

# A micro-analytical approach to partition coefficients in plagioclase and clinopyroxene of basaltic sills in Serra Geral Formation, Paraná Basin, Brazil

*Leonardo Cardoso Renner<sup>1</sup>, Léo Afraneo Hartmann<sup>2</sup>, Wilson Wildner<sup>3</sup>, Hans-Joachim Massonne<sup>4</sup> & Thomas Theye<sup>4</sup>*

**Abstract** In the present study, detailed micro-analytical data (major and trace elements) in minerals from sills of Serra Geral Formation represents the first attempt to evaluate the partition coefficient of Sc, Ti, V, Cr, Co, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Mo, Cs, Ba, REE, Hf, Ta, Pb, Th and U among coexisting plagioclase, augite and pigeonite from 12 tholeiitic basaltic sills along the eastern border of the Paraná Basin, Brazil. These 12 samples come from Rio Grande do Sul, Paraná, São Paulo and Goiás states. Plagioclase/melt partition coefficients  $K_D$  were determined in core, intermediate and rim zones using LA-ICP-MS and the results were compared with the variations of the elements Ca, Al, Na and Si in those areas. Partition coefficients of Sr and Eu decrease with the increase of Ca concentration in plagioclase. A reduction in temperature and consequently a rise in Na and Si levels increase the partition coefficients of Sr and Eu at the rim of mineral. The  $K_D$  of Ni and V in clinopyroxene has positive correlations with  $Mg^{2+}$ ,  $Ca^{2+}$  and  $Al^{3+}$  due to higher concentrations of those elements at the core and lower concentration towards the rim and negative correlations of  $K_D$  of Sc, Cr, Co, Pb and Lu, due to preferential substitutions by  $Fe^{2+}$  in sixfold coordination. The use of LA-ICP-MS and EPMA made possible the chemical quantification of major and trace elements and the determination of partition coefficients in coexisting plagioclase and clinopyroxene in basaltic melts.

**Keywords:** Partition coefficient, plagioclase, clinopyroxene, Serra Geral Formation, LA-ICP-MS.

**Resumo** Uma abordagem microanalítica dos coeficientes de partição em plagioclásio e clinopiroxênio de soleiras basálticas da Formação Serra Geral, Bacia do Paraná, Brasil. No presente estudo, dados detalhados de microanálise (elementos maiores e traços) em minerais de *sills* da Formação Serra Geral representam a primeira tentativa de avaliar o coeficiente de partição de Sc, Ti, V, Cr, Co, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Mo, Cs, Ba, REE, Hf, Ta, Pb, Th e U entre plagioclásio, augita e pigeonita coexistentes em 12 *sills* basálticos toleíticos ao longo da borda leste da Bacia do Paraná, Brasil. Estas 12 amostras são dos estados do Rio Grande do Sul, Paraná, São Paulo e Goiás. Coeficientes de partição em plagioclásio foram comparados em zonas de núcleo, intermédio e borda, usando LA-ICP-MS, e os resultados foram comparados com as variações das áreas de distribuição dos elementos Ca, Al, Na e Si. O coeficiente de partição do Sr e Eu diminui com o aumento da concentração de Ca no plagioclásio e aumenta para a borda, relacionado com o aumento da concentração de Na e Si e diminuição da temperatura. O  $K_D$  do Ni e V nos clinopiroxênios analisados possui correlações positivas para  $Mg^{2+}$ ,  $Ca^{2+}$  e  $Al^{3+}$  devido a altas concentrações destes elementos no núcleo, diminuindo para a borda, e correlações negativas de  $K_D$  Sc, Cr, Co, Pb e Lu, devido a substituição preferencial pelo  $Fe^{2+}$  em coordenações octaédricas. O uso de LA-ICP-MS e EPMA possibilitam a caracterização química de elementos maiores e traços e a determinação do coeficiente de partição em minerais coexistentes e o líquido.

**Palavras-chave:** Coeficiente de partição, plagioclásio, clinopiroxênio, Formação Serra Geral, LA-ICP-MS.

**INTRODUCTION** Continental flood basalts are major igneous units and their investigation enables to understand the processes active in the mantle and crust. The Serra Geral Formation of South America comprises the Paraná magmatic province, constituted by basalts, basaltic andesites and acid rocks. Recent technological advances, such as the spot LA-ICP-MS analysis of selected small portions of crystals, have opened a window into processes related to the crystallization

of the magmas. This is a detailed investigation of trace element content and zonation in plagioclase and clinopyroxene from Paraná province sills and dykes and of their partition coefficients with the magmatic liquid.

Micro-analytical data (EPMA and LA-ICP-MS) and partitioning coefficients of Sc, Ti, V, Cr, Co, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Mo, Cs, Ba, REE, Hf, Ta, Pb, Th and U between plagioclase and clinopyroxene in basalt and andesite basaltic melts is here intensively

1 - Programa de Pós-graduação em Geociências, Instituto de Geociências, Universidade Federal do Rio Grande do Sul, Porto Alegre (RS), Brazil. E-mail: leorenner@gmail.com

2 - Instituto de Geociências, Universidade Federal do Rio Grande do Sul, Porto Alegre (RS), Brazil. E-mail: leo.hartmann@ufrgs.br

3 - CPRM/SUREG-PA, Serviço Geológico do Brasil, Porto Alegre (RS), Brazil. E-mail: wwildner@pa.cprm.gov.br

4 - Institut für Mineralogie und Kristallchemie, Stuttgart Universität, Germany. E-mail: h-j.massone@mineralogie.uni-stuttgart.de, thomas.theye@mineralogie.uni-stuttgart.de

used in sills of Serra Geral Formation, Paraná Basin, Brazil (Fig. 1). We also utilize chemical data to examine how variations in major components in the plagioclase and clinopyroxene influence  $K_D$  in basalt and basaltic andesite melts.

Plagioclase is affected by rebalancing of  $Ca^{2+}Al^{3+} \leftrightarrow Na^{+}Si^{4+}$  influencing in partition coefficients such as Sr, Eu and Pb in core, intermediate and rim zones. Clinopyroxene exerts major control on distributions of some trace elements and REE in substitutions of Fe and Mg (sixfold coordination), Ca (eightfold coordination) and Al (six or fourfold coordination).

The recent advances of micro-analytical techniques allow to determine the concentration of major, trace and REE elements in diversity zones of minerals, characterizing the chemical order of crystallization and observing probable variables that can influence in mineral/melt partition coefficients such as  $T$ ,  $P$ ,  $fO_2$  and bulk composition.

**ANALYTICAL METHODS** Samples were collected during field trips in Rio Grande do Sul state, Brazil by CPRM (Geological Survey of Brazil) projects in the last five years. These samples are representative of sills in

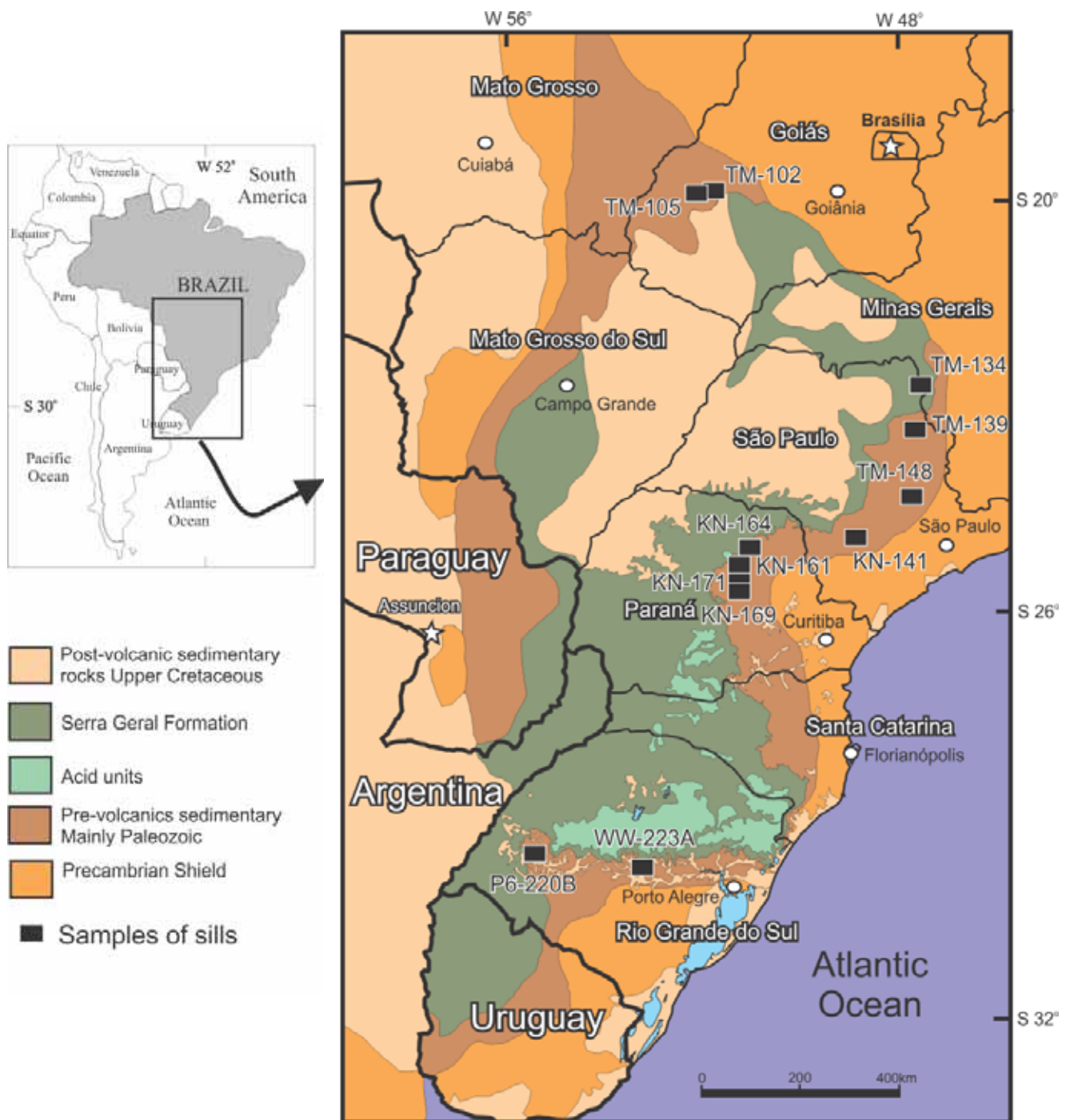


Figure 1 - Geological map showing the general rocks and distribution of sill samples of Paraná Basin in South America. Modified from Bellieni et al. (1986), Melfi et al. (1988), Nardy et al. (2002) and Machado (2003).

the Santiago and Alegrete regions, west of Rio Grande do Sul and sills of Agudo region, central Rio Grande do Sul. The selected samples consist of two sills from Rio Grande do Sul. The other ten sills samples collected by CPRM are from Paraná, São Paulo and Goiás states in the Paraná Basin (Fig. 1).

Major and trace elements were analyzed in whole rocks by Inductively Coupled Plasma - Emission Spectrometer (ICP-ES) and Inductively Coupled Plasma - Mass Spectrometer (ICP-MS) at *Acme Analytical Laboratories Ltd.* in Canada. Chemical data of the analyzed samples are shown in table 1.

Micro-analyses of plagioclase and clinopyroxene of 12 samples of sills of the Paraná Basin were performed at the Institut für Mineralogie und Kristallchemie of the Universität Stuttgart in Germany. These analyses were conducted in thin sections coated by a thin film of carbon and inserted two by two and analyzed by a CAMEXA SX-100 probe, which is equipped with five high-resolution wavelength-dispersive spectrometer (WDS) and one low-resolution energy-dispersive spectrometer (EDS). Analytical conditions for plagioclase and clinopyroxene included a focused beam (beam diameter ~1 µm), Faraday current of 15 nA with an acceleration voltage of 15 kV. Spectrometer SP1 - TAP (Al and Si), SP2 - LLIF (Fe, Ni and Mn), SP3 - LPET (Ti and Cr), SP4 - TAP (Na and Mg) and SP5 - PET (K and Ca) and calibration standards were: Diopside for Si, Mg and Ca; Cordierite for Al; Fe<sub>2</sub>O<sub>3</sub> for Fe; NiO for Ni; Rutile for Ti; Cr<sub>2</sub>O<sub>3</sub> for Cr; Rhodochrosite for Mn; Albite for Na and Orthoclase for K. Counting times were 20 s on the peak for all these elements except Na with 10 s. Background counting time was 20 s for Al, Si, Mn and Fe, 10 s for Ni, Ti, Cr, Mg, K and Ca and 10 s for Na on each side of the background.

For the 12 samples, over 184 backscattering electron images were acquired by CAMECA SX-100 in the same laboratory. The images were acquired at 15 kV and 15 nA, modifying the brightness, gamma and contrast to accentuate the textures in individual crystals. The elements analyzed in plagioclase and clinopyroxene crystals, including portions of the matrix, were: K<sub>2</sub>O, FeO, Na<sub>2</sub>O, Al<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, MnO, CaO, Cr<sub>2</sub>O<sub>3</sub>, NiO, MgO and SiO<sub>2</sub>, totalizing about 680 analyzed points, time analyses 2.5 min each.

Also characteristic X-ray distribution maps were made of K, Na, Ca, Fe, Mg, Ti and Al for plagioclase and clinopyroxene in each sample, totaling 18 maps.

A total of 217 spots were analyzed for Sc, Ti, V, Cr, Mn, Co, Ni, Cu, Zn, Rb, Sr, Y, Zr, Nb, Mo, Cs, Ba, Hf, Ta, Pb, Th and U and REE in the Laser Ablation Inductively Coupled Plasma Mass Spectrometry (LA-ICP-MS) at the Institut für Mineralogie, Universität Würzburg in Germany. In this analysis of trace elements, two crystals of plagioclase and two crystals of clinopyroxene were collected from each sample. The laser ablation microprobe was a New Wave (Merchantek) 266 LUV. A Quadrupole MS Agilent 7500i, plasma power 1250 W, carrier gas (Ar) 1.3 L/min, plasma gas (Ar) 14.9 L/min and auxiliary gas (Ar) 0.9 L/min.

Ablations were performed in single spots with 50 µm diameter, repetition rate 10 Hz, laser energy = 0.87<sup>-1</sup>.10 mJ, laser density 44-53 J/cm<sup>2</sup>. Time-resolved analyses were obtained on the maximum peak 18 s, measuring the instrumental + gas background and 22 s measuring the mineral (0 s on delay) for each analysis.

External calibration via NIST 612, 50 ppm glass-standard, values after Pearce *et al.* (1997). Other standard used for reproducibility and accuracy: NIST 614, and another chip NIST 612 glass-standard. Internal standard is SiO<sub>2</sub> for EPMA measurements. The result was calculated via GLITTER version 3.0 On-line Interactive Data Reduction for the LA-ICP-MS, *Macquarie Research Ltd.* 2000.

The partition coefficient is used for calculating the distribution of trace elements found between the mineral and a melt. It is defined by:  $K_{D_i}^{\text{mineral}} = C_i^{\text{mineral}} / C_i^{\text{melt}}$  where  $K_{D_i}$  is the partition coefficient, and  $C_i$  is the concentration of element  $i$  in ppm or wt.%.

The backscattering images (BSE) by EPMA were used for selecting the best places for analyses of trace elements and REE. Internal zonation was characterized by differences in composition of these elements according to the growth of the plagioclase and clinopyroxene crystals.

**PETROGRAPHY AND GEOCHEMISTRY** The sills were macroscopically analyzed and displayed a color range from light gray to dark gray, aphanitic fine-grained (P6220B) to phaneritic coarse-grained. Rocks are holocrystalline (75-99%) with primary mineralogy which includes phenocrysts and microphenocrysts of plagioclase, clinopyroxene, opaque minerals and apatite as accessory. The secondary mineralogy comprehends clay minerals and carbonate, presence of alteration products of groundmass replaced by brown clay and Fe oxyhydroxide. The sills from Rio Grande do Sul comprise grains of chalcopyrite (WW223A) and native copper (P6220B), both reaching the maximum size of 0.3 mm. The sills from the other states comprise grains of chalcopyrite and pyrite, sizes ranging from 0.7 up to 2.7 mm. Outside Rio Grande do Sul, the Fe-Ti oxides present larger size, as well as the grains of plagioclase, reaching up to 10.2 mm and of clinopyroxene, up to 10 mm. The estimated modal distribution of minerals for the sills from Rio Grande do Sul state is: plagioclase 7.5~26%, clinopyroxene 30~28%, Fe-Ti oxides 2~6% and groundmass 40~60% to WW223A and P6220B, respectively. The estimated modal for the sills from Paraná is: plagioclase 30~42%, clinopyroxene 23~30%, Fe-Ti oxides 4~9% and groundmass 20~37%. The modal for sills from São Paulo is: plagioclase 18~48%, clinopyroxene 20~28%, Fe-Ti oxides 3.5~15% and groundmass 8~41%. The estimated modal for sills from Goiás is: plagioclase 50~51%, clinopyroxene 23~36.5%, Fe-Ti oxides 7~15% and groundmass 1~20%.

Backscattered electrons images (BSE) of two clinopyroxene samples (Figs. 2A and 2B) and two plagioclase samples (Figs. 2C and 2D) show the analyzed

spots (laser ablation craters in A). The red circles indicate the area of laser ablation analyses and small filled circles show EPMA analyses. Clinopyroxene and plagioclase display chemical zoning in all crystals examined, but this zoning is either faded or banded in BSE images. In the BSE of clinopyroxene crystals (Fig. 2A), we can see a brighter rim and a diminishing intensity towards its core, identifying Fe increase from the core towards the rim, Ca and Mg decreasing concentration from core towards the rim. Thus we classify the clinopyroxene core as a high temperature augite and rim as being pigeonite of lower temperature (Fig. 2B). Figure 2D shows plagioclase oscillatory zonations identified in some of the studied plagioclase samples. The zonations vary according to their chemical composition, core is Ca rich and darker rim is Na rich. The rim portions display higher Si, Na and K concentrations and lower Al and Ca concentrations. In most of the samples, plagioclase

presents Ca concentration at the core higher than at rim identifying a normal crystallization from high to low temperature.

Bulk rock analyses (Tab. 1) led to chemical classification of the studied rock into basalt and basaltic andesite based on alkali-silica diagram, Zanettin (1984).  $\text{Na}_2\text{O}+\text{K}_2\text{O}$  versus  $\text{SiO}_2$  wt.%.  $\text{SiO}_2$  concentrations varying from 48.6 to 54.7 wt.% and concentrations of  $\text{Na}_2\text{O}+\text{K}_2\text{O}$  between 3 to 5 wt.%. Using as well the major elements A = ( $\text{Na}_2\text{O}+\text{K}_2\text{O}$ ), F = (FeO) and M = (MgO) in wt.% as shown in the AFM diagram (Irvine & Baragar 1971), all samples plot in tholeiitic magmas.

