

Toxicity of herbicides on *Escherichia coli* growth

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(With 4 figures)

Abstract

Agriculture uses a huge variety and quantity of chemicals. If, on one hand, the goal is to increase productivity, on the other hand these products contaminate aquatic environments. Among these products, herbicides deserve greater attention in relation to contamination of aquatic environments due to their extensive use to weed control. This study was carried out because the effects of these molecules on aquatic microorganisms such as *Escherichia coli*, is still unclear. Using microdilution plate assays, *Escherichia coli* were exposed to various commercial formulations of herbicides widely used in Brazil. The herbicide paraquat was the only one able to prevent the growth of *Escherichia coli* and is characterized as bacteriostatic.

Keywords: aquatic ecosystem, bacteriostatic, paraquat.

Toxicidade de herbicidas sobre o crescimento de *Escherichia coli*

Resumo

A agricultura utiliza uma enorme variedade de produtos químicos, e se por um lado a meta é aumentar a produtividade, por outro lado estes produtos contaminam ambientes aquáticos. Entre estes produtos, herbicidas merecem maior atenção em relação à contaminação de ecossistemas aquáticos, devido a seu amplo emprego para controlar plantas daninhas. Este estudo foi realizado porque os efeitos destas moléculas em micro-organismos aquáticos, tais como *Escherichia coli*, ainda são pouco estudados. Através de ensaios de microdiluição, *E. coli* foi exposta a várias formulações comerciais de herbicidas amplamente utilizados no Brasil e foi verificado que o herbicida paraquat foi o único capaz de impedir o crescimento de *E. coli*, caracterizando-se assim como bacteriostático.

Palavras-chave: ecossistemas aquáticos, bacteriostático, paraquat.

1. Introduction

Aquatic ecosystems experience changes most often associated with industrial, urban and agricultural activities. Among these activities, the latter has been particularly responsible for more contamination to non-target organisms, both vertebrates and invertebrates (Mozeto and Zagatto, 2006). Thus, although pesticides fulfill the role of protecting crops from pests, diseases and weeds, increasing productivity, they pose risks to aquatic environments (Mozeto and Zagatto, 2006).

Herbicides are a heterogeneous category of chemical products especially synthesized for weed control, but they can reach aquatic ecosystems in several ways from the area where they are applied, reaching aquatic organisms (Tomita and Beyruth, 2002).

Aquatic organisms are exposed to these molecules simply by contact with water or sediment, and toxic effects may include both mortality and sublethal effects, such as changes in growth, development, reproduction, physiology, behavior and appearance of pathologies (Rand and Petrocelli, 1985).

Atrazine (2-chloro-4-ethylamino-6-isopropyl-amino-1,3,5-triazine) and picloram (4-Amino-3,5,6-trichloro-2-pyridinecarboxylic acid) herbicides are among the most used in Brazil agriculture. The first is especially recommended for crops of maize, sorghum, sugar cane, among others, and has high leaching potential (Rodrigues and Almeida, 2005). As for picloram, it is used in Brazilian pastures and is known for having high persistence in soil (Close et al.,

2003b; Berisford et al., 2006) low sorption, high water solubility and high leaching potential and can reach groundwater aquifers (Bovey and Richardson, 1991; Close et al., 2003a). Due to these characteristics, atrazine and picloran are of great concern for aquatic ecosystems.

Historically, studies on the effects of such chemicals on aquatic life have focused on macroscopic organisms, with particular emphasis on vertebrates (for example, Poleksic and Karan, 1999; Sarikaya and Yilmaz, 2003; Botelho et al., 2009). However, bacteria represent important members of the fresh water flora and are presumably vulnerable to alterations in their growth and reproduction due to exposure to chemical contaminants (Higgins and Hohn, 2007).

Aquatic bacteria such as *Escherichia coli* are extremely important to the aquatic environment since they participate in the organic matter decomposition, nutrient cycling and energy flow (Bernhard et al., 2005). Although various studies have evaluated the toxicity promoted by pesticides to the environment using bacteria as a test organism, reports on the *Escherichia coli* use are minimal (Bálague et al., 2001; Higgins and Hohn, 2007).

