

## Antioxidant and antibacterial activity of extracts, fractions and isolated substances from the flowers of *Acacia podalyriifolia* A. Cunn. ex G. Don

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The extracts and fractions from the flowers of *A. podalyriifolia* were analyzed previously for antibacterial activity using diffusion in disk. Antioxidant properties were evaluated by determining radical scavenging power (DPPH test) and total phenol content was measured (Folin method). The present study describes the *in vitro* antibacterial (determining minimum inhibitory concentration) and antioxidant activities (by thiobarbituric acid reactive species – TBARS method) for the ethanol extract, dichloromethane and ethyl acetate fractions and two flavanones (naringenin and 5-β-D-glycosyl-naringenin) isolated from the flowers of *Acacia podalyriifolia* A. Cunn. ex G. Don. The flavanones naringenin and 5-β-D-glycosyl-naringenin had not previously been obtained from this species. The most effective antibacterial activity was observed in the ethyl acetate fraction (MIC=0.25 mg mL<sup>-1</sup> against *Staphylococcus aureus* ATCC 6538, MIC = 0.125 mg mL<sup>-1</sup> against *Staphylococcus epidermidis* ATCC 12229, MIC=0.5 mg mL<sup>-1</sup> against *Streptococcus pyogenes* ATCC 19615, *Klebsiella pneumoniae* ATCC 13883 and *Proteus mirabilis* ATCC 43071). The evaluated samples showed antioxidant activity on the TBARS test, especially for ethanol extract (1000 ppm), which was the most active (29.43% ± 0.65) followed by ethyl acetate fraction (1000 ppm, 24.84% ± 1.28), both demonstrating higher activity than that presented by ascorbic acid (1000 ppm, 21.73% ± 1.77), although lower than the BHT (1000 ppm 35.15% ± 3.42), both reference compounds. Naringenin and 5-β-D-glycosyl-naringenin demonstrated antioxidant action, but only naringenin inhibited the growth of gram-positive and gram-negative bacteria.

**Uniterms:** *Acacia podalyriifolia*/qualitative analysis. *Acacia podalyriifolia*/phytochemistry. *Acacia podalyriifolia*/antibacterial activity. *Acacia podalyriifolia*/antioxidant properties. Flavanones. 5-β-D-glycosyl-naringenin. Naringenin.

Os extratos e frações de *Acacia podalyriifolia* foram analisados previamente para a atividade antibacteriana através da difusão em disco e as propriedades antioxidantes foram verificadas pela determinação da capacidade removedora do radical livre DPPH e pela mensuração do conteúdo de fenólicos totais (Método de Folin). O presente estudo descreve as atividades antibacteriana (determinação da concentração inibitória mínima) e antioxidante (espécies reativas do ácido tiobarbitúrico – teste TBARS) para o extrato etanólico e as frações diclorometano e acetato de etila e para duas flavanonas (naringenina e 5-β-D-glicosil-naringenina) isoladas das flores de *Acacia podalyriifolia* A. Cunn. ex G. Don. As flavanonas naringenina e 5-β-D-glicosil-naringenina ainda não haviam sido obtidas desta espécie. A atividade antibacteriana mais efetiva foi observada com a fração acetato de etila (CIM=0,25 mg/mL contra *Staphylococcus aureus* ATCC 6538; CIM=0,125 mg/mL, contra *Staphylococcus epidermidis* ATCC 12229; CIM=0,5 mg/mL contra *Streptococcus pyogenes* ATCC 19615, *Klebsiella pneumoniae* ATCC 13883 e *Proteus mirabilis* ATCC 43071). As amostras avaliadas demonstraram atividade pelo teste TBARS, especialmente o extrato

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etanólico (1000 ppm), que foi o mais ativo ( $29,43\% \pm 0,65$ ), seguido pela fração acetato de etila (1000 ppm,  $24,84\% \pm 1,28$ ), ambos demonstrando atividade mais elevada que a apresentada pelo ácido ascórbico (1000 ppm,  $21,73\% \pm 1,77$ ), ainda que menor que a do BHT (1000 ppm,  $35,15\% \pm 3,42$ ), ambas substâncias de referência. Naringenina e 5- $\beta$ -D-glicosil-naringenina demonstraram ação antioxidante, porém somente a naringenina inibiu o crescimento de bactérias gram-positivas e gram-negativas.