The EPMA analyses of clinopyroxene and plagioclase crystals were done on the same areas where LA-ICP-MS analyses were carried out. We also performed analyses showing lines along minerals highlighting the chemical differences in core, intermediate and rim zones (Fig. 2).

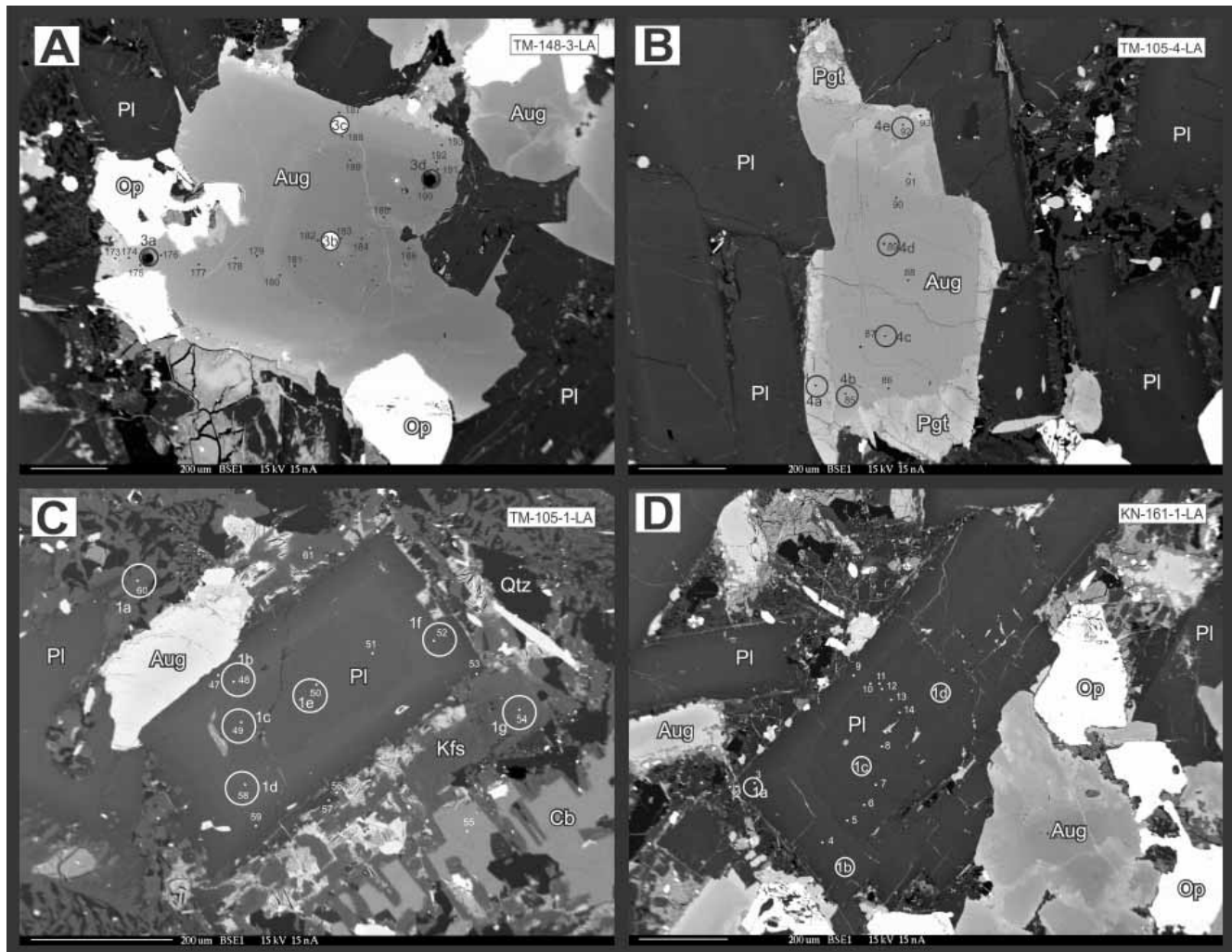


Figure 2 - Backscattered electron images of clinopyroxene (A-) and plagioclase (C-D) analyzed in this study. Open circles show LA-ICP-MS analyses locations. The filled circles show EPMA analyses locations. (A) showing laser ablation craters. Recommendations by the IUGS Subcommittee on the Systematics of Metamorphic Rocks: Pl = plagioclase; Qtz = quartz; Kfs = feldspar-K; Aug = augite; Pgt = pigeonite; Op = opaque minerals and Cb = carbonate minerals.

Table 1 - Bulk rock chemical composition of major (wt.%) and trace elements (ppm) data for sills (basalt and basaltic andesite) in Serra Geral Formation, Paraná Basin, Brazil. Blank = not analyzed.

State	Rio Grande do Sul			São Paulo				Paraná			Goiás	
Sample	WW223A	P6220B	TM139	KN141	TM134	TM148	KN161	KN164	KN169	KN171	TM102	TM105
Type	Basaltic andesite						Basalt					
SiO <sub>2</sub>	54.76	52.59	54.28	50.02	50.93	51.16	51.96	51.35	52.06	51.67	48.86	50.18
TiO <sub>2</sub>	1.92	1.25	2.716	2.570	3.96	3.03	2.15	1.72	2.15	2.59	2.96	3.96
Al <sub>2</sub> O <sub>3</sub>	12.29	13.22	12.31	13.00	12.47	13.37	13.09	13.16	12.81	12.85	14.02	12.01
Fe <sub>2</sub> O <sub>3</sub>	14.99	13.49	13.47	15.05	14.80	13.37	14.81	13.35	15.22	15.14	14.32	16.28
MnO	0.19	0.18	0.18	0.21	0.17	0.20	0.21	0.20	0.22	0.21	0.18	0.21
MgO	3.32	5.25	2.63	5.19	4.23	3.06	4.62	6.27	4.62	4.60	5.36	3.81
CaO	7.38	8.63	6.57	9.44	7.89	7.22	8.81	10.12	8.71	8.88	9.82	7.80
Na <sub>2</sub> O	2.72	2.21	2.83	2.36	2.56	3.28	2.54	2.23	2.47	2.43	2.48	2.66
K <sub>2</sub> O	1.15	1.53	2.06	1.07	1.44	1.80	0.99	0.83	0.99	0.95	0.86	1.42
P <sub>2</sub> O <sub>5</sub>	0.19	0.15	0.66	0.34	0.43	0.65	0.28	0.20	0.26	0.31	0.30	0.48
LOI	0.9	1.30	2.26	1.13	1.39	1.37	0.59	0.51	0.26	0.56	0.62	1.31
Total	99.82	99.81	99.96	100.37	100.27	98.52	100.04	99.95	99.78	100.20	99.80	100.12
FeO	13.34	12.00	11.98	13.39	13.17	11.89	13.18	11.88	13.54	13.47	12.74	14.48
Sc	39	42	23	37	33	24	39	42	41	41	33	32
Ti	11520	7500	16296	15420	23772	18216	12918	10356	12930	15558	17772	23760
V	550	399	228	481	461	238	377	378	409	420	581	445
Cr			< 20	95	24	< 20	51	131	51	47	106	< 20
Co	47.8	50.2	31	49	43	31	48	40	43	42	48	41
Ni	6.1	14.9	< 1	34	39	< 1	20	49	18	16	55	< 1
Cu	140	203	157	202	136	16	208	167	197	229	102	262
Zn	51	39	109	99	89	99	102	91	102	105	93	125
Rb	61.7	45.3	42	23	30	38	25	17	24	21	17	29
Sr	228	306	485	371	434	462	302	245	234	230	480	460
Y	32.4	24.6	53.4	34.6	39.3	43.6	36.8	30.5	40.0	45.0	25.8	41.4
Zr	179	119	360	176	249	268	174	137	187	206	151	260
Nb	12.5	8.1	31.3	18.9	23.6	29.9	15.5	11.2	17.0	16.6	23.0	26.8
Mo	1.6	0.9	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2	< 2
Cs	2.3	1.1	0.5	0.2	0.3	0.4	0.2	0.3	0.5	0.2	0.2	0.3
Ba	454	281	668	377	481	570	356	265	306	272	329	464
Ga	19	18	28	24	25	26	24	18	22	22	25	26
La	23.50	14.00	52.54	23.98	33.42	43.47	22.13	16.85	23.92	23.88	24.98	38.34
Ce	53.00	32.70	110.90	52.77	71.85	90.09	48.40	36.46	51.23	52.87	53.49	81.02
Pr	6.87	4.31	14.03	6.30	9.06	11.34	5.84	4.43	6.17	6.59	6.64	10.07
Nd	28.80	18.60	60.48	27.50	40.02	46.94	25.24	19.39	26.69	29.59	28.99	43.52
Sm	6.32	4.36	12.80	6.05	9.05	9.86	5.73	4.66	6.42	7.29	6.43	9.54
Eu	1.55	1.22	3.95	2.07	2.87	3.26	1.94	1.61	2.08	2.39	2.26	3.19
Gd	6.25	4.42	12.44	6.37	8.76	9.89	6.04	5.23	7.36	8.28	6.11	9.18
Tb	1.1	0.8	1.9	1.0	1.3	1.5	1.0	0.9	1.2	1.4	0.9	1.4
Dy	6.09	4.63	10.09	5.85	7.50	8.20	5.91	5.47	7.19	8.22	5.16	7.83
Ho	1.13	0.87	1.85	1.17	1.40	1.52	1.15	1.06	1.44	1.62	0.94	1.45
Er	3.20	2.46	5.29	3.44	4.15	4.40	3.58	3.29	4.41	4.94	2.78	4.29
Tm	0.45	0.36	0.71	0.48	0.56	0.61	0.51	0.47	0.64	0.69	0.36	0.58
Yb	3.12	2.41	4.44	3.05	3.40	3.64	3.21	2.86	3.96	4.30	2.22	3.49
Lu	0.44	0.34	0.58	0.42	0.47	0.50	0.46	0.41	0.55	0.62	0.30	0.47
Hf	4.8	3.4	9.4	4.6	6.7	7.1	4.4	3.7	5.2	5.5	4.3	6.9
Ta	0.70	0.50	2.00	1.08	1.58	1.86	0.91	0.71	1.00	1.02	1.04	1.76
W	0.4	1.1	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5	< 0.5
Pb	1.4	2	19	< 3	23	< 3	13	11	5	15	9	13
Th	7.70	3.90	4.57	2.67	3.10	3.66	2.43	1.89	2.57	2.26	1.96	3.38
U	1.50	1.10	0.92	0.56	0.64	0.73	0.46	0.36	0.48	0.45	0.40	0.70



**Plagioclase** The microanalyses classified plagioclase according to Or-Ab-An diagram (Deer *et al.* 2003) (Fig. 3) in portions which vary from oligoclase to labradorite ( $An_{20-68} Ab_{31-74} Or_{1-8}$ ). Table 2 shows the oxides Si, Al, Fe, Mg, Ca, Na, K and Ti mean values analyzed by EPMA in plagioclase crystals. Plagioclase fractionation of crystals in those samples present the expected results, increasing values from core towards rim for Si, K and Na oxides and decreasing Al, Mg and Ca values from core towards rim. In most samples FeO and  $TiO_2$  wt.% values have a decrease from core towards rim except WW223A and KN161 samples results which present different values. The plagioclase rims present values of those oxides with large variations related to core and intermediate zones, those variations can be related to the end of mineral crystallization and last stages of chemical/thermal balance of those elements.

The analyses of some major elements by EPMA (Fig. 4) made it possible to identify chemical variations related to crystallization of plagioclase at rim, intermediate and core zones. Negative correlations of CaO and  $SiO_2$  show a normal fractioning of Ca oxides with larger concentrations in the core portions of the mineral and silica at the rim zones (Figs. 4a, 4d and 4g). The binary diagrams show a positive correlation between  $TiO_2$  and  $Al_2O_3$  in plagioclase, high concentrations of  $TiO_2$  wt.% and  $Al_2O_3$  wt.% at the core and low concentrations at the rim (Figs. 4b, 4e and 4h). Figures 4c, 4f and 4i show lower values for FeO wt.% and An% at rim and increase of concentration at the core, identifying positive correlations of those elements.

Figure 5 shows binary diagrams of An% versus FeO,  $TiO_2$ ,  $K_2O$  and  $Al_2O_3$  wt.%. The plagioclase

fractionation is considered normal, according to the An% with  $Al_2O_3$  and FeO wt.% positive correlations (Figs. 5a and 5d). In relation to Na negative correlations, we highlight the low values  $TiO_2$  for the plagioclase sills collected in Rio Grande do Sul compared to those in other states (Fig. 5c).

**Clinopyroxene** The microanalyses of clinopyroxene in sills classified the mineral into two chemical types: high-Ca (augite) and low-Ca (pigeonite), according to table 3. According to quadrilateral diagram from Morimoto (1988) and isotherms modified from Lindsley (1983) to  $P = 1$  kbar, the clinopyroxene core, intermediate and rim zones results were plotted (Fig. 6). Most clinopyroxene has pigeonite rims of lower temperatures ( $\sim 700^\circ C$  to  $950^\circ C$ ), however, some clinopyroxene have augite rims, of temperature equivalent to those of pigeonite rims, showing that differences between the two chemical terms do not influence temperature in the several sills we studied. The core of the respective clinopyroxenes (augite and pigeonite) have the same behavior, different nuclei composition but similar temperatures ( $\sim 1150^\circ C$  to  $1100^\circ C$ ), are very well represented by sills samples from Paraná state. No pigeonite nuclei were identified in sills samples from São Paulo, only low temperature rims.

Figure 7 shows a positive correlation between  $Al_2O_3$  wt.% and Wo%,  $TiO_2$  wt.% and  $Al_2O_3$  wt.% and negative correlation between MgO wt.% and Fs% from clinopyroxene microanalyses. According to Wo% values, we can separate clinopyroxene into two groups (augite  $> 22$  Wo% and pigeonite  $< 22$  Wo%) (Figs. 7a, 7d and 7g). Values above 1.0 wt.%  $Al_2O_3$

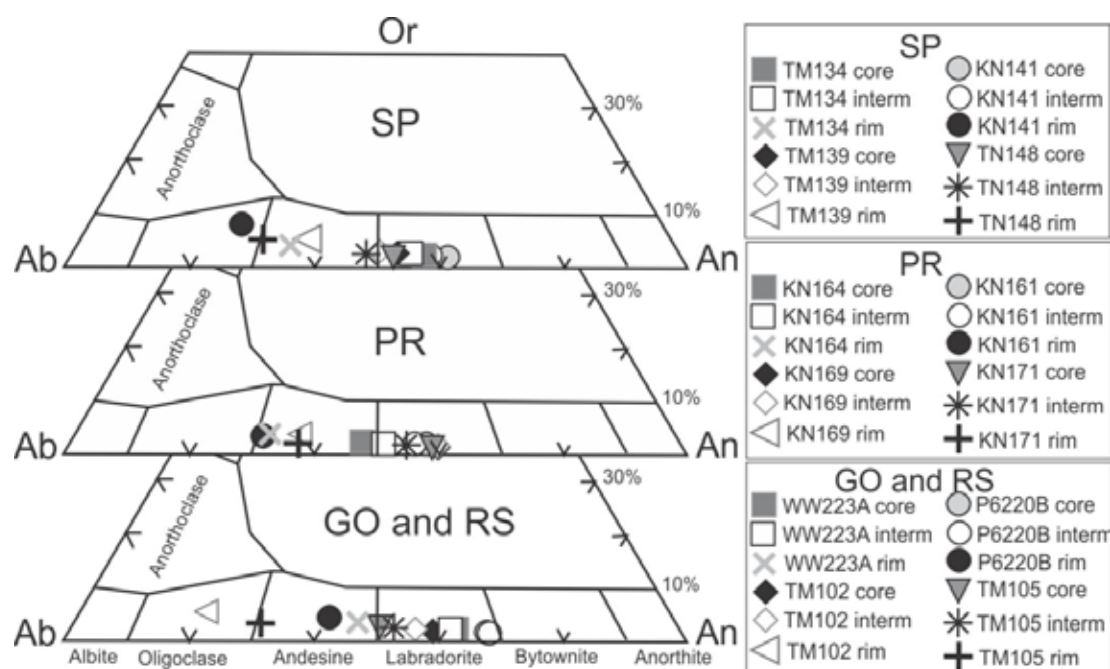


Figure 3 - Plagioclase composition ternary diagram (core to rim) Deer *et al.* (2003) from São Paulo (SP), Paraná (PR), Goiás (GO) and Rio Grande do Sul (RS). Aug = augite and pgt = pigeonite.

Table 2 - Selected plagioclase major microanalysis EPMA (wt.%) in core, intermediate and rim zones.