In order to comprehend the effects of chemical agents on aquatic bacteria better, an experiment in vitro was carried out, where the toxicity of many common herbicides was evaluated using *Escherichia coli* as a test-organism.

2. Material and Methods

To evaluate herbicides effect on the growth of *Escherichia coli* the ATCC 25922 strain was used obtained from the Microbiology Laboratory at the University Vale do Rio Doce – UNIVALE. The herbicides were obtained from the Department of Agronomy at the University Vale do Rio Doce. The herbicide concentrations used in each well

in this study were in mg/L: DMA (0.23), Agimix (0.10), Siptran, AtraneX and Herbitrim (0.10), Fusilade (0.23), Gliz and Roundup (0.09), Dual (0.06), Fortex (0.232), MSMA (0.12), Gramoxone (0.08), Padron (0.03), Tordon (0.10), Combine (0.11) and Trifluralina (0.045). Some technical characteristics are shown in Table 1.

The definition of concentrations used was based on the commercially recommended dose taking into account the average lethal concentration (LC50) for aquatic wildlife (fishes) (Rodrigues and Almeida, 2005).

All herbicide solutions were prepared prior to their use, and were subsequently sterilized by filtration (Millipore filter, 0.22 mm) in a laminar flow hood. The broth microdilution assay was done according to the National Committee for Clinical Laboratory Standards (NCCLS) M7-A6 (2003). Microdilution plates were used for inoculation and evaluation by adding 180 µL BHI and inoculum broth and 20 µL of each herbicide with three replicates each. Negative (200 µL BHI medium) and positive controls (bacteria + 200 µL BHI medium) were also used with three replicates.

After microdilution plate preparation, the bacterial population was incubated at 37 °C and the colony forming units density (CFU) was measured by optical density at 655 nm on a spectrophotometer.

Growth curves were prepared from the relationship CFU and time in each herbicide tested. Specifically for the range between 120 and 210 minutes after incubation (exponential growth value for *Escherichia coli*, considering the control), linear equations were established to estimate the angular coefficient, relating CFU and time. These values and those obtained in 600 minutes reading from the original growth equations were analysed for variance and averages, when significant, were compared using the Tukey test at 5% probability.

Table 1. Technical information about herbicides suggested toxicity evaluation on non – target organisms.

Commercial product	Technical product	Chemical group	Concentration of technical product	Concentration used in this study (mg/L)
DMA	2,4-D	Phenoacetic acid	670 g/L	0.23
Agimix	Alachlor + Atrazine	Chloroacetanilide + Triazine	260 + 260 g/L	0.10
Siptran, AtraneX and Herbitrim	Atrazine	Triazine	500 g/L	0.10
Fusilade	Fluazifop p-butyl	Aryloxyphenoxy propionate	125 g/L	0.23
Gliz and roundup original	Glyphosate	Glicine derivate	480 g/L	0.09
Dual	S-metalachlor	Acetanilides	960 g/L	0.06
Fortex	Msma + Diuron	Organoarsenic + Urea	360 + 140 g/L	0.232
MSMA	Msma	Organoarsenic	720 g /L	0.12
Gramoxone	Paraquat	Bypyridyls	200 g/L	0.08
Padron	Picloran	Pyridinecarboxylic acid	388 g/L	0.03
Tordon	Picloram + 2,4 – D	Picolinic and Phenoxyacetic acid	64 + 240 g/L	0.10
Combine	Tebuthiuron	Urea derivate	500 g/L	0.11
Trifluralina	Trifluralin	Dinitroaniline	445 g/L	0.045

3. Results

The growth of *Escherichia coli* in the range between 100-300 minutes incubation was observed when exposed to the herbicides Agimix, Atranex, Herbitrin and Siptran (all with technical product atrazine). Regardless of herbicide, *Escherichia coli* growth was higher in the control treatment (free of herbicides) (Figure 1).