**Unitermos:** *Acacia podalyriifolia*/análise qualitativa. *Acacia podalyriifolia*/atividade antibacteriana. *Acacia podalyriifolia*/propiedades antioxidantes. Flavanonas. 5- $\beta$ -D-Glicosil-naringenina. Naringenina.

## INTRODUCTION

*Acacia podalyriifolia* A. Cunn. Ex G. Don, Leguminosae-Mimosoideae, is an exotic plant cultivated in the south of Brazil as an ornamental tree (Burkart, 1979). Several species in the gender *Acacia* present important medicinal applications (Hagos, Samuelsson, 1988; Chhabra *et al.*, 1990; Nabi *et al.*, 1992; Sekine *et al.*, 1997; Kambizi, Afolayan, 2001; Saleem *et al.*, 2001; Andrade *et al.*, 2003; Meera *et al.*, 2005; Wu *et al.*, 2005) and possess phenolic compounds, for which countless biological activities are described (Kerber, Silva, 1993; Andrade *et al.*, 2003; Meera *et al.*, 2005). However, to date, studies of *A. podalyriifolia* are scarce (White, 1943; Ballandrin *et al.*, 1978; Anderson, Bell, 1976; Churms *et al.*, 1970; Andrade *et al.*, 2003; Andrade *et al.*, 2005; Andrade *et al.*, 2007), motivating research on this species.

The extracts and fractions from *A. podalyriifolia* were analyzed previously for antibacterial and antioxidant actions and have demonstrated activities. For the evaluation of the antibacterial action, a selection was made against two gram-positive and two gram-negative bacteria, using the diffusion in disk (Andrade *et al.*, 2005). The antioxidant action was verified with the free radical 2,2-diphenyl-1-picrylhydrazyl (DPPH) (Andrade *et al.*, 2007).

Previous results from this study have shown that determination of the minimum inhibitory concentration (MIC), adapted from Brasileiro *et al.* (2006) with modifications, allows quantitative estimation of the antibacterial activity. For this test, several sequential dilutions of the samples with antibacterial activity were incorporated into the medium, added to the bacterium and incubated for 24 hours. MIC is the lowest concentration of antibacterial agent that is capable of inhibiting the growth of the microorganism *in vitro* (Brasileiro *et al.*, 2006).

Lipid peroxidation, which occurs by increased oxidative stress caused by reactive oxygen species (ROS), is a known mechanism of cellular damage that contributes to aging and many pathological processes such as chronic inflammation, respiratory disorders, neurodegenerative diseases, diabetes mellitus, atherosclerosis, autoimmu-

ne diseases of the endocrine glands and carcinogenesis (Dawn-Linsley *et al.*, 2005; Chanwitheesuk *et al.*, 2005; Andrade *et al.*, 2007).

Antioxidants are substances that can significantly delay or prevent the oxidation of substrates as well as prevent or repair damage caused to cells by reactive oxygen species (Chanwitheesuk *et al.*, 2005).

TBARS assays measure the end-point oxidative damage, being useful to evaluate the effects of induced oxidative stress and protection from lipid peroxidation by antioxidants which may be present in the analyzed material (Dawn-Linsley *et al.*, 2005).

## MATERIAL AND METHODS

Flowers of *A. podalyriifolia* were collected in Curitiba – Brazil, from June to September, 2007. The material was identified by the botanist Gert Hatschbach of the Municipal Botanical Museum of Curitiba and the exsiccate deposited under the number 268.219.

After drying in the shade, 300 g of flowers of *A. podalyriifolia* were submitted to ethanol extraction, following by liquid-liquid partition with hexane, dichloromethane and ethyl acetate, using Soxhlet, as per methodology described in previous studies (Andrade *et al.*, 2005; Andrade *et al.*, 2007; Carvalho *et al.*, 2009). After the evaporation of the solvents under reduced pressure and temperature of 40 °C, a 0.7 g of dichloromethane fraction and 6.3 g of ethyl acetate fraction were obtained.

The dichloromethane fraction (0.5 g) was submitted to chromatography in a column of silicagel 60 (0.063 to 0.200 mm) Merck® with a mixture of solvents, beginning with hexane, followed by gradually increasing polarity (hexane:ethyl acetate and ethyl acetate:methanol). The crystallization obtained among the sub-fractions 19 to 57 (142.2 mg) and subsequent analysis by thin-layer chromatography (Aluminum Sheet F254 - Merck®) with mobile phase chloroform:methanol (90:10), demonstrated a mixture of compounds that were gathered and submitted to chromatographic separation in a column of silicagel 60 (0.063 to 0.200 mm) Merck® with the same sequence of previous solvents. Crystals in yellow needles (15.2 mg),

were obtained corresponding to the group of sub-fractions 2 to 4, designated as substance 1.

