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Galactans: An Overview of their Most Important Sourcing and Applications as Natural Polysaccharides

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ABSTRACT

Since last decades, lot of biological and rheological properties of polysaccharides and oligosaccharides were described. Among them, galactans and more especially sulfated galactans from seaweeds have shown interesting and specific properties not only as texturing agents but also as biological active compounds on several organisms. This class of polysaccharides includes classical sulfated galactans extracted from seaweeds and classified as agar and carrageenans. However, some galactans are more complex and their specific structural features have been characterized after their extraction from terrestrial plants, seaweeds but also animals and microoragnisms. This review catalogues the origins, structural characteristics and potentialities of these polysaccharides and their oligosaccharides derivatives.

Key words: Polysaccharide, seaweed, oligosaccharides, galactan

INTRODUCTION

Polysaccharides are natural biopolymers isolated from animals, plants, fungi, bacteria but also from Union seaweeds. The International Biochemistry and Molecular Biology (IUBMB) defined them as macromolecules consisting of a large number of monosaccharide residues joined to each other by glycosidic linkages. They have a large variety of structures and as a consequence, varied biological functions and properties such as: storage of energy (starch, glycogen), cell wall architecture (cellulose, pectins) or gel formation (mucilage). However, they present shared features such as their high density of -OH functions leading to a hydrophilic character and a high capacity to establish H-bond network. So, except some of them, polysaccharides are generally hydrosoluble and they extensively modify the rheology of aqueous media into which they are introduced, even at low concentrations. This is the basis of their functional properties as thickening and gelling agents. For this, they are actually exploited and valorised, notably in food industries. More, lot of studies have suggested that poly- and oligosaccharides revealed biological functions and were directly implicated in biological responses. So in this context, the identification of new and original polysaccharidic structures constitutes a new field of applications, even if at this time a very low part of natural polysaccharides is exploited. More, the increasing of knowlegdes

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relative to poly- and oligosaccharidic structures with biological functions could help significantly the glycocode decoding. Effectively in an "omic" era, glycomic appears as the next step after genomic and proteomic and the concept of glycome has been validated by glycobiologists (Turnbul and Field, 2007). Polysaccharides can be homopolymers and heteropolymers of neutral (pentoses and hexose) and/or anionic sugars (hexoses), substituted or not by non-sugars compounds, linear, or ramified. A part but not all of them have a regular chemical structure based on 2 up 8 sugars repeat unit. They are stereoregular and may adopt ordered conformation. These features lead to specific behaviours in solution. Resulting conformations are spirals, sheets, single, double and triple helices (Rinaudo, 2004).

To be developed for industrial applications, each new polysaccharidic structure has to find a niche market unexplored by traditional polysaccharides or their chemical derivates. Therefore, the search for new and/or competitive polysaccharidic agents from natural origins has led to developments in the field of agars, carrageenans and others attractive galactans. All these galactans defined as polymers of galactose, firstly detected in 1892 (Schulze and Steiger, 1892), were isolated from marine and terrestrial ecosystems. They have a lot of applications as stabilizer, viscosifier, gelling and emulsifying agents but only agar and carrageenans have been agreed as food adjuncts.

These sulfated galactans have no equivalent in terrestrial plants and may constitute up to the 70 % of the dry matter of some red seaweeds. At this prospect, these last 2 decades all the research efforts were done to optimize the conditions for their large scale up production from lot of original sources, for their structural investigations and for the characterization of their rheological and biological properties. This review catalogues the last advances in the topic of production of galactan polysaccharides and their potential applications.

GALACTANS FROM SEAWEED

Generalities

Many marine *algae* were characterized by their content in sulfated polysaccharides, which have no equivalent in land plants (Kloareg and Quatrano, 1988). Different classifications were proposed for seaweeds. However, the most useful is based on

pigmentation and food reserve. Species employed for extraction of commercial polysaccharide belong to the division Rhodophyta (red algae) and to the division Phaephyta (brown algae). Red algae (Rhodophyta) produced sulfated galactans, agars and carrageenans, which were the main components of their cell walls (Craigie, 1990). Note that other polysaccharides (floridean, mannan, xylan complex cellulose, and extracellular mucilages) were also described for their functions in the cell wall architecture or as food reserve. From more than 4000 red algae (Rhodophyta) species known at present, only about 70–80 species are used for industrial production of gelling galactans. The most important ones belonged to the 2nd subclass (Florideophyceae) of Rhodophyta division, orders Gelidiales and Gigartinales and mainly genera Ahnfeltia, Chondrus, Eucheuma, Furcellaria, Gelidiella, Gelidium, Gigartina, Gracilaria and Pterocladia (Witvrouvw and De Clercq, 1997). Their galactans were abundantly described (Table 1), with a great attention for these of Gracilaria known as a producer of agar (Table 2).

They are made up of linear chains of galactoses and except some of them, their backbone is a linear chain of β-D-galactopyranose residues linked through positions 1 and 3 (A units) and α -D/L-galactopyranose residues linked through positions 1 and 4 (B units) arranged in an alternated sequence (AB)n (Fig. 1). They can be ramified by other neutral sugars (Xyl, Glc, Ara, Man), can carry methyl ether groups, sulfate hemiester groups and pyruvic acid linked as a cyclic ketal. Some of these galactans have their B units totally or partially replaced with 3,6anhydro-L/D-galactopyranose after elimination of sulfate from C6 by enzymatic or alkalin treatment. In relation with all these structural features, sulfated galactans from seaweeds are classified in three classes: agarocolloides, carrageenans and complex galactans.