Comparing the toxicity of auxin mimicking herbicides, *Escherichia coli* were more sensitive to the Padron herbicide than to Tordon and DMA (Figure 2). Compared to the control, the Tordon herbicide promoted greater growth, unlike DMA and Padron herbicides.

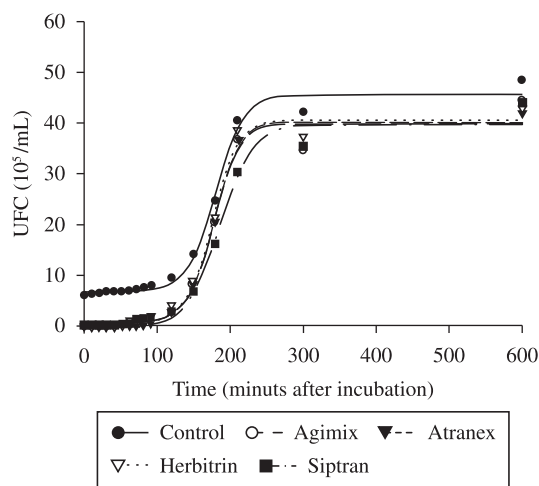


Figure 1. *Escherichia coli* growth under BHI containing different atrazine formulations, incubated for 600 minutes. Non-linear regressions tested by “t” at 5% of probability, with coefficient of determination higher than 90%.

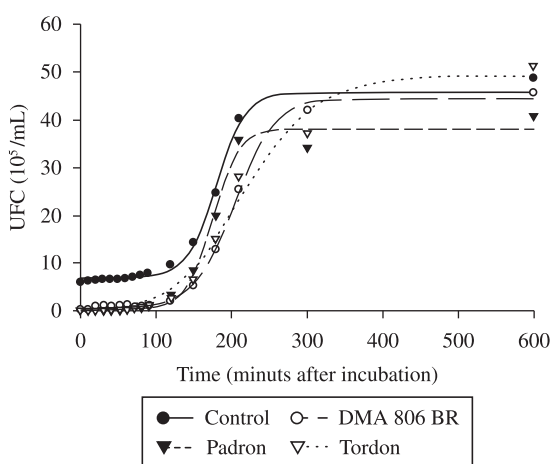


Figure 2. *Escherichia coli* growth under BHI containing DMA, Padron and Tordon (auxin mimicking herbicides), incubated for 600 minutes. Non-linear regressions tested by “t” at 5% of probability, with coefficient of determination higher than 90%.

Regarding non-selective herbicides, recommended for plant desiccation, there is growth of *Escherichia coli* from 100 to approximately 320 minutes incubation when exposed to Roundup and Gliz. However, in relation to the Gramoxone herbicide, there was no bacterial growth during the first 300 minutes of incubation, showing extreme sensitivity of this organism to the product (Figure 3).

Among Dual, Fortex, Combine, Fusilade, MSMA and Trifluralin herbicides, it was observed that *Escherichia coli* has similar sensitivity to all these products. However, it was lower when compared to the control (Figure 4).

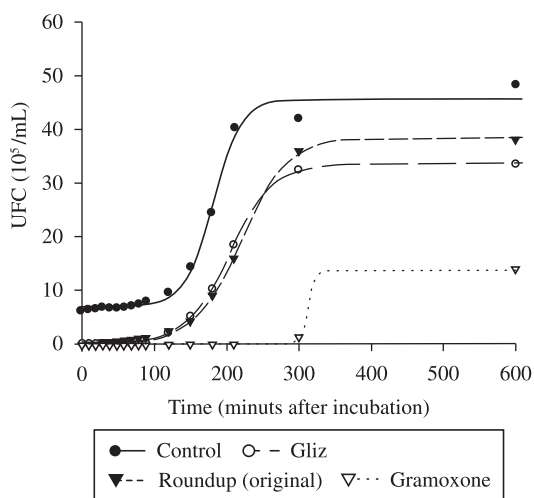


Figure 3. *Escherichia coli* growth under BHI containing Gliz, Roundup and Gramoxone, incubated for 600 minutes. Non-linear regressions tested by “t” at 5% of probability, with coefficient of determination superior to 90%.

