

ISSN 1516-635X Jan - Mar 2008 / v.10 / n.1 / 45 - 52

Author(s)

Móri C¹ Garcia EA² Ducatti C³ Denadai JC⁴ Gottmann R⁴ Mituo MAO⁴

- ¹ Ph.D. in Animal Production and Nutrition Animal/FMVZ – UNESP/Botucatu.
- ² Prof. Doctor of the Department of Animal Production/FMVZ – UNESP/Botucatu.
- ³ Prof. Doctor Supervisor of the Center of Stable Enviromental Isotopes – Biosciences Institute UNESP/Botucatu.
- ⁴ Graduate students FMVZ UNESP/ Botucatu.

Mail Address

Cleusa Móri Rodovia BR 116, Km 460 Caixa Postal 122 11.900-000. Registro, SP, Brazil.

E-mail: cleusamori@uol.com.br

Keywords

Coturnix coturnix coturnix, diets, stable isotopes, traceability.

Arrived: August / 2007 Approved: February / 2008 Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon (¹³C/¹²C) and Nitrogen (¹⁵N/¹⁴N) Isotopes

ABSTRACT

Studies on the detection of animal by-products in poultry meat are rare, and non-existent on quail meat. This study aimed at detectiong increasing levels of poultry offal meal (POM) in quail meat, using carbon (${}^{13}C/{}^{12}C$) and nitrogen (${}^{15}N/{}^{14}N$) stable isotopes technique. Sixty four on-day-old male quails derived from a commercial farm were randomly distributed into seven different groups, which were fed experimental diets containing 0, 1.5, 3.0, 4.5, 6.0, 7.5, and 15% of POM. Diets were formulated to contain equal energy, protein, and amino acid levels. Four individuals per treatment were sacrificed at 42 days of age for breast muscle (*Pectoralis major*), keel, and tibia collection, which were subsequently submitted to analyses. Isotopic δ^{13} C and δ^{15} N enrichment was observed in all analyzed tissues, with the lowest detection level of 3% dietary inclusion of poultry offal meal.

INTRODUCTION

Increasing public demands for quality products, specifically for safe products, have required farmers and agribusinesses to change their production systems in order to supply products with proven safety, particularly after the outbreaks of Bovine Spongiform Encephalopathy (BSE) or "mad cow disease". The problems caused by this zoonosis in Europe, Japan, and Canada aroused the interest of consumers on aspects related to the origin of the consumed products and to production methods. This resulted in the development of new technologies and reinforced the concept of traceability, integrating all the links of the production chain, from farm to retail. The adoption of mechanisms that may ensure the quality of the final product through the identification and detailed knowledge on the different production processes is now an essential requirement, i.e., an inherent condition of the production process (Antunes, 2003). Moreover, traceability systems are the best tool to provide information on product quality to the consumers, allowing transparency of the different production chains (Bertolini et al., 2006).

Some systems have allowed the identification of animal by-products, such as microchips, bar codes, labels, tags, etc. However, these systems rely on farmers' integrity, and are based on recorded information, but records are not sufficient to ensure product intrinsic quality. According to Ilbery *et al.* (2000), an independent meat certification technology needs to be developed to comply with consumers' demands.

The analysis of isotopic ratio for the chemical element carbon (¹³C/ ¹²C) by mass spectrometry has been successfully used to test the authenticity, the quality, and the geographic origin of several products, such as fruit juice (Bricout & Koziet, 1987), wine (Martin *et al.*, 1988), dairy products (Rossmann *et al.*, 2000; Manca *et al.*, 2001), beer (Sleiman, 2006), and orange beverages (Queiroz, 2005).



 CO_2 is the main chemical compound of terrestrial plants, with a value of $\delta^{13}C$ of approximately -7.7‰ (Kennedy & Krouse, 1990). As a function of their photosynthetic cycle, plants can be divided into two different groups: those belonging to C_3 photosynthetic cycle or to C_4 photosynthetic cycle. The first are called C_3 cycle plants because the first synthesized organic compound in their biochemical pathway contains three carbon atoms. On the other hand, those which first organic compound contains four carbon atoms are called C_4 cycle plants (Ducatti, 2004).

During photosynthetic assimilation, C₃ plants fix CO₂ from the atmosphere through the *Calvin-Benson* cycle, with δ^{13} C values between -22 and -34‰ (modal value = -26.7%). CO₂ fixation by C₄ plants uses the *Hatch*-Slack cycle, and presents values between -9 and -16‰ (modal value = -12.6%). As the concentration of C₃ and C₄ plants is lower than the international standard (Pee Dee Belemnit - PDB), their values are always negative. Therefore, C_3 and C_4 have different isotopic fingerprints due to the fractioning during photosynthetic fixation of carbon (Smith & Epstein, 1971; O'Leary, 1981; Kennedy & Krouse, 1990; Vogel, 1993; Ducatti, 2004). The natural differences in carbon isotopic ratio among plant species provides the opportunity to use part of these plants as tracers in animal, with no need to change their natural state or to synthesize specific markers, in addition to prevent human health and environmental safety problems related to the use of radioactive isotopes (Jones et al., 1979; Tieszen, 1978; Tyrrell et al., 1984; Boutton et al., 1988; Metges et al., 1990).