The agarocolloids

The agarocolloids have been divided into two groups: agars and agaroids (Craigie, 1990). The main difference between them lies in the fact that among agars, the part of $4-\alpha$ -D-galactopyranose (B unit) in the form of derived 3,6-L/D-anhydrogalactose is very high and the percentage of sulfate groups are weaker (around 2%) than among agaroids (20%).

Table 1 - Main red algae producers of galactans.

Sources	Sugar units	Substitutions	Linkages	Yield	References
Callophyllis variegata	D-Gal, D-AnGal Xyl, Glc	Sul	β -(1,4) α -(1,3)	16.3%	Rodriguez et al., 2005
Phacelocarpus peperocarpos	D-Gal, D-AnGal Xyl, Glc	Sul	β-(1,4) α-(1,3)	24-45%	Liao et al., 1996
Eucheuma denticulatum	D-Gal, D-AnGal	Sul	β -(1,4) α -(1,3)	24%	Viana et al., 2004
Schizymenia binderi	D-Gal, Glc, Xyl	Pyr, Sul	α -(1,3) β -(1,4)	17.6%	Matsuhiro et al., 2005
Claviclonium ovatum	Gal, Glc, Xyl	Met, Sul	-	32%	Chiovitti et al., 2004
Sarconema filiforme	Gal, Xyl, Ara Man, Glc	-	=	6-13%	Chiovitti et al., 1998
Ahnfeltia tobuchiensis	Gal, An-Gal	Met, Sul	-	-	Truus et al., 2006
Grateloupia indica	Gal, Glc, Xyl, Fuc	Sul, Pyr	α -(1,3) β -(1,4)	13%	Chattopadhyay et al, 2007
Bostrychia montagnei	D-Gal, L-Gal, AnGal, Xyl	-	α -(1,3) β -(1,4)	-	Duarte et al., 2002
Pachymenia lusoria	D-Gal, L-Gal, AnGal, Xyl	Sul, Met, Pyr	α -(1,3) β -(1,4)	-	Miller et al., 1995
Codium yezoense	D-Gal, Arab, Xyl	Sul, Pyr	β -(1,6), α -(1,3)	-	Bilan et al., 2007
Gymnogongrus torulosus	D-Gal, D-Gal, AnGal, Xyl, Glc	Sul	α -(1,3) β -(1,4)		Estevez et al., 2001

Table 2 - Gracilaria agar producers.

Species	Sugar units	Linkages	Substitution	Yield	References
Gracilaria edulis	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	-	Villanueva
	AnGal	• , ,			et al., 1999
Gracilaria salicornia	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	2.9+/-0.1% - 15.7+/-1.3%	Calumpong
	AnGal	•			et al., 1999
Gracilaria arcuata	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	2.9+/-7% - 21.7+/-0.7%	Calumpong
	AnGal	•			et al., 1999
Gracilaria blodgettii	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	20.7+/-5.9%	Calumpong
	AnGal	•			et al.,1999
Gracilaria cornea	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	20.1+/-1.5%	Freile-
	AnGal	• , ,			Pelegrim, 2000
Gracilaria sp	D-Gal, L-Gal,	β -(1,3) α -(1,4)	Sul	-	Friedlander
	AnGal	•			et al.,1984

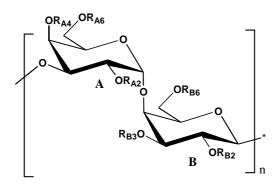


Figure 1 - Schematic and global representation of sulfated galactans from red seaweeds. B unit is in D configuration. R_{A2}: H, SO₃⁻; R_{A4}: H, SO₃⁻; Pyruvic acid (cyclic ketal with 0₆); R_{A6}: H, CH₃, SO₃⁻; Pyruvic acid (cyclic ketal with 0₄), R_{B2}: H, CH₃, SO₃⁻; R_{B3}: H, R_{A6}: H, SO₃⁻.

Agar

Agar, currently known as "agar-agar", a Malaysian name for seaweeds, was extracted from red

seaweeds belonging to *Gelidium*, *Pterocladia*, *Petrocladiella*, *Gracilaria* and *Hypnea* species (Rinaudo, 2007). Structurally, agar is made up of a

complex series of polysaccharides going from a neutral one, the agarose, to a highly charged galactan: the agaropectin. Agarose (sometimes called agaran) is a linear polymer with the AB alterning sequence called agarobiose where the B unit is 3,6-α-L-anhydrogalactopyranose (Fig. 2). This polymer with a molecular weight of 120000 Da has low sulfatation degrees (about 2%) but the C6 of the A unit may be highly methylated (up to 20%). The major differences between agarose and carrageenans was described as the presence of 3,6- α -L-anhydrogalactopyranose rather than 3,6- α -Danhydrogalactopyranose and the lack of sulfate groups (Rees, 1969). Agaropectin heterogeneous mixture of smaller molecules that occur in lesser amounts. They were structurally characterized by the same backbone that agarose with many acid groups such as sulfate, pyruvate but also glycuronate (Furneaux and Stevenson, 1990; Knutsen et al., 1994). Agaropectin were also methylated (Akari, 1996; Rinaudo, 2007).

Agar and agarose are not soluble in water at room temperature and they are abundantly exploited for their gelling properties. Agar is extracted from red algae in boiling water with or without alkali leading to polysaccharides with various rates of $3.6-\alpha$ -L-anhydrogalactopyranose.

