Oxidative stress: molecular perception and transduction of signals triggering antioxidant gene defenses

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Abstract

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Received January 31, 2005 Accepted April 18, 2005 Molecular oxygen (O_2) is the premier biological electron acceptor that serves vital roles in fundamental cellular functions. However, with the beneficial properties of O₂ comes the inadvertent formation of reactive oxygen species (ROS) such as superoxide $(O_2^{\bullet-})$, hydrogen peroxide, and hydroxyl radical (OH*). If unabated, ROS pose a serious threat to or cause the death of aerobic cells. To minimize the damaging effects of ROS, aerobic organisms evolved non-enzymatic and enzymatic antioxidant defenses. The latter include catalases, peroxidases, superoxide dismutases, and glutathione S-transferases (GST). Cellular ROS-sensing mechanisms are not well understood, but a number of transcription factors that regulate the expression of antioxidant genes are well characterized in prokaryotes and in yeast. In higher eukaryotes, oxidative stress responses are more complex and modulated by several regulators. In mammalian systems, two classes of transcription factors, nuclear factor kB and activator protein-1, are involved in the oxidative stress response. Antioxidant-specific gene induction, involved in xenobiotic metabolism, is mediated by the "antioxidant responsive element" (ARE) commonly found in the promoter region of such genes. ARE is present in mammalian GST, metallothioneine-I and MnSod genes, but has not been found in plant Gst genes. However, ARE is present in the promoter region of the three maize catalase (Cat) genes. In plants, ROS have been implicated in the damaging effects of various environmental stress conditions. Many plant defense genes are activated in response to these conditions, including the three maize Cat and some of the superoxide dismutase (Sod) genes.

Key words

- Catalase
- Aging
- Telomeres
- Gene regulation
- Superoxide dismutase

Genomics

Introduction

The environment in which they exist affects all living organisms. Whether internal or external to the organism, the environment is continually changing, and the organism must adapt if it is to survive. An organism apparently well adapted to its environment at any one time may be poorly adapted only

a short time later if it cannot modify its physiology or behavior in response to changing environmental or metabolic conditions. Organisms that can adjust to changes in the environment are likely to exhibit a greater degree of adaptiveness than those that cannot. Environmental changes, irrespective of source, cause a variety of "stresses" or "shocks" that a cell must face repeatedly,

and to which its genome must respond in a programmed manner for the organism to survive. How the genome perceives and transduces environmental signals to effect the expression and/or repression of pertinent genes in a selective manner remains a key question. Examples are responses to light, oxidative stress, pathogenicity, wounding, anaerobiosis, thermal shock, and the "SOS" response in microorganisms (1). Some sensing mechanism(s) must be present to alert the cell to imminent danger, and to trigger the orderly sequence of events that will mitigate this danger. In addition, there are genomic responses to unanticipated, unprogrammed challenges for which the genome is unprepared, but to which it responds in discernible though initially unforeseen and unpredictable ways (2). Many, though not all, signals are perceived at the cell surface by plasma membrane receptors. Activation of such receptors by mechanisms such as ligand binding may lead to alterations in other cellular components, ultimately resulting in alterations in cell shape, ion conductivity, gene activity, and other cellular functions (3). Identification and isolation of mutants that are unable to respond, or that respond abnormally to a particular signal, may provide ways to decipher the mechanisms by which a particular signal is transduced into a given response. Long before humans began manipulating and altering their environment, organisms from the simplest to the most complex began evolving methods to cope with stressful stimuli. Consequently, most living cells possess an amazing capacity to cope with a wide diversity of environmental challenges, including natural and synthetic toxins, pathogens, extreme temperatures, high metal levels, and radiation. Many studies in the past have demonstrated clear "cause-effect" relationships upon exposure of a given organism or cell to a particular environmental factor or stressor. But only recently have certain environmental insults been shown to elicit specific genomic responses (4). At present, relatively little is known of the underlying molecular mechanisms by which the genome perceives environmental signals and mobilizes the organism to respond. Such information is not only interesting in and of itself, but is also essential in any future attempts to engineer organisms for increased tolerance to environmental adversity. Recent dramatic advances in molecular biology and genomics have made it possible to investigate the underlying mechanisms utilized by organisms to cope with environmental stresses. Investigations of genomic responses to challenge are shedding light on unique DNA sequences capable of perceiving stress signals, thus allowing the cell or organism to mobilize its defenses (5). The general picture emerging from recent studies involves the sensing of a signal and the transduction of the signal to the transcription apparatus to catalyze transcription initiation. A signal transduction pathway contains elements that enable a signal to be transmitted within and between cells and to be translated into an appropriate response. Cells can respond to a variety of environmental, physical, and chemical stimuli using a diverse range of transduction and response mechanisms. The essential features of a signal transduction pathway comprise a receptor (recognition element) capable of detecting a stimulus, second messengers (transmission elements) such as calcium or phosphorylation cascades, and response elements (e.g., gene transcription). Such signaling networks are now amenable to study and dissection by biochemical and genetic approaches that may elucidate the underlying mechanisms and lead to the identification of the molecules responsible. A thorough understanding of how organisms perceive, respond, and adapt to changing environmental and developmental stimuli is now attainable.

Gene responses

Terminally differentiated cells express

an array of genes required for their stable functioning and precise metabolic roles. A genome can respond in a rapid and specific manner by selectively decreasing or increasing the expression of specific genes. Genes whose expression is increased during times of stress presumably are critical to the organism's survival under adverse conditions. Examination and study of such "stressresponsive" genes have implications for human health and well being, for agricultural productivity, and for furthering basic biological knowledge. In addition to aiding the organism under stress, genomes that are modified by stress can be utilized to study the molecular events that occur during periods of increased or decreased gene expression. The mechanisms by which an organism recognizes a signal to alter gene expression and responds to fill that need are important physiologically and render possible the examination of gene regulation under various environmental regimens. The mechanisms of induction of stress response genes are similar among various organisms examined. Similarities in stress-induced changes in gene expression have been observed for a variety of stressors (3,6-8). Some that have been studied in detail include radiation, thermal shock, pathogenic infections, anaerobiosis, trauma, photostress, physical wounding, oxidative stress, water stress, and heavy metals. In all cases, specific changes in transcript and/or protein expression have been observed in various organisms subjected to such challenges.

Recent studies by several laboratories, using a variety of organisms, indicate that oxidative stress is a common denominator underlying many diseases and environmental insults which can lead to cell death in virtually all aerobes (9). It is also becoming clear that a variety of different biotic and abiotic stresses cause their deleterious effects, directly or indirectly, via reactive oxygen species (ROS) generation.

For example, numerous toxic environ-

mental chemicals such as xenobiotics, pesticides, herbicides, fungicides, ozone, cigarette smoke, and radiation cause their harmful environmental effects via generation of free radicals and other ROS.