Different from carbon, the ${}^{15}N/{}^{14}N$ isotopic ratio in plants does not depend on their photosynthetic cycle. In legumes that use atmospheric air as nitrogen fixation source, this isotopic ratio is similar to the international standard (N₂ atm). In other plants, such as grasses, the ratio varies as a function of the isotopic ratio of the specific nitrogen source in each soil, and depends on many other factors, such as weather and soil fertilization.

Similarly to carbon, the natural abundance of ¹⁵N can be used as a tracer in animal nutrition studies. The abundance of naturally marked substrates, with no restriction to its use in terms of environmental contamination, indicates that these tracers can be used in studies on aquatic organisms (Schroeder, 1983), sharks (Domi *et al.*, 2005), oysters (Piola *et al.*, 2005), and seals (Zhao *et al.*, 2006). The team of the Center of Stable Environmental Isotopes (CIE/IB/UNESP Botucatu) suggests that these natural tracers can be

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

successfully used in studies on layers (Denadai *et al.*, 2007; Carrijo *et al.*, 2000), broilers (Gottmann, 2007; Oliveira, 2005; Carrijo *et al.*, 2006; Cruz, 2002), fish (Zuanon *et al.*, 2006), and sheep (Silva, 2003).

Agribusinesses strive to comply with the requirements of modern consumers, who use their power of choice, and therefore certificated products, containing consistent information on food safety, nutritional quality, feeding regime, and origin, are essential.

Considering the scope of the use of stable isotope technique, and the need to differentiate Brazilian poultry products in the global market, the present study aimed at detecting the inclusion of poultry offal meal (POM) in the breast muscle, keel, and tibia of 42-day-old meat quails by analyzing ¹³C/¹²C and ¹⁵N/¹⁴N isotopic ratios in these tissues.

MATERIAL AND METHODS

The experiment was carried out in the facilities of the Poultry sector of Edgárdia Experimental Farm of the School of Veterinary Medicine and Animal Science of the State University of São Paulo (UNESP), Botucatu campus, in 2006. A total number of 56 one-day-old male European quails (*Coturnix coturnix coturnix*) was acquired from a commercial farm. Birds were housed from one to 42 days of age in a 15 x 4m rearing house, with asbestos tiles and lateral plastic curtains. Seven 100 cm x 80 cm x 35 cm metal cages, used for rearing laying quails, housed eight birds each. Each cage was equipped with a brooder with a 550 watts infrared lamp. Chick cup drinkers, with 0.5L capacity, were used, and water was changed twice daily. These drinkers were replaced by a trough drinker, placed in back of the cage, when birds were 14 days of age. During the first 10 days, birds were fed in a pan feeder, covered with a 1cm plastic mesh, with the aim of reducing feed wastage. On day 11, feed was provided in trough feeders placed in the front of the cages. Feed and water were offered ad libitum during the entire experimental period. Lighting program applied during the first two weeks was 24 hours of light, using 100watts incandescent light bulbs. Only natural light was provided thereafter.

The experimental treatments were T1, T2, T3, T4, T5, T6, and T7, which corresponded to the inclusions of 0, 1.5, 3.0, 4.5, 6.0, 7.5, and 15% POM, respectively. Feeding schedule was divided into starter (one to 21 days of age) and finished (22 a 42 days of age) phase. Feeds were formulated to supply the birds' nutritional



requirements (Tables 1 and 2). Both basal diets contained the equal energy, protein, phosphorus, and amino acid (methionine) levels. Each ingredient used for feed manufacturing derived from the same batch. According to chemical analysis, the poultry offal meal contained 96.14% dry matter (DM), 65.54% crude protein (CP), 12.47% ether extract (EE), 14.49% mineral matter (MM), and mean isotopic values of δ^{13} C = - 16.28 ± 0.07‰ and δ^{15} N = 4.30 ± 0.03 ‰.

Table 1 - Percentage composition of ingredients, calculated				
nutritional levels, and mean isotopic values of the starter				
experimental diets (one to 21 days of age), on fresh matter basis.				