State	Rio Grande do Sul												São Paulo																	
	WW223A				P6220B				TM139				KNI141				TM148				KNI164									
Sample	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$		
	Core	n=5	Rim	n=1	Core	n=2	Interm	n=3	Core	n=2	Rim	n=8	Interm	n=11	Rim	n=7	Core	n=4	Interm	n=8	Rim	n=2	Core	n=9	Interm	n=12	Rim	n=9		
SiO <sub>2</sub>	51.51	3.68	51.71	0.12	55.44	49.62	1.14	49.4	1.45	56.42	7.57	53.85	1.01	54.41	1.50	57.63	0.16	50.56	3.11	51.38	3.42	59.96	0.72	51.51	3.68	51.71	0.12	55.44		
Al <sub>2</sub> O <sub>3</sub>	28.66	2.20	28.28	0.01	26.34	29.27	0.25	29.31	0.42	25.04	3.62	27.04	0.14	26.56	0.31	24.91	0.25	28.33	1.85	27.94	1.98	22.74	0.40	28.66	2.20	28.28	0.01	26.34		
FeO	0.71	0.15	0.84	0.16	0.85	0.75	0.07	0.8	0.1	0.69	0.34	0.69	0.00	0.65	0.07	0.56	0.15	0.74	0.14	0.75	0.17	0.42	0.03	0.71	0.15	0.84	0.16	0.85		
MgO	0.11	0.01	0.12	0.03	0.04	0.18	0.08	0.14	0.05	0.05	0.05	0.13	0.02	0.10	0.03	0.03	0.00	0.12	0.03	0.11	0.05	0.03	0.03	0.11	0.01	0.12	0.03	0.04		
CaO	12.97	2.86	12.73	0.22	9.68	13.98	0.76	14.13	1.04	8.38	5.57	11.02	0.43	10.41	0.88	7.73	0.15	12.76	2.72	12.00	2.91	5.12	0.57	12.97	2.86	12.73	0.22	9.68		
Na <sub>2</sub> O	4.15	1.55	4.27	0.16	6.03	3.61	0.4	3.58	0.6	6.32	2.57	5.26	0.30	5.54	0.47	6.82	0.21	4.32	1.45	4.57	1.41	7.77	0.01	4.15	1.55	4.27	0.16	6.03		
K <sub>2</sub> O	0.39	0.20	0.35	0.00	0.61	0.22	0.02	0.22	0.08	0.71	0.56	0.45	0.02	0.50	0.07	0.91	0.12	0.28	0.13	0.33	0.17	1.38	0.71	0.39	0.20	0.35	0.00	0.61		
TiO <sub>2</sub>	0.05	0.02	0.05	0.01	0.06	0.04	0.02	0.03	0.01	0.05	0.02	0.10	0.01	0.09	0.02	0.06	0.00	0.08	0.02	0.08	0.01	0.05	0.00	0.05	0.02	0.05	0.01	0.06		
H <sub>2</sub> O	1.41	1.64	1.64	0.94	2.3	2.3	2.36	2.36	2.31	1.43	1.71	1.31	2.78	2.84	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00		
Total	100.00		100.00		100.00	100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00
Ab (%)	35.90		37.00		51.20	31.40		31.00		55.30		45.20		47.70		58.30		37.40		40.00		67.50		35.90		37.00		51.20		51.20
An (%)	61.90		61.00		45.40	67.30		67.70		40.60		52.30		49.50		36.50		61.00		58.10		24.60		61.90		61.00		45.40		45.40
Or (%)	2.20		2.00		3.40	1.30		1.30		4.10		2.50		2.80		5.10		1.60		1.90		7.90		2.20		2.00		3.40		3.40
State	São Paulo												Paraná																	
Sample	TM134				TM148				KNI161				KNI164																	
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$		
Core	n=7	Interm	n=11	Rim	n=7	Core	n=13	Interm	n=25	Rim	n=14	Core	n=4	Interm	n=21	Rim	n=4	Core	n=9	Interm	n=12	Rim	n=9	Core	n=9	Interm	n=12	Rim	n=9	
SiO <sub>2</sub>	52.86	1.14	53.39	1.76	58.67	0.95	54.32	0.43	55.47	0.61	60.39	0.95	52.31	1.14	52.62	0.13	58.84	0.72	56.15	5.99	55.25	3.43	60.37	2.29	52.86	1.14	53.39	1.76	58.67	
Al <sub>2</sub> O <sub>3</sub>	28.07	0.87	27.83	1.10	24.76	0.81	27.52	0.11	26.91	0.02	24.00	0.10	28.13	0.90	27.81	0.16	24.05	0.45	26.95	3.51	27.33	2.03	24.50	0.99	28.07	0.87	27.83	1.10	24.76	
FeO	0.79	0.04	0.71	0.05	0.64	0.03	0.68	0.01	0.62	0.00	0.54	0.17	0.46	0.09	0.54	0.02	0.54	0.08	0.59	0.27	0.65	0.18	0.52	0.20	0.79	0.04	0.71	0.05	0.64	
MgO	0.13	0.01	0.11	0.04	0.05	0.01	0.08	0.01	0.07	0.00	0.04	0.02	0.09	0.03	0.09	0.01	0.01	0.06	0.05	0.06	0.04	0.02	0.01	0.13	0.01	0.11	0.04	0.05		
CaO	11.94	0.86	11.53	1.41	7.11	0.99	10.80	0.08	9.88	0.14	6.12	0.29	12.03	0.88	11.72	0.15	6.51	0.61	9.77	4.67	10.39	2.77	6.55	1.71	11.94	0.86	11.53	1.41	7.11	
Na <sub>2</sub> O	4.74	0.39	4.98	0.77	7.01	0.65	5.27	0.02	5.76	0.10	7.59	0.17	4.70	0.54	4.95	0.08	7.84	0.19	5.83	2.59	5.44	1.53	7.51	0.77	4.74	0.39	4.98	0.77	7.01	

$\sigma$  = standard deviation, Interm = intermediary and n = number of analyses.  
To be continued

Table 2 continuation

State	Paraná																									
	São Paulo				TM148				KN161				KN164													
Sample	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$										
	Core	n=7	Interm	n=11	Rim	n=11	Core	n=7	Core	n=13	Interm	n=25	Rim	n=14	Core	n=4	Interm	n=21	Rim	n=4	Core	n=4	Interm	n=12	Rim	n=9
K <sub>2</sub> O	0.36	0.07	0.40	0.09	0.68	0.20	0.32	0.02	0.41	0.04	0.84	0.15	0.35	0.09	0.34	0.01	0.65	0.18	0.41	0.25	0.39	0.18	0.68	0.14		
TiO <sub>2</sub>	0.10	0.01	0.09	0.02	0.05	0.02	0.07	0.00	0.05	0.00	0.04	0.00	0.05	0.01	0.07	0.00	0.03	0.00	0.05	0.00	0.07	0.00	0.05	0.02		
H <sub>2</sub> O	0.99	0.93	1.06	1.06	1.06	0.90	0.90	0.82	0.82	0.49	1.85	1.84	1.51	0.24	0.39	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		
Total	100.00	100.00	100.05	100.05	100.00	100.00	100.00	100.00	100.00	100.07	100.00	100.00	100.00	100.09	100.03	100.00	100.00	100.00	100.00	100.09	100.03	100.00	100.00	100.22		
Ab (%)	40.90	42.90	61.60	61.60	46.00	46.00	50.10	50.10	65.90	65.90	40.60	40.60	66.10	66.10	47.60	47.60	50.70	50.70	46.90	46.90	50.20	50.20	31.30	31.30		
An (%)	57.00	54.80	34.50	34.50	52.10	52.10	47.50	47.50	29.30	29.30	57.40	57.40	30.30	30.30	46.90	46.90	50.20	50.20	46.90	46.90	50.20	50.20	31.30	31.30		
Or (%)	2.00	2.30	3.90	3.90	1.80	1.80	2.30	2.30	4.80	4.80	2.00	2.00	3.60	3.60	2.20	2.20	2.40	2.40	2.40	2.40	2.20	2.20	3.90	3.90		
State	Paraná												Goiás													
Sample	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$										
	Core	n=5	Interm	n=14	Rim	n=14	Core	n=7	Core	n=11	Interm	n=31	Rim	n=10	Core	n=4	Interm	n=14	Rim	n=1	Core	n=5	Interm	n=9	Rim	n=7
SiO <sub>2</sub>	53.02	0.18	53.14	0.40	59.74	3.16	52.84	0.92	53.59	0.11	58.32	1.80	51.59	0.11	52.41	0.43	62.19	62.19	53.98	0.22	53.26	0.35	59.03	0.96		
Al <sub>2</sub> O <sub>3</sub>	28.75	0.09	28.60	0.26	25.10	1.75	28.28	0.43	27.67	0.01	25.07	1.03	28.34	0.11	27.83	0.14	22.66	22.66	26.81	0.34	27.26	0.27	24.04	0.11		
FeO	0.73	0.03	0.72	0.03	0.52	0.13	0.73	0.02	0.75	0.05	0.54	0.11	0.55	0.08	0.62	0.06	0.35	0.35	0.63	0.00	0.65	0.01	0.43	0.10		
MgO	0.14	0.00	0.13	0.01	0.03	0.02	0.13	0.01	0.11	0.02	0.03	0.01	0.09	0.01	0.06	0.00	0.01	0.01	0.09	0.00	0.06	0.01	0.02	0.00		
CaO	12.40	0.05	12.31	0.32	7.52	2.31	12.07	0.62	11.29	0.02	7.56	1.49	12.26	0.03	11.69	0.20	4.36	4.36	10.51	0.12	10.91	0.00	6.40	0.43		
Na <sub>2</sub> O	4.48	0.06	4.46	0.09	6.92	1.08	4.56	0.30	5.03	0.01	7.06	0.75	4.61	0.09	4.96	0.01	8.84	8.84	5.59	0.01	5.31	0.05	7.91	0.19		
K <sub>2</sub> O	0.27	0.00	0.27	0.02	0.69	0.23	0.26	0.04	0.32	0.00	0.54	0.06	0.32	0.02	0.36	0.01	0.98	0.98	0.43	0.00	0.38	0.03	0.63	0.04		
TiO <sub>2</sub>	0.07	0.00	0.07	0.00	0.04	0.02	0.08	0.01	0.08	0.00	0.05	0.00	0.10	0.01	0.10	0.01	0.02	0.02	0.10	0.00	0.09	0.01	0.04	0.01		
H <sub>2</sub> O	0.12	0.28	0.05	0.05	0.05	0.02	0.02	1.12	1.12	0.81	2.12	1.95	0.59	1.84	2.06	1.47	1.47	1.84	1.84	2.06	2.06	100.00	100.00			
Total	100.00	100.01	100.14	100.14	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00			
Ab (%)	38.90	39.00	60.00	60.00	40.00	40.00	43.80	43.80	60.90	60.90	39.80	39.80	74.30	74.30	45.80	45.80	66.70	66.70	47.90	47.90	52.00	52.00	29.80	29.80		
An (%)	59.50	59.50	36.00	36.00	58.50	58.50	54.30	54.30	36.00	36.00	58.40	58.40	20.30	20.30	49.70	49.70	29.80	29.80	49.70	49.70	52.00	52.00	29.80	29.80		
Or (%)	1.50	1.60	3.90	3.90	1.50	1.50	1.80	1.80	3.10	3.10	1.80	1.80	5.40	5.40	2.20	2.20	3.50	3.50	2.40	2.40	2.20	2.20	3.50	3.50		

$\sigma$  = standard deviation, Interm = intermediary and n = number of analyses.



classify the core and intermediate portions of the mineral, either augite or pigeonite. Figure 7b shows the values below 1 for TiO and Al<sub>2</sub>O<sub>3</sub> wt.% in rims of augite and pigeonite in Goiás and Rio Grande do Sul states. Negative correlation is identified in diagrams of MgO wt.% versus Fs% in figures 7c, 7f and 7i. Figure 7c shows that higher MgO wt.% values are related to clinopyroxene nuclei and pigeonite nuclei to higher values. In figures 7c and 7f, there are two trends related to chemical compositions of core, intermediate and rim zones for augite and pigeonite. Figure 7i of São Paulo state does not represent two trends, but presents the highest Fs% values for augite and pigeonite.

The EPMA mean microanalyses of clinopyroxene crystals (Fig. 8) shows TiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> wt.% variations versus En, Fs and Wo% contents for samples from

Rio Grande do Sul, Paraná, São Paulo and Goiás. The highest concentration values for TiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> wt.% found in clinopyroxene are in São Paulo, followed by Goiás, Paraná and the lowest values are in Rio Grande do Sul. Positive correlations are found in Al<sub>2</sub>O<sub>3</sub> wt.% versus En and Wo% diagram, figures 8b and 8f and TiO<sub>2</sub> wt.% versus Wo%, figures 8a and 8b. Lowest TiO<sub>2</sub> wt.% concentrations in clinopyroxene are evidence of low concentration of that oxide in other minerals as plagioclase and Ti-magnetite and ilmenite.

**RESULTS - PARTITION COEFFICIENTS** The partition coefficients of trace elements in plagioclase and clinopyroxene crystals in sills of Serra Geral Formation in Paraná Basin describe the mineral/melt preference. The total rock analyses made it possible to

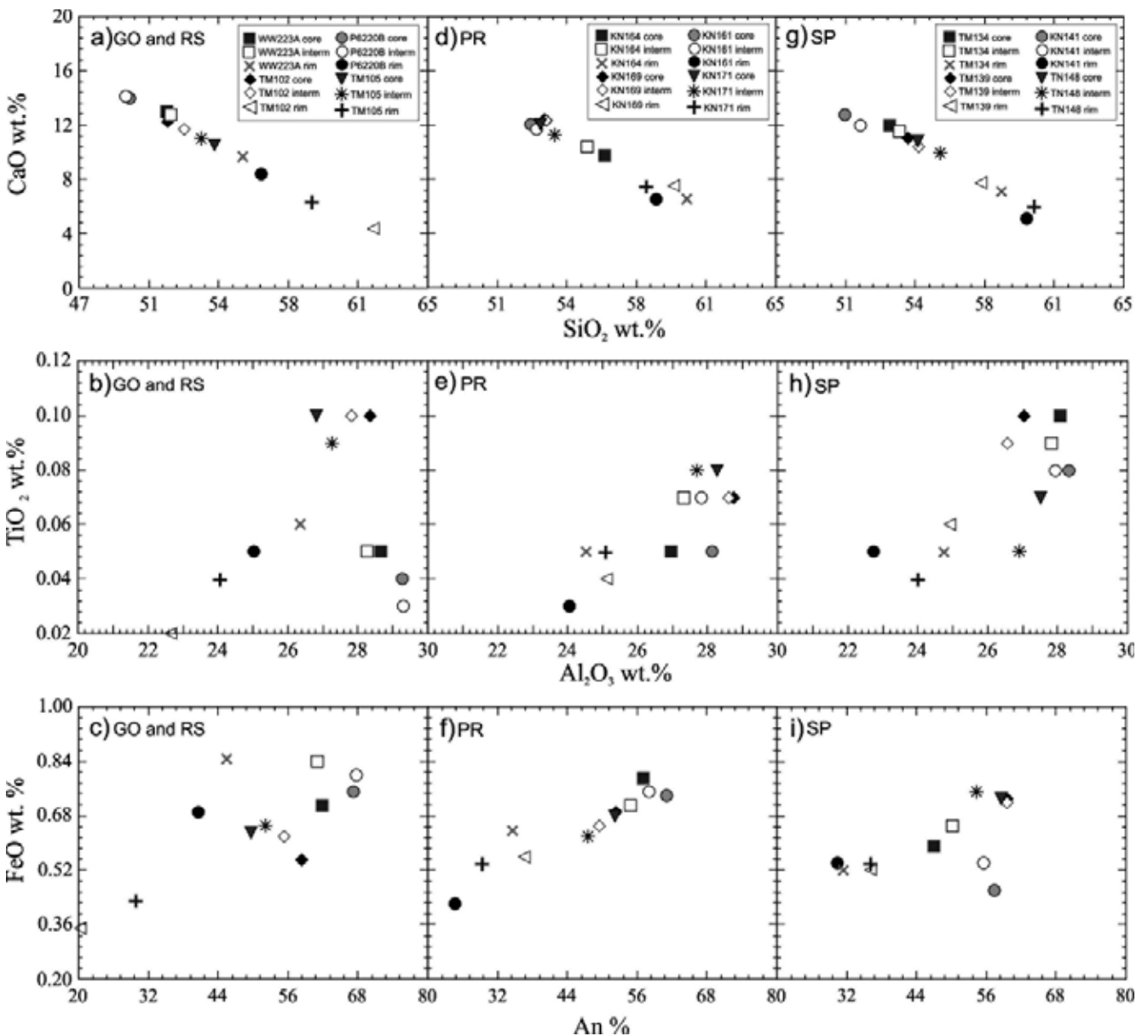


Figure 4 - CaO wt.% versus SiO<sub>2</sub> wt.%, TiO<sub>2</sub> wt.% versus Al<sub>2</sub>O<sub>3</sub> wt.% and FeO wt.% versus An% for plagioclase average microanalyses from EPMA in core, intermediate and rim zones. Aug = augite and pgt = pigeonite.

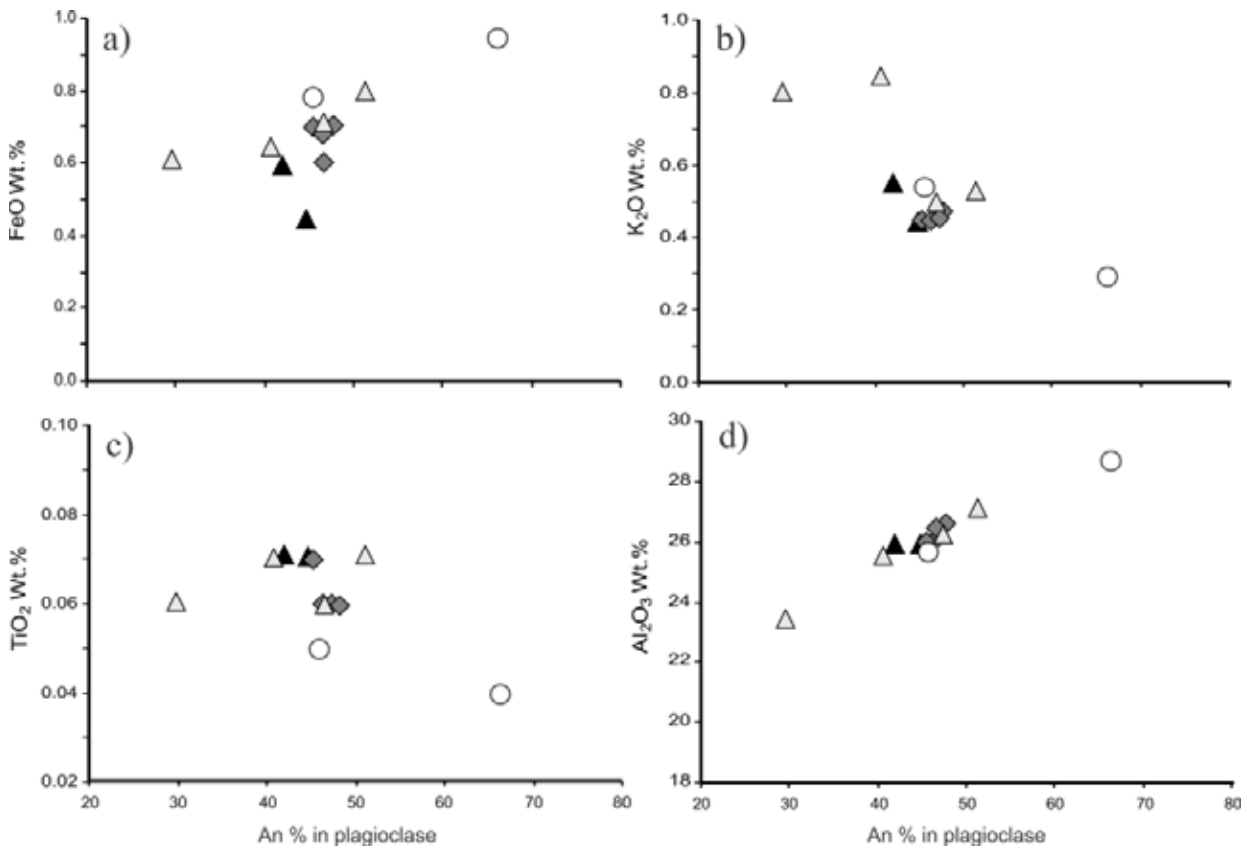


Figure 5 - Binary diagram show the distribution between FeO, K<sub>2</sub>O, TiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> wt.% and anorthite content in plagioclase average microanalyses from EPMA in sills of Serra Geral Formation, Paraná Basin. Black triangle: plagioclase from Goiás; gray triangle: São Paulo; gray lozenge: Paraná; white circle: Rio Grande do Sul.