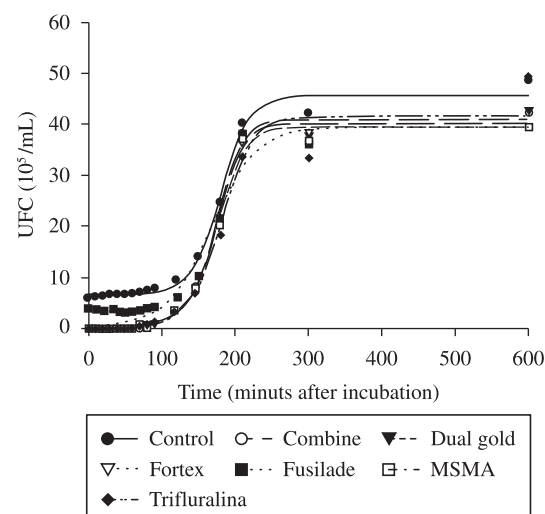


Figure 4. *Escherichia coli* growth under BHI containing different herbicide formulation, incubated for 600 minutes. Non-linear regressions tested by “t” at 5% of probability, with coefficient of determination higher than 90%.

4. Discussion

Considering the exponential growth phase of *Escherichia coli* (observed between approximately 120-210 minutes in this work), it is possible to estimate the following decreasing scale of herbicide toxicity on *Escherichia coli*: Gramoxone > Roundup = Gliz > DMA = Tordon = Siptran > Trifluralin = Fusilade = Padron = Fortex > MSMA = Agimix = Atranex = Dual = Herbitrim = Combine (Table 2). This relationship establishes the cell division intensity during the growth log phase.

After 600 minutes of incubation, the following decreasing scale herbicides toxicity on *Escherichia coli* based on the CFU number was established: Gramoxone > Gliz > Roundup = Fusilade = MSMA = Fortex = Padron = Atranex = Combine = herbitrim = Dual = > Siptran = Agimix > DMA > Tordon = Trifluralin (Table 2). In this case, absolute values for the total CFU formed in each treatment were compared between each other.

The first bacterial growth phase is the lag phase, which according to Trabulsi and Alterthum (2008) is an adaptation period to the environment which they were exposed to, with intense metabolic activity and enzyme production that are prerequisites for their division. In this study, the lag phase corresponds to the period from 0 to 100 minutes incubation. It can be noted that there was no bacterial growth in the control treatment, suggesting

faster adaptation as the medium was free of herbicides. After 100 minutes of adaptation, the bacterium started the log or exponential growth phase, characterized by cell multiplication. At this stage there is cell division at a maximum and constant speed reaching a point where there is lack of nutrients, and then the number of dividing cells becomes equal to the number of death cells (Trabulsi and Alterthum, 2008), then starting the stationary phase. All these three phases can be observed in Figures 1-4.

According to Finkel (2006) when bacterial isolates of environmental origin are successfully cultured in the laboratory, they tend to adhere to a model of growth that has been well characterized in the bacteriological literature, as mentioned above by Trabulsi and Alterthum (2008). According to this model, the period between the lag phase and nutrients depletion followed by bacterial death occurs in less than 24 hours for most coliform bacteria such as *Escherichia coli*. However, this could explain the difference between bacterial growth from 110 to 220 minutes and 600 minutes of exposure to the same herbicide. In this case, 110 to 220 minutes correspond to the stage in which bacteria are more active and thus at the height of its reproduction. By reaching the final period of exposure (600 minutes), slower growth can be observed due to cellular stress caused by intense multiplication in the previous phase.

Table 2. Values of *Escherichia coli* growth (estimated between 120 and 210 and at 600 minutes of incubation) in linear function with respective determination coefficient (r^2) and total number of colony-forming units (CFU) at the end of incubation.