	Experimental diets						
Ingredients (%)	T1	T2	T3	T4	T5	Т6	T7
Ground corn	44.81	46.00	47.21	48.50	49.50	50.34	53.32
Soybean meal	48.60	46.45	44.35	42.20	40.10	38.05	27.98
Poultry offal meal	-	1.50	3.00	4.50	6.00	7.50	15.00
Soybean oil	2.93	2.52	2.10	1.66	1.32	1.03	-
Calcitic limestone	1.03	0.98	0.98	0.94	0.90	0.90	0.61
Dicalcium phosphate	1.79	1.60	1.38	1.18	1.00	0.77	-
DL-Methionine	0.05	0.05	0.04	0.04	0.03	0.03	-
L-Lysine	-	-	-	-	-	-	-
Kaolin	-	0.15	0.19	0.23	0.40	0.63	2.34
Salt	0.39	0.35	0.35	0.35	0.35	0.35	0.35
Vitamin-mineral							
premix ¹	0.40	0.40	0.40	0.40	0.40	0.40	0.40
Total	100	100	100	100	100	100	100
Calculated nutrition	nal lev	els					
Metabolizable energy	,						
(kcal/kg)	2900	2900	2900	2900	2900	2900	2900
Crude protein (%)	26.00	26.00	26.00	26.00	26.00	26.00	26.00
Crude fiber (%)	3.94	3.84	3.75	3.65	3.56	3.46	2.97
Calcium (%)	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Avail. phosphorus (%)0.45	0.45	0.45	0.45	0.45	0.45	0.49
Methionine (%)	0.44	0.45	0.44	0.45	0.44	0.45	0.44
Methionine +							
Cystine (%)	0.86	0.86	0.86	0.86	0.86	0.86	0.86
Lysine (%)	1.50	1.49	1.48	1.47	1.46	1.45	1.41
Mean Isotopic Values ²							
δ ¹³ C	-21.11	-20.58	-20.12	-19.90	-19.80	-18.71-	17.41
δ ¹⁵ N	0.73	0.83	1.20	1.30	1.39	1.57	2.28

1 - Composition of the vitamin-mineral premix from Nutron[®]/kg feed: folic acid 200 mg; pantothenic acid 3,120 mg; choline 75,500 mg; biotin 10,000 mcg; niacin 8,400 mg; Vit. A 1,680 Ul; Vit. B1 436.50 mg; Vit. B12 2,400 mcg; Vit B2 1,200 mg; Vit. B6 624 mg; Vit. D3 400,000 Ul; Vit. E 3,500 mg; Vit. K3 360 mg; Cu 2,000 ppm; Fe.12,500 ppm; I. 187.50 ppm; Mn.18,750 ppm; Zn. 17,500 ppm; Se 75.00 ppm. 2 - Mean isotopic values expressed as d ¹³C relative to the standard *Peedee Belemnite (PDB)* and d ¹⁵N relative to the standard atmospheric N₂.

On day 42, four birds per treatment (n = 4) were randomly selected and sacrificed by neck dislocation to collect breast muscle, keel, and tibia samples for isotopic analyses. Breast muscles samples were collected by cutting a section of approximately 20 g in transversal direction of the intermediate longitudinal third of the muscle. In order to collect keel samples, the cartilage of the sternum was dissected, and its insertion in the bone was transversally cut in a right angle to the dorsal surface. Bone samples were

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

obtained by collecting the intermediate longitudinal third of the right tibia. Bone marrow was removed by washing with distilled water. All tissue samples were duly identified and frozen at -20°C. At the time of analysis, tissue samples were thawed, washed in distilled water, placed on Petri dishes, and dried in a force-ventilation oven (Marconi – model MA 035) at 55°C for 48 hours. After drying, samples were ground in a cryogenic mill (Spex – model 6700 *freezer/mill*) at -196°C at maximum frequency for three minutes in order to obtain homogenous material with very fine particle size, with the appearance of talcum (Licatti, 1997; Ducatti, 2004).

Table 2 - Percentage composition of ingredients, calculated					
nutritional levels, and mean isotopic values of the finisher					
experimental diets (22 to 42 days of age), on fresh matter basis.					