Also, the ethyl acetate fraction (5 g) was submitted to chromatographic separation in a column of silicagel 60 (0.063 to 0.200 mm) Merck® with a mixture of solvents, beginning with ethyl acetate 100%, followed by gradually increasing polarity (ethyl acetate:formic acid and ethyl acetate:methanol:water:formic acid), yielding crystals in white needles (482 mg), corresponding to the group of sub-fractions 20-33, designated as substance 2.

Substances 1 and 2 were exposed to ultraviolet (200-400 nm) using the reagents diagnoses, according to Mabry *et al.* (1970), <sup>1</sup>H RMN (300 MHz CD<sub>2</sub>Cl<sub>2</sub>), <sup>13</sup>C NMR (300 MHz CDCl<sub>3</sub>) and Infrared (400-4000 cm<sup>-1</sup>). Substance 2 hydrolyzed with trifluoroacetic acid 4N (TFA), following thin-layer chromatography (Aluminum Sheet F254 - Merck®) with mobile phase isopropanol:ethyl acetate:nitroethane:water (30:5:5:5), and the standard monosaccharides (glucose, galactose, rhamnose, mannose, xylose) and naringenin, using orcinol 1% in methanol:H<sub>2</sub>SO<sub>4</sub> (90:10), with heating as disclosed. For substance 1, thin-layer chromatography was performed with authentic sample of naringenin and chloroform: methanol (90:10) as mobile phase.

**Naringenin (1):** crystals in yellow needles, 3.04% of production in relation to the dichloromethane fraction (0.0071% in relation to the initial material); M.P. 250-252 °C, HRMS (CIMS) *m/z*: 294.5 [M + Na]<sup>+</sup> (272.068 made calculations for C<sub>15</sub>H<sub>12</sub>O<sub>5</sub>); IR (KBr) 3035-3280 (O-H), 2831-2976 (C-H), 1639 (C=O), 1602 (C=C), 1313-1520 (C-OH), UV λ<sub>max</sub> (MeOH) 288 (3.92), 320 (sh) (3.58), (NaOMe) 244 (4.00), 322 (4.18), (NaOAc) 251 (3.43), 320 (4.05), (NaOAc/H<sub>3</sub>BO<sub>3</sub>) 291 (4.06), 320 (3.27), (AlCl<sub>3</sub>) 310 (4.04), 363 (3.20), (AlCl<sub>3</sub> / HCl) 311 (4.11), 371 (3.41) nm (log ε); <sup>1</sup>H NMR (300 MHz CD<sub>3</sub>OD) δ 2.99 (1H, *dd*, *J* 1.8, 9.6 Hz, H-3α), 3.03 (1 H, *dd*, *J* 9.6, 12.6 Hz, H-3β), 5.24 (1H, *dd*, *J* 1.8, 9.6 Hz, H-2α), 5.81 (2H, *s*, H-6, 8), 6.74 (2H, *d*, *J* 6.5 Hz, H-3', 5'), 7.23 (2H, *d*, *J* 6.5 Hz, H-2', 6'); <sup>13</sup>C NMR (100 MHz CD<sub>3</sub>OD) δ 44.05(C-3), 80.50 (C-2), 96.21 (C-6), 97.09 (C-8), 103.37 (C-10), 116.36 (C-3', C-5'), 129.07 (C-2', C-6'), 131.10 (C-1'), 159.02 (C-4'), 165.47 (C-5), 168.39 (C-7), 164.90 (C-9), 197.81 (C-4).

**5-β-D-glycosyl-naringenin (2):** crystals in white needles, 9.64% of production in relation to the ethyl acetate fraction (0.2% in relation to the initial; M.P. 172-173 °C; [α]<sub>D</sub><sup>20</sup> - 121° (*c* 1.3, MeOH); HRMS (CIMS) *m/z*: 457.1187 [M.+ Na]<sup>+</sup> (434.1213 made calculation for C<sub>21</sub>H<sub>22</sub>O<sub>10</sub>); IR (KBr) 3362(O-H), 2891-2978 (C-H), 1685 (C=O), 1585-