After its solubilisation in hot water (85 to 100 °C), agar solidifies in the range from 30 to 40°C. Gelation of agar involves conversion from a fluctuating disordered coil conformation in water solution to a rigid ordered structure (co-axial double helix) which forms the junction zones of the gel network. Agarose gives the stronger gels.

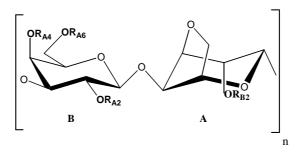


Figure 2 - Schematic representation of agarobiose and agaropectin. R_{A2} : H, SO_3^- ; R_{A4} : H, SO_3^- ; Pyruvic acid (cyclic ketal with 0_6); R_{A6} : H, CH_3 , SO_3^- ; Pyruvic acid (cyclic ketal with 0_4), R_{B2} : H, CH_3 , SO_3^- .

The agaroids

The agaroids polysaccharides are close to the ideal structure of agarobiose (Fig. 2), but they include fewer units of 3,6-anhydrogalactose, replaced by 4- α -L-galactopyranose. The rate of their sulfate groups can reach up to 20% generally placed on C6 of α -L-galactopyranose. Among these agaroids two examples can be cited: (i) that of funorans whose proposal A and B units are, as for agars, D-galactose and 3,6-L-anhydrogalactose, but whose C6 and C2 hydroxyl of 3- β -D-galactopyranose and

4-α-L-galactopyranose are substituted by sulfates or methyl groups (Fig. 3); (ii) that of porphyrans, produced by *Porphyra umbilicalis*. Porphyran is a highly substituted agarose-type. It was described constituted of 4-linked 6-O-sulfo-α-Lgalactopyranose (the precursor of the 3,6- anhydro unit) and 3-linked 6-O-methyl-β-Dgalactopyranose (Fig. 4) (Morrice et al., 1983). Note that funoran extracted from the kind Gloiopeltis was used as adhesive (Hirase and Watanabe, 1972).

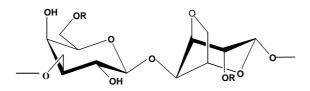


Figure 3 - Structure of furonan R: CH₃; SO₃.

Figure 4 - Structure of porphyran.

CARRAGEENANS

Kappa and Lambda families

Carrageenans are linear galactans with D-galactose residues α -(1,4) and β -(1,3) linked alternately. In this repeating unit called carrabiose the galactose residues α -(1,4) linked (A unit) often occur as 3,6-anhydro-D-galactose (Fig. 5). Sulfate ester groups were found on some or all galactoses (Knutsen et al., 1994; Rees, 1969). It is evident that since carrageenan is a large molecule being made up of

some 1000 residues, the possibility for structural variations is enormous. The carrageenans have been categorized in several families and subfamilies according to the number and position of sulfate ester, rate of (3,6) anhydro-bridges in the α -related residues and on their solubility in solutions of potassium chloride (Smith and Cook, 1953). They were characterized as high molecular weight polysaccharides with 20% to 40% sulfate-ester content (Kloareg and Quatrano, 1988).

Figure 5- Schematic representations of carrabioses.

Higher levels of sulfate ester meant higher level of solubility and lower gel strength. The first group forming gels has been defined as Kappa family and included iota (ι) and kappa (κ) carrageenans as well as their precursors: the X (μ) and Y(ν) carrageenans respectively (Anderson et al., 1973). All the members of this family were sulfated in C4 position of the A unit. The main differences

between the κ and ι subfamilies were a C2 sulfated B unit in the case of ι and an higher 3,6-anhydrogalactose content (28-35%) for κ carrageenans compared to ι one (25-30%). The precursors forms of κ (μ) and ι (ν) subfamilies had the same features except the presence of 3,6-anhydrogalactose residues replaced by a 6-O-sulfated B unit. The second family, called λ

carrageenans is not sulfated at C4 position of the B unit and without 3,6-anhydrogalactopyranose. It is the most sulfated carrageenan with 3 sulfate groups per repeating unit. These ester sulfates are located on C2 of A unit and in C2 and C6 of B units. This polysaccharide with highly anionic load owned thickening properties but not gelling ones (Rees, 1969; Greer and Yaphe, 1984). The λ family contains subfamilies including xi (ξ), theta (θ) and pi (π)-carrageenans, where only the 4- α -D-galactopyranose unit varies (Fig. 6). The theta structure has a B unit only sulfated at C2 position.

It was obtained by slow alkaline transformation of the lambda carrageenan (Ciancia et al., 1993). This structure was also detected naturally present in some algae such as *Callophyllis hambroniana* (Miller, 2003; Falshaw et al., 2005).

The other structures of this family as desulfated ξ carrageenan (Penman and Rees, 1973) and pyruvated π carrageenan (DiNinno et al., 1979) have been observed in tetrasporophytes from different species of Gigartinaceae (McCandless et al., 1983; Falshaw and Furneaux, 1995; Falshaw and Furneaux, 1998).

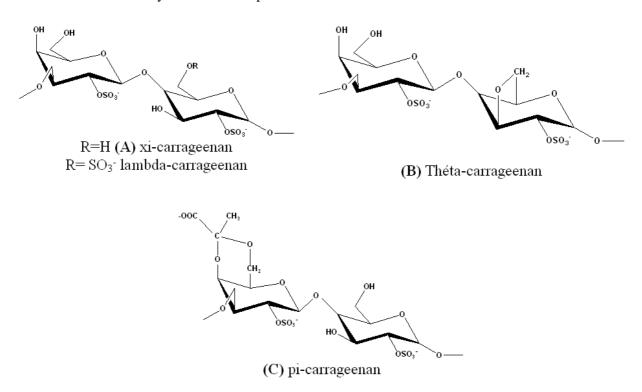


Figure 6 - Schematic representation of lambda-carrageenans. (A): xi-carrageenan, (B): theta-carrageenan, (C): pi-carrageenan

Carrageenan hybrids.