	Experimental diets						
Ingredients (%)	T1	T2	Т3	T4	T5	Т6	T7
Ground corn	50.42	52.38	53.45	54.19	54.50	55.19	60.70
Soybean meal	40.91	38.63	36.52	34.50	32.53	30.50	19.83
Poultry offal meal	0	1.50	3.00	4.50	6.00	7.50	15.00
Soybean oil	4.99	4.32	3.95	3.70	3.59	3.35	1.46
Calcitic limestone	0.90	0.86	0.87	0.82	0.82	0.79	0.78
Dicalcium phosphate	1.67	1.47	1.26	1.08	0.85	0.65	0.01
DL-Methionine	0.10	0.09	0.09	0.08	0.08	0.07	0.05
L-Lysine	-	-	-	-	-	-	0.07
Kaolin	0.26	-	0.11	0.38	0.88	1.20	1.35
Salt	0.35	0.35	0.35	0.35	0.35	0.35	0.35
Vitamin-mineral							
premix ¹	0.40	0.40	0.40	0.40	0.40	0.40	0.40
Total	100	100	100	100	100	100	100
Calculated nutrition	nal lev	els					
Metabolizable energy							
(kcal/kg)	3100	3100	3100	3100	3100	3100	3100
Crude protein (%)	23.00	23.00	23.00	23.00	23.00	23.00	23.00
Crude fiber (%)	3.54	3.45	3.35	3.26	3.16	3.06	2.55
Calcium (%)	0.90	0.90	0.90	0.90	0.90	0.90	1.03
Avail. phosphorus (%) 0.42	0.42	0.42	0.42	0.42	0.42	0.48
Methionine (%)	0.45	0.45	0.45	0.45	0.45	0.45	0.45
Methionine +							
Cystine (%)	0.82	0.82	0.82	0.82	0.82	0.82	0.82
Lysine (%)	1.29	1.28	1.27	1.26	1.25	1.25	1.25
Mean Isotopic Values ²							
δ ¹³ C	-20.25	-19.80	-19.40	-19.15	-18.42	-18.15-	16.77
δ ¹⁵ N	0.96	1.19	1.28	1.36	1.60	1.80	2.47

1 - Composition of the vitamin-mineral premix from Nutron®/kg feed: folic acid 162.50 mg; pantothenic acid 2,600 mg; choline 65,250; niacin 7,000 mg; Se 75.00 ppm; Vit. A 1,400 UI; Vit. B1 388.00 mg; Vit. B12 2,000 mcg; Vit B2 1,000 mg; Vit. B6 520 mg; Vit. D3 300,000 UI; Vit. E 2,500 mg; Vit. K3 300 mg; Cu 2,000 ppm; Fe.12,500 ppm; I. 187.50 ppm; Mn.18,750 ppm; Zn. 17,500 ppm. 2 - Mean isotopic values expressed as d ¹³C relative to the standard *Peedee Belemnite (PDB)* and d ¹⁵N relative to the standard atmospheric N₂.

Isotopic analyses of feed ingredients, feeds, and tissues were carried out at the Center of Stable Environmental Isotopes of the Biosciences Institute (CIE/IB), UNESP, Botucatu campus. Isotopic carbon (¹³C/¹²C) and nitrogen (¹⁵N/¹⁴N) ratios were determined in a isotopic ratio mass spectrometer (IRMS) type DELTA – S (Finnigan Mat) coupled to an Elementary Analyzer



(EA 1108 CHN), according to the method described by Ducatti (2004). Analyses results were expressed as *delta per thousand* of the sample isotopic ratio relative to the international standards *PeeDee Belemnite* (PDB) and atmospheric nitrogen (N_2), for the elements carbon and nitrogen, respectively, according to the formula:

$\delta\%_{(sample, standard)} = [(R_{sample} - R_{standard}) / R_{standard}] \times 10^{3}$

where R represents the ratio between the heaviest and the lightest isotope, specifically ${}^{13}C/{}^{12}C$ and ${}^{15}N/{}^{14}N$, of the sample and the standard.

The obtained isotopic results were submitted to multivariate analysis of variance (MANOVA) using GLM (General Linear Model) procedures of SAS (1999) statistical software. Data were generated by the error matrices for each tissue, which were later graphically distributed in regions (ellipses) with 95% confidence of observing possible differences between experimental treatment means and control treatment means. This software allows to verify if the values of the isotopic pair ($\delta^{13}C$ and $\delta^{15}N$) of the control treatment (vegetable feed), are statistically different from the values of the isotopic pair of the treatments which included animal protein. In order to determine the estimated inclusion percentage of poultry offal meal (POM) in breast muscle, keel, and tibia composition, the model of isotopic dilution of two sources and two isotopes in product synthesis (Ducatti, 2004). This model allows measuring the index of participation of each source based on the analyses of the breast muscle, keel, and tibia of the quails fed exclusively plant ingredients (source 1) and of the quails fed increasing POM levels (source 2). The following equation system was applied:

Product = Source 1 + Source 2

$$\delta^{13}C = a\delta F_1 + b\delta F_2$$
 (1)
 $\delta^{15}N = a\delta'F_1 + b\delta'F_2$ (2)

where:

 δ^{13} C and δ^{15} N = relative isotopic enrichment value of the product (breast muscle, keel, or tibia) for carbon-13 and nitrogen-15, respectively. $\delta F_1 e \delta F_2 =$ relative isotopic enrichment value

 $\delta' F_1 = \delta' F_2 = 0$ relative isotopic enhemittent value of carbon of source 1 (vegetable) and source 2 (animal), respectively. relative isotopic enrichment value of nitrogen of source 1 (vegetable) and source 2 (animal), respectively.