1646 (C=C), 1458-1531 (C-OH), UV λ<sub>max</sub> (MeOH): 283 (4.11), 320 (sh) (3.78), (NaOMe): 249 (4.02), 325 (4.31); (NaOAc): 252 (4.02), 325 (4.31); (NaOAc/H<sub>3</sub>BO<sub>3</sub>): 283 (4.15), 321 (sh) (3.88); (AlCl<sub>3</sub>): 226 (4.36), 283 (4.15), 320 (sh) (3.83); (AlCl<sub>3</sub>/HCl): 226 (4.34), 283 (4.11), 320 (sh) (3.81); nm (log ε); <sup>1</sup>H NMR (300 MHz CD<sub>3</sub>OD) δ 2.56 (1H, *dd*, *J* 2.7, 17.10 Hz, H-3α), 2.99 (1 H, *dd*, *J* 12.9, 17.10 Hz, H-3β), 3.47 a 3.53 (4H of the glucose, *m*, H-2'', 3'', 4'', 5''), 3.71 (1H, *d*, *J* 7.2 Hz, H-6''α), 3.73 (1H, *d*, *J* 7.2 Hz, H-6''β), 4.67 (1H, *d*, *J* 7.5 Hz, H-1''), 5.33 (1H, *dd*, *J* 2.7, 12.9 Hz, H-2α), 6.06 (1H, *d*, *J* 2.1 Hz, H-6), 6.37 (1H, *d*, *J* 2.1 Hz, H-8), 6.75 (2H, *dd*, *J* 2.7, 8.4 Hz, H-3', 5'), 7.26 (2H, *dd*, *J* 6.0, 8.7 Hz, H-2', 6'); <sup>13</sup>C NMR (300 MHz CD<sub>3</sub>OD) δ 44.51(C-3), 60.69 (C-6''), 69.61 (C-4''), 73.47 (C-2''), 75.57 (C-5''), 77.57 (C-3''), 78.16 (C-2), 97.78 (C-8), 98.96 (C-6), 103.49 (C-1''), 105.38 (C-10), 115.15 (C-3', C-5'), 128.33 (C-2', C-6'), 128.98 (C-1'), 157.69 (C-4'), 160.74 (C-5), 164.24 (C-7), 165.14 (C-9), 190.10 (C-4).

**Antioxidant activity:** The samples obtained from the flowers of *Acacia podalyriifolia* (ethanol extract, dichloromethane and ethyl acetate fractions and isolated substances naringenin and 5-β-D-glycosyl-naringenin) were submitted to the antioxidant test thiobarbituric acid reactive species, using concentrations of 100, 500 and 1000 ppm (parts per million), according to Morais *et al.* (2006) with modifications. Ascorbic acid and BHT (butylated hydroxy toluene) were used as reference compounds. The whole procedure was performed in triplicate.

A 0.5 mL volume of egg yolk solution (10% w/v), and 0.1 mL of each sample or reference compound was added to test tubes, and the volume completed to 1 mL with distilled water. Each of the test tubes then received 0.05 mL of solution of 2,2'-azobis(2-amidinopropane) dihydrochloride - AAPH (0.07 mol/L), 1.5 mL of acetic acid 20% (pH 3.5) and 1.5 mL of thiobarbituric acid - TBA (0.8% w/v) in solution of sodium dodecil sulfate - SDS (1.1% w/v). The material thus prepared was subjected to a water bath (95 °C) for 1 hour, while stirring. After cooling, each tube received 5 mL of n-butanol, centrifuged for 10 minutes at 3000 rpm, and the supernatants were measured by spectrophotometer (Shimadzu) at 532 nm. The same process was carried out with control tubes to which all reagents were added except the samples.

The antioxidant activity was determined by the Antioxidant Index (AI) obtained as a percentage, according to the equation: **AI (%) = 1 - (A/C) X 100**, where *A* is the absorbance of the sample and *C* is the absorbance of fully oxidized control. Results are expressed as mean and standard deviation. The statistical examination of the data

was performed using the R Project for Statistical Computing (Gnu Operating System, 2010). Mean values were compared by using analysis of the variance (ANOVA) test and differences between means were detected using Tukey's test ( $p < 0.05$ ).