Each carrageenan extracted from algae could be considered as a hybrid, comprising a regular sequence of the same diholoside interrupted by segments of another type of carrageenan (Bellion et al., 1982). It was sometimes difficult to estimate whether a polysaccharide is a single molecule or a hybrid of two distinct carrageenans (Craigie, 1990). It is also interesting to note that in general, the nature of carrageenan produced by algae depends on the different generations of biological cycle. For example, the species belonging to *Gigartinales* order may contain several types of

carrageenans and in one case, the structural variability associated with the alternation of generations especially considering levels of sulfate groups and 3,6-anhydrogalactoses, has been highlighted. McCandless et al. (1982, 1983) have studied the carrageenans of *Phyllophoraceae* and *Gigartinaceae* and have shown that the parietal polysaccharides generated by gametophyte were a hybrid form of t/κ or κ/t depending on the species, while carrageenan from tetrasporophyte were rather λ subfamilies forms (λ , ξ and π). This trend was confirmed since some types of *Gigartinaceae* such as *Chondrus*, *Gigartina*, *Iridea* or

Rhodoglossum were gametophytes producing κ or ι carrageenan, while their tetrasporophytes contained λ or ξ types (Chen and McLachlan, 1972; Matsuhiro and Urzua, 1992; Matulewicz et al., 1989; 1990; McCandless and Craigie, 1974; McCandless et al., 1973, 1983; Pickmere et al.,

1973). More recently, Falshaw et al. (2001) have also demonstrated for the Chilean red *algae* Sarcothalia crispata and Gigartina skottsbergii the presence of hybrid carrageenans. The various species which produce the hybrid carrageenans are summaries in Table 3.

Table 3 - Carrageenan hybrid producers.

Carrageenan type following the biologic cycle					
Species	Gametophyte	Tetrasporophyte	References		
Gigartina skottsbergii	κ and ι or (μ and ν)	λ	Falshaw et al., 2001		
Sarcothalia crispata	κ and ι or (μ and ν)	λ	Falshaw et al., 2001		
Gigartina atropurpurea	κ and ι or (μ and ν)	λ	Falshaw et al., 2003		
Gigartina clavifera	κ and ι	ξ	Falshaw et al., 1995		
Gigartina alveata	κ and ι	ξ	Falshaw et al., 1995		
Gigartina pistillata	κ , ι and ν	λ , ξ and π	Falshaw, et al., 1998		

COMPLEX GALACTANS

Can be classified as "complex galactans" linear backbone of galactoses arranged with the previous described alterning sequence AB and ramified by other hexoses and/or pentoses. Complex galactans have been extracted, purified ans characterized Halymeniales from the red alga Cryptonemiales) families such as: Grateloupia elliptica, Pachymenia carnosa and Pachymenia lusoria (Hirase et al., 1967, Farrant et al., 1972; Miller et al., 1995). Apart the Halymeniales family, certain genus of other families of red alga contained complex galactans such as Asparagopsis armata (Garon-Lardiere, 2004). A highly complex structure has been also isolated from Kappaphycus alvarezii (Estevez et al., 2000). In this case, the polysaccharide contained D and galactopyranoses as B units and was ramified by galactose, glucose and xylose linked to C2 and/or C4 of the 3-linked β -D-galactose units and C3 and/or C2 of the 4-linked α-D- and/or α-Lgalactose residues (Estevez et al., 2000). The second well documented complex galactan was the sulfated, pyruvilated and methylated polysaccharide from the red algae Pachymenia Lusoria. Its structural analysis showed a backbone of 3-1inked D-galactopyranoses alternating with 4-1inked D- or L-galactopyranosyl residues. The 3linked residues were nearly all 2-O-sulfated. Considering this A unit, it was 6-O-methylated and 4,6 pyruvylated all the 3 and 6 residues respectively. Analysis of 4-1inked galactopyranoses revealed that half were 2-Omethylated. The rest of them were galactopyranosyl residues in D (50%) and L (50%) configurations (Miller et al., 1995). Other structures of complex galactans were summarized in the Table 4.

Table 4 - Structures of complex galactans.

Species	Gal Configuration	Structure	Substituent monosaccharide	Substituent group	References
Cryptonemia	DL-hydrid	3)- β -D-Galp-(1,4)- α -	β-D-Xyl, β-D-	Sul Met Pyr	Zibetti et al.,
crenulata		Galp-(1,	Gal		2005
Kappaphycus	DL-hydrid	3)- β -D-Galp-(1,4)- α -	β-D-Xyl, β-D-	Sul Met	Estevez et al.,
alvarezii		Galp-(1,	Glu		2004
Porphyra	DL-hydrid	3)- β -D-Galp-(1,4)- α -		Sul Met	Brasch et al.,
columbina	•	Galp-(1,			1981
Gymnogongrus	DL-hydrid	3)- β -D-Galp-(1,4)- α -	β-D-Xyl	Sul	Estevez et al.,
torulosus	•	AnGalp-(1,			2008
Gloiopeltis	DL-hydrid	3)- β -D-Galp-(1,4)- α -		Sul Met	Takano et al.,
furcata	•	Galp-(1,			1998
Grateloupia	D	3)- α -D-Galp-(1,4)- α -	Fuc, Glu, Gal,	Sul	Chattopadhyay
indica: Î		Galp-(1,	Xyl		et al., 2007
Pachymenia	DL-hydrid	3)- β -D-Galp-(1,4)- α -	Xyl	Sul Met Pyr	Miller et al.,
lusoria		Galp-(1,			1995