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

By isolating value (b) of equation (1) and equation (2), and making them equal, value (a) is obtained, i.e., the index of participation of the vegetable source.

$$\frac{\delta^{13}\mathbf{C} - \mathbf{a}\delta\mathbf{F}_{1}}{\delta\mathbf{F}_{2}} = \frac{\delta^{15}\mathbf{N} - \mathbf{a}\delta'\mathbf{F}_{1}}{\delta'\mathbf{F}_{2}}$$
$$\mathbf{a} = \frac{\delta^{13}\mathbf{C}\delta'\mathbf{F}_{2} - \delta^{15}\mathbf{N}\delta\mathbf{F}_{2}}{\delta\mathbf{F}_{1}\delta'\mathbf{F}_{2} - \delta'\mathbf{F}_{1}\delta\mathbf{F}_{2}}$$
(3)

By replacing value (a) in equation (1) or equation (2), value (b) is obtained, i.e., the index of participation of the animal source.

$$\mathbf{b} = \frac{\delta^{15} \mathbf{N} \delta \mathbf{F}_1 - \delta^{13} \mathbf{C} \delta' \mathbf{F}_1}{\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2}$$
(4)

By summing equation (3) with equation (4), and making it equal to one, the following equation is obtained:

$$\frac{\delta^{13} \mathbf{C} \delta' \mathbf{F}_2 - \delta^{15} \mathbf{N} \delta \mathbf{F}_2}{\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2} + \frac{\delta^{15} \mathbf{N} \delta \mathbf{F}_1 - \delta^{13} \mathbf{C} \delta' \mathbf{F}_1}{\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2} = 1$$

$$(\delta^{13} \mathbf{C} \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2) + (\delta^{15} \mathbf{N} \delta \mathbf{F}_1 - \delta^{13} \mathbf{C} \delta' \mathbf{F}_1) =$$

$$(\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2)$$
or:

$$\mathbf{a}(\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2) + \mathbf{b}(\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2) = (\delta \mathbf{F}_1 \delta' \mathbf{F}_2 - \delta' \mathbf{F}_1 \delta \mathbf{F}_2)$$

The result is a + b = 1, as there are only two sources of participation and only one product.

In order to avoid a possible fractioning factor between the diet (source) and the studied tissue (product), and different carbon/nitrogen ratios, the same tissue is used when applying the equation system (source: tibia/product: tibia).

For source 2, isotopic values of the breast muscle, keel, and tibia derived from the treatment with 15% inclusion (maximum inclusion level) were used, where b values were corrected by multiplying the obtained values by 0.15.



RESULTS AND DISCUSSION

The results of the isotopic analyses (δ^{13} C and δ^{15} N) in the starter and finisher diets used in the present study are presented in Tables 1 e 2, respectively. As POM inclusion percentage increased in the experimental diets, an enrichment of δ^{13} C and δ^{15} N values was observed. This enrichment possibly occurred as a function differences in diet composition: diets were formulated to contain equal energy and protein levels in both rearing phases, but as POM dietary levels increased, soybean meal and oil percentages increased, whereas corn percentage decreased.

 δ^{13} C and δ^{15} N means for the studied tissues of 42day-old quails are presented in Table 3. Mean values reveal the isotopic enrichment of the breast muscle, keel, and tibia of the birds fed increasing POM levels as compared to those that received the diet with no POM inclusion (control treatment). Although each tissue has its own isotopic value, fractioning factor, and isotopic turnover (Hobson & Clark, 1992b; 1992a), the animal is what it isotopically consumes, i.e., $\pm 1\%$ ¹³C and ±3‰ ¹⁵N (DeNiro & Epstein, 1976; 1978). This behavior is shown in Figures 1, 2, and 3, where one can observe that the confidence ellipses are placed in a linear and increasing form as a function of POM dietary level. This is only possible to visualize when diets are carefully formulated and analyzed, and allows its use for the certification process of meat quails.

In the breast muscle, with POM inclusion levels equal or above 3.0% (treatments T4, T5, T6 e T7), confidence ellipses increased the distance from the graph axes, which represent the control treatment (Figure 1). Confidence regions of treatments T4, T5, T6 (POM inclusions of 4.5, 6.0, and 7.5%) present overlapping, demonstrating that these treatments were not different, and reinforcing the notion that any inclusion level above this interval can be isotopically detected. There was a slight dislocation of T2 confidence region towards the carbon axis. This behavior probably

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

indicated the low efficiency of the breast muscle of quails in the detection of dietary POM inclusion levels below 3% T3).

The differences between $\delta^{13}C$ and $\delta^{15}N$ means (Figures 1, 2, and 3) obtained in the error matrices are negative, and therefore the ellipses are located in the third quadrant. Confidence regions that overlap on any of the graph axes possibly present some positive values, and are not adequate to detect POM in quail tissues.



Figure 1 – Confidence regions for differences between delta % ¹³C and % ¹⁵N means of the breast muscle of 42-day-old meat quails in treatments T2, T3, T4, T5, T6, and T7.