Herein, I will focus on oxidative stress, its causes and consequences, and mechanisms employed by organisms to cope with it.

Oxygenation of the Earth

Earth is the only planet in our solar system known to contain molecular oxygen (O_2) in its atmosphere and to support aerobic life. However, when Earth was formed about 4.5 billion years ago, its atmosphere was unlike the present, being primarily reducing and essentially free of oxygen. Most likely, the earliest living organisms were anaerobic heterotrophs living in the primitive ocean depths, shielded from the damaging effects of solar ionizing radiation. The earliest relatively low levels of oxygen were probably the result of photolytic dissociation of water by the sun's ionizing radiation. The bulk of Earth's present oxygen concentration (21% O₂) is derived from the photosynthetic activities of cyanobacteria and plants. It has been estimated that Earth contains about 410 $\times 10^3$ Erda moles (Emol = 10^8 mol) of oxygen, and of this, 38.4 x 103 Emol is in the hydrosphere as water. Molecular oxygen is present in the atmosphere (37 Emol) and in the hydrosphere (0.4 Emol) and undergoes continuous turnover, with the total oxygen exchange estimated at ~15 x 103 Emol/106 years. Aerobic life is responsible for the major portion of oxygen turnover, with photosynthesis being the main input into the oxygen reservoir, and respiration the main output. The two processes are in approximate equilibrium, and fossil fuel combustion is the major source of oxygen loss from the reservoir (10,11).

The accumulation of dioxygen (O_2) in Earth's atmosphere permitted the evolution of the enormous variety of aerobic organ-

isms that use O_2 as the electron acceptor, thus providing a higher yield of energy compared with fermentation and anaerobic respiration.

O₂, ROS, and oxygen toxicity

In its ground state (its normal configuration, O_2) molecular oxygen is relatively unreactive. However, during normal metabolic activity, and as a consequence of various environmental perturbations (e.g., extreme temperatures, radiation, xenobiotics, toxins, air pollutants, various biotic and abiotic stresses, and diseases) O_2 is capable of giving rise to frightfully reactive excited states such as free radicals and derivatives (9,12).

The oxidation powers of O_2 are restricted because electrons can only be absorbed from another species whose electron spin is antiparallel to the two unpaired, parallel-spin electrons in diatomic oxygen. This spin restriction renders ground state molecular oxygen sufficiently unreactive, so that it cannot abstract electrons from other species. However, removal of the spin restriction by adding a single e-, or upon transfer of energy to oxygen from a photosensitizer (e.g., chlorophyll, flavin-containing compounds), increases the reactivity of oxygen. Photosensitizers can harvest light and energize O2 to form singlet oxygen (${}^{1}\Delta_{g}O_{2}$), which can interact directly with another molecule, transferring the additional energy to the target

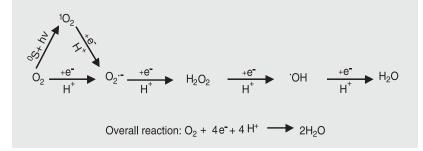


Figure 1. Pathways in the univalent reduction of oxygen to water leading to generation of various intermediate reactive oxygen species (ROS).

molecule.

The complete reduction of O_2 to water requires four electrons. O₂ has a preference for a stepwise univalent pathway of reduction resulting in partially reduced intermediates (Figure 1). The reactive species of reduced dioxygen include the superoxide radical $(O_2^{\bullet-})$, hydrogen peroxide (H_2O_2) , and the hydroxyl radical (OH•). The latter can also be generated by the interaction of O₂•and H₂O₂ in the presence of metal ions. Both O₂•- and OH• are extremely reactive and can cause molecular damage, leading to cell death. The hydroxyl radical reacts with virtually anything, inflicting indiscriminate and extensive intracellular damage. The O₂•- is the conjugate base of a weak acid, the perhydroxyl radical (HO₂), whose pK_a is 4.69 ± 0.08 . Thus, under acidic conditions, the very reactive perhydroxyl radical may predominate following a one electron reduction of dioxygen, while at higher pH values the O₂•- is predominant. These and the physically energized form of dioxygen, singlet oxygen (1O2), are the biologically most important ROS (Table 1). An activation energy of ~22 kcal/mol is required to raise molecular oxygen (O₂) from its ground state to its first singlet state. In higher plants, this energy is readily obtained from light quanta via such transfer molecules as chlorophyll. Unless abated, all of these intermediate oxygen species are extremely reactive and cytotoxic in all organisms (9,13). ROS can interact with proteins, lipids, and nucleic acids to cause severe molecular damage (Table 2). Thus, oxygen provides a paradox, in that it is essential for aerobic life, yet in its reduced forms is one of the most toxic substances with which life on Earth must cope. ROS are found in virtually all intracellular organelles or compartments as a consequence of normal metabolic activity. Each organelle or compartment has potential targets for oxidative damage, as well as mechanisms for the elimination of excess ROS accumulation (Figure 2).

Defenses against reactive oxygen

To minimize the damaging effects of ROS, aerobic organisms evolved both nonenzymatic and enzymatic antioxidant defenses (Table 3). Non-enzymatic defenses include compounds of intrinsic antioxidant properties, such as vitamins C and E, glutathione, and ß-carotene. Purely enzymatic defenses, such as superoxide dismutases

Name	Notation	Some comments and basic sources
Molecular oxygen (triplet ground state)	O ₂ ; ³ Σ	Common form of dioxygen gas
Singlet oxygen (1st excited singlet state)	¹ O ₂ ; ¹ Δ	Photoinhibition; UV irradiation; PS II e ⁻ transfer reactions (chloroplasts)
Superoxide anion	O ₂ •-	Formed in many photooxidation reactions (flavoprotein, redox cycling); Mehler reaction in chloroplasts; mitochondrial e ⁻ transfer reactions; glyoxysomal photorespiration; peroxisomal activity; nitrogen fixation; reactions of O ₃ and OH [•] in apoplastic space; defense against pathogens; oxidation of xenobiotics
Hydrogen peroxide	H ₂ O ₂	Formed from O ₂ *- by dismutation; photorespiration; β-oxidation; proton-induced decomposition of O ₂ *-; defense against pathogens
Hydroxyl radical	OH•	Decomposition of O ₃ in apoplastic space; defense against pathogens; reactions of H ₂ O ₂ with O ₂ *- (Haber-Weiss); reactions of H ₂ O ₂ with Fe ²⁺ (Fenton) highly reactive with all macromolecules
Perhydroxyl radical	O ₂ H•	Protonated form of O ₂ •-; reactions of O ₃ and OH• in apoplastic space
Ozone	O ₃	UV radiation or electrical discharge in stratosphere; reactions involving combustion products of fossil fuels and UV radiation in troposphere

Table 2. Examples of reactive oxygen species (ROS) damage to lipids, proteins and DNA.