Physico-chemical properties of seaweed galactans

The sulfated galactans such as agarose or carrageenans are extensively exploited imputable their industrial applications. These biopolymers are considered as the major hydrocolloids used as texturing agents for food industrial application. The wide employments of these marine sulfated polysaccharides are essentially based on their unique properties to form strong aqueous gels (Ress et al., 1982; Lahaye, 2001). It was well known that modifications in the backbone of sulfated galactans allowed to greatly changes in their physico-chemical and biological properties (Lahaye, 2001). For example, just a small conformation/configuration change from α-Dgalactopyranoses in carrageenans galactopyranoses in agaran could promote very important modifications in these polysaccharides (Lahaye, 2001). Moreover, it was largely defined that a higher levels of 3,6-anhydro-α-Lgalactopyranosyl units in agarans and a lower sulfate contents were the major structural requirements for gelling (Lahaye, 2001).

In gelling applications, agarose, κ -, and t-carrageenans form thermoreversible gels in aqueous solutions. Agar is known to form a very porous gel and the pore size can be roughly measured by assessing the size of particulates that are excluded from the gel in a gel permeation experiment. It has been shown that agar gels could allow to molecules up to 30000 daltons to percolate through their gelling network. An agar gel has the unusual property of behaving like a sponge. It can be dried and upon rehydration it will swell to its original size and shape (Ress et al.,

1982). The carrageenan would cause excessive flocculation due to the protein reactivity. The reactivity with proteins depends on gelling properties of carragenans but also on their structural regularity. In most, cases ion-ion interactions between the sulfate groups of polysaccharides and the cationic character of protein are involved. For this, in food systems, one of the most important properties that truly differentiate carrageenan (ι and κ) from other hydrocolloids is its ability to complex or interacts with proteins. In milk protein systems, at peripheral locations on the casein micelle there is a concentration of positive charges attracting the negatively-charged sulfate groups of carrageenans to form linkages among the dispersed casein micelles (Fig. 7). This reaction, in combination with the normal water gelling capabilities of carrageenan, can increase the gel strength about 10-fold. Thus, cationic groups allowed to determine the physical properties of carrageenans as κ - and ι - but not λ -carrageenans formed gels in presence of potassium or calcium ions (Morris et al., 1980; Rees et al., 1982).

As well described for many years, gelling in carrageenan is caused by helix formation occurring in repeat structures where the B residue is in a ${}^{1}C_{4}$ conformation. Lambda carrageenan with its two sugar residues in a ${}^{4}C_{1}$ conformation and does not form gels. All the gelling types of carrageenan which include κ , β and τ contain a 3,6 anhydro bridge on the B unit which forces the sugar to flip from a ${}^{4}C_{1}$ conformation to a ${}^{1}C_{4}$ one forming then cross-link networks and gels (Morris et al., 1980; Rees et al., 1982).

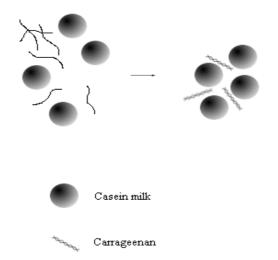


Figure 7 - Interaction of milk protein with carrageenan.

GALACTANS FROM PLANTS

It was well defined that plants could store a very large number of carbohydrate polymers. Starch is the principal polysaccharide, followed by fructans and cell wall storage polysaccharides defined as CWSP (McCann and Roberts, 1991; Carpita and Gibeaut, 1993). As mentioned in the plant cell wall model, it was shown that CWSP were mainly present in seeds (Reid, 1985; McCann and Roberts, 1991; Carpita and Gibeaut, 1993). Therefore, a classification in CWSPs revealed 3 polysaccharide families: mannans, xyloglucans and galactans (Table 5). The later were mainly carbohydrate chains branched rhamnogalacturonan belonging to the matrix of pectins (McCann and Roberts, 1991; Carpita and Gibeaut, 1993). Historically, seed cell wall galactans were first detected in 1892 but the characterization of β -(1,4)-galactan as a distinct polysaccharide in seeds of Lupinus albus was demonstrated by Hirst in 1947 (Hirst et al., 1947). By studying four species of lupins, Al-Kaisey and Wilkie (Al-Kaisey and Wilkie, demonstrated the presence of two types of galactan polysaccharides in their seeds. So, a α -(1,3), (1,6)linked galactan with some 4-linkages and a α -(1,4) galactan substituted by L-arabinofuranose were characterised. Lupinus polyphyllus and Lupinus

angustifolius L. seeds contained a galacto-arabino rhamnogalacturonan (Al-Kaisey and Wilkie, 1992; Cheetham et al., 1993). Consequently, Lupinus polyphyllus seeds were currently used to produce larger amount of galactans. Finally, note to mention that linear β -(1,4)-galactans were isolated from the lemon peels (Labavitch et al., 1976), from potato tubers (Wood and Siddiqui, 1972), from Norwegian spruce compresssion wood (Bouveng and Meier, 1959) and from Strychnosnux-vomica L. seeds (Andrews et al., 1954).