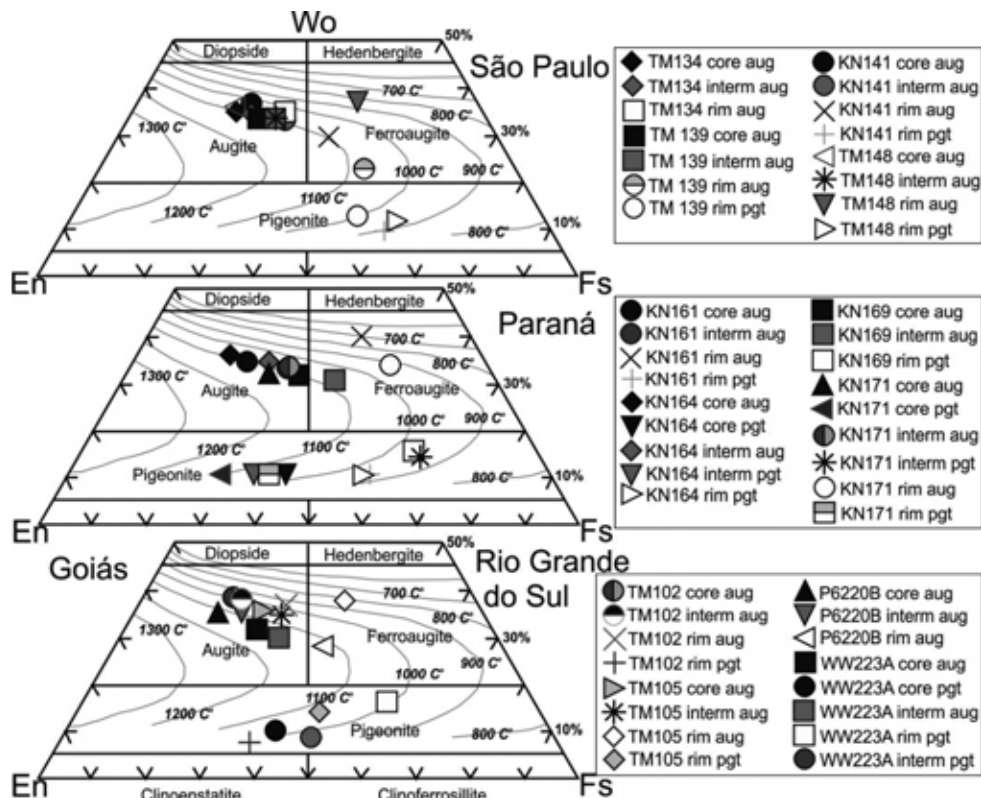


Figure 6 - Pyroxene classification quadrilateral diagram [Ca, Mg, Fe (Fe + Mn) in mol%] from Morimoto (1988). Isotherms modified Lindsley (1983) to P = 1 kbar. Aug = augite and pgt = pigeonite.

Table 3 - Representative major elements EPMA (wt.%) compositions of clinopyroxene in core, intermediate and rim zones.

State		Rio Grande do Sul																		
Sample		WW223A									P6220B									
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$		
	Core Aug	n=3	Core Pgt	n=2	Interm Aug	n=1	Interm Pgt	n=4	Rim Pgt	n=4	Core Aug	n=4	Interm Aug	n=5	Rim Aug	n=4				
SiO <sub>2</sub>	50.12	0.31	51.51	0.46	51.01		49.73	1.39	48.27	0.62	50.73	0.29	49.03	0.52	47.4	0.68				
Al <sub>2</sub> O <sub>3</sub>	1.58	0.31	0.87	0.00	1.54		0.79	0.07	0.66	0.08	1.62	0.26	1.58	0.12	1.15	0.3				
FeO	15.22	1.79	24.28	0.78	18.22		27.60	4.39	32.94	1.22	9.71	1.51	12.35	0.54	23.13	0.65				
MnO	0.34	0.02	0.50	0.07	0.37		0.56	0.08	0.66	0.01	0.25	0.00	0.26	0.05	0.45	0.15				
MgO	15.64	0.55	18.28	1.03	14.27		15.57	3.46	9.21	0.75	17.84	1.25	15.74	0.07	11.47	0.98				
CaO	15.65	1.97	4.95	0.28	14.75		4.03	0.22	7.48	2.14	17.89	0.03	17.38	0.78	13.45	3.98				
Na <sub>2</sub> O	0.14	0.04	0.05	0.02	0.12		0.04	0.02	0.07	0.04	0.18	0.04	0.55	0.47	0.18	0.02				
TiO <sub>2</sub>	0.59	0.05	0.36	0.03	0.61		0.36	0.05	0.48	0.03	0.33	0.12	0.41	0.1	0.72	0.04				
NiO	0.02	0.03	0.00	0.00	0.06		0.02	0.02	0.01	0.00	0.03	0.00	0.02	0.01	0.02	0.03				
Cr <sub>2</sub> O <sub>3</sub>	0.03	0.01	0.01	0.01	0.03		0.01	0.01	0.00	0.00	0.05	0.00	0.05	0.01	0.01	0.01				
H <sub>2</sub> O	0.65		0.00		0.00		0.46		0.43		1.37		1.73		1.98					
Total	100.00		100.81		101.00		100.07		100.23		100.00		100.00		100.00					
Wo (%)	31.58		9.95		30.03		8.45		16.08		35.43		35.38		28.13					
En (%)	43.91		51.14		40.42		45.44		27.54		49.16		44.58		33.37					
Fs (%)	24.51		38.90		29.55		46.11		56.38		15.40		20.04		38.50					
State		São Paulo																		
Sample		TM139									KN141									
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$
	Core Aug	n=8	Interm Aug	n=9	Rim Aug	n=1	Rim Pgt	n=3	Core Aug	n=6	Interm Aug	n=9	Rim Aug	n=1	Rim Pgt	n=1				
SiO <sub>2</sub>	49.07	0.72	49.42	0.50	47.50		48.56	0.51	49.56	0.29	49.15	0.52	47.80		48.39					
Al <sub>2</sub> O <sub>3</sub>	2.23	0.06	1.65	0.11	0.74		0.67	0.16	1.78	0.19	1.31	0.07	0.86		0.23					
FeO	14.86	0.22	16.64	0.58	28.93		31.35	2.75	13.27	0.20	17.97	0.31	23.11		34.32					
MnO	0.36	0.01	0.43	0.02	0.76		0.83	0.04	0.32	0.02	0.47	0.01	0.66		0.95					
MgO	14.79	0.21	13.87	0.19	9.84		11.82	1.24	14.86	0.34	13.01	0.07	10.56		10.35					
CaO	16.36	0.49	16.13	0.23	10.42		5.85	1.03	17.77	0.27	15.97	0.37	14.07		4.46					
Na <sub>2</sub> O	0.23	0.03	0.22	0.01	0.13		0.08	0.01	0.26	0.03	0.21	0.00	0.12		0.07					
TiO <sub>2</sub>	1.29	0.07	0.97	0.06	0.53		0.49	0.10	1.02	0.12	0.85	0.03	0.65		0.35					
NiO	0.02	0.01	0.01	0.02	0.00		0.01	0.02	0.01	0.00	0.01	0.01	0.04		0.03					
Cr <sub>2</sub> O <sub>3</sub>	0.01	0.00	0.01	0.01	0.00		0.00	0.01	0.00	0.00	0.01	0.00	0.01		0.00					
H <sub>2</sub> O	0.82		0.70		1.12		0.43		1.14		1.04		2.10		0.84					
Total	100.03		100.07		100.00		100.09		100.00		100.00		100.00		100.00					
Wo (%)	33.51		33.08		22.03		12.34		36.22		32.95		29.73		9.61					
En (%)	42.15		39.58		28.95		34.68		42.15		37.35		31.05		31.04					
Fs (%)	24.34		27.34		49.02		52.98		21.63		29.71		39.22		59.35					

$\sigma$  = standart deviation, Interm = intermediary, Aug = augite and Pgt = pigeonite and n = number of analyses.

To be continued

*Table 3 continuation*

State		Paraná																	
Sample		KN169						KN171											
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	
	Core Aug	n=14	Interm Aug	n=14	Rim Pgt	n=10	Core Aug	n=10	Core Pgt	n=3	Interm Aug	n=20	Interm Pgt	n=2	Rim Aug	n=3	Rim Pgt	n=4	
SiO <sub>2</sub>	50.35	0.29	49.61	0.38	48.01	0.44	50.30	0.35	52.33	0.45	49.87	0.47	51.20	0.09	48.49	0.80	47.51	0.23	
Al <sub>2</sub> O <sub>3</sub>	1.16	0.09	1.05	0.10	0.57	0.11	1.51	0.21	0.95	0.24	1.40	0.26	0.81	0.07	0.48	0.25	0.50	0.16	
FeO	19.59	1.12	23.42	2.17	35.57	2.84	16.35	0.61	18.04	0.74	17.51	2.43	23.28	0.25	27.70	1.13	35.90	2.49	
MnO	0.44	0.03	0.55	0.04	0.78	0.06	0.38	0.04	0.39	0.03	0.41	0.08	0.48	0.02	0.62	0.03	0.78	0.05	
MgO	12.63	0.43	10.37	1.22	7.65	1.26	14.35	0.35	22.16	0.67	13.11	1.45	18.55	0.33	6.21	0.18	7.50	0.82	
CaO	15.03	0.88	14.35	1.02	6.95	1.45	15.54	0.60	5.20	1.40	16.00	0.97	4.98	0.31	15.52	1.01	6.77	1.84	
Na <sub>2</sub> O	0.16	0.03	0.16	0.04	0.07	0.03	0.20	0.04	0.05	0.04	0.19	0.04	0.06	0.04	0.15	0.06	0.08	0.03	
TiO <sub>2</sub>	0.74	0.06	0.71	0.07	0.49	0.06	0.89	0.12	0.40	0.06	0.89	0.10	0.48	0.01	0.36	0.21	0.50	0.05	
NiO	0.02	0.02	0.01	0.02	0.02	0.02	0.01	0.02	0.04	0.01	0.02	0.03	0.03	0.02	0.01	0.01	0.01	0.01	
Cr <sub>2</sub> O <sub>3</sub>	0.01	0.01	0.00	0.00	0.01	0.01	0.01	0.01	0.05	0.01	0.01	0.01	0.01	0.01	0.00	0.01	0.00	0.00	
H <sub>2</sub> O	0.05		0.02		0.05		0.46		0.47		0.60		0.12		0.43		0.45		
Total	100.19		100.25		100.16		100.00		100.10		100.01		100.00		100.00		100.00		
Wo (%)	31.16		30.21		15.12		32.00		10.31		33.17		10.09		33.54		14.77		
En (%)	36.43		30.38		23.15		41.11		61.15		37.82		52.31		18.67		22.76		
Fs (%)	32.42		39.41		61.73		26.90		28.54		29.01		37.60		47.79		62.47		

State		Goiás																	
Sample		TM102						TM105											
	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	Mean	$\sigma$	
	Core Aug	n=8	Interm Aug	n=10	Rim Aug	n=4	Rim Pgt	n=1	Core Aug	n=7	Interm Aug	n=8	Rim Aug	n=1	Rim Pgt	n=2			
SiO <sub>2</sub>	49.71	0.55	49.64	0.28	49.82	0.34	50.32		49.98	0.15	49.78	0.13	48.35		49.28	0.69			
Al <sub>2</sub> O <sub>3</sub>	2.26	0.24	2.04	0.32	1.23	0.53	0.58		1.42	0.02	1.22	0.04	0.42		0.66	0.23			
FeO	10.34	0.47	11.32	0.86	16.09	3.42	23.46		14.62	0.12	16.62	0.57	22.50		26.95	4.65			
MnO	0.25	0.03	0.27	0.04	0.37	0.05	0.55		0.39	0.02	0.43	0.01	0.55		0.71	0.13			
MgO	15.81	0.40	15.34	0.36	12.30	2.70	20.17		14.17	0.22	13.15	0.31	8.31		14.09	2.52			
CaO	18.95	0.47	18.71	0.34	17.64	1.21	3.13		17.51	0.23	16.93	0.25	18.11		6.78	1.36			
Na <sub>2</sub> O	0.27	0.02	0.29	0.02	0.57	0.63	0.04		0.24	0.02	0.21	0.01	0.24		0.09	0.07			
TiO <sub>2</sub>	1.10	0.12	1.13	0.09	0.67	0.34	0.44		0.87	0.01	0.78	0.01	0.38		0.44	0.09			
NiO	0.04	0.03	0.02	0.03	0.04	0.03	0.05		0.01	0.01	0.02	0.01	0.00		0.02	0.02			
Cr <sub>2</sub> O <sub>3</sub>	0.14	0.07	0.05	0.06	0.01	0.01	0.02		0.00	0.00	0.01	0.00	0.00		0.00	0.00			
H <sub>2</sub> O	1.12		1.18		1.25		1.25		0.77		0.84		1.12		0.97				
Total	100.00		100.00		100.00		100.00		100.00		100.00		100.00		100.00				
Wo (%)	38.50		38.10		37.05		6.27		35.77		34.88		37.99		14.13				
En (%)	44.70		43.47		35.95		56.20		40.28		37.70		24.26		40.86				
Fs (%)	16.80		18.43		27.00		37.54		23.95		27.43		37.75		45.01				

$\sigma$  = standart deviation, Interm = intermediary, Aug = augite and Pgt = pigeonite and n = number of analyses.