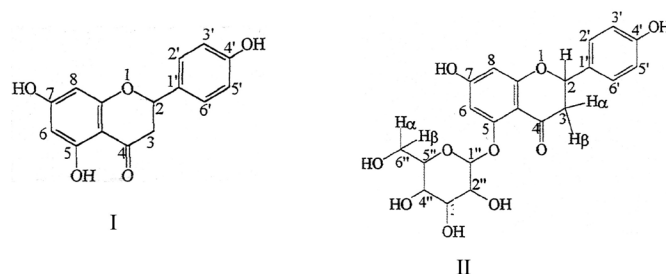
**Antibacterial activity:** The antibacterial action was ascertained by determination of the minimum inhibitory concentration (MIC), according to Brasileiro *et al.* (2006) with modifications. The gram - positive strains used were *Staphylococcus aureus* (ATCC 6538), *Staphylococcus epidermidis* (ATCC 12229) and *Streptococcus pyogenes* (ATCC 19615) while the gram-negative strains used were *Escherichia coli* (ATCC 25922), *Klebsiella pneumoniae* (ATCC 13883), *Proteus mirabilis* (ATCC 43071), *Pseudomonas aeruginosa* (ATCC 27857) and *Salmonella typhimurium* (ATCC 14028), all acquired by Newprov® in lyophilized disk. The bacterial suspensions were adjusted, using a sterile solution of sodium chloride 0.9%, to conform to the turbidity standard 0.5 of the Mac Farland scale ( $10^8$  CFC  $\text{mL}^{-1}$ ), according to the National Committee For Clinical Laboratory Standards (1997). A 0.1 mL aliquot of bacterial suspension containing  $10^8$  CFC  $\text{mL}^{-1}$  was mixed with 100 mL of sterilized solution of Tween 80 (2%) and this mixture was then used in the assay. A sequential dilution of the samples was used (2, 1, 0.5, 0.25, 0.125 and 0.075  $\text{mg mL}^{-1}$ ) and added to test tubes containing 1 mL of the TSB (Tryptic Soybean Broth). Subsequently, 1 mL of bacterial suspension, previously adjusted, was then applied to each tube. The negative control tube contained only the microorganisms without the samples, while the positive control tube contained the microorganisms and chloramphenicol (2  $\text{mg mL}^{-1}$ ). The whole process was repeated twice. The material was incubated to 37 °C for 24 hours. The minimum inhibitory concentration was defined as the lowest concentration where no microbial development occurred, verified by absence of turbidity in the tube containing the sample.

## RESULTS AND DISCUSSION

The dichloromethane fraction submitted to the silicagel column chromatography produced 15.2 mg of crystals in the form of needles that had yellow coloration (substance 1), while the ethyl acetate fraction produced 482 mg of crystals in the form of needles with white coloration (substance 2).

The structures of the isolated substances were determined using classic spectroscopic methods (UV,  $^1\text{H}$  RMN,  $^{13}\text{C}$  RMN, IR and mass spectrometric), analytical thin-layer chromatography with standards and comparisons to

the literature (Mabry *et al.*, 1970; Agrawal, 1992; Kerber, Silva, 1993; Saito *et al.*, 1994; Ogundaini *et al.*, 1996; Zapesochnaya *et al.*, 2002; Morimura *et al.*, 2006), which identified substance 1 as naringenin and substance 2 as 5- $\beta$ -D-glycosyl-naringenin (Figure 1).



**FIGURE 1** - Chemical structure of naringenin (I) and 5- $\beta$ -D-glycosyl-naringenin (II).

These compounds had not previously been isolated from *Acacia podalyriifolia*, but their presence had been noted in the flowers of *Acacia longifolia* (Kerber, Silva, 1993).

The evaluated samples obtained from the flowers of *A. podalyriifolia* showed antioxidant activity on the TBARS test, especially for ethanol extract (1000 ppm), which was the most active, followed by ethyl acetate fraction (1000 ppm), both demonstrating higher activity than that presented by reference compounds, ascorbic acid (1000 ppm), although were lower than the BHT (1000 ppm). Both isolated substances also showed activity, albeit less intense than those reported by their original fractions (Table I).

For the antibacterial activity, absence of turbidity in the test tubes containing TSB, sample and appraised bacterium, indicates that the sample material at the given concentration, showed an inhibitory effect for the growth of strains. As demonstrated in Table II, the ethanolic extract and the dichloromethane fraction became only slightly active, demonstrating inhibition to the strains of *Staphylococcus aureus* and *Staphylococcus epidermidis* with MIC of 1  $\text{mg mL}^{-1}$ . The ethyl acetate fraction inhibited the growth of *Staphylococcus epidermidis* with MIC of 0.125  $\text{mg mL}^{-1}$ , *Staphylococcus aureus* with MIC of 0.25  $\text{mg mL}^{-1}$ , *Streptococcus pyogenes*, *Klebsiella pneumoniae* and *Proteus mirabilis* with MIC of 0.5  $\text{mg mL}^{-1}$ . For isolated substances, there was inhibitory effect only with naringenin (MIC of 2  $\text{mg mL}^{-1}$ ) to *Staphylococcus aureus*, *Staphylococcus epidermidis*, *Klebsiella pneumoniae* and *Proteus mirabilis*. 5- $\beta$ -D-glycosyl-naringenin caused no inhibitory effect at the given concentrations.