Brillouet and Riochet (Brillouet and Riochet, 1983) have described the correlation between the galactose amount and rhamnose:galacturonic acid ratio in the cell wall polysaccharides of lupin cotyledons. This correlation was easily attributed to an increase in sites (rhamnosyl residues) for anchoring galactan side chains to the pectic backbone. In fact, galactans associated with arabinans, as arabinogalactan, constituted the neutral pectic substance present at low amounts in plant cell wall. Pectic galactans were characterized as β -(1,4)-linked linear polymers which could possess a small number of 6-linked residues (Ghosh and Das, 1984). The arabinogalactans are present in two distinct types of macromolecules in plant cell walls.

Table 5 - Galactan plants producers.

Species	Sugar units	Linkages	Substitutions	References
Acacia glomerosa	D-Gal, L-Ara	β-(1,3)	Met	León de pinto et al., 2001
Acacia tortuosa	Gal, Ara, Xyl, AcGlu	β-(1,3)	Met	León de pinto et al., 1998
Strychnos sp	Ara, Gal, Rha	β -(1,3)-Galp, β -(1,4)-Galp α-(1,4)-Araf, α-(1,5)-Araf	-	Corsaro et al., 1995
Acacia macracantiha	Gal, Ara, Rha, AcGlu	β-(1,3)	Met	Martinez et al., 1996
Raphanus sativus	Ara, Gal, Glc, Xyl	β -(1,3), β -(1,6)	-	Tsumuraya et al., 1988
Pinus radiata	Gal, Ara	-	-	Fenemor, 1984
Atractylodes lancea	Ara, Gal, Xyl, Rha, Fuc, Man, Glc	β-(1,3), β-(1,6)	-	Yu et al., 2001
Angelica acutiloba	Ara, Rha, Gal, Glc	β -(1,3), β -(1,4)	-	Zhang et al., 1996

Arabinogalactan I

Perez in 2000 described type I arabinogalactans (AGs) which are abundant in primary walls of dicotyledonous plant tissues including many fruits (Perez et al., 2000). They comprise a main backbone of (1-4)-linked β -D-Galp to which short

side chains of (1-5)-linked α -Araf are attached at the O3 position. Ferulic acid has been found attached to some arabinose and galactose residues in some tissues such as sugar beet (Ralet et al., 2005).

Arabinogalactan II

Type II Arabinogalactans are highly complex polysaccharides which are generally associated with proteins and described as arabinogalactan proteins (AGPs). They are widely distributed in the plant kingdom, and they are one of the most structurally complex macromolecules in nature. These proteoglycans showed typically molecular weights between 60 to 300 kDa, a composition including more than 90% of carbohydrates and a core polypeptide that was usually rich in Hyp, Ala, Ser, and Thr (Serpe and Nothnagel, 1999).

Type II Arabinogalactans habe been extracted along with pectic polymers with they were probably covalently linked on rhamnogalacturonan (Vincken et al., 2003). They exhibit highly complex structures comprising a highly branched galactan core of (1,3)- and (1,6)-linked β -D-Galp residues. Short side chains of (1,6)-linked β-D-Galp including between one and three residues in length are present. Galactosyl residues in these side chains are often substituted with terminal α-L-Araf attached at O3 or O6. Small amounts of GlcpA may also be present. A low molecular weight type II arabinogalactan associated with a hydroxyproline-rich peptide (arabinogalactan proteins, AGPs) was isolated from wheat flour (Fincher and Stone, 1974).

Finally, sulfated galactan have been isolated from marine plants such as the marine sea grass *Rupia maritima* (Aquino et al., 2005). This family of marine plant is a vascular flowering plant (Angiospermae, Spermatophyta) known to grow in highly saline marine environments. As described by Aquino et al. (2005), the chemical backbone structure of *Rupia maritima* sulfated D-galactan was only composed of tetrasaccharide repeating units made up of [,3)-β-D-Galp-2(OSO₃)-(1,4)-α-

D-Galp-(1,4)- α -D-Galp-(1,3)- β -D-Galp- $4(OSO_3^-)$ -(1]n. As mentioned by authors, this galactan is a regular structure contrary to red algae galactan sulfate where glycosidic units are distributed in an alternating sequence. It was suggested that the sulfo ester of this galactan could influence the caption of salt ions from sea water in order to physiologically regulate the osmosis in the marine plante (Aquino et al., 2005).

GALACTANS FROM ANIMAL ORIGINS

As presented previously, sulfated galactans are largely distributed in nature in a great variety of organisms. In the animal kingdom, sulfated polysaccharides such as glycosaminoglycans have been identified in large amounts in vertebrate tissues (Assaro and Dietrich, 1977; Mathews, 1975) and invertebrate species as *Chordata* tunicata. Ciona intestinalis and Herdmania monus which constituted a rich source of sulfated polysaccharides with original structures (Mourao and Perlin, 1987; Pavao et al., 1998). Therefore, lot of studies have described that the marine invertebrates ascidians families contained a very high quantity of sulfated galactans (Mourao and Perlin, 1987; Pavao et al., 1989; Santos et al., 1992; Pavao et al., 1998). Note that these sulfated galactans from invertebrates showed the same structural pattern of simple and well-defined units found in invertebrate sulfated fucans from seacucumber and from the egg jelly coat of seaurchins (Pomin, 2009). Then, the regulated backbone structures of theses sulfated galactans was characterized as $[,3)-\alpha-L-Galp-2(SO_3)-(1,]n$ from Echinometra lucunter (Alves et al., 1997); [4)- α -L-Galp-2[1)- α -L-Galp-3 (OSO_3)]-3 (OSO_3) -(1,]n from Styela plicata (Mourao et al., 1987); [,4)- α -L-Galp-3(SO₃)-(1,]n from Herdmania monus (Santos et al., 1992) and sulfated [3)-β-D-Galp-(1,4)-α-Gal-(1,]n from both *Botryocladia* occidentalis and Gelidium crinale (Fonseca et al., 2008).

