 kJ mm⁻¹

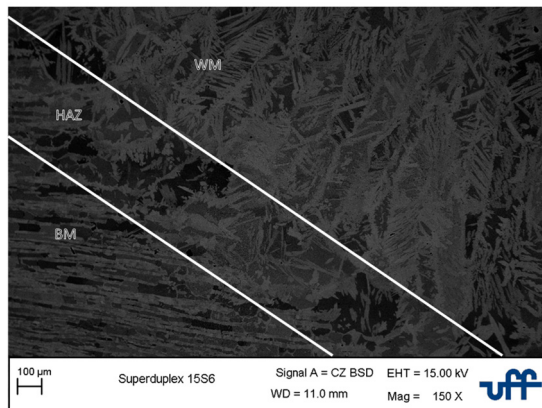


Figure 15. SEM image of the welding zones for heat input of 1.2 kJ mm⁻¹.

Therefore the heat input and thickness are important parameters to take into account during the adequate selection of the welding procedure depending on the application^{29,30}. The model presented in this study is an important tool to correctly design the suitable welding

procedure and control of the changes on the HAZ. The simulation results indicated that, under the welding conditions considered, very small amount of sigma phase is dispersed formed on the HAZ.

Therefore, the predicted amount is negligible, as can be seen in Figure 13 and Table 4. These results are in agreement with microscopic observation on the welded samples, which could not be observed for all cases considered. Figure 14, 15 and 16 show SEM images of the welding regions used to estimate the grain size and phase changes evolution. The results of the micrographs and the calculations present same trend. As can be observed, the ferritic grain size increased as the heat input increased. Same pattern is observed for the austenite volume fractions, which increased with the heat input. The average values for the calculated and measured results for grain growth and austenite volume fractions are in good agreement for the heat input and parameters developed during the SDSS welding procedure. With the proposed procedure we can select the suitable operational conditions for welding this material according with the available facilities.

4. Conclusions

We proposed a coupled model for dynamically estimating the thermophysical properties during welding procedures of the SDSS. The model was verified comparing experimental measurements of the transient temperatures during welding and the final microstructure. Local effects of heat input, thickness and interpass temperatures were evaluated using the mathematical model. We found that very small amount of deleterious phases could be formed on the HAZ during welding of SDSS sheets (less than 15 of volume fraction). The amount of deleterious phases increased for interpass temperatures of 600°C and smaller thickness for heat input of 1.2 kJ. mm⁻¹.

5. Acknowledgements

The first author thanks to CNPq and FAPERJ for financial support.

6. References

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