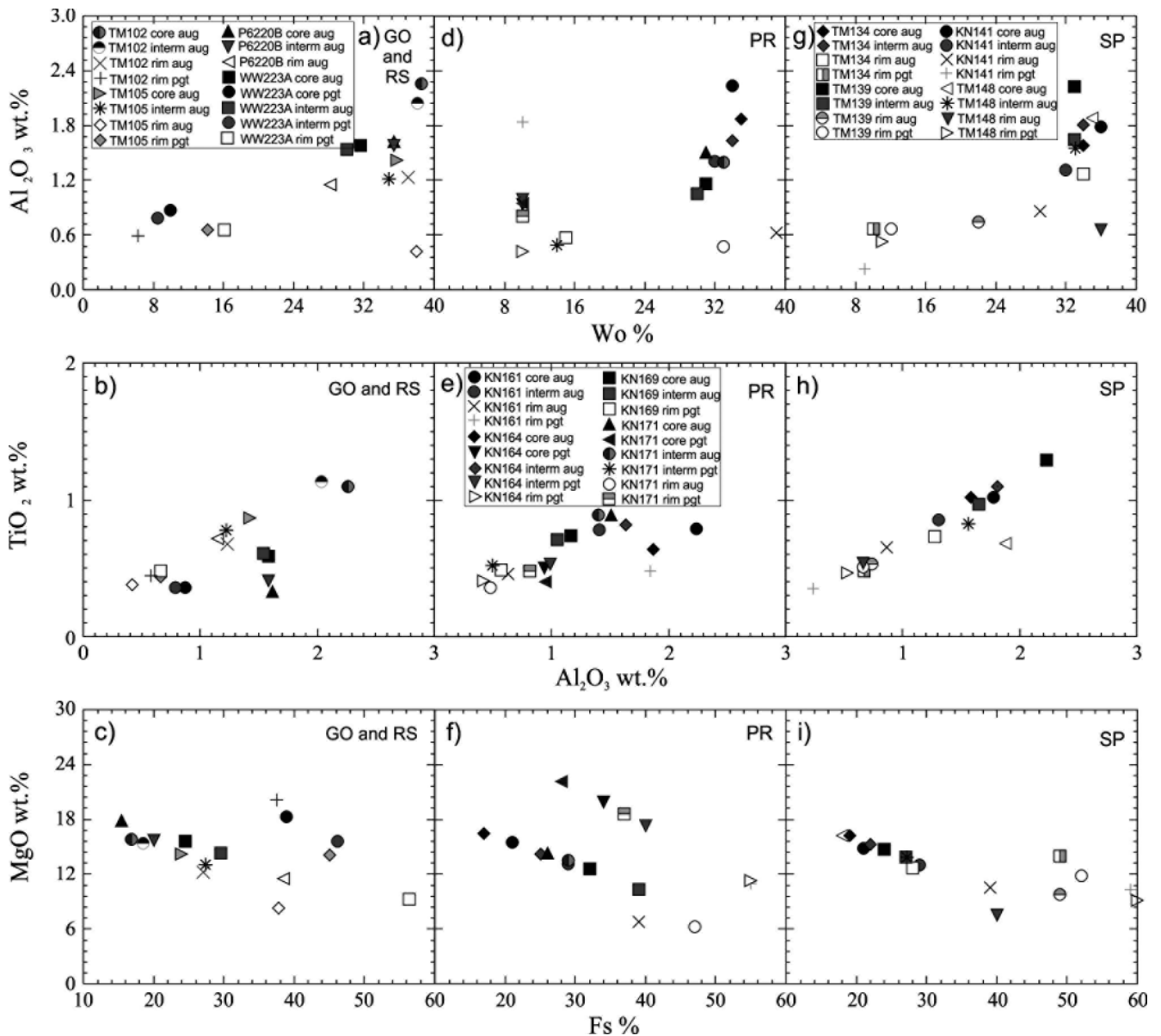


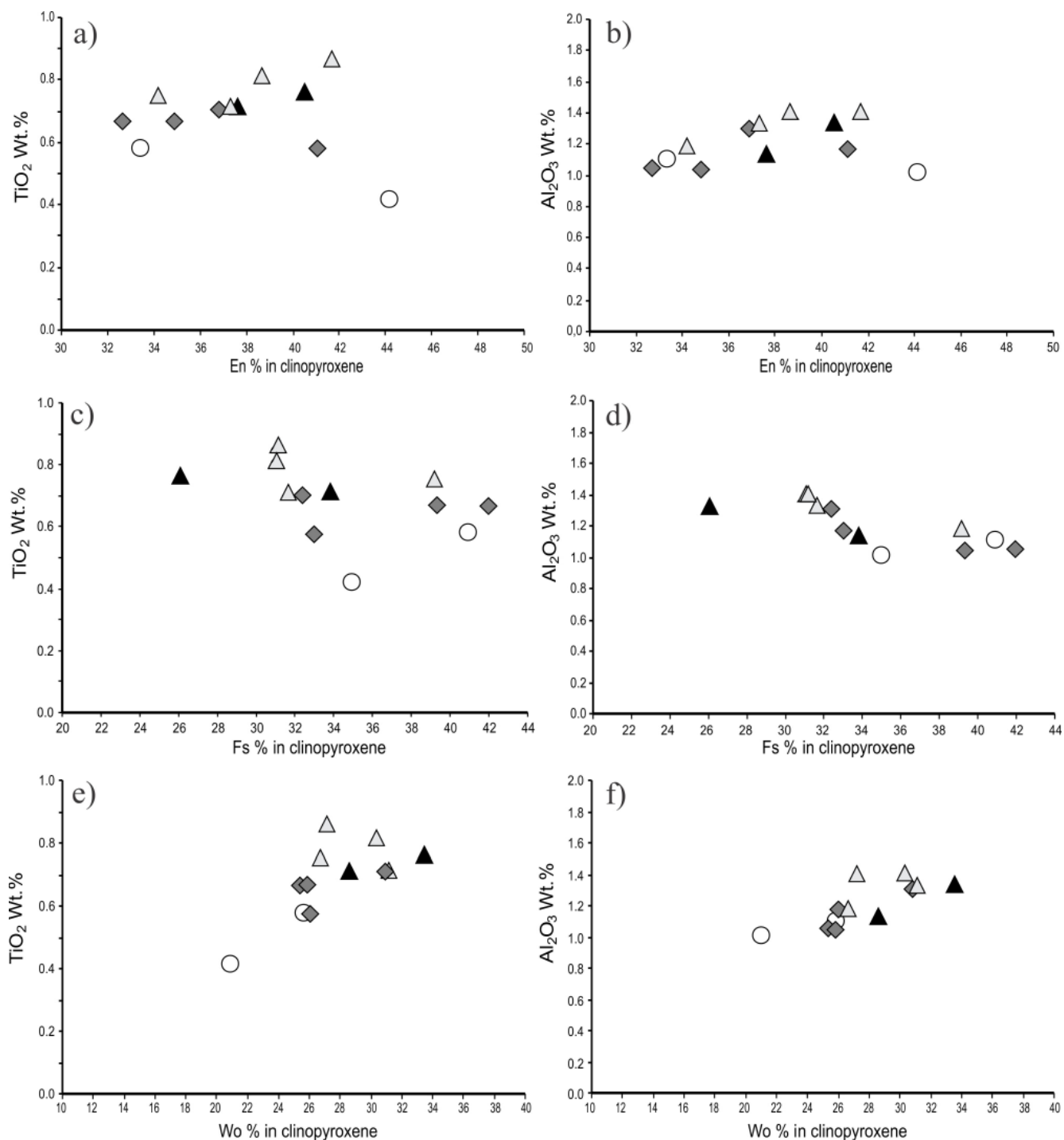
Figure 7 -  $Al_2O_3$  wt.% versus  $Wo\%$ ,  $TiO_2$  wt.% versus  $Al_2O_3$  wt.% and  $MgO$  wt.% versus  $Fs\%$  for clinopyroxene average microanalyses from EPMA in core, intermediate and rim zones from São Paulo (SP), Paraná (PR), Goiás (GO) and Rio Grande do Sul (RS). Aug = augite and pgt = pigeonite.

quantify the elements concentration of the melt and the analyses results of LA-ICP-MS quantified the values in ppm of the trace elements of core, intermediate and rim zones in those minerals. Table 4 shows data compilation of partition coefficients for elements in basaltic andesite and basalt which were studied by several authors in the 60 s and 90 s. It is worth mentioning that the  $K_D$  determination in those decades was carried out by separating minerals of interest from the rocks for further analyses by means of AAS and XRF according to available technology in that time.

The coefficient of partition in plagioclase and clinopyroxene crystals in this study are based on the use of advanced spot techniques of analyses, avoiding inclusions, zones and fractures which may interfere in

the values, thus obtaining  $K_D$  not only in the grain as a whole but in several portions of the crystal, allowing the understanding of the grain genesis and trace elements interaction in the temperature and crystallization ranges in these minerals.

Characteristic X-ray maps were done with the help of EPMA which resulted in different maps: K, Na, Ca, Fe, Mg, Ti and Al. In the present study, we selected one clinopyroxene crystal (KN141 - Ca and Mg elements) and one plagioclase crystal (KN161 - Ca and Na) to illustrate these maps. The characteristic X-ray map and backscattering images showing the EPMA analyses and LA-ICP-MS for different trace elements are shown in figure 9.



*Figure 8 - Clinopyroxene average values of TiO<sub>2</sub>, and Al<sub>2</sub>O<sub>3</sub> wt.% versus En, Fs and Wo% content in sills of Serra Geral Formation, Paraná Basin. Black triangle: clinopyroxene of Goiás; gray triangle: São Paulo; gray lozenge: Paraná; white circle: Rio Grande do Sul.*

**Plagioclase** In the studied rocks, plagioclase displays zonation in most of the analyzed crystals, identifying core, intermediate and rim. Those zonation were previously identified by microscope and backscattering images.

The LA-ICP-MS microanalyses in core, intermediate and rim zones in plagioclase crystals of Serra Geral Formation sills, Paraná Basin, were able

to identify variations in the partition coefficient shown in table 5. In this table, only the mean of two analyzed plagioclase crystals of KN161 sample from Paraná state and the mean of two augite crystals (KN141) from São Paulo state are discussed.

The most compatible element with plagioclase is Sr, followed by Eu and Pb (LILE) with compatible values at rim in some of the plagioclase.



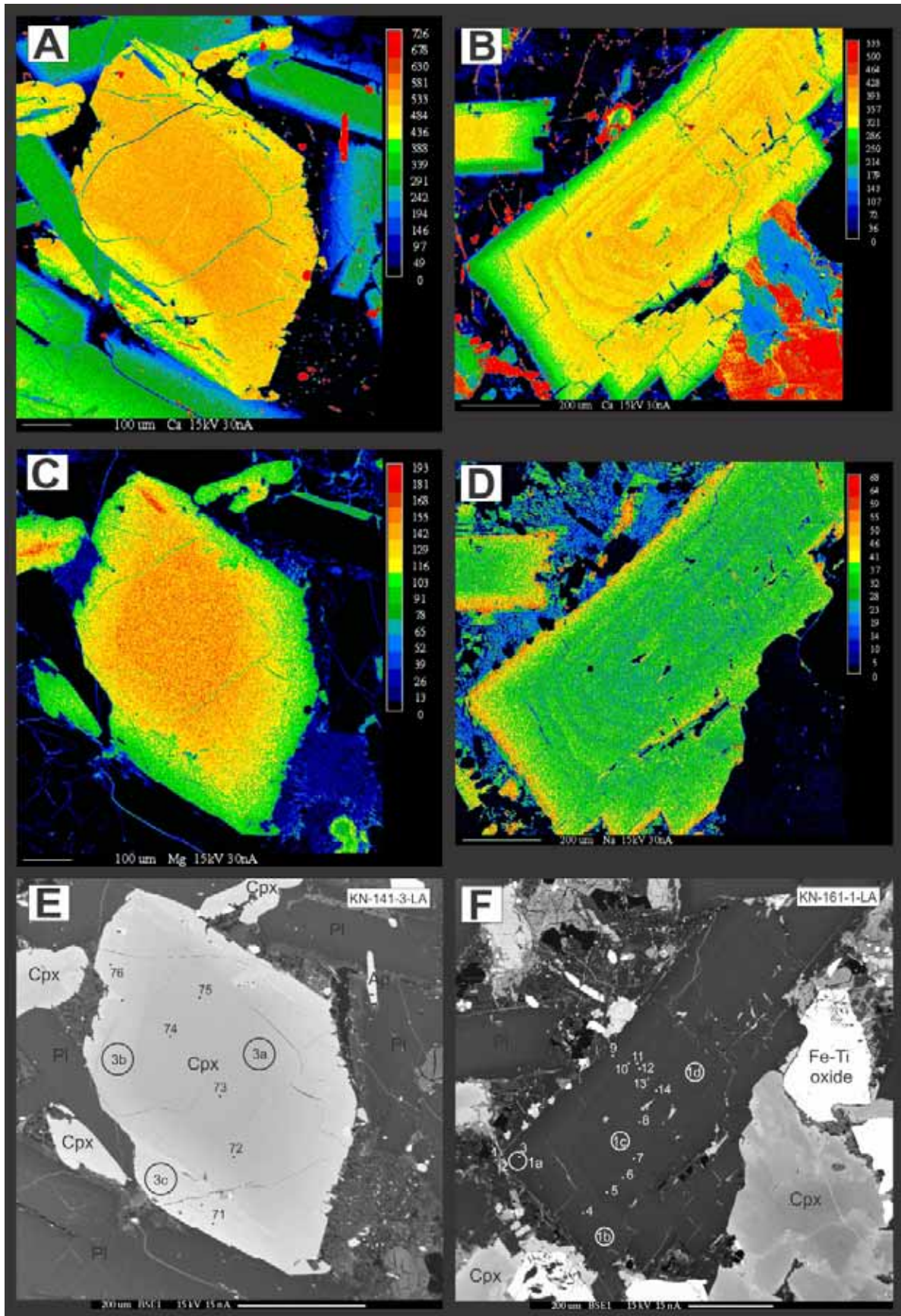


Figure 9 - Characteristic X-ray map composition of clinopyroxene (A) = Ca and (C) = Mg and plagioclase (B) = Ca and (D) = Na before EPMA and LA-ICP-MS microanalyses. Backscattered electron images of clinopyroxene (E) and plagioclase (F). Open circles show LA-ICP-MS analyses locations. The filled circles show EPMA analyses locations. Recommendations by the IUGS Subcommittee on the Systematics of Metamorphic Rocks: Pl = plagioclase; Cpx = clinopyroxene; Op = opaque minerals and Ap = apatite.

*Table 4 - Mineral/melt partition coefficients for basalt and basaltic andesite liquids compiled from 37 published experimental studies. Blank = not found.*

	Clinopyroxene		Plagioclase			Clinopyroxene		Plagioclase	
	Basalt	Basaltic andesite	Basalt	Basaltic andesite		Basalt	Basaltic andesite	Basalt	Basaltic andesite
Sc	1.42 <sup>(1)</sup> -3.3 <sup>(2)</sup>	2.2-3.5 <sup>(3)</sup>	0.008 <sup>(2)</sup> -0.1 <sup>(4)</sup>		Ce	0.017 <sup>(26)</sup> -1.05 <sup>(28)</sup>	0.09 <sup>(29)</sup> -0.58 <sup>(20)</sup>	0.016 <sup>(17)</sup> -0.2 <sup>(11)</sup>	0.06 <sup>(8)</sup> -0.14 <sup>(7)</sup>
Ti	0.37 <sup>(1)</sup> -0.45 <sup>(5)</sup>	0.19-0.34 <sup>(3)</sup>	0.038 <sup>(6)</sup>	0.04-0.057 <sup>(7)</sup>	Pr	0.04 <sup>(30)</sup> -0.626 <sup>(9)</sup>	0.635-1.012 <sup>(20)</sup>	0.063 <sup>(19)</sup> -0.17 <sup>(11)</sup>	0.035-0.13 <sup>(7)</sup>
V	0.74 <sup>(6)</sup> -4.82 <sup>(1)</sup>	0.5-7.2 <sup>(8)</sup>	0.01 <sup>(6)</sup>	0.022-0.032 <sup>(7)</sup>	Nd	0.06 <sup>(30)</sup> -0.69 <sup>(28)</sup>	0.5 <sup>(8)</sup> -1.31 <sup>(20)</sup>	0.014 <sup>(17)</sup> -0.168 <sup>(19)</sup>	0.026 <sup>(7)</sup> -0.15 <sup>(8)</sup>
Cr	2.94-26.91 <sup>(9)</sup>	9.7 <sup>(3)</sup> -70 <sup>(8)</sup>	0.02 <sup>(6)</sup> -0.6 <sup>(4)</sup>	0.075 <sup>(7)</sup>	Sm	0.462 <sup>(5)</sup>	0.50 <sup>(36)</sup> -0.445 <sup>(37)</sup>	0.067 <sup>(36)</sup> -0.072 <sup>(37)</sup>	0.0394 <sup>(37)</sup>
Mn	0.55-1.31 <sup>(12)</sup>	1.2-1.8 <sup>(3)</sup>	0.04 <sup>(13)</sup> -0.07 <sup>(6)</sup>		Eu	0.2 <sup>(30)</sup> -0.87 <sup>(18)</sup>	0.09 <sup>(8)</sup> -1.2 <sup>(20)</sup>	0.062 <sup>(19)</sup> -1.526 <sup>(18)</sup>	0.06 <sup>(8)</sup> -0.79 <sup>(7)</sup>
Co	0.68-1.22 <sup>(12)</sup>	1.2 <sup>(10)</sup> -2.4 <sup>(3)</sup>	0.07-0.5 <sup>(4)</sup>		Gd	0.24-0.84 <sup>(31)</sup>	0.778-1.24 <sup>(20)</sup>	0.004 <sup>(7)</sup> -0.066 <sup>(11)</sup>	0.016 <sup>(7)</sup> -0.08 <sup>(8)</sup>
Ni	1.2-10 <sup>(14)</sup>	4 <sup>(10)</sup> -10 <sup>(3)</sup>	0.04-0.5 <sup>(4)</sup>		Tb	0.28-1.124 <sup>(9)</sup>	0.57 <sup>(10)</sup> -1.33 <sup>(20)</sup>	0.013 <sup>(2)</sup> -0.11 <sup>(4)</sup>	0.04-0.061 <sup>(7)</sup>
Cu	0.071 <sup>(2)</sup> -0.36 <sup>(15)</sup>	0.05 <sup>(10)</sup> -0.69 <sup>(3)</sup>	0.004 <sup>(2)</sup> -0.17 <sup>(6)</sup>	0.07-0.38 <sup>(3)</sup>	Dy	0.256 <sup>(32)</sup> -1.18 <sup>(28)</sup>	0.764-1.09 <sup>(20)</sup>	0.018 <sup>(33)</sup> -0.314 <sup>(19)</sup>	0.011 <sup>(7)</sup> -0.06 <sup>(8)</sup>
Zn	0.5 <sup>(6)</sup>	0.24-0.31 <sup>(10)</sup>	0.11 <sup>(6)</sup> -0.18 <sup>(13)</sup>		Ho	0.3-1 <sup>(30)</sup>	0.712-1.13 <sup>(20)</sup>	0.011 <sup>(2)</sup> -0.048 <sup>(11)</sup>	
Rb	0.011 <sup>(11)</sup> -0.13 <sup>(4)</sup>	0.01-0.04 <sup>(8)</sup>	0.016 <sup>(16)</sup> -0.3 <sup>(4)</sup>	0.008 <sup>(7)</sup> -0.19 <sup>(8)</sup>	Er	0.259 <sup>(32)</sup> -1.17 <sup>(28)</sup>	0.615-0.922 <sup>(20)</sup>	0.018 <sup>(19)</sup> -0.041 <sup>(11)</sup>	0.005 <sup>(7)</sup> -0.06 <sup>(27)</sup>
Sr	0.04 <sup>(17)</sup> -0.44 <sup>(18)</sup>	0.6-0.9 <sup>(8)</sup>	1.55 <sup>(17)</sup> -10 <sup>(4)</sup>	1.3 <sup>(8)</sup> -3.5 <sup>(7)</sup>	Tm	0.2 <sup>(30)</sup> -1.047 <sup>(9)</sup>		0.036 <sup>(11)</sup>	
Y	0.29-1.71 <sup>(9)</sup>	0.66-1.12 <sup>(20)</sup>	0.023-0.03 <sup>(19)</sup>	0.01-0.038 <sup>(7)</sup>	Yb	0.178 <sup>(23)</sup> -1.14 <sup>(31)</sup>	0.09-1.03 <sup>(20)</sup>	0.004 <sup>(17)</sup> -0.04 <sup>(34)</sup>	0.004 <sup>(7)</sup> -0.3 <sup>(8)</sup>
Zr	0.001 <sup>(9)</sup> -0.7 <sup>(21)</sup>	0.235-0.382 <sup>(20)</sup>	0.01 <sup>(22)</sup> -0.27 <sup>(4)</sup>	0.0009 <sup>(7)</sup>	Lu	0.006 <sup>(9)</sup> -1.12 <sup>(28)</sup>	0.55-0.6 <sup>(10)</sup>	0.007 <sup>(19)</sup> -0.037 <sup>(33)</sup>	0.031-0.043 <sup>(27)</sup>
Nb	0.004-0.065 <sup>(21)</sup>	0.025-0.121 <sup>(20)</sup>	0.01 <sup>(11)</sup>	0.008-0.045 <sup>(7)</sup>	Hf	0.004 <sup>(9)</sup> -0.48 <sup>(4)</sup>	0.3 <sup>(10)</sup> -0.589 <sup>(20)</sup>	0.01 <sup>(11)</sup> -0.13 <sup>(4)</sup>	0.02 <sup>(10)</sup>
Mo				0.39 <sup>(7)</sup>	Ta	0.011-0.261 <sup>(21)</sup>	0.013 <sup>(36)</sup>	0.04-0.08 <sup>(4)</sup>	0.027-0.11 <sup>(7)</sup>
Cs	0.001 <sup>(23)</sup> -0.13 <sup>(4)</sup>	0.026-0.048 <sup>(20)</sup>	0.13 <sup>(4)</sup> -1 <sup>(13)</sup>	0.022-0.067 <sup>(7)</sup>	Pb	0.005 <sup>(24)</sup> -0.014 <sup>(5)</sup>	0.102-0.126 <sup>(20)</sup>	0.36 <sup>(11)</sup> -0.76 <sup>(13)</sup>	0.18-1.07 <sup>(7)</sup>
Ba	0.0001 <sup>(24)</sup> -0.05 <sup>(5)</sup>	0.002 <sup>(25)</sup> -0.2 <sup>(8)</sup>	0.183 <sup>(19)</sup> -3.6 <sup>(4)</sup>	0.03 <sup>(8)</sup> -1.45 <sup>(7)</sup>	Th	0.0003 <sup>(11)</sup> -0.04 <sup>(4)</sup>	0.019-0.038 <sup>(20)</sup>	0.05-0.09 <sup>(4)</sup>	0.01 <sup>(10)</sup> -0.19 <sup>(7)</sup>
La	0.002 <sup>(26)</sup> -0.21 <sup>(9)</sup>	0.12 <sup>(10)</sup> -0.4 <sup>(20)</sup>	0.02 <sup>(17)</sup> -0.3 <sup>(11)</sup>	0.075 <sup>(7)</sup> -0.227 <sup>(27)</sup>	U	0.0003 <sup>(23)</sup> -0.06 <sup>(35)</sup>	0.02 <sup>(20)</sup> -0.04 <sup>(10)</sup>	0.06-0.1 <sup>(4)</sup>	0.01 <sup>(10)</sup> -0.34 <sup>(7)</sup>

(1)=Jenner *et al.* 1993; (2)=Paster *et al.* 1974; (3)=Ewart *et al.* 1973; (4)=Villemant *et al.* 1981; (5)=Hauri *et al.* 1994; (6)=Bougault & Hekinian, 1974; (7)=Dunn & Sen, 1994; (8)=Reid, 1983; (9)=Skulski *et al.* 1994; (10)=Dostal *et al.* 1983; (11)=McKenzie & O'Nions, 1991; (12)=Dale & Henderson, 1972; (13)=Kravuchuk *et al.* 1981; (14)=Duke, 1976; (15)=Hart & Dunn, 1993; (16)=Matsui *et al.* 1977; (17)=McKay *et al.* 1994; (18)=Sun *et al.* 1974; (19)=Bindeman *et al.* 1998; (20)=Larsen, 1979; (21)=Forsythe *et al.* 1994; (22)=McCallum & Charette, 1978; (23)=Watson *et al.* 1987; (24)=Beattie, 1993; (25)=Hart & Brooks, 1974; (26)=Frey, 1969; (27)=Drake & Weill, 1975; (28)=Nagasawa, 1973; (29)=Gaetani & Grove, 1995; (30)=Irving & Frey, 1984; (31)=Hack *et al.* 1994; (32)=Sobolev *et al.* 1996; (33)=Schnetzler & Philpotts, 1970; (34)=Ringwood, 1970; (35)=Benjamin *et al.* 1978; (36)=Arth, 1976 and (37)=Fujimaki *et al.* 1984.