GALACTANS FROM MICROBIAL ORIGINS

In the light of the precedent paragraphs and as many of us know, the polysaccharides find origin in all the nature. The polysaccharides market has been booming in terms of biotechnological operations and value-addition. To meet the constantly increasing demand and to produce the newer types/properties, the production of biopolymers by the bacterial fermentation appears as an important alternative to the animal, plant and seaweed exploitation. Some important advantages

of the bacterial polysaccharides are: (i) easy way of production without any contraint of seasons, extraction and purification without use of the drastic conditions, etc. (Delattre et al., 2009). These polysaccharides from microorganisms have been described for their biological activities (Sekina et al., 1985). Consequently, to adapt to today's needs in biotechnology field, screening of new and original microorganism from a wide part of living organism environments has become the new challenge of modern microbiotechnologists. Some bacteria are known as producer of galactans. For example Methylobacterium sp. producing a 100% galactan (D-galactopyranose) was isolated by Verhoef et al. (2003). Other galactan could be produced by Lactococcus lactis subspecies cremoris H414 (Gruter et al., 1992) or Lactococcus lactis subsp. cremoris B891 (Van Casteren et al., 2000). Finally, it is interesting to mention that as galactan from seaweed, the bacterial galactan from Bifidobacteria catenulatum presented biological activities and notably effect in healing gastric ulcers (Nagaoka et al., 1994).

GALACTANS FROM FUNGAL ORIGINS

Literature speaking about galactan from fungi is little but exists. The genus *Pleurotus* is known as a producer of this polysaccharide. Looking at structures, the main chains consisted of 6- α -D-galactopyranoses in the case of *Pleurotus eryngii* (Carbonero, 2008) and *Inonotus levis* (Vinogradov and Wasser, 2005), but 4- α -D-galactopyranoses were detected for *P. ostreatoroseus* (Rosado et al., 2003). In all cases, parts of galactose were methylated in position 3.

MAIN APPLICATIONS OF POLYSACCHARIDES IN BIOTECHNOLOGY

Red algae polysaccharide in biotechnology: example of sulfated galactans as bioactive compounds

The seaweed, including red *algae* is used since a long time in the world for food, medical, and more recently, biotechnology. An assessment was conducted in 1999 (Zemke-White and Ohno, 1999) concerning the production and use of

seaweed around the world for the years 1994-1995. They determined that about the 221 species of seaweed used, 145 were for food and 101 for the production of phycocolloids. 32 were Chlorophyceae (green algae), 124 were Rhodophyceae (red algae) and 65 were Pheophyceae (brown algae). The origin of the algae is twofold: either they are derived from natural production worldwide (2 million tons of dry weight), or they come from culture, which tonnage represents 1 million tons of dry weight. It is interesting to note that since 1984 the global production of algae grew by 119%, thus underscoring the continued growth in the use of algae. Apart the industrial uses, some of these algae were directly used for human consumption for their great nutritional interest (Zemke-White 1999). Ohno. Regarding industrial applications of galactans extracts from algae, they are generally around the texture (gelling and thickening properties), but now they are also used as cosmetic and biomedical products.

Unconnected to their natural biological implication in animal, algae and plants, galactans could act as potential pharmacological product in mammalian systems. For example, sulfated galactan could modulate: antiviral (Harrop et al., 1992; Pujol, 1996; Carlucci et al., 1997; Caceres et al., 2000), antimetastatic (Coombe et al.. 1987) antiangiogenic (Cumashi 2007). et al., antiinflammatory (Berteau and Mulloy, 2003); antiadhesive (Berteau and Mulloy, 2003), anticoagulant (Pereira et al., 2002) antithrombotic (Mourao and Pereira, 1999; Berteau and Mulloy, 2003) activities.

Nevertheless, in some cases, the polysaccharide application on animal tissues induced only a weak biological activity, without doubt due to the polymer high molecular weight (Delattre et al., 2005). In fact, the large size of polysaccharides locally applied on animal tissues could be a problem for their diffusion inside them (Delattre et 2005). So. low molecular polysaccharides sequences such as oligosaccharide structures could interact more efficiently with specific membrane receptors as described for example in the case of heparan sulfate (Yu et al., 2000). Consequently, oligosaccharides from depolymerization of galactans were used because of their high biological activity on cells (Delattre et al., 2005).

Oligogalactans from polysaccharides and their applications in biotechnology

It is now well established that carbohydrateprotein interactions play a major rule in numerous biological processes (Delattre et al., 2005; Varki, 1993). Then, it appeared practical to exploit the potential of bioactive polysaccharides and their derivatives such as oligosaccharides and low molecular weigh (LMWs) polysaccharides. Generally, oligosaccharides are mostly considered for their potentialities as cellular signalling agents. Therefore, some oligosaccharides acting like signalling molecules (the term hormone is sometime employed) on various organisms arouse actually a great interest. Then, their role in cellular development is very important notably in cellular and subcellular communication, signalisation and interaction (Delattre et al., 2005; Varki, 1993). Lot of oligosaccharides have been described as being implicated in a wide variety of biological activities such as anticoagulant, cofactor for growth factor, production. cvtokines and chemokines tumorigenesis, signalling molecules in responses to infection or other cellular damages, regulator of blood coagulation, against viral and bacterial infections (Delattre et al., 2005). Then, according the important relationship between oligosaccharides structures biological and functions, a lot of industrial applications can be investigated. This context has led to worldwide research in the area of oligosaccharides production. However, even if prospective polysaccharides applications of were demonstrated, very few studies and biological applications have been developed oligosaccharides produced by depolymerisation of polysaccharides and more especially sulfated galactans.