Oxidative damage to lipids

- Occurs via several mechanisms of ROS reacting with fatty acids in the membrane lipid bilayer, leading to membrane leakage and cell death.
- · In foods, lipid peroxidation causes rancidity and development of undesirable odors and flavors.

Oxidative damage to proteins

- · Site-specific amino acid modifications (specific amino acids differ in their susceptibility to ROS attack)
- · Fragmentation of the peptide chain
- · Aggregation of cross-linked reaction products
- · Altered electrical charge
- · Increased susceptibility to proteolysis
- Oxidation of Fe-S centers by $O_2^{\bullet-}$ destroys enzymatic function
- · Oxidation of specific amino acids "marks" proteins for degradation by specific proteases
- · Oxidation of specific amino acids (e.g., Try) leads to cross-linking

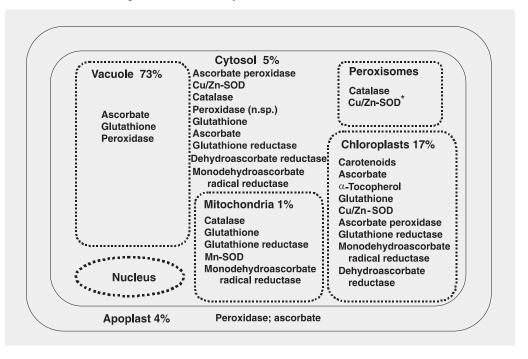
Oxidative damage to DNA

- · DNA deletions, mutations, translocations
- Base degradation, single-strand breakage
- · Cross-linking of DNA to proteins

(SOD), catalases (CAT) and peroxidases, protect by directly scavenging superoxide radicals and hydrogen peroxide, converting them to less reactive species. SODs catalyze

the dismutation of $O_2^{\bullet-}$ to H_2O_2 , and CAT and peroxidases reduce H_2O_2 to $2H_2O$. The similarity between the SOD and CAT reactions is that each is an oxidation-reduction in

Figure 2. Intracellular antioxidant resources in plant cells. SOD = superoxide dismutase.



Non-enzymatic antioxidant molecules				
Antioxidant molecule	Subcellular location			
Ascorbate (vitamin C)		Plastid; apoplast; cytosol; vacuole		
B-Carotene		Plastid		
Glutathione, reduced (GSH)		Plastid; mitochondrion; cytosol		
Polyamines (e.g., putrescine, spermine	e)	Nucleus; plastid; mitochondrion; cytosol		
α-Tocopherol (vitamin E)		Cell and plastid membranes		
Zeaxanthin		Chloroplast		
Antioxidant enzymes				
Enzyme	EC number	Subcellular location		
Ascorbate peroxidase	1.11.1.11	Plastid stroma and membranes		
Peroxidases (non-specific)	1.11.1.7	Cytosol; cell wall-bound		
Catalase	1.11.1.6	Glyoxysome; peroxisome; cytosol; mitochondria		
Superoxide dismutase (SOD)	1.15.1.1	Cytosol (Cu/ZnSOD); plastid (Cu/ZnSOD; FeSOD); mitochondrion (MnSOD); peroxisome		
Dehydroascorbate reductase	1.8.5.1	Cytosol; plastid		
Glutathione reductase	1.6.4.2	Mitochondrion; cytosol; plastid		
Monodehydroascorbate reductase	1.6.5.4	Plastid stroma		
Glutathione S-transferases	2.5.1.18	Cytosol; microsomal		

which the substrate, $O_2^{\bullet-}$ for SOD and H_2O_2 for CAT, is both reductant and oxidant, whereas different reductants are required for the peroxidases, depending upon their specificities. Under some conditions CAT can act as an efficient peroxidase. SODs deal with the first product of the univalent reduction of O_2 , converting it to H_2O_2 , which must then be destroyed by CAT and/or peroxidases. Thus, the SOD and CAT serve, in tandem, as front-line antioxidant defenses:

$$O_2^{\bullet-} + O_2^{\bullet-} + 2H^+ \xrightarrow{SOD} O_2 + H_2O_2 (K_2 = 2.4 \times 10^9 \text{ M}^{-1} \text{ s}^{-1})$$
 $H_2O_2 + H_2O_2 \xrightarrow{CAT} 2 H_2O + O_2 (K_1 = 1.7 \times 10^7 \text{ M}^{-1} \text{ s}^{-1})$
 $H_2O_2 + R(OH)_2 \xrightarrow{Px} 2H_2O + R(O)_2 (K_4 = 0.2-1 \times 10^3 \text{ M}^{-1} \text{ s}^{-1})$

Superoxide dismutase

SODs have been isolated and characterized from a wide variety of organisms. One class consists of SODs with Cu(II) plus Zn(II) at the active site (Cu/ZnSOD), another with Mn(III) (MnSOD), a third with Fe(III) (FeSOD), and a fourth with Ni(II/III) (NiSOD). Cu/ZnSODs are generally found in the cytosol of eukaryotic cells, in chloroplasts, and in some prokaryotes; MnSODs are found in prokaryotes and in mitochon-

dria; FeSODs are generally found in prokaryotes, in algae and in some higher plant chloroplasts; NiSODs have been found in Streptomyces. Unlike most other organisms that have only one of each type of SOD in the various cellular compartments, plants have multiple forms of each type encoded by more than one gene (Figure 3), indicative that plants have far more complex antioxidant defenses (14,15). Plants also produce a large variety of small non-enzymatic antioxidant compounds as second tier defenses, such as glutathione, ascorbate, tocopherols, flavonoids, alkaloids, and carotenoids in high concentrations that are capable of quenching ROS. The dismutation of $O_2^{\bullet-}$ to $O_2 +$ H₂O₂ by SOD is hardly a bargain, as the resulting H₂O₂ can react with metal ions, giving rise to the highly toxic OH. Fortunately, CAT come to the rescue by degrading H_2O_2 to O_2 and H_2O . Most aerobes, including mammals, possess at least one form of homotetrameric CAT with ferriheme at the active sites.

Catalase

CAT is largely, but not exclusively, localized in peroxisomes, wherein many H₂O₂producing enzymes reside. Thus CAT, which

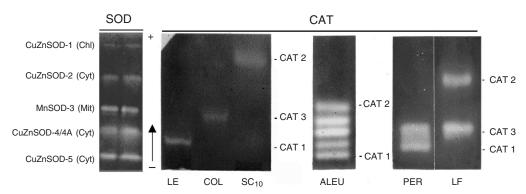


Figure 3. Zymograms showing multiplicity of superoxide dismutase (SOD, left) and catalase (CAT, right) in maize. The cytosolic CuZnSOD-4 and SOD-4A co-migrate to the same position, as do all members of the *Sod3* multigene family (14). Intracellular location is indicated in parentheses. Catalases are tissue-specific in their expression: LE = milky endosperm; COL = coleoptile; SC = scutellum; ALEU = aleurone; PER = pericarp; LF = green leaf. When CAT-1 and CAT-2 are co-expressed (ALEU) their subunits interact to generate intergenic heterotetramers. CAT-3 does not form heterotetramers *in vivo* when co-expressed with either CAT-1 or CAT-2 (16). Arrow shows direction of migration in gel.

exhibits a high $K_{\rm m}$ for ${\rm H_2O_2}$, can act upon the ${\rm H_2O_2}$ produced before it diffuses to other parts of the cell. CAT is a tetrameric hemecontaining enzyme that is found in all aerobic organisms. Because of its wide distribution, evolutionary conservation, and capacity to rapidly degrade hydrogen peroxide, it has been proposed that CAT plays an important role in systems which have evolved to allow organisms to live in aerobic environments.