Table 6 shows the plagioclase compatible elements in the different analyzed samples. The mean values for partition coefficients for plagioclase analyses are shown in table 7. Only two samples (KN164 and KN161) present negative anomalies of REE, different from the other samples which show low and positive REE concentrations.

The concentrations in variation in elements Sr, Ba, Eu and Pb from core to rim are common for all the plagioclases with a normal crystallization. This fact occurs because of the valence similarity and ionic radius between the Ca<sup>2+</sup> (0.99 Å in eightfold coordination) and Na<sup>+</sup> (0.95 Å in eightfold coordination) with this elements. Ca low concentrations at rim of plagioclase

minerals (Fig. 9B) would correlate to low  $K_D$  for Sr and Eu, but in analyses of those rims we have found opposite, we found high Na concentrations (Fig. 9D) related to larger  $K_D$  at rim.

Aiming to identify the zones to which the element  $K_D$  is related, binary diagrams were used (Fig. 10) for showing the distribution in core, intermediate and rim for Ca and Na. The partition coefficients of Sr, Eu, Ba and Pb versus An% content in plagioclase crystals, and higher values of An are in the core of the crystal, the lowest values of An are related to the rim of the mineral. Figure 10a shows the  $K_D$  Sr, consistent element in all samples with negative correlation in most of them, that is, the  $K_D$  Sr increase from core to rim, except from TM105 sample, this suggests that for Sr the substitution element is related mostly by replacement of Na instead of Ca. The partition coefficient of Eu (Fig. 10b) shows that only half of the samples are compatible with this element in plagioclase, showing negative correlation with the An

content, and the higher values of  $K_D$  Eu are related to the rim of this mineral, with affinity for Na. This case resembles the diagram  $K_D$  Ba versus An%, with higher values for the rim. Pb is compatible in only five samples (core to rim), with negative to neutral correlations (Fig. 10d). Because the valence of  $Pb^{4+}$  (0.84 Å in sixfold coordination) is very different from the physical and chemical characteristics of Ca and Na, no satisfactory correlation of these elements was found in plagioclase, only with melt composition. Other variables, such as temperature, pressure, liquid composition, melt composition and  $fO_2$ , can modify the partition coefficient in plagioclase and clinopyroxene.

**Clinopyroxene** Clinopyroxene has a larger number of elements compared with plagioclase in the same sample. This is because of the variety between the various stages of clinopyroxene and diverse elements of its chemical formula ( $Mg_2Si_2O_6$ - $CaMgSi_2O_6$ - $CaFeSi_2O_6$ - $Fe_2Si_2O_6$ ) in quadrilateral representation.

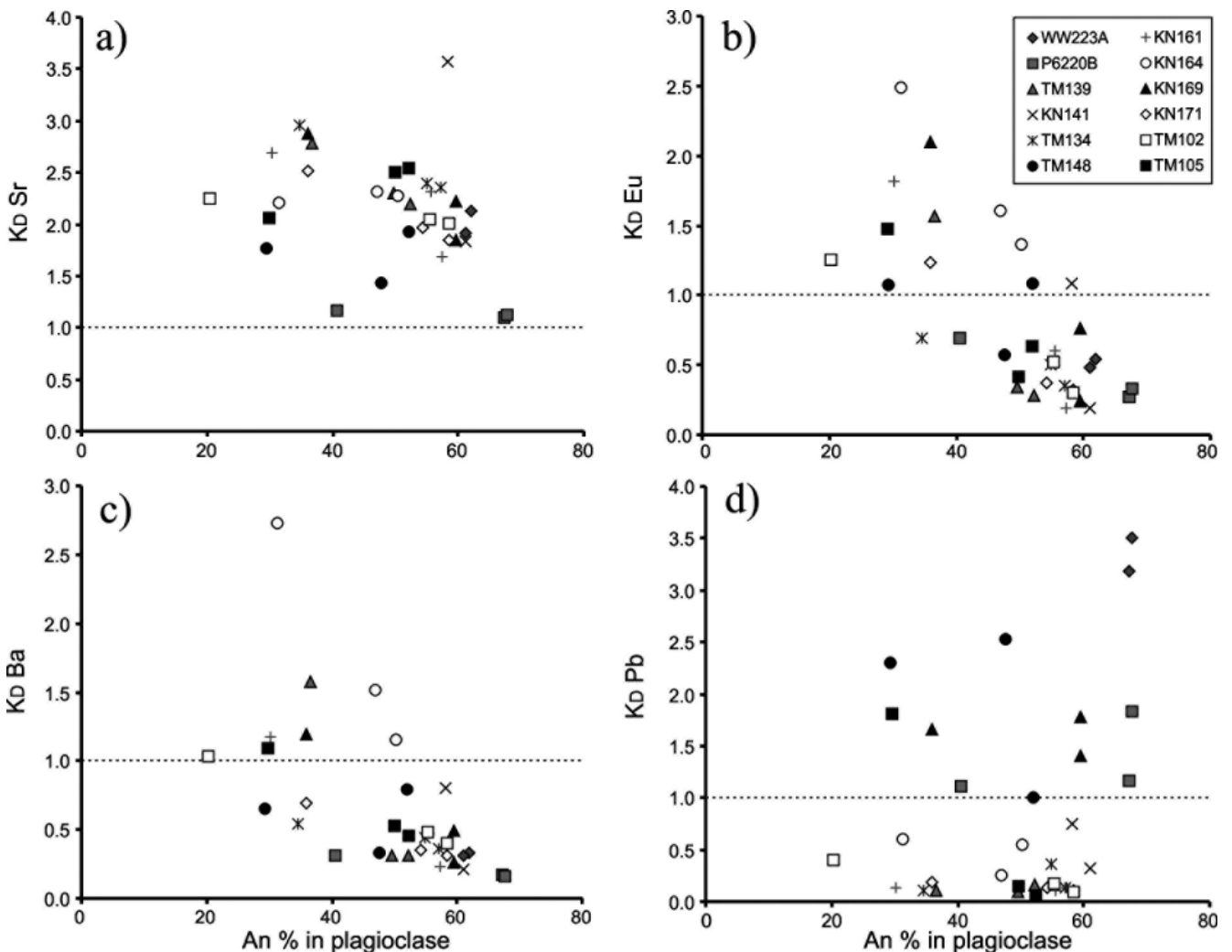


Figure 10 - Calculated partition coefficient (LA-ICP-MS) for a) Sr, b) Eu, c) Ba and d) Pb versus An% plagioclase microanalyses (EPMA). The dashed line delineates the field among the compatible element zone and incompatible element zone.

Table 5 - Laser Ablation ICP-MS (ppm) analyses and partition coefficients ( $K_D$ ) of core, intermediate and rim zones in plagioclases minerals from Paraná State and augites from São Paulo States.

Sample	Paraná - plagioclase										São Paulo - augite									
	Analyses					Partition Coefficient ( $K_D$ )					Analyses (ppm)					Partition Coefficient ( $K_D$ )				
	Core (n=3)	Interm (n=3)	Mean	$\sigma$	Core	Interm	Rim	Mean	$\sigma$	Core (n=2)	Interm (n=2)	Rim (n=1)	Mean	$\sigma$	Core	Interm	Rim	Mean	$\sigma$	
Sc	0.81	0.78	1.37	1.00	0.02	0.02	0.06	0.03	0.03	127.00	127.00	130.00	128.00	1.73	3.43	3.43	3.51	3.46	0.05	
Ti	497.33	648.33	565.56	76.55	0.04	0.05	0.04	0.01	0.01	7239.50	7128.00	4673.00	6346.83	1450.65	0.47	0.46	0.30	0.41	0.09	
V	5.75	4.01	3.78	2.10	0.02	0.01	0.00	0.01	0.01	552.00	520.00	153.00	408.33	221.70	1.15	1.08	0.32	0.85	0.46	
Cr	0.95	1.40	1.22	0.24	0.02	0.03	0.03	0.02	0.00	2.25	29.25	2.10	11.20	15.63	0.02	0.31	0.02	0.12	0.16	
Co	0.67	0.88	0.87	0.19	0.01	0.02	0.02	0.02	0.00	54.55	55.60	55.40	55.18	0.56	1.11	1.13	1.13	1.13	0.01	
Ni	0.26	0.32	0.29	0.03	0.01	0.02	0.01	0.00	0.00	80.75	83.25	33.60	65.87	27.97	2.77	2.45	0.99	2.07	0.95	
Cu	2.89	1.96	2.34	0.48	0.01	0.01	0.01	0.01	0.00	0.89	1.40	1.00	1.10	0.27	0.01	0.01	0.00	0.01	0.00	
Zn	4.67	7.68	7.05	2.13	0.05	0.07	0.09	0.07	0.02	54.30	64.30	129.00	82.53	40.55	0.53	0.65	1.30	0.83	0.42	
Rb	0.59	0.65	0.75	0.23	0.02	0.03	0.04	0.03	0.01	0.10	0.08	0.07	0.08	0.01	0.00	0.00	0.00	0.00	0.00	
Sr	496.67	695.33	668.00	159.43	1.68	2.32	2.69	2.23	0.51	17.45	19.00	13.80	16.75	2.67	0.05	0.05	0.04	0.04	0.01	
Y	0.14	0.17	1.06	1.57	0.00	0.00	0.08	0.03	0.04	17.90	21.90	73.80	37.87	31.18	0.49	0.63	2.13	1.08	0.91	
Zr	0.03	0.05	0.52	0.83	0.00	0.00	0.01	0.00	0.00	17.10	20.45	53.40	30.32	20.06	0.09	0.12	0.30	0.17	0.12	
Nb	0.03	0.04	0.05	0.02	0.00	0.00	0.00	0.00	0.00	0.07	0.10	0.14	0.10	0.04	0.00	0.01	0.01	0.01	0.00	
Mo	0.12	0.11	0.13	0.03	0.06	0.06	0.08	0.07	0.01	0.24	0.30	0.12	0.22	0.09	0.10	0.15	0.06	0.10	0.04	
Cs	0.03	0.02	0.03	0.01	0.15	0.10	0.15	0.13	0.03	0.05	0.06	0.04	0.05	0.01	0.23	0.28	0.20	0.23	0.04	
Ba	80.40	174.33	224.24	174.25	0.23	0.47	1.17	0.62	0.49	0.36	0.35	0.35	0.35	0.00	0.00	0.00	0.00	0.00	0.00	
La	0.82	1.40	2.11	1.76	0.04	0.06	0.19	0.09	0.08	1.17	1.28	5.96	2.80	2.74	0.04	0.05	0.25	0.11	0.12	
Ce	1.39	2.43	4.06	3.77	0.03	0.05	0.17	0.08	0.08	4.14	5.15	20.30	9.86	9.05	0.07	0.10	0.38	0.19	0.17	
Pr	0.15	0.22	0.38	0.34	0.02	0.04	0.13	0.06	0.06	0.72	1.10	3.73	1.85	1.64	0.12	0.17	0.59	0.30	0.26	
Nd	0.61	0.73	1.49	1.43	0.02	0.03	0.12	0.06	0.06	5.55	6.70	22.50	11.58	9.47	0.18	0.24	0.82	0.41	0.35	
Sm	0.12	0.15	0.37	0.41	0.02	0.03	0.15	0.06	0.07	2.27	2.97	7.07	4.10	2.59	0.35	0.49	1.17	0.67	0.44	
Eu	0.37	1.28	1.73	1.63	0.19	0.60	1.82	0.87	0.85	0.60	0.77	1.52	0.96	0.49	0.27	0.37	0.73	0.46	0.24	
Gd	0.10	0.12	0.35	0.42	0.01	0.02	0.14	0.06	0.07	2.50	4.33	10.80	5.88	4.36	0.40	0.68	1.70	0.92	0.68	
Tb	0.01	0.02	0.04	0.05	0.01	0.01	0.10	0.04	0.05	0.43	0.63	1.95	1.00	0.83	0.45	0.62	1.90	0.99	0.79	
Dy	0.06	0.07	0.21	0.26	0.01	0.01	0.09	0.04	0.04	3.71	4.12	11.70	6.51	4.50	0.51	0.70	2.00	1.07	0.81	
Ho	0.01	0.01	0.04	0.06	0.01	0.01	0.10	0.04	0.05	0.65	0.94	2.77	1.46	1.15	0.52	0.81	2.37	1.23	1.00	
Er	0.03	0.04	0.12	0.14	0.01	0.01	0.08	0.03	0.04	1.93	2.47	7.45	3.95	3.05	0.50	0.72	2.16	1.13	0.91	

$\sigma$  = standart deviation, Interm = intermediary and n = number of analyses.  
To be continued

Table 5 continuation

State	Paraná - plagioclase										São Paulo - augite									
	Analyses					Partition Coefficient ( $K_D$ )					Analyses (ppm)					Partition Coefficient ( $K_D$ )				
Sample	Core (n=3)	Interm (n=3)	Mean	$\sigma$	Core	Interm	Rim	Mean	$\sigma$	Core (n=2)	Interm (n=2)	Rim (n=1)	Mean	$\sigma$	Core	Interm	Rim	Mean	$\sigma$	
Tm	0.01	0.01	0.01	0.01	0.02	0.02	0.05	0.03	0.02	0.26	0.30	1.17	0.57	0.52	0.46	0.61	2.40	1.16	1.08	
Yb	0.05	0.07	0.12	0.10	0.02	0.02	0.07	0.04	0.03	1.68	2.43	6.99	3.70	2.88	0.55	0.80	2.29	1.21	0.94	
Lu	0.02	0.01	0.03	0.02	0.04	0.03	0.10	0.05	0.04	0.28	0.33	1.49	0.70	0.68	0.63	0.77	3.47	1.62	1.60	
Hf	0.05	0.04	0.06	0.02	0.01	0.01	0.02	0.01	0.00	0.97	1.00	1.89	1.29	0.52	0.18	0.22	0.41	0.27	0.12	
Ta	0.02	0.01	0.01	0.00	0.02	0.01	0.01	0.02	0.01	0.03	0.04	0.02	0.03	0.01	0.02	0.03	0.02	0.03	0.01	
Pb	2.14	1.45	1.77	0.35	0.14	0.11	0.13	0.13	0.02	1.19	1.31	4.82	2.44	2.06	0.31	0.44	1.61	0.78	0.72	
Th	0.03	0.01	0.02	0.01	0.01	0.01	0.01	0.01	0.00	0.03	0.05	0.16	0.08	0.07	0.01	0.02	0.06	0.03	0.03	
U	0.02	0.02	0.01	0.00	0.04	0.03	0.02	0.03	0.01	0.02	0.03	0.03	0.03	0.01	0.04	0.04	0.05	0.04	0.01	

$\sigma$  = standard deviation, Interm = intermediary and n = number of analyses.

The substitution of trace elements is mainly for  $Fe^{2+}$  (0.64 Å in sixfold coordination),  $Mg^{2+}$  (0.65 Å in sixfold coordination),  $Ca^{2+}$  (0.99 Å in eightfold coordination) and  $Al^{3+}$  (0.50 Å in six or fourfold coordination). Different concentrations of Ca and Mg in the crystal of clinopyroxene are shown in figures 9A and 9C where the core of the clinopyroxene has the highest concentrations of Mg and Ca with reduction to the rim. Figure 9E shows the difference in brightness, dark to the core and lighter for the rim due to the larger concentrations of Fe. In the same figure, the LA-ICP-MS and EPMA spots are also shown.

Table 8 shows the analyzed compatible elements in augite crystals.

A few crystals of clinopyroxene have rim of pigeonite and rare cases of core and intermediate zones, with values of CaO wt.% < 7. These areas of pigeonite were analyzed for trace elements, so it was possible to calculate the  $K_D$  of these elements. However, the difficulty in analyzing the rim of pigeonite is due to the diameter of the laser ablation spot (~50 µm). Figure 2A shows a very thin rim of pigeonite and figure 2B shows a thicker rim.

Observing tables 9 and 10 it is possible to identify which clinopyroxene samples have higher concentrations of HREE than LREE. Table 10 shows the compatible elements values for these pigeonite zones. However, the higher values of  $K_D$  of REE are found in the analyzed augite and smaller values of  $K_D$  for the pigeonite.