Lot of evidences have indicated that the molecular weight (Mw), structural features and content in sulfur groups of seaweed galactans were important factors which influence their biological behaviors (Franz and Alban, 1995; Sugawara and Ishizaka 1982). Therefore sulfated galactans such as carrageenans and their depolymerized products, have been shown to possess biological activities such as immunomodulation and antitumor activity galactan (Yuan et al., 2006). Sulfated oligosaccharides were described for their antitumor activities on Ehrlich ascites cells, Meth-A tumor, and mammary adenocarcinoma (Coombe et al., 1987; Noda et al., 1989).

Among mammalian, glycosaminoglycans (GAGs) and more essentially oligo-glycosaminoglycans from heparin, heparan sufate, hyaluronate and chondroitin sulphate have been identified as implicated in a wide variety of biological processes as, regulator of blood coagulation, growth factor, tumorigenesis, cofactor for cytokines and chemokines production, assisting viral and bacterial infections, and signalling molecules in responses to infection or other cellular damages (Pineo and Hull, 1997). Consequently, biological activities oligosaccharides from plant and algae suggest that these compounds may have a variety of actions on animal cells. In fact, in regard to their homologies with GAG structure, sulfated polysaccharides (galactan, ulvan and fucoidan) extracted from marine algae were considered as sources of marine molecules with potential applications in medicine (Farias et al., 2000).

MAIN STRATEGIES DEVELOPED FOR THE PRODUCTION OF OLIGOSACCHARIDES

Even if the potentialities of oligosaccharides are unquestionable, in most cases, the lack of oligosaccharides scale up production strategies is the main drawback limiting their applications. Consequently, the improvement oligosaccharides engineering strategies arouses currently a great interest. Different industrial ways can be investigated for the scale up production of oligosaccharidic structures. Generally, bioactive oligosaccharides come from either (i) synthesis (using enzymatic or chemical engineering): or (ii) polysaccharide depolymerization (using physical, chemical or enzymatic methods). The best way to degrade the polysaccharides in homogenous oligosaccharides structures corresponds to the bioconversion of polysaccharide by enzymic degradation. So, polysaccharide cleavage enzymes widely exploited to degrade various polysaccharides, as polysaccharide hydrolases used to degrade neutral polysaccharides, and polysaccharide hydrolases and lyases used to degrade anionic polysaccharides.

Various physical methods are commonly used to depolymerize a large amount of polysaccharides. From them, microwave irradiation was developed, firstly in the extraction of plant polysaccharides and secondly in oligomers production strategies. The main advantages of microwaves irradiations are to produce an efficient thermal energy source for shorter reaction times and so, to cleave the polysaccharides. Nevertheless, a low proportion of oligosaccharidic families are generally produced. In this way microwaves treatment applied to *algae* galactans conducted to the production of various low molecular weight polysaccharides with antitumor activities (Zhou et al., 2004).

Many polysaccharides could be chemically depolymerised by acidic treatment to produce a large amount of oligosaccharides. The depolymerization of polysaccharides is generally realized by heating in diluted or concentrated acid (2M) such as HCl, H₂SO₄, TFA, formic or nitrous acid. That's the way, a lot of oligogalactans such as oligo-carrageenans (Hjerde et al., 1994), oligoarabinogalactans from the wood of western larch (Ponder, 1998), and smalls galactans chains from Lupin seeds (Vogl et al., 2000) were produced and isolated.

Even if all these physico-chemical methods are very efficient to cleave glycosidic linkages, there are some drawbacks. In fact, polymers cannot be totally converted into oligomers (but rather LMW) and these last are in majority randomly produced. Consequently, for all these disadvantages, enzymatic depolymerization is generally a good alternative to resolve the specific oligosaccharides productions. Then, many genera of marine bacteria have been employed to produce galactanases, such as *Alteromonas agarlyticus* (Potin et al., 1993) or *Vibrio* (Araki et al., 1998).

CONCLUSION

To conclude, often described as very simple macromolecules extracted from seaweeds and used in industry for their rheological properties, galactans have been identified as very complex polysaccharides. Their natural functions are not always clear and are probably not only limited to structural function in plant cell wall. Effectively, more and more papers have detailed galactans and more especially sulfated galactans as biological active compounds in various cell types or about full organisms. Even if the sufate content appeard as an essential structural feature associated with biological activity, it is probably not the sole as not sulfated galactans from terrestrial plants showed interesting potentialities. The nature of glycosidic bonds between two galactose residues,

their isomeric specificities, their degrees of substitution by other monosaccharides and no sugars compounds such as pyruvate or methyl seem to play a major role in their biological function. Recent progresses in structural characterization of these polysaccharides will conduct in a next future to a better understanding of their structural evolution during the different steps of life of an organism. This will lead to a tentative of correlation between structure and function but also to a more interesting exploitation of terrestrial plant and seaweeds for industrial application. This point as the development of processes for specific extraction and purification of galactans with well defined features will be essential for future developments of these compounds in markets with higher values to those actually exploited.

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