In the keel of 42-day-old meat quails, dietary POM inclusions of 3.0% or more of POM were also different from the control treatment (Figure 2). Confidence regions in this region were very distinct as compared to the breast muscle. Confidence ellipses were "flatter", presenting higher amplitude of the points generated by the error matrix, which is possibly a result of the higher standard deviations. Higher than expected standard error are common in meat quails, as these birds were not submitted to high genetic improvement.

The detection level of POM in the tibia was similar to those found in the breast muscle and keel. In the tibia, ellipses presented visible distance and a more linear behavior (Figure 3).

Table 3 - Mean δ^{13} C and δ^{15} N isotopic values and respective standard deviations in the breast muscle, keel, and tibia of 42-day-old quails, according to different poultry offal meal inclusion levels in the diet.

Dietary POM		Sampled tissue						
inclusion, %	Breast	Breast muscle		Ceel	Tibia			
	δ ¹³ C	δ¹⁵N	δ¹³C	δ¹⁵N	δ ¹³ C	δ ¹⁵ N		
0.0	-21.70 ± 0.25	2.53 ± 0.07	-19.35 ± 0.31	3.57 ± 0.22	-18.61 ± 0.06	2.71 ± 0.07		
1.5	-21.41 ± 0.13	2.70 ± 0.09	-18.82 ± 0.15	3.68 ± 0.03	-18.16 ± 0.40	2.90 ± 0.16		
3.0	-20.99 ± 0.44	2.81 ± 0.08	-18.35 ± 0.65	3.80 ± 0.14	-17.91 ± 0.23	3.00 ± 0.10		
4.5	-20.72 ± 0.38	3.04 ± 0.08	-18.17 ± 0.27	3.96 ± 0.13	-17.62 ± 0.40	3.20 ± 0.13		
6.0	-20.52 ± 0.51	3.17 ± 0.15	-17.46 ± 0.44	4.13 ± 0.11	-17.12 ± 0.15	3.31 ± 0.18		
7.5	-19.86 ± 0.50	3.21 ± 0.22	-16.59 ± 0.46	4.25 ± 0.20	-16.25 ± 0.50	3.52 ± 0.16		
15.0	-18.27 ± 0.46	3.87 ± 0.15	-15.56 ± 0.42	4.9 0± 0.23	-14.84 ± 0.28	4.16 ± 0.18		





Figure 2 – Confidence regions for differences between delta % ¹³C and %¹⁵N means of the keel of 42-day-old meat quails in treatments T2, T3, T4, T5, T6, and T7.



Figure 3 – Confidence regions for differences between delta ‰ ¹³C and ‰¹⁵N means of the tibia of 42-day-old meat quails in treatments T2, T3, T4, T5, T6, and T7..

Isotopic ¹³C and ¹⁵N enrichment was observed in the three studied tissues relative to control treatment means and treatments with increasing inclusions of POM. This similar enrichment of both isotopes in the breast muscle, keel, and tibia is probably explained by the high rates of basal metabolism of quails (Hobson & Clark, 1992a), as these birds present high activity, and consequently high turnover rate. Although isotopic enrichment was similar in the studied tissue, the keel of 42-day-old guails presented higher ¹⁵N enrichment. This may be due to differences in the composition of essential and non-essential amino acids of these tissues. According to Moran Jr. (1999), most of the amino acids in the breast muscles are essential amino acids, which exhibit little change in their isotopic ratio when incorporated in the tissue (Pinnegar & Polunin, 1999). On the other hand, type-I collagen, which makes up approximately 95% of the organic bone matrix (Pizauro Jr., 2002), and therefore is the largest nitrogen source

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

of the bone, consists mainly of non-essential amino acids. In this sense, diets containing POM supply a higher amount of ¹⁵N for the endogenous synthesis of non-essential amino acids, as well as intact nonessential amino acids in the dietary protein. Some authors report that the primary sources responsible for the isotopic fractioning of nitrogen are metabolic reactions involved in the processes of deamination and transamination of amino acids (Gaelber *et al.*, 1996; Minagawa & Wada, 1984).

The variations found among tissue are still not well understood. According to Tiezen *et al.* (1983), the main biochemical fractions are isotopically different, and the isotopic differences in the body may be a reflex of their different biochemical composition. Tissue with lower lipid content would probably have a higher δ^{13} C level as compared to a tissue with higher lipid content, which is relatively poor in carbon-13 (Tiezen *et al.*, 1983; Piasentier *et a*l., 2003).

The results obtained by the equation system are shown in Table 4. These results suggest that the error in the comparison between experimental (used in the diet) and calculated values (equation system) are lower than 1%.

Table 4 - Dietary inclusion level of poultry offal meal determinedby the equation system in the different tissues of 42-day-oldmeat quails.