CAT is one of the most active catalysts produced by nature. It decomposes hydrogen peroxide at an extremely rapid rate, corresponding to a catalytic center activity of about 10^7 min⁻¹. Depending upon the concentration of H_2O_2 , it exerts a dual function. At low concentrations (<1 μ M) of H_2O_2 , it acts "peroxidatically", i.e., a variety of hydrogen donors (e.g., ethanol, ascorbic acid) can be oxidized in the following manner: $RH_2 + H_2O_2 \rightarrow R + 2H_2O$.

At high concentrations of substrate, CAT decomposes toxic hydrogen peroxide at an extremely rapid rate using the "catalatic" reaction in which H_2O_2 acts as both an acceptor and donor of hydrogen molecules: $2H_2O_2 \rightarrow 2H_2O + O_2$.

CAT is unique among H₂O₂ degrading enzymes in that it degrades H₂O₂ without consuming cellular reducing equivalents. Hence, CAT provides the cell with a very energy efficient mechanism to remove hydrogen peroxide. Therefore, when cells are stressed for energy and are rapidly generating H₂O₂ through "emergency" catabolic processes, H₂O₂ is degraded by CAT in an energy-efficient manner. This results in a net gain of reducing equivalents and, therefore, cellular energy. It has been proposed that CAT may be uniquely suited to regulate the homeostasis of H₂O₂ in the cell. In the catalatic mode, CAT has a very high apparent Michaelis constant and, therefore, is not easily saturated with substrate. Thus, the enzyme activity increases linearly over a wide range of H₂O₂ concentrations, thereby maintaining a controlled intracellular H_2O_2 concentration. In mammalian systems, organs with high concentrations of CAT (i.e., liver and kidney) have low levels of endogenous H_2O_2 , and organs with low concentrations of CAT (i.e., lung and heart) have high endogenous levels of H_2O_2 . Further, if CAT activity is inhibited, H_2O_2 concentrations rise in the liver. As in the case of SOD, multiple CATs (isozymes; Figure 3) encoded by specific genes are found in plants, whereas animals exhibit one form of CAT (16,17). Both Cat and Sod genes respond differentially to various stresses known to generate ROS (Figure 4).

ROS, telomeres, and aging

Many hypotheses have been proposed to explain the root cause of aging. One broadbased hypothesis is that generalized homeostatic failure leads to age-related decline. Another is the "free-radical theory of aging" suggesting that endogenous ROS continually damage cellular macromolecules, including DNA (18). Incomplete repair of such damage would lead to its accumulation over time resulting in age-related deterioration.

The evidence implicating the generation of ROS as key factors in determining longevity is accumulating (19). Much of the earlier evidence was correlative. However, recent evidence has identified longevity-influencing genes responsive to ROS. It has further been shown that continual induction of DNA damage during normal aging results in genomic instability due to persistent DNA lesions, mutations, stalled repair, and transcription interference. The involvement of ROS in limiting lifespan has been suggested by analyses of transgenic Drosophila, which systematically overexpress both Sod and Cat genes. Some strains of these transgenic flies live up to 30% longer than their natural counterparts, whereas flies carrying only one of these constitutively expressed transgenes do not live longer, suggesting that ROS toxicity plays an important role in longevity (20,21).

Much attention has also recently been given to the role of telomere shortening as a causative factor in aging/senescence (22,23). Telomeres are stretches of repetitive DNA of tandem short sequence repeats (TTAGGG) that "cap" the ends of eukaryotic chromosomes to protect against degradation. Telomeres in most human cells shorten with each round of DNA replication, because they lack the enzyme telomerase. Telomerase is a specialized reverse transcriptase, which helps to replicate the telomere ends of chromosomes. However, telomerase is normally expressed only in germ-line cells and is derepressed in tumor cells in which telomeres are stabilized. It has recently been demonstrated that oxidative stress is the main reason for telomere shortening (24). It has been observed that H₂O₂ plus Cu(II) induced 8-oxo-7,8-dihydro-2'-deoxyguanosine formation in the telomere sequences more efficiently than in non-telomere sequences. Oxidative damage is not repaired as well in telomeric DNA as elsewhere in the chromosome. Oxidative stress accelerates telomere loss, while antioxidants decelerate it. Thus, oxidative stress is a critical modulator of telomere loss and telomere-driven replicative senescence is primarily a stress response. In a recent study, individuals under generalized stress were found to have shorter telomeres and less telomerase activity, and more oxidative stress (25). Given the available information, aging is likely to be a multifac-

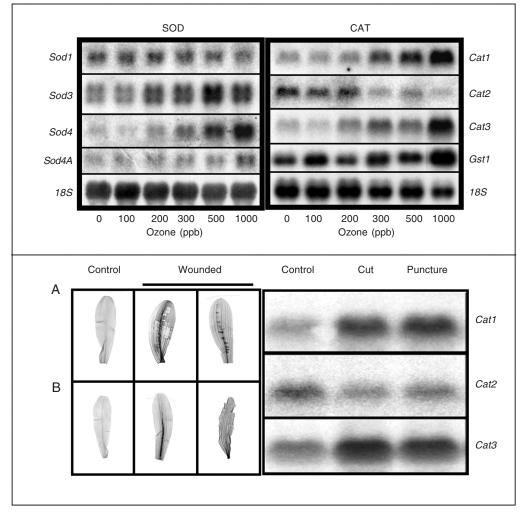


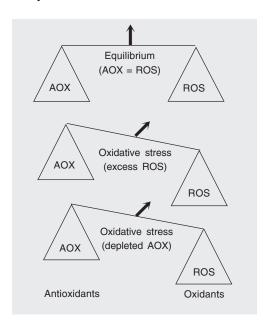
Figure 4. Response to acute ozone exposure of the Sod (topleft) and Cat (top-right) genes in maize leaves after 6 h of O₃ fumigation. As noted, some of the transcripts are upregulated (e.g., Sod3, Sod4, Cat1, Cat3), while others are downregulated (e.g., Sod1, Cat2). The 18S rRNA is a loading control. The production of H2O2 in maize leaves (bottom-left) in response to wounding. Seven-day-old plants were excised at the base of the stems and supplied with diaminobenzidine (DAB) for 6 h. The plants then were wounded and continuously supplied with DAB for 4 h. The production of H₂O₂ can be visualized by the deposition of the brown-red (dark areas) color products in the leaves. A, Control and wounded leaves (wounding was conducted near the main vein). B, Control and wounded leaves (wounding was conducted at the main vein and cut at the leaf edges). Cat transcript accumulation in response to wounding (bottom-right). RNA was isolated from similarly treated leaves, indicative that differential transcript accumulation is effected by induction of H_2O_2 .

torial process; whether ROS are peripheral targets that correlate with longevity or central regulators of human aging remains to be resolved.