The results obtained on these tests have demonstrated the ethyl acetate fraction was more active. Although the flavanone 5- $\beta$ -D-glycosyl-naringenin, the substance

**TABLE I** - Antioxidant Index (%), obtained by TBARS test, of reference compounds and extracts, fractions and isolated substances from the flowers of *Acacia podalyriifolia* A. Cunn. ex G. Don

Concentration	1000 ppm		500 ppm		100 ppm	
	AI	SD	AI	SD	AI	SD
BHT	35.15	3.42 ac	20.25	1.07 bc	10.73	0.59 cab
Ascorbic acid	21.73	1.77 af	14.87	1.44 bd	3.67	0.70 cdef
Ethanol extract	29.43	0.65 acde	18.05	2.01 bcd	13.07	1.91 ca
Ethyl acetate fraction	24.84	1.28 aef	16.45	1.73 bcd	5.98	1.38 ccde
Dichloromethane fraction	19.75	2.31 afg	7.60	1.42 bf	4.55	1.60 ccde
Naringenin	14.60	2.42 ag	8.60	1.31 bef	0.00	0.00 cf
5-β-D-glycosyl naringenin	27.74	2.50 ade	14.19	2.02 bd	1.63	1.55 cef

NOTE: AI = antioxidant index (%), data presented as mean of three experiments with medium values; SD = standard deviation. Differences between means indicated by the same letters are not statistically significant (Tukey's test,  $p < 0.05$ ).

**TABLE II** - Antibacterial activity, by determination of MIC, of extracts, fractions and isolated substances from the flowers of *Acacia podalyriifolia* A. Cunn. ex G. Don

SAMPLES (mg mL <sup>-1</sup> )	MIC (mg mL <sup>-1</sup> )							
	<i>S. aureus</i>	<i>S. epidermidis</i>	<i>S. pyogenes</i>	<i>E. coli</i>	<i>K. pneumoniae</i>	<i>P. aeruginosa</i>	<i>P. mirabilis</i>	<i>S. typhimurium</i>
EE	1	1	>2	>2	>2	>2	>2	>2
DCMF	1	1	>2	>2	>2	>2	>2	>2
EAF	0.25	0.125	0.5	>2	0.5	>2	0.5	>2
NAR	2	2	>2	>2	2	>2	2	>2
NARGLY	>2	>2	>2	>2	>2	>2	>2	>2

NOTE: >2 = absence of inhibition, EE = ethanolic extract, DCMF = dichloromethane fraction, EAF = ethyl acetate fraction, NAR = naringenin and NARGLY = 5-β-D-glycosyl naringenin, MIC = minimum inhibitory concentration, the lowest concentration where no microbial development occurs, verified by absence of turbidity in the tube with the sample.

isolated in greatest abundance from this fraction, has demonstrated antioxidant action, it showed no activity against the tested bacteria. It is possible that the presence of flavonoids and phenolic compounds in the ethyl acetate fraction, as demonstrated in the phytochemistry studies carried out by Andrade *et al.* (2003), or a possible synergism among these, could be responsible for the biological effects observed with these samples.

The amount of flavonoids 5-β-D-glycosyl-naringenin and naringenin in the flowers of *A. podalyriifolia* (2000 μg g<sup>-1</sup> and 71 μg g<sup>-1</sup>, respectively) was considerable. Numerous studies describe the pharmacological properties of these flavanones are an antioxidant effect (Acker, 2000), hepatoprotective (Lee *et al.*, 2004; Salgado *et al.*, 2007), anti-inflammatory (Bodet *et al.*, 2008), antiviral (Paredes *et al.*, 2003; Nahmias *et al.*, 2008), antihypertensive (Sapnara *et al.*, 2006) and antimutagenic (Choi *et al.*, 1994; Renugadevi, Prabu, 2009) effects. These studies demonstrate the many potential applications of these compounds, with

*A. podalyriifolia* an important source of these substances.

Further investigation of the ethyl acetate fraction extracted from the flowers of *A. podalyriifolia* may yield further discoveries.

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