POM dietary	Inclusion level (equation system)					
inclusion level	Sampled tissues					
(%)	Breast meat	Keel	Tibia			
0	0.0	0.0	0.0			
1.5	1.7	1.5	1.9			
3.0	3.1	3.0	2.9			
4.5	5.4	4.5	4.6			
6.0	6.7	6.5	6.1			
7.5	7.7	8.7	8.8			
15.0	15.0	15.0	15.0			

CONCLUSIONS

The technique of stable isotopes is able to detect the inclusion of poultry offal meal in the breast muscle, keel, and tibia of 42-day-old meat quails, when inclusion levels are equal or higher than 3%.

REFERENCES

Antunes R. Da granja à mesa. Avicultura Industrial 2003; 94(110): 24.

Bertolini M, Bevilacqua M, Massini R. FMECA approach to product traceability in the food industry. Food Control 2006; 17:137-145.



Boutton TW, Tyrrell HF, Patterson BW. Carbon kinetics of milk formation in Holstein cows in late lactation. Journal of Animal Science 1988; 66:2636-2645.

Bricout J, Koziet J. Control of authenticity of orange juice by isotopic analysis. Journal of Agricultural Food Chemistry 1987; 35:758-760.

Carrijo AS, Pezzato AC, Ducatti C. Avaliação do metabolismo nutricional em poedeiras pela técnica dos isótopos estáveis do carbono (¹³C/¹²C). Revista Brasileira de Ciência Avícola 2000; 3: 209-218.

Carrijo AS, Pezzato AC, Ducatti C, Sartori JR, Trinca L, Silva ET. Traceability of Bovine Meat Bone Meal in Poultry by Stable Isotope Analysis. Revista Brasileira de Ciência Avícola 2006; 8:37-42.

Denadai JC. Avaliação Metabólica de dietas C₃ e C₄ na formação do ovo (gema e albúmen) pelo uso da técnica dos isótopos estáveis de carbono [dissertação]. Botucatu (SP): Universidade Estadual Paulista; 2004.

Cruz VC. Livre escolha de alimentos dos ciclos fotossintéticos C₃ e C₄, fracionamento e turnover dos isótopos estáveis do carbono nos tecidos de frangos de corte [dissertação]. Botucatu (SP): Universidade Estadual Paulista; 2002.

DeNiro MJ, Epstein S. You are what you eat (plus a few) the carbon isotope cycle in food chains. Geological Society of America 1976; 6:834.

DeNiro MJ, Epstein S. Influence of diet on the distribution of carbon isotopes in animals. Geochimica et Cosmochimica Acta 1978; 42:495-506.

Domi N, Bouquegneau K, Das K. Feeding ecology of five commercial shark species of the Celtic Sea through stable isotope and trace metal analysis. Marine Environmental Research 2005; 60:551-569.

Ducatti C. Isótopos estáveis ambientais [apostila]. Botucatu (SP): Universidade Estadual Paulista; 2004.

Gaelber OH, Vitt TG, Vukmirovich R. Isotope effects in metabolism in ¹⁵N and ¹⁴N from unlabeled dietary. Journal Biochemistry 1966; 44:1249-1257.

Hobson KA, Clark RG. Assessing avian diets using stable isotopes I: Turnover of ¹³C in tissues. The Condor 1992a; 94:181-188.

Hobson KA, Clark RG. Assessing avian diets using stable isotopes II: Factors influencing diet-tissue fractionation. The Condor 1992b; 94:189-197.

Ilbery B, Kneafsey M, Bamford M. Protecting and promoting regional specialty food and drink products in the European Union. Outlook on Agriculture 2000; 29:31-37.

Jones R, Ludlow M, Troughton J. Estimation of the proportion of C_3 and C_4 plant species in diet of animals from the ratio of natural ¹²C and ¹³C isotopes in the faeces. Journal of Agricultural Science 1979; 92:91-100.

Poultry Offal Meal Traceability in Meat Quail Tissues using the Technique of Stable Carbon ($^{13}C/^{12}C$) and Nitrogen ($^{15}N/^{14}N$) Isotopes

Kennedy BV, Krouse HR. Isotope fractionation by plants and animals: Implications for nutrition research. Canadian Journal Physiology and Pharmacology. 1990; 68:960-972.

Licatti F. Isótopos estáveis do carbono (${}^{13}C/{}^{12}C$) em plantas do ciclo bioquímico C₃ e C₄ [monografia]. Botucatu (SP): Universidade Estadual Paulista; 1997.

Manca G, Camin F, Coloru G, Del Caro A, Detentori D, Franco MA, Versini G. Characterization of the geographical origin of *Pecorino Sardo* cheese by casein stable isotope (¹³C/¹²C and ¹⁵N/¹⁴N) ration and free amino acid ratios. Journal of Agricultural and Food Chemistry 2001; 49:1404-1409.