Oxidative stress

ROS such as O₂•-, H₂O₂, and OH• are produced in all aerobic organisms and normally exist in the cell in balance with anti-

Figure 5. Oxidative stress results from imbalance between the levels of antioxidants (AOX) and reactive oxygen species (ROS). Cells are normally able to balance the production of oxidants and antioxidants to maintain redox equilibrium. Oxidative stress occurs when this equilibrium is upset by excess levels of ROS, or depletion of antioxidant defenses.



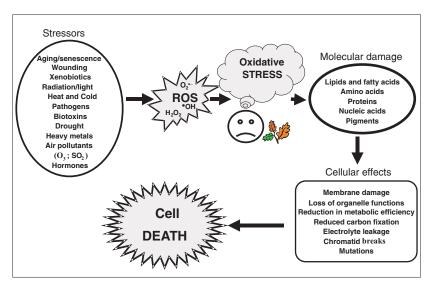


Figure 6. Scheme showing some of the initiators (stressors) of reactive oxygen species (ROS) and the biological consequences leading to a variety of physiological dysfunctions that can lead to cell death.

oxidant molecules. Oxidative stress occurs when this critical balance is disrupted due to depletion of antioxidants or excess accumulation of ROS, or both (Figure 5). That is, when antioxidants are depleted and/or if the formation of ROS increases beyond the ability of the defenses to cope, then oxidative stress and its detrimental consequences ensue. Such stress occurs when severely adverse environments or physiologic conditions overwhelm biological systems. One rapid and clear indicator of oxidative stress is the induction of antioxidant defenses and/ or increases in endogenous ROS levels. The formation of ROS can be accelerated as a consequence of various environmental stress conditions, including UV-radiation, high light intensities, exposure to herbicides, extreme temperatures, toxins such as cercosporin and aflatoxin, air pollutants, metals, wounding, and xenobiotics. Many inducers of oxidative stress are known carcinogens, mutagens, and toxins. ROS production and accumulation is a common denominator in many diseases and environmental insults and can lead to severe cellular damage leading to physiological dysfunction and cell death in virtually all aerobes (Figure 6).

When oxidative stress occurs, cells function to counteract the oxidant effects and to restore redox balance by resetting critical homeostatic parameters. Such cellular activity leads to activation or silencing of genes encoding defensive enzymes, transcription factors, and structural proteins (26,27).

ROS perform essential cellular functions

It has recently become apparent that ROS are not always harmful metabolic byproducts as generally believed (27). When tightly regulated, ROS perform critical functions in the cell. In fact, a significant body of evidence indicates that ROS, particularly $O_2^{\bullet-}$ and H_2O_2 , act as intracellular signaling molecules. In bacteria, the transcription factor

OxyR activates a number of genes inducible by H₂O₂, while the transcription factors SoxR/ SoxS mediate responses to $O_2^{\bullet-}(28,29)$. Yeast also has two distinct adaptive stress responses, one directed towards H₂O₂ and one towards $O_2^{\bullet-}$ (30-32). In higher eukaryotes, both animals and plants, oxidative stress responses are more complex and are modulated by several regulators (9). ROS-dependent redox cycling of cysteinyl thiols is critical for establishing protein-protein and protein-DNA interactions that determine many aspects of signal transduction pathways by regulating the activity of many transcription factors (26,32). For example, activation of nuclear factor κB (NF-κB) and activator protein-1 (AP-1), known to have critical roles in proliferation, differentiation and morphogenesis, can result from stimulation by diverse agents proceeding through a common pathway involving ROS generation. The pathway leading to H₂O₂ production and subsequent redox activation of NF-κB has been shown to involve the Rho family of small GTP-binding proteins (33).

Both intracellular and extracellular sources of ROS are capable of modulating gene expression. Low doses of H_2O_2 (<20 μ M) can elicit changes in phosphorylation of specific regulatory proteins including protein kinase B. Direct signaling action of

H₂O₂ in the differential regulation of antioxidant genes in plants likely occurs via protein-DNA interactions in the region of the antioxidant-responsive element (ARE; TGACTCA), NF-κB, and abscisic acid-responsive element 2 (ACGT) in the promoters of these genes. In addition to induction of defense gene expression, other roles of ROS in plants include direct killing of pathogens, involvement in cell wall structure, and promotion of programmed cell death (9). In yeast and animal cells, ROS have been shown to arrest cell division, and cell cycle progression is under negative ROS control (34). A clear and powerful example of how ROS are put to constructive uses was the observation that O₂ plays an important role against invading microbes, in effect serving as a broad-spectrum antibiotic (35). In response to invasion by pathogens, plants also mount a broad range of defense responses, including a rapid and transient production of large amounts of ROS ("oxidative burst") (36).

Thus, ROS play different, or even opposing roles, during different cellular processes (Figure 7). For example, under physiologic conditions, H_2O_2 may play an important role in signal transduction pathways and in activation of the transcription factor NF- κ B, while under pathologic (stress) conditions, H_2O_2 can lead to apoptosis or necrosis. Con-

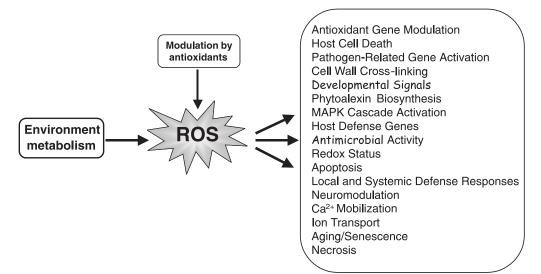


Figure 7. Scheme showing some of the useful pleiotropic roles of reactive oxygen species (ROS) known to occur in most higher organisms, indicative of the fact that ROS are not always harmful to cells. MAPK = mitogen-activated protein kinase.

sequently, the steady-state level of ROS within cells is critical and is determined by the interplay between ROS-generating mechanisms and the subtle modulating roles played by cellular antioxidants. In plants, H₂O₂ is generated under a diverse range of biotic and abiotic conditions, and its accumulation in specific tissues at specific developmental times, in the appropriate quantities, benefits plants and can mediate crosstolerance to other stresses (37). H₂O₂ affects gene expression and activates MAP kinases (MAPK), which in turn function as regulators of transcription (3).