Figure 11 shows  $K_D$  values of Sc, Cr, Ni, Co, V, Zn, Pb and Lu versus En% concentrations in clinopyroxene crystals. En% shows positive correlation with Ca and Al and negative correlation with Fe. Thus, it was identified the probable replacement of these elements with the Mg, Ca and Al concentrations according to the crystal core, intermediate and rim. The partition coefficients have positive correlations with En concentrations in clinopyroxene:  $K_D$  Ni and  $K_D$  V (Figs. 11c and 11e). The negative correlations are:  $K_D$  Sc,  $K_D$  Cr,  $K_D$  Co,  $K_D$  Pb and  $K_D$  Lu (Figs. 11a, 11b, 11d, 11g and 11h). The above values of  $K_D > 1$  for Sc, Ni and Co show that these elements are compatible with the substitutions studied in clinopyroxene with Mg and Fe. The values of  $K_D$  Sc ( $Sc^{3+}$  0.81 Å in sixfold coordination) suggest substitution for Fe, related to an increase in  $K_D$  Sc to the rim mineral, except the TM139 and TM105 samples suggest that Mg substitution identified by the increase in  $K_D$  Sc values in the core (Fig. 11a). Distributions of partition coefficients are similar to figure 11d when compared to figure 11a. Figure 11e ( $K_D$  V) has positive correlations with En% (except P6220B, TM102 and TM134 samples negative correlations). Given the physical and chemical properties of  $V^{5+}$  (0.52 Å in sixfold coordination), this element has a higher probability of substitution for  $Al^{3+}$  (0.50 Å in sixfold coordination) than Fe and Mg. The diagram in figure 11f shows increase in the  $K_D$   $Zn^{2+}$  (0.74 Å in sixfold coordination) from the core to the rim, due to probable

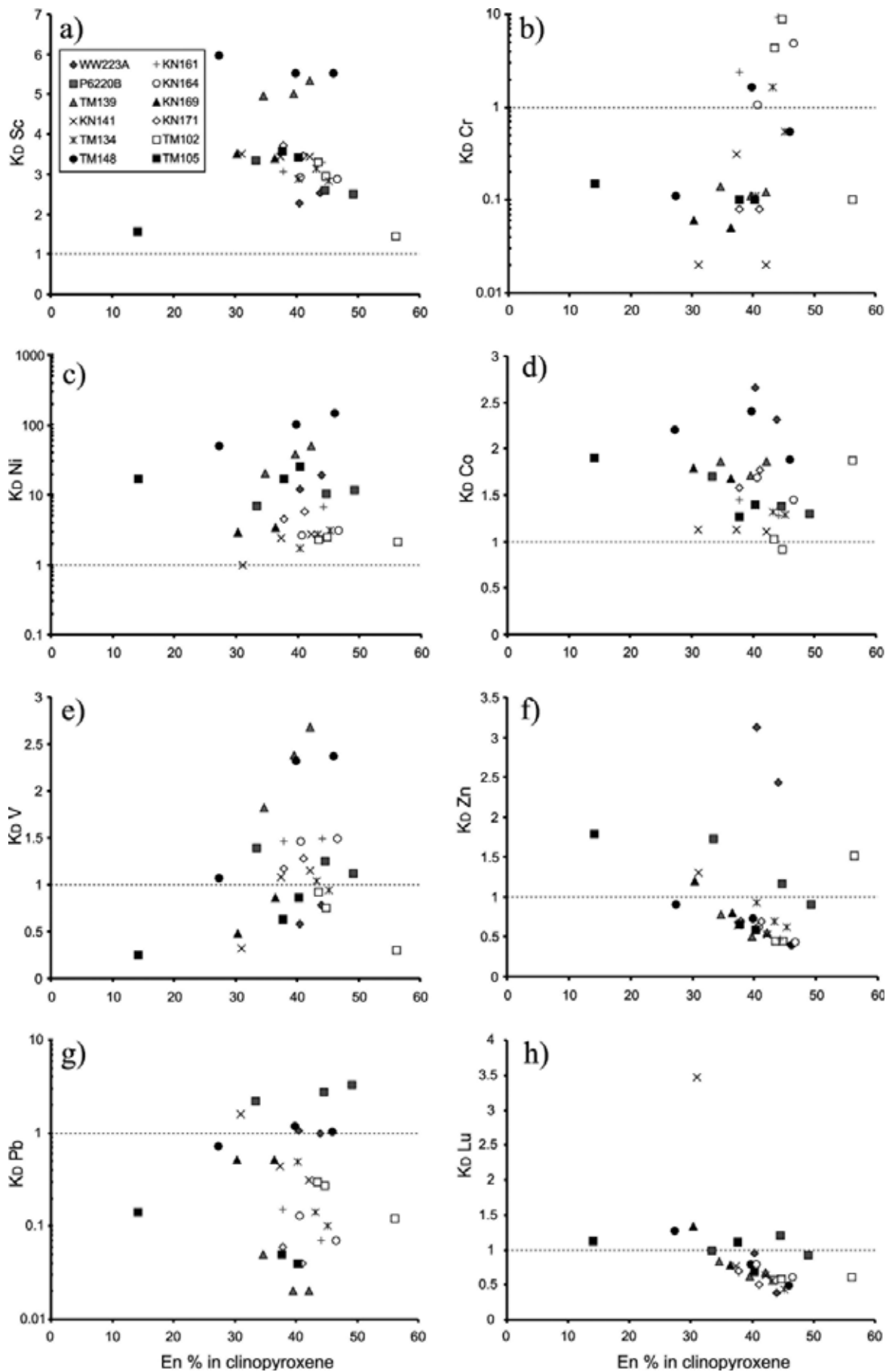


Figure 11 - Calculated partition coefficient (LA-ICP-MS) for a) Sc, b) Cr, c) Ni, d) Co, e) V, f) Zn, g) Pb and h) Lu versus En% clinopyroxene microanalyses (EPMA). The dashed line delineates the field among the compatible element zone and incompatible element zone.



Table 6 - Compatible elements zone in plagioclase crystals. Blank = not analyzed.

Element	Sr			Eu			Ba			Pb			
	Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A		2.13	1.91		< 0.5			< 0.3			3.18	3.51	
P6220B		1.1	1.13	1.16		< 0.7		< 0.3			1.16	1.83	1.11
KN141		1.83	3.57		0.19	1.08		< 0.8			< 0.7		
TM134		2.57	0.33	2.95		< 0.7		< 0.5			< 0.3		
TM139		2.2	2.3	2.78	0.28	0.34	1.57	0.31	0.31	1.58	< 0.1		
TM148		1.92	1.43	1.77	1.08	0.57	1.07		< 0.8		1	2.53	2.3
KN161		1.68	2.32	2.69	0.19	0.60	1.82	0.22	0.46	1.17	< 1.3		
KN164*		2.31	2.27	2.21	1.61	1.36	2.49	1.52	1.15	2.73	< 0.6		
KN169		1.84	2.22	2.88	0.24	0.76	2.1	0.26	0.49	1.19	1.78	1.4	1.66
KN171		1.84	1.97	2.51	0.32	0.37	1.23		< 0.9		< 0.2		
TM102		2.01	2.05	2.25	0.3	0.52	1.25	0.4	0.48	1.03	< 0.4		
TM105		2.5	2.54	2.06	0.42	0.64	1.48	0.54	0.47	1.11	0.16	0.1	1.8

\*KN164 (La core = 0.24, La interm = 0.55, La rim = 1.76; Ce core = 0.18, Ce interm = 0.48, Ce rim = 1.62; Pr core = 0.14, Pr interm = 0.42, Pr rim = 1.44; Nd core = 0.1, Nd interm = 0.37, Nd rim = 1.53; Sm core = 0.05, Sm interm = 0.28, Sm rim = 1.02).

substitution for the Fe (except TM102 sample with significant increase in rim areas and intermediate for the core, substitution with Mg). The diagram  $K_D$  Lu<sup>3+</sup> (0.85 in sixfold coordination) versus En% represents a similarity with the other elements HREE, suggesting negative correlations with the substitution for Fe.

**CONCLUSIONS** Based on geochemical relationships and *in situ* partition coefficients using Laser-ablation ICP-MS of plagioclase and clinopyroxene from sills in Serra Geral Formation, Paraná Basin, Brazil, we have identified change in composition of major and trace elements in core, intermediate and rim zones of these minerals.

Plagioclase is generally regarded to control partition coefficient of elements such as Sr, Eu and Pb. The crystallization of this mineral reflects the variations of major elements such as K<sub>2</sub>O, Al<sub>2</sub>O<sub>3</sub>, CaO and Na<sub>2</sub>O wt.% showing typical concentrations of basic and intermediate tholeiitic magmas.

According to mineral cooling the Ca<sup>2+</sup>Al<sup>3+</sup> ↔ Na<sup>+</sup>Si<sup>4+</sup> cation change by means of slow diffusion aiming at a rebalancing in NaAlSi<sub>3</sub>O<sub>8</sub>-CaAl<sub>2</sub>Si<sub>2</sub>O<sub>8</sub> system, starting the crystallization in *liquidus* until its entire crystallization in the *solidus* line in binary system (1 atm) from Bowen (1928). The more Ca terms (anorthite, bytownite and labradorite) are formed in higher temperatures while the more Na (albite, oligoclase and andesine) in lower temperatures. The differences in composition of Ca and Na are followed

by the partition coefficient values of several trace elements and REE.

Plagioclase has positive correlations for Na<sub>2</sub>O to the  $K_D$  Sr and Eu with higher values at the rim, suggesting a greater proportion of substitution of these elements for Na. An explanation for that fact would be high Na and K concentrations at plagioclase minerals rim. Explanation could be that the magma oxidation state can affect the partition coefficient of Eu, low  $fO_2$  increasing, because the ratio Eu<sup>2+</sup>/Eu<sup>3+</sup> increasing with fusion, and the Eu<sup>2+</sup> is much more compatible than Eu<sup>3+</sup> in plagioclase basaltic (Best 2003).

The compositional variations of major elements such as Fe<sup>2+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup> and Al<sup>3+</sup> in clinopyroxene influence the  $K_D$  of elements such as Sc, V, Cr, Co, Ni, Zn, Y, HREE and Pb in eightfold, sixfold and fourfold coordination.

The separation of the sills into high-Ti (TiO<sub>2</sub> ≥ 2%) and low-Ti (TiO<sub>2</sub> ≤ 2%) is not only due to the abundance of Ti-magnetites and ilmenites, but also by lower concentrations of this element in plagioclase, as shown in this work.

The major elements and trace elements in core, intermediate and rim zones when combined with detailed chemical mapping provides new insights into how plagioclase and clinopyroxene grew in basic and intermediate sills in Paraná Basin, Brazil.

Additional studies in plagioclase and clinopyroxene are required to evaluate their genesis. We suggest the analyses of variables, such as  $T$ ,  $P$ ,  $fO_2$ .

*Table 7 - Average partition coefficients from plagioclase sills in Serra Geral Formation, Paraná Basin, Brazil. Blank = not analyzed.*

Plagioclase												
State	Rio Grande do Sul			São Paulo				Paraná			Goiás	
Sample	WW223A	P6220B	TM139	KN141	TM134	TM148	KN161	KN164	KN169	KN171	TM102	TM105
Type	Basaltic andesite						Basalt					
Sc	0.02	0.02	0.07	0.03	0.01	0.03	0.03	0.02	0.02	0.02	0.02	0.02
Ti	0.04	0.04	0.04	0.05	0.02	0.03	0.04	0.05	0.04	0.04	0.04	0.03
V	0.01	0.02	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.01
Cr			0.06	0.02	0.04	0.06	0.02	0.01	0.03	0.03	0.01	0.05
Co	0.04	0.05	0.04	0.02	0.02	0.02	0.02	0.04	0.02	0.02	0.01	0.04
Ni	0.06	0.09	0.28	0.01	0.02	0.37	0.01	0.02	0.01	0.01	0.01	0.40
Cu	0.07	0.01	0.01	0.01	0.04	0.12	0.01	0.02	0.01	0.01	0.02	0.01
Zn	0.18	0.19	0.11	0.07	0.06	0.09	0.07	0.09	0.07	0.07	0.07	0.10
Rb	0.15	0.01	0.06	0.02	0.04	0.02	0.03	0.42	0.02	0.02	0.03	0.05
Sr	2.02	1.13	2.43	2.70	1.95	1.71	2.23	2.26	2.32	2.10	2.11	2.37
Y	0.02	0.01	0.07	0.01	0.01	0.00	0.03	0.21	0.01	0.00	0.01	0.01
Zr	0.03	0.00	0.02	0.00	0.01	0.00	0.00	0.07	0.00	0.00	0.00	0.00
Nb	0.05	0.00	0.01	0.00	0.01	0.00	0.00	0.03	0.00	0.00	0.00	0.00
Mo	0.07	0.15	0.06	0.07	0.04	0.06	0.07	0.09	0.05	0.06	0.05	0.06
Cs	0.15	0.02	0.03	0.14	0.05	0.04	0.13	0.27	0.03	0.07	0.12	0.09
Ba	0.32	0.21	0.73	0.51	0.36	0.59	0.62	1.80	0.65	0.45	0.64	0.71
La	0.11	0.06	0.10	0.08	0.06	0.07	0.09	0.85	0.07	0.05	0.10	0.07
Ce	0.11	0.05	0.09	0.06	0.05	0.05	0.08	0.76	0.06	0.04	0.08	0.06
Pr	0.08	0.03	0.08	0.05	0.04	0.03	0.06	0.67	0.04	0.03	0.06	0.04
Nd	0.06	0.03	0.07	0.04	0.03	0.03	0.06	0.67	0.03	0.03	0.05	0.04
Sm	0.05	0.02	0.07	0.03	0.02	0.02	0.06	0.45	0.02	0.02	0.03	0.02
Eu	0.51	0.43	0.73	0.64	0.46	0.90	0.87	1.82	1.03	0.64	0.69	0.85
Gd	0.03	0.03	0.07	0.03	0.02	0.01	0.06	0.36	0.01	0.01	0.02	0.01
Tb	0.02	0.02	0.06	0.02	0.01	0.01	0.04	0.27	0.01	0.01	0.01	0.01
Dy	0.02	0.01	0.07	0.01	0.01	0.01	0.04	0.28	0.01	0.01	0.01	0.01
Ho	0.02	0.02	0.07	0.02	0.01	0.01	0.04	0.24	0.01	0.01	0.02	0.01
Er	0.02	0.01	0.07	0.02	0.01	0.01	0.03	0.20	0.01	0.01	0.02	0.01
Tm	0.03	0.04	0.08	0.04	0.01	0.02	0.03	0.16	0.02	0.01	0.04	0.02
Yb	0.02	0.02	0.07	0.04	0.02	0.01	0.04	0.13	0.01	0.01	0.03	0.02
Lu	0.03	0.04	0.09	0.03	0.02	0.02	0.05	0.13	0.02	0.02	0.04	0.02
Hf	0.02	0.01	0.02	0.01	0.01	0.01	0.01	0.08	0.01	0.01	0.01	0.01
Ta	0.05	0.03	0.10	0.02	0.01	0.01	0.02	0.07	0.01	0.01	0.02	0.01
Pb	3.34	1.36	0.12	0.53	0.15	1.94	0.13	0.47	1.61	0.14	0.23	0.69
Th	0.03	0.00	0.02	0.01	0.02	0.00	0.01	0.14	0.01	0.01	0.01	0.00
U	0.13	0.01	0.02	0.04	0.03	0.02	0.03	0.14	0.03	0.02	0.04	0.02

Table 8 - Average partition coefficients from clinopyroxenes sills in Serra Geral Formation, Paraná Basin, Brazil. Blank = not analyzed.

Pyroxene												
State	Rio Grande do Sul			São Paulo				Paraná			Goiás	
Sample	WW223A	P6220B	TM139	KN141	TM134	TM148	KN161	KN164	KN169	KN171	TM102	TM105
Type	Basaltic andesite						Basalt					
Sc	2.52	2.82	4.97	3.46	2.95	5.67	3.18	2.89	3.45	3.60	3.17	3.50
Ti	0.26	0.39	0.50	0.41	0.29	0.28	0.41	0.49	0.39	0.41	0.43	0.24
V	0.78	1.26	2.29	0.85	0.95	1.92	1.45	1.47	0.67	1.23	0.84	0.75
Cr			0.12	0.12	0.76	8.56	5.81	3.00	0.05	0.08	4.51	0.10
Co	2.32	1.46	1.81	1.13	1.34	2.17	1.37	1.57	1.73	1.67	0.97	1.33
Ni	19.18	9.67	35.73	2.07	2.51	98.45	5.67	2.86	3.13	5.10	2.19	20.83
Cu	0.04	0.10	0.01	0.01	0.01	0.06	0.01	0.00	0.00	0.01	0.01	0.00
Zn	2.43	1.26	0.61	0.83	0.75	0.68	0.59	0.53	1.00	0.69	0.75	0.62
Rb	0.01	0.02	0.00	0.00	0.00	0.00	0.02	0.01	0.00	0.00	0.17	0.00
Sr	0.03	0.07	0.05	0.04	0.05	0.03	0.04	0.06	0.04	0.05	0.07	0.06
Y	0.35	0.77	0.57	1.08	0.51	0.54	0.53	0.58	0.87	0.56	0.60	0.73
Zr	0.04	0.15	0.12	0.17	0.09	0.07	0.11	0.11	0.13	0.09	0.13	0.11
Nb	0.01	0.10	0.00	0.01	0.00	0.00	0.02	0.03	0.00	0.00	0.00	0.00
Mo	0.24	0.26	0.10	0.10	0.09	0.15	0.11	0.09	0.12	0.13	0.09	0.09
Cs	0.03	0.09	0.06	0.23	0.10	0.08	0.22	0.11	0.06	0.16	0.17	0.09
Ba	0.00	0.02	0.00	0.00	0.00	0.00	0.01	0.02	0.00	0.00	0.11	0.00
La	0.02	0.09	0.04	0.11	0.04	0.03	0.08	0.06	0.06	0.04	0.09	0.06
Ce	0.04	0.15	0.08	0.19	0.08	0.05	0.11	0.10	0.10	0.07	0.14	0.11
Pr	0.07	0.19	0.13	0.30	0.13	0.08	0.16	0.17	0.16	0.13	0.19	0.19
Nd	0.08	0.24	0.19	0.41	0.19	0.11	0.22	0.21	0.28	0.19	0.26	0.28
Sm	0.11	0.44	0.35	0.67	0.32	0.26	0.35	0.40	0.47	0.31	0.43	0.45
Eu	0.11	0.32	0.31	0.46	0.31	0.21	0.31	0.30	0.31	0.28	0.41	0.35
Gd	0.28	0.60	0.48	0.92	0.43	0.36	0.51	0.51	0.63	0.45	0.50	0.60
Tb	0.20	0.59	0.49	0.99	0.43	0.41	0.53	0.49	0.64	0.45	0.50	0.61
Dy	0.36	0.71	0.60	1.07	0.54	0.52	0.61	0.58	0.84	0.60	0.61	0.77
Ho	0.34	0.83	0.67	1.23	0.58	0.68	0.63	0.68	0.90	0.62	0.67	0.81
Er	0.38	0.89	0.61	1.13	0.53	0.69	0.62	0.59	0.84	0.57	0.55	0.74
Tm	0.28	0.91	0.57	1.16	0.51	0.68	0.65	0.56	0.88	0.57	0.60	0.74
Yb	0.48	1.05	0.62	1.21	0.58	0.66	0.62	0.63	0.93	0.60	0.61	0.79
Lu	0.38	1.04	0.71	1.62	0.57	0.84	0.62	0.70	1.06	0.60	0.64	0.90
Hf	0.12	0.19	0.21	0.27	0.18	0.13	0.21	0.18	0.21	0.16	0.20	0.18
Ta	0.04	0.08	0.01	0.03	0.02	0.01	0.04	0.04	0.02	0.03	0.02	0.02
Pb	0.99	2.76	0.03	0.78	0.25	0.99	0.11	0.10	0.51	0.05	0.36	0.04
Th	0.01	0.03	0.01	0.03	0.01	0.01	0.04	0.03	0.01	0.02	0.02	0.01
U	0.01	0.19	0.02	0.04	0.03	0.03	0.07	0.04	0.05	0.04	0.05	0.03