Martin GJ, Guillou C, Martin ML, Cabanis MT, Tep X, Aerny J. Natural factors of isotope fractionation and the characterization of wines. Journal of Agricultural and Food Chemistry 1988; 36:316-322.

Metges C, Kempe K, Schimidt HL. Dependence of the carbon isotope contents of breath carbon dioxide, milk, serum and rumen fermentation products on the delta ¹³C value of food in dairy cows. British Journal of Nutrition 1990; 63:187-196.

Minagawa M, Wada E. Stepwise enrichment of ¹⁵N along food chains: Further evidence and relation between ¹⁵N and animal age. Geochimica et Cosmochimica Acta 1984; 48:1135-1140.

Moran Jr ET. Live production factors influencing yield and quality of poultry meat. In: Richardson, R.I., Mead, G.C., editors. Poultry Meat Science. Wallingford: CABI Publishing; 1999. p.175-195.

O'Leary MH. Carbon isotope fractionation in plants. Phytochemistry 1981; 20(4):553-567.

Oliveira RP. Rastreabilidade da Farinha de Vísceras de aves na alimentação de frangos de corte pela técnica dos isótopos estáveis (δ^{13} C e δ^{15} N) [tese]. Botucatu (SP): Universidade Estadual Paulista; 2005.

Piasentier E, Valusso R, Camin F, Versini G. Stable isotope ratio analysis for authentication of lamb meat. Meat Science 2003; 64: 239-247.

Pinnegar JK, Polunin VC. Differential fractionation of δ^{13} C and δ^{15} N among fish tissue: Implications for the study of trophic interactions. Funcional Ecology 1999; 13:225-231.

Piola RF, Moore SK, Suthers IM. Carbon and nitrogen stable isotope analisys of three types of oyster tissue in an impacted estuary. Estuarine Coastal and Shelf Science 2005; 1-12.

Pizauro Jr JM. Estrutura e função do tecido ósseo. Fisiologia aviária aplicada a frangos de corte. In: Macari, M. editor. Fisiologia aviária aplicada à frangos de corte. Jaboticabal: FUNEP-UNESP; 2002. p. 375.

Queiroz CE. Utilização dos isótopos estáveis de carbono e nitrogênio na detecção de adulteração e avaliação energética de bebidas de laranja [tese]. Botucatu (SP): Universidade Estadual Paulista; 2005.

Rossmann A, Haberhauer G, Holzl S, Horn P, Pichlmayer F, Voerkelius S. The potential of multielement stable isotope analysis for regional origin assignment of butter. European Food Research & Technology 2000; 211:32-40.



Sas Institute. SAS/STAT[™]. SAS user's guide for windows environment. 8.0th ed. Cary (NC): SAS Institute; 1999.

Schroeder GL. Stable isotope rations as naturally occurring tracers in the aquaculture food web Aquaculture 1983; 203-210.

Sleiman M. Determinação do percentual de malte e cevada em cervejas tipo pilsen utilizando os isótopos estáveis de carbono e nitrogênio na detecção de adulteração e avaliação energética de bebidas de laranja [tese]. Botucatu (SP): Universidade Estadual Paulista; 2006.

Silva JJ. Determinação da fase lactente – ruminante em cordeiros pelas técnicas dos δ^{13} C e micro –histologia fecal [dissertação]. Botucatu (SP): Universidade Estadual Paulista; 2003.

Smith BM, Epstein S. Two categories of ¹³C/¹²C rations of higher plants. Plant Physiology 1971; 47:380-384.

Tieszen LL, Boutton TW, Tesdahl KG, Slade NA. Fractionation and turnover of stable carbon isotopes in animal tissues: implications for δ^{13} C analysis of diet. Oecologia 1983; 57:32-37.

Tieszen LL. Carbon isotope fractionation in biological material Nature 1978; 276:97-98.

Tyrrel HF, Pelletier G, Chevalier R, Hillaire-Marcell C, Gagnon M. Use of carbon 13 as tracer in metabolism studies. Canadian Journal of Animal Science 1984; 129 (suppl):127-129.

Vogel JC. Variability of carbon isotope fractionation during photosynthesis. In: Ehleringer JR, Hall AE, Farquhar GD. (ed). Stable isotopes and plant carbon-water relations. San Diego: Academic Press 1993; 29-46.

Zhao L, Schell DM, Castellini MA. Dietary macronutrients influence ¹³C e ¹⁵N signatures of pinnipeds: Captive feeding studies with harbor seals (*Phoca vitulina*) Part A. Comparative Biochimistry and Phisiology 2006; 143: 469-478.

Zuanon JAS, Pezzato AC, Pezzato LE, Passos JRS, Barros MM, Ducatti C. Muscle δ^{13} C change in Nile tilapia (*Oreochromis niloticus*) Effects of growth and carbon turnover Part B. Comparative Biochimistry and Phisiology 2006; 145: 101-107.