ROS and gene expression

Numerous studies indicate that cells have the means to sense ROS and to induce specific responses, but the underlying mechanisms are not fully understood (9,26). The transcriptional network that responds to ROS in eukaryotes is currently being deciphered, whereas the prokaryotic system is better understood.

Nearly three decades ago, it was shown that the expression of ~30 proteins was induced by H₂O₂ in bacteria (38). Of these 30 proteins, 12 were maximally induced within 10 min and 18 between 10-30 min. The OxyR regulatory protein was subsequently shown to regulate expression of 9 of the 12 rapidly induced proteins. The tetrameric OxyR protein is a member of the LysR family of transcription activators and exists in two forms, reduced and oxidized; only the oxidized form is able to activate transcription. Further studies led to the identification of a number of OxyR-activated genes (39). Similarly, the SoxRS regulatory proteins were found to regulate expression of O2 •-responsive proteins in bacteria (40). Regulation of the SoxRS regulon occurs by conversion of SoxR to an active form that enhances soxS transcription. The enhanced levels of SoxS in turn activate expression of the regulon (40). In addition to SoxR and OxyR,

several other transcriptional regulators modulate the expression of antioxidant genes in bacteria, indicative of the complexity and connectivity of overlapping regulatory networks. No apparent homologs of OxyR, SoxR, or SoxS have been found in eukaryotes (28), but a number of other transcription factors have been found to play a role in regulating the expression of antioxidant genes in eukaryotes. In yeast, transcription regulators of antioxidant genes include ACE1, MAC1, YAP1, YAP2, HAP1, and HAP2/3/ 4 (30). In higher eukaryotes, oxidative stress responses are more complex and are modulated by several different regulators. In mammalian systems, NF-kB and AP-1 are involved in regulating the oxidative stress response. The ARE, present in the promoter region of mammalian glutathione S-transferase, metallothionein-I, and MnSod genes, causes induction of these genes in response to oxidants (41). NF-κB, AP-1, and ARE have also been found in promoters of antioxidant genes in higher plants (Figure 8) (9). The role of these factors is not unique to activation of antioxidant genes, as they are known, particularly NF-κB, to play central roles in regulating cellular responses to other stresses as well as regulating normal growth and metabolism.

There is substantial evidence that a variety of biotic and abiotic stresses induce ROS, which serve as a common factor in regulating various signaling pathways (Figure 9) (9,42). Similar stresses also activate MAPKs with kinetics that either precede or parallel H₂O₂ production, indicating that MAPKs may be one of several converging points in the defense-signaling network (43). In addition, exogenous application of several plant hormones and toxins has been shown to induce $O_2^{\bullet-}$ and H_2O_2 synthesis, leading to differential induction of some antioxidant genes and isozymes (Figure 10) (9,37,41,44). Thus, the identification of all genes and proteins regulated by H₂O₂ is an important step toward treatments that might confer

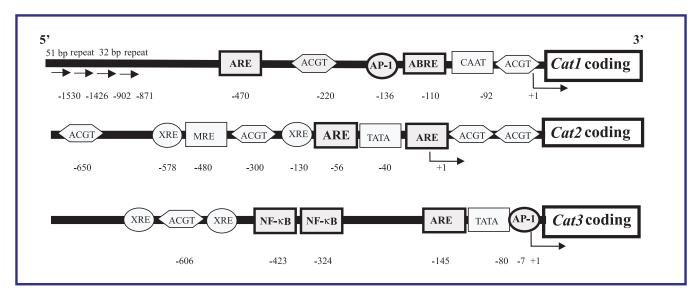


Figure 8. Schematic representation of the promoter region of each of the three catalase (Cat) genes of maize indicating the locations of the NF- κ B, ARE, AP-1, and ABRE motifs relative to the transcription start site (+1) of each gene. NF- κ B = nuclear factor κ B; ARE = antioxidant-responsive element: ABRE = abscisic acid-responsive element; AP-1 = activator protein-1.

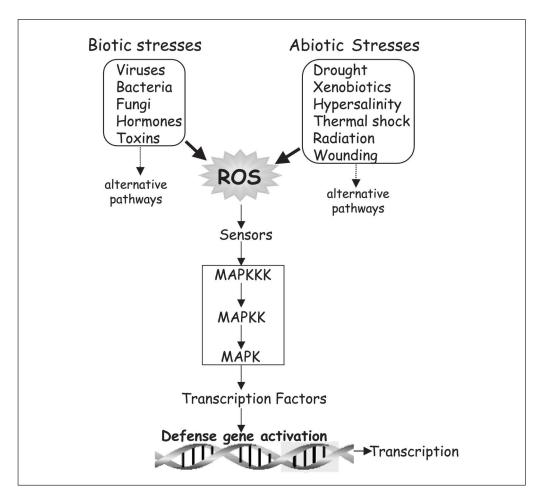


Figure 9. Convergence of various biotic and abiotic stress stimuli onto mitogen-activated protein kinase (MAPK) pathways via reactive oxygen species (ROS) as a common factor, leading to activation of antioxidant defense genes. Alternative pathways have also been implicated. MAPKKK = MAPK kinase kinases.

tolerance to multiple, but interrelated, stresses. In addition to induction/repression of antioxidant defense genes, ROS are known to similarly affect expression of a variety of other genes involved in different signaling pathways in microbes (28), yeast (45), plants (46), and animals (19).

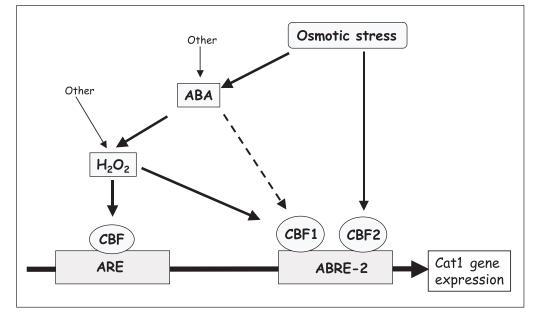
Genomic scale ROS-responsive gene expression

The advent of microarray expression analysis makes possible the assessment of gene expression on a genomic scale, rendering tens of thousands of genes assayable in a single experiment (47). Thus, identification of ROS-responsive genes on a global scale is now tenable. DNA microarrays can comprehensively examine gene expression networks during oxidative stress. There is now significant progress being made in surveying gene expression in response to H₂O₂ in *Escherichia coli* (48), yeast (4,45), animals (19), and higher plants (46).