Herbicide (commercial product®)	Growth rate 120-210 minutes after incubation (optical density at 655 nm)	Coefficient of determination	CFU ($\times 10^5$) at the end of incubation (600 minutes) (optical density at 655 nm)
Control (without herbicide)	0.3415 ^b	0.98	48.41 ^a
Agimix	0.3786 ^a	0.97	44.07 ^b
Atranex	0.3825 ^a	0.95	41.94 ^{bc}
Combine	0.3928 ^a	0.96	42.16 ^{bc}
DMA	0.2572 ^c	0.93	45.68 ^{ab}
Dual	0.3830 ^a	0.95	42.88 ^{bc}
Fortex	0.3680 ^{ab}	0.96	39.35 ^c
Fusilade	0.3560 ^b	0.94	39.18 ^c
Gliz	0.1823 ^d	0.95	33.40 ^d
Gramoxone	0.0000 ^c	0.00	13.90 ^c
Herbitrin	0.3867 ^a	0.95	42.67 ^{bc}
MSMA	0.3778 ^a	0.95	39.22 ^c
Padron	0.3655 ^{ab}	0.95	40.75 ^{bc}
Roundup	0.1517 ^d	0.95	38.08 ^c
Siptran	0.3054 ^c	0.94	43.86 ^b
Tordon	0.2810 ^c	0.95	51.29 ^a
Trifluralina	0.3412 ^b	0.95	49.00 ^a
C.V. (%)	6.75	5.24	9.68

Averages followed by letters equal in each column do not differ among themselves by Tukey (5% of probability).

Escherichia coli is routinely used as an indicator of water quality, however, its concentration can be influenced by stressors, which can inhibit (Collier et al., 1990; Bálague et al., 2001) or stimulate (Yurovskaya, 1975) its growth, as seen in this study.

In this study, *Escherichia coli* were not affected by atrazine concentrations. These same results were also found by Koutsotoli et al. (2005) and Higgins and Hohn (2007). In the work by Koutsotoli et al. (2005), the growth of NCTC 9001 *Escherichia coli* and *Enterococcus faecalis* was not affected by up to 0.1 mg/L atrazine concentrations. Higgins and Hohn (2007) also exposed *Escherichia coli* to atrazine concentrations and did not observe bacterial growth inhibition. In this study, the same atrazine concentration (in all commercial formulations) used in Koutsotoli et al. (2005) was applied (0.1 mg/L) and although authors have used different strains of *Escherichia coli* in these two studies, the herbicide was not toxic to their growth. In another study involving a commonly used herbicide, Bálague et al. (2001) reported abolition of *E. coli* growth at 2 mm concentration of 2,4-dichlorophenoxyacetic acid (2,4-D).

During the first 300 minutes, the paraquat herbicide (Gramoxone commercial formulation) did not allow bacterial growth. After this period, there is a reasonable exponential growth phase, indicating high toxicity of the product to *Escherichia coli*, however discarding the bactericidal action hypothesis. The number of CFU in the final period of incubation (600 minutes) was much lower compared to other herbicides, suggesting a high toxicity of this product for this bacterium, especially in 300 minutes exposure.

In *Escherichia coli*, paraquat has been reported as a bacteriostatic agent (Davison and Papirmeister, 1971) and this information was confirmed in our study. Farrington et al. (1973) demonstrated that this product reacts rapidly with molecular oxygen generating the toxic radical (O_2^-). The enzymatic antioxidant defenses are fundamental to reverse such toxicity and thus the superoxide dismutase enzyme acts on the O_2^- dismuting it into hydrogen peroxide (H_2O_2) and water (H_2O). As the hydrogen peroxide (H_2O_2) is also toxic to biological systems, catalysis, another important antioxidant enzyme, promotes the dissociation of this compound into $H_2O + O_2$ (Halliwell and Gutteridge, 2002) and thus paraquat toxicity ceases to exist. The complete bacterial growth inhibition by paraquat action in the first 300 minutes may be due to the O_2^- -toxic effect. After this period, antioxidant enzymes may have eliminated part of the toxicity and then bacterial growth was observed.

In conclusion, our data suggest that with the exception of paraquat, all herbicides may not influence bacterial replication.

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