*Table 9 - Compatible elements zone in augite crystals. Blank = not analyzed.*

Element	Sc			Co			Ni			Zn		
Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A	2.52			2.32			19.18			2.43		
P6220B	2.51	2.61	3.34	1.3	1.38	1.7	11.56	10.44	7	0.9	1.16	1.73
TM134	2.83	3.13	2.89	1.29	1.32	1.39	3.11	2.71	1.7	< 0.9		
TM139	5.33	5	4.59	1.86	1.71	1.86	49.4	37.45	20.33	< 0.7		
TM148	5.52	5.52	5.96	1.88	2.41	2.2	145.5	100.05	49.8	< 0.9		
KN161	3.3	3.06		1.28	1.45		6.66	4.68		< 0.7		
KN164	2.87	2.92		1.45	1.69		3.07	2.64		< 0.6		
KN169*	3.39	3.51		1.68	1.79		3.34	2.91		0.8	1.2	
KN171	3.47	3.73		1.77	1.58		5.68	4.52	5.1	< 0.7		
TM102	2.95	3.31	3.24	0.92	1.03	0.94	2.51	2.28	1.78	0.45	0.45	1.34
TM105	3.41	3.58		1.4	1.27		25	16.67		< 0.6		
Element	V			Cr			Pb			Tm		
Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A	< 0.8						0.99			< 0.3		
P6220B	1.12	1.25	1.39				3.3	2.77	2.2	0.74	0.96	1.04
TM134	0.94	1.04	0.86	0.55	1.64	0.11	< 0.5			< 0.6		
TM139	2.68	2.38	1.82	< 0.1			< 0.03			< 0.6		
TM148	2.37	2.32	1.07	23.39	2.12	0.17	1.05	1.19	0.73	< 0.9		
KN161	1.56	1.33		9.26	1.45		< 0.1			< 0.7		
KN164	1.49	1.46		4.92	1.07		< 0.1			< 0.6		
KN169*	< 0.8			< 0.06			< 0.5			0.65	1.1	
KN171	1.28	1.17		< 0.09			< 0.06			< 0.6		
TM102	< 0.9			8.83	4.32	0.39	< 0.5			< 0.8		
TM105	< 0.8			< 0.1			< 0.05			< 0.9		
Element	Lu			Yb			Ho			Er		
Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A	< 0.4			< 0.5			< 0.3			< 0.4		
P6220B	0.93	1.2	0.99	0.98	1.18	0.98	< 1			0.76	0.87	1.05
TM134	< 0.7			< 0.7			< 0.7			< 0.7		
TM139	< 0.8			< 0.7			< 0.7			< 0.7		
TM148	0.49	0.76	1.27	< 0.9			0.39	0.54	1.1	0.43	0.57	1.07
KN161	< 0.7			< 0.7			< 0.7			< 0.7		
KN164	< 0.8			< 0.6			< 0.7			< 0.7		
KN169*	0.78	1.34		0.72	1.14		0.71	1.1		0.69	1	
KN171	< 0.7			< 0.6			< 0.7			< 0.6		
TM102	< 0.8			< 0.7			< 0.8			< 0.6		
TM105	0.69	1.11		< 0.9			< 0.9			< 0.9		

\*KN169 (Y core = 0.65, Y interm = 1.10, Y rim = 0.87; Dy core = 0.62, Dy interm = 1.07, Dy rim = 0.84).

Table 10 - Compatible elements zone in pigeonite crystals. Blank = not analyzed.

Element	Sc			Co			Ni			Zn		
Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A	2.16	2.36		2.74	2.66		13.48	12.08		2.92	3.12	
KN164	1.42	1.5		2.56	2.57		4.11	3.58		1.26	1.38	
KN171	1.23			2.02			13.25			< 0.9		
TM102			1.45						1.87			2.11
TM105			1.56						1.9			17

Element	Pb			Cr			Lu		
Zone	Core	Interm	Rim	Core	Interm	Rim	Core	Interm	Rim
WW223A	1.29	1.06							< 1
KN164	< 0.3				< 0.8				< 0.4
KN171	< 0.2			3.47					< 0.1
TM102			< 0.1						< 0.6
TM105									1.13

**Acknowledgements** The first author wishes to thank CNPq and CAPES for the doctoral scholarship in Brazil and Germany, PRONEX-FAPERGS/CNPq (“Strategic Minerals”) for financial support.

CPRM (Geological Survey Brazil) provided support and most of the sill samples. Stuttgart University provided Electron Microprobe facility and overall support.

## References

- Arth J.G. 1976. Behaviour of trace elements during magmatic processes - a summary of theoretical models and their applications. *J. Res. U.S. Geol. Surv.*, **4**:41-47.
- Beattie P. 1993. The effect of partial melting of spinel peridotite on uranium series disequilibria: constraints from partitioning studies. *Earth and Planetary Science Letters*, **117**:379-391. DOI: 10.1016/0012-821X(93)90091-M.
- Bellieni G., Comin-Chiaromonte P., Marques L.S., Melfi A.J., Nardy A.J.R., Paratrechas C., Le Bas, M.J., Le Maitre R.W., Streckeis A. Zannetin B. 1986. A chemical classification of volcanic rocks based on total alkali-silica diagram. *Journal of Petrology*, **27**:745-750.
- Benjamin T., Heuser W.R., Burnett D.S. 1978. Laboratory studies of Actinide Partitioning Relevant to 244 Pu Chronometry. In: Lunar and Planetary Institute, Lunar and Planetary Science Conference, 9, *Proceedings Paper*, p. 1.393-1.406.
- Best M.G. 2003. *Igneous and Metamorphic Petrology*. 2.ed., Massachusetts, Blackwell Science Ltd., 729 p.
- Bindeman I.N., Davis A.M., Drake M.J. 1998. Ion microprobe study of plagioclases-basalt partition experiments at natural concentration levels of trace elements. *Geochimica et Cosmochimica Acta.*, **62**(7):1.175-1.193. DOI: 10.1016/S0016-7037(98)00047-7.
- Bougault H. & Hekinian R. 1974. Rift valley in the Atlantic Ocean near 36 degrees 50'N; petrology and geochemistry of basalt rocks. *Earth and Planetary Science Letters*, **24**(2):249-261. DOI: 10.1016/0012-821X(74)90103-4.
- Bowen N.L. 1928. *The evolution of the igneous rocks*. Princeton, Princeton University Press, 334 p.
- Dale I.M. & Henderson P. 1972. The Partition of Transition Elements in Phenocryst-bearing Basalts and the Implications about Melt Structure. In: Geological Survey of Canada, International Geological Congress, 24, *Sector 10*, p. 105-111.
- Deer A.A.W., Howie R.A., Zussman J. 2003. *Minerais constituintes das rochas - uma introdução*. Lisboa, Longmans, Green and Co. Ltda., 652 p.
- Dostal J., Dupuy C., Carron J.P., Dekerneizon M., Maury R.C. 1983. Partition-Coefficients of Trace-Elements - Application to Volcanic-Rocks of St-Vincent, West-Indies. *Geochimica et Cosmochimica Acta*, **47**(3):525-533. DOI: 10.1016/0016-7037(83)90275-2.
- Drake M.J. & Weill D.F. 1975. Partition of Sr, Ba, Ca, Y, Eu<sup>2+</sup>, Eu<sup>3+</sup>, and Other Rees between Plagioclases Feldspar and Magmatic Liquid - Experimental Study. *Geochimica et Cosmochimica Acta*, **39**(5):689-712. DOI: 10.1016/0016-7037(75)90011-3.
- Duke J.M. 1976. Distribution of the period four transition elements among olivine, calcic clinopyroxene and mafic silicate liquid; experimental results. *Journal of Petrology*, **17**(4):499-521.
- Dunn T. & Sen C. 1994. Mineral/Matrix Partition-Coefficients for Ortho-Pyroxene, Plagioclases, and Olivine in Basaltic to Andesitic Systems - a Combined Analytical and Experimental-Study. *Geochimica et Cosmochimica Acta*, **58**(2):717-733. DOI: 10.1016/0016-7037(94)90501-0.
- Ewart A., Bryan W.B., Gill J.B. 1973. Mineralogy and Geochemistry of the Younger Volcanic Islands of Tonga, S. W. Pacific. *Journal of Petrology*, **14**(3):429-465.
- Forsythe L.M., Nielsen R.L., Fisk M.R. 1994. High-Field-Strength Element Partitioning between Pyroxene and

- Basaltic to Dacitic Magmas. *Chemical Geology*, **117**(1-4):107-125. DOI: 10.1016/0009-2541(94)90124-4.
- Frey F.A. 1969. Rare earth abundances in a high-temperature peridotite intrusion. *Geochimica et Cosmochimica Acta*, **33**(11):1.429-1.447. DOI: 10.1016/0016-7037(69)90183-5.
- Fujimaki H., Tatsumoto M., Aoki K. 1984. Partition coefficients of Hf, Zr, and REE between phenocrysts and groundmasses. In: Lunar and Planetary Institute, Lunar and Planetary Science Conference, 14, *Proceedings Paper*, p. B662-B672.
- Gaetani G.A. & Grove T.L. 1995. Partitioning of Rare-Earth Elements between Clinopyroxene and Silicate Melt - Crystal-Chemical Controls. *Geochimica et Cosmochimica Acta*, **59**(10):1.951-1.962. DOI: 10.1016/0016-7037(95)01190-.
- Hack P.J, Nielsen R.L., Johnston A.D. 1994. Experimentally determined rare-Earth element and Y partitioning behavior between clinopyroxene and basaltic liquids at pressures up to 20 kbar. *Chemical Geology*, **117**:89-105. DOI: 10.1016/0009-2541(94)90123-6.
- Hart S.R. & Brooks C. 1974. Clinopyroxene-matrix partitioning of K, Rb, Cs, Sr and Ba. *Geochimica et Cosmochimica Acta*, **38**:1.799-1.806. DOI: 10.1016/0016-7037(74)90163-X.
- Hart S.R. & Dunn T. 1993. Experimental cpx/melt partitioning of 24 trace elements. *Contributions to Mineralogy and Petrology*, **113**:1-8.
- Hauri E.H., Wagner T.P., Grove T.L. 1994. Experimental and natural partitioning of Th, U, Pb and other trace elements between garnet, clinopyroxene and basaltic melts. *Chemical Geology*, **117**:149-166. DOI: 10.1016/0009-2541(94)90126-0.
- Irving A.J. & Frey F.A. 1984. Trace-Element Abundances in Megacrysts and Their Host Basalts - Constraints on Partition-Coefficients and Megacryst Genesis. *Geochimica et Cosmochimica Acta*, **48**(6):1.201-1.221. DOI: 10.1016/0016-7037(84)90056-5.
- Irvine T.N. & Baragar W.R.A. 1971. A guide to the chemical classification of the common volcanic rocks. *Canadian Journal of Earth Sciences*, **8**:523-548.
- Jenner G.A., Foley S.F., Jackson S.E., Green T.H., Fryer B.J., Longerich H.P. 1993. Determination of partition coefficients for trace elements in high pressure-temperature experimental run products by laser ablation microprobe-inductively coupled plasma-mass spectrometry (LAM-ICP-MS). *Geochimica et Cosmochimica Acta*, **57**(23-24):5.099-5.103.
- Kravuchuk I.K., Chernysheva I., Urosov S. 1981. Element distribution between plagioclases and groundmass as an indicator for crystallization conditions of the basalts in the southern vent of Tolbachik. *Geochemistry International*, **17**:18-24.
- Larsen L.M. 1979. Distribution of Ree and Other Trace-Elements between Phenocrysts and Peralkaline Undersaturated Magmas, Exemplified by Rocks from the Gardar Igneous Province, South Greenland. *Lithos*, **12**(4):303-315. DOI: 10.1016/0024-4937(79)90022-7.
- Lindsley D. 1983. Pyroxene thermometry. *American Mineralogist*, **68**:477-493.
- Machado F.B. 2003. *Geologia e possíveis zonas de efusão do magmatismo ácido cretácico da Bacia do Paraná*. Monografia (TCC), Instituto de Geociências e Ciências Exatas, Universidade Estadual Paulista, Rio Claro, 124 p.
- Matsui Y., Onuma N., Nagasawa H., Higuchi H., Banno S. 1977. Crystal structure control in trace element partition between crystal and magma. *Tectonics*, **100**:315-324.
- McCallum I.S. & Charette M.P. 1978. Zr and Nb partition coefficients: implications for the genesis of mare basalts, krep, and sea floor basalts. *Geochimica et Cosmochimica Acta*, **42**:859-869. DOI: 10.1016/0016-7037(78)90098-4.
- McKay G., Le L., Wagstaff J., Crozaz G. 1994. Experimental partitioning of rare Earth elements and strontium: constraints on petrogenesis and redox conditions during crystallization of Antarctic angrite Lewis Cliff 86010. *Geochimica et Cosmochimica Acta*, **58**:2.911-2.919. DOI: 10.1016/0016-7037(94)90124-4.
- McKenzie D. & O'Nions R.K. 1991. Partial melt distributions from inversion of rare Earth element concentrations. *Journal of Petrology*, **32**:1.021-1.091.
- Melfi A.J., Piccirillo E.M., Nardy A.J.R. 1988. Geological and magmatic aspects of the Paraná Basin - an introduction. In: Piccirillo E.M. & Melfi A.J. (eds.) *The Mesozoic flood volcanism of the Paraná Basin: petrogenetic and geophysical aspects*. São Paulo, Instituto Astronômico e Geofísico, p. 1-14.
- Morimoto N. 1988. Nomenclature of Pyroxene. *Min. Mag.*, **52**:535-50.
- Nagasawa H. 1973. Rare-Earth distribution in alkali rocks from Oki-Dogo Island, Japan. *Contributions to Mineralogy and Petrology*, **39**:301-308.
- Nardy A.J.R., Oliveira M.A.F., De Betancourt R.H.S., Verdugo D.R.H., Machado F.B. 2002. Geologia e estratigrafia da Formação Serra Geral. *Rev. Geociências*, **21**:15-32.
- Paster T.P., Schauwecker D.S., Haskin L.A. 1974. The behavior of some trace elements during solidification of the Skaergaard layered series. *Geochimica et Cosmochimica Acta*, **38**(10):1.549-1.577. DOI: 10.1016/0016-7037(74)90174-4.
- Pearce N.J.G., Perkins W.T., Westgate J.A., Gorton M.P., Jackson S.E., Neal C.R., Chenery S.P. 1997. A compilation of new and published major and trace element data for NIST SRM 610 and NIST SRM 612 glass reference materials. *Geostandards Newsletter*, **21**(1):115-144.
- Reid F. 1983. Origin of the Rhyolitic Rocks of the Taupo Volcanic Zone, New-Zealand. *Journal of Volcanology and Geothermal Research*, **15**(4):315-338.
- Ringwood A.E. 1970. Petrogenesis of Apollo 11 basalts and implications for lunar origin. *Journal of Geophysical Research*, **75**(32):6.453-6.479.
- Schnetzler C.C. & Philpotts J.A. 1970. Partition coefficients of rare-earth elements between igneous matrix material and rock-forming mineral phenocrysts; II. *Geochimica et Cosmochimica Acta*, **34**(3):331-340. DOI: 10.1016/0016-7037(70)90110-9.
- Skulski T., Minarik W., Watson E.B. 1994. High-Pressure Experimental Trace-Element Partitioning



- between Clinopyroxene and Basaltic Melts. *Chemical Geology*, **117**(1-4):127-147. DOI: 10.1016/0009-2541(94)90125-2.
- Sobolev A.V., Migdisov A.A., Portnyagin M.V. 1996. Incompatible element partitioning between clinopyroxene and basalt liquid revealed by the study of melt inclusions in minerals from Troodos lavas, Cyprus. *Petrology*, **4**(3):307-317.
- Sun C.O., Williams R.J., Sun S.S. 1974. Distribution coefficients of Eu and Sr for plagioclases-liquid and clinopyroxene-liquid equilibria in oceanic ridge basalt; an experimental study. *Geochimica et Cosmochimica Acta*, **38**(9):1.415-1.433. DOI: 10.1016/0016-7037(74)90096-9.
- Villemant B., Jaffrezic H., Joron J.L., Treuil M. 1981. Distribution Coefficients of Major and Trace-Elements - Fractional Crystallization in the Alkali Basalt Series of Chaine-Des-Puys (Massif Central, France). *Geochimica et Cosmochimica Acta*, **45**(11):1.997-2.016. DOI: 10.1016/0016-7037(81)90055-7.
- Watson E.B., Othman D.B., Luck J.M., Hofmann A.W. 1987. Partitioning of U, Pb, Cs, Yb, Hf, Re and Os between Chromian Diopsidic Pyroxene and Haplobasaltic Liquid. *Chemical Geology*, **62**(3-4):191-208. DOI: 10.1016/0009-2541(87)90085-4.
- Zanettin B. 1984. Proposed new chemical classification of volcanic rocks. *Episodes*, **7**:19-20.

**Manuscrito ID 16503**

**Submetido em 31 de dezembro de 2009**

**Aceito em 04 de outubro de 2011**