A genome-wide transcription profile of $E.\ coli$ cells exposed to H_2O_2 was examined with a DNA microarray composed of 4169 $E.\ coli$ open reading frames (48). Gene ex-

pression was measured in isogenic wildtype and oxyR deletion mutants $(\Delta oxyR)$ to confirm that the H₂O₂-response regulator OxyR activates most of the H₂O₂-inducible genes. A very rapid and strong induction was observed of a set of OxyR-regulated genes in the wild type but not in the $\Delta oxyR$, providing internal validation of the experiment and confirmation for the induction of the oxidative stress genes identified earlier by other means (38). Several new H₂O₂inducible genes were also identified: some were members of the OxyR regulon and some induced by an OxyR-independent mechanism suggestive of other H₂O₂ sensors and regulators in E. coli (38). Several genes repressed by OxyR were highly expressed in the $\Delta oxyR$ mutant. Overall, the mRNA of 140 genes in the wild type, and 167 genes in the $\Delta oxyR$ were significantly induced after H₂O₂ treatment. It was also found that soxS was induced by H₂O₂, indicating an overlap with other regulatory pathways. Two genes, *Fpr* and *sodA*, known to be members of the SoxRS regulon were also highly induced by H_2O_2 in both wild type and $\Delta oxyR$. The microarray data also showed an overlap between oxidative stress, heat

Figure 10. Schematic representation of the possible signal transduction pathways of *Cat1* gene expression in response to the plant hormone abscisic acid (ABA) and osmotic stress in maize. ARE = antioxidant-responsive element; ABRE-2 = ABA responsive element 2; CBF1 = *Cat1* binding factor-1; CBF2 = *Cat1* binding factor-2.



shock and SOS responses (38). The results from the *E. coli* microarrays clearly indicate that the activities of transcription factors in addition to OxyR and SoxRS are likely modulated by oxidative stress.

Whole-genome expression patterns in Saccharomyces cerevisiae cells exposed to H₂O₂, in addition to other stresses, indicated that ~2/3 of the genome is involved in the response to environmental changes, and the global set of genes induced/repressed by each environmental signal were identified (4,45). The response to oxidative stress involves ~1/3 of the yeast genome and the maximal effects on gene expression occur slightly later relative to other stresses examined during similar time-courses, with most of the transcriptome returning to prestress levels within 2 h following exposure to H₂O₂ (4). Genes that are repressed for ~60 min after exposure to H₂O₂ are only transiently repressed in other stress time courses. Thus, genes encoding the translation apparatus and its regulators are remarkably coordinated for each environmental change, although the dynamics of each response are different. The expression programs following H₂O₂ or O₂•- treatment were essentially identical despite the fact that different ROS are involved. There was strong induction of genes known to be involved in detoxification of both H₂O₂ and O₂•-, such as CAT, SOD, and glutathione peroxidase, as well as genes involved in oxidative and reductive reactions (e.g., thioredoxin, glutathione reductase, glutaredoxin). The genes most strongly induced in response to H₂O₂ and O₂•- were dependent on the transcription factor Yap1p for their induction. Genes moderately induced by ROS and other signals are regulated by different transcription factors, depending on the conditions, and different upstream signaling pathways may govern their response. It has also been demonstrated that in Schizosaccharomyces pombe H₂O₂ activates the Sty1 (stress-activated MAPK) pathway in a dose-dependent manner via two

sensing mechanisms (49). At low H₂O₂ levels, a two-component signaling pathway, which feeds into either of the two (Wak1 or Win1) stress-activated MAPK kinase kinases, regulates Sty1. At high H₂O₂ levels, however, Sty1 activation is controlled mainly by an independent two-component mechanism, which requires the function of both Wak1 and Win1. In addition, the individual bZip transcription factors, Pap1 and Atf1, were found to function within a limited range of [H₂O₂]: Pap1 activates target genes at low [H₂O₂], whereas Atf1 controls transcriptional responses to high [H₂O₂], with some minor overlap. Some apparent cross talk among Sty1, Atf1, and Pap1 has been detected (50). Thus, S. pombe deploys a combination of stress-responsive regulatory proteins to gauge and trigger the appropriate transcriptional response to increasing H₂O₂ concentrations (49). This organism mounts two separate responses to oxidative stress: an adaptive response to low-level H₂O₂ exposure that protects it from subsequent exposures to higher [H₂O₂], and an acute response that allows the cell to survive a sudden, potentially lethal dose of H₂O₂.

Large-scale cDNA microarray analysis of the Arabidopsis transcriptome during oxidative stress identified 175 non-redundant expressed sequence tags from a sample of 11,000, which are regulated by H₂O₂. Of these, 62 are repressed and 113 are induced. In addition, RNA blots showed that some of the H₂O₂-regulated genes are also modulated by other signals known to involve oxidative stress (51). Furthermore, a substantial portion of these genes have predicted functions in defense response, cell rescue and signaling, and transcription, underscoring the pleiotropic effects of H₂O₂ in the response of plants to stress. Overall, the microarray used was estimated to represent only ~30% of the Arabidopsis genome, depending on redundancy, and 1 to 2% of the genes represented in the array are affected by H₂O₂-imposed oxidative stress, being

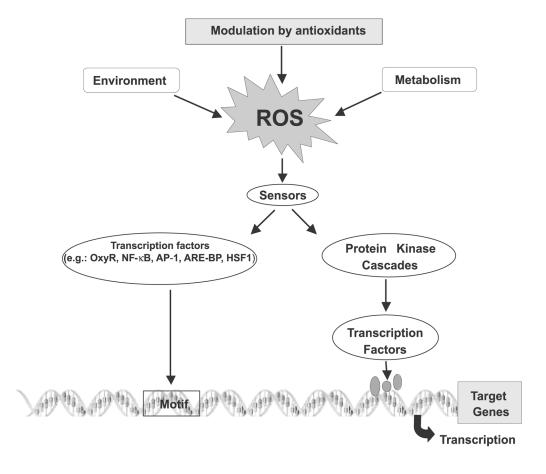
comparable to the situation in yeast (4). Of the 175 genes identified as H_2O_2 -responsive, most have no obvious direct role in oxidative stress but may be linked to oxidative stress indirectly, as a consequence of other biotic and abiotic stresses, explaining their sensitivity to H_2O_2 . Among the genes induced by H_2O_2 were genes encoding transcription factors, suggesting that they may mediate downstream H_2O_2 responses. As in other organisms, expression of the MAPKs in *Arabidopsis* is induced by oxidative stress, which in turn can mediate the induction of oxidative stress-responsive genes (52).

Toward an integrated view of oxy-stress responses

The traditional view of ROS as mere indiscriminate reactive byproducts of cellular metabolism has recently undergone a metamorphosis. This view came about by the discovery that ROS may act as signal-transducing molecules and that activation of intracellular transcription factors such as OxyR, SoxRS, NF-κB, and AP-1 occur via interaction with ROS, leading to gene transcription (Figure 11).

Genome sequencing and expression profiling using DNA or oligonucleotide microarrays, and related technologies, have been used effectively in the study of global gene expression patterns in response to different growth and environmental conditions to which organisms are exposed. Subsequent hierarchical clustering methods allow for the allocation of genes, coregulated temporally or in response to a given signal, into specific expression groups or regulons (48, 53). The numbers of genes that can be detected by these methods in response to any given environmental or developmental sig-

Figure 11. The major signaling pathways activated in response to oxidative stress. Reactive oxygen species (ROS) originating from environmental signals or from metabolic activity are modulated by antioxidants to nontoxic levels, at which point they serve as signaling molecules. ROS can activate gene transcription in two ways: a) via transcription factors, such as NF-κB. AP-1. and ARE-binding proteins (ARE-BP) that can interact directly with specific DNA motifs on promoters of target genes, or b) via activation of MAPK cascades, which in turn activate transcription factors that trigger target gene transcription. The degree to which a given pathway is activated depends on the nature and duration of the stress, as well as on cell type and developmental stage. NF- κB = nuclear factor κB ; AP-1 = activator protein-1; ARE-BP = antioxidant-responsive element binding proteins; HSF1 = heat shock transcription factor 1.



nal far exceed the limited number that could have been detected only a few years ago. With the current methods the expression of tens of thousands of genes can be detected in a single experiment in response to ROS, or to any given signal. Since transcription of genes into mRNA is governed by transcription factors which bind to cis-regulatory regions of the DNA in the vicinity of the target gene, the question arises as to whether large co-regulated groups of genes share cis-regulatory elements that bind to common transcription factors. The data available with respect to oxidative stress seem to suggest this. Cis-acting elements within the promoters of ROS-activated genes are being defined as well as their cognate trans-acting factors. A comparative analysis of promoter sequences of genes with similar expression profiles should provide a basis for unraveling common regulatory sequences and overlapping gene expression networks modulating ROSresponsive genes. The antioxidant enzymes CAT and SOD play key roles in modulating the levels of endogenous H_2O_2 and $O_2^{\bullet-}$, which in turn, at specific concentrations, act to modulate the expression of other ROSresponsive genes.

The use of microarrays and future derivations thereof, to examine global gene responses to ROS is assured. It has become clear that there are far more genes and gene clusters responding to ROS than previously thought, and that ROS likely play far more key roles in cellular activities than anticipated. As global gene responses to ROS are examined temporally and spatially along with different time-courses, and varied oxidant concentrations, even more complex regulons and regulatory cascades will emerge. It will also be instructive to examine gene expression profiles in specific mutants to help delineate the roles of specific regulators and of ROS. The identification of alternate pathways of ROS-dependent gene expression and characterization of the redox sensing mechanisms involved should point to new

insights and directions. Some genes have been identified whose transcription is responsive to a variety of stresses in addition to oxystress, while others appear to be responsive only to ROS or other specific signals. Although some regulatory systems have been implicated in modulating such responses, the complete network of regulators of ROS responses that activate such genes remains unclear.

Microarray results from different organisms clearly underscore the fluidity of genomes to reorganize and respond to changes in the cellular and extracellular environment. However, characterization of global gene expression programs at the transcript level is only the first step toward defining the role or function of each ROS-responsive gene. It is not unlikely that small changes in gene expression could lead to large alterations in protein levels. Consequently, proteomic analyses are essential to correlate mRNA changes with protein levels. There are in fact ample data demonstrating that ROS can alter the activity of cellular proteins.

A large and growing number of sequenced genomes and emerging technologies present us with enormous opportunities to advance biological science, and our knowledge of how genomes perceive signals to respond to variable environments. Oxidative stress and the responses to it seem to be central to many of the key biological questions. However, as the euphoria of new sequencing and technological advances begins to fade, we must recognize the magnitude of the problems that still need to be solved. Knowing the sequences of tens of thousands of ROSresponsive genes only reminds us that we still do not know the many proteins they encode, nor the biochemical or biological function of the great majority of such proteins. How such proteins interact with ROS to drive the various physiological processes in aerobic organisms remains a great puzzle. For the future, the fundamental challenge will be to integrate the information now

being obtained on gene expression patterns with structural, topological, and functional parameters and interactions of the various proteins encoded by ROS-responsive regulons, and to view the cell in which they function holistically.

The paradox

The oxygen paradox is indeed the paradox of evolution itself. Evolutionary pressures have made the best of a bad situation by generating mechanisms to curtail the undesirable toxic effects of ROS, which are an unavoidable consequence of the aerobic lifestyle, and to put them to constructive uses. Indeed evolution has co-opted ROS to serve necessary and useful purposes in the maintenance of cellular homeostasis and in the communication of cells with the external environment. The focus must now be placed on a more thorough understanding of how ROS-mediated signals are perceived, transduced, and interpreted by the cell's genetic machinery. Perhaps the most noteworthy observation to date concerning oxidative stress and the negative and positive roles of ROS is their universality among aerobic organisms and the similarities emerging in the regulatory mechanisms underlying their roles in all species.

The classical view of ROS as villains that indiscriminately destroy biomolecules has undergone a shift, in which positive biological roles are considered as well. It is now accepted that ROS, particularly H₂O₂ and O₂•-, are carefully regulated metabolites capable of signaling and communicating critical information to the cell's genetic machinery. Redox regulation of gene expression by oxidants and antioxidants is emerging as a vital mechanism in the health of all eukaryotes, including man.

It would indeed be interesting and challenging to identify all the changes in gene expression regulated by oxidative stress, and to determine their commonality among diverse species. Such a global analysis of the effects of H₂O₂ and O₂• on the transcriptome of eukaryotes has not yet been attained, but with the emergence of post-genomic technologies it will likely be not far off.

Conclusions

The survival of organisms on earth depends upon the interactions of their genomes with the environments in which they exist. In the course of evolution, organisms evolved a complex array of mechanisms for adapting to both minor and major fluctuations in the environment. The emergence of oxygenic photosynthesis presented early life forms with the greatest environmental challenge and an opportunity. The challenge was to develop antioxidant defenses in order to survive; the opportunity was to exploit the reactivity of oxygen for energy yielding and biosynthetic reactions. The opportunity led to the highly diversified life forms that evolved sufficient defenses and managed to exploit the aerobic lifestyle. Oxygen toxicity likely led to massive extinctions of those organisms unable to cope with it, unless they took refuge in isolated anaerobic niches. Thus, oxygen is a "double-edged sword" in that it makes life on earth possible, but in its reduced forms (ROS), it is highly toxic and lethal. Oxidative stress can arise from an imbalance between generation and elimination of ROS leading to excess ROS levels inflicting indiscriminate damage to virtually all biomolecules, leading in turn to various diseases and cell death. The notion that ROS are merely toxic byproducts of O₂ metabolism has recently been altered by experimental evidence indicating that ROS are carefully regulated metabolites capable of signaling and communicating information to the cell's genetic machinery. Redox regulation of gene expression by oxidants and antioxidants is emerging as a vital mechanism in the health and disease of all organisms. Irrespective of how or when ROS are generated, an increase in intracellular oxidants results in two critical effects: damage to various cell components and activation of specific signaling pathways, influencing various cellular processes leading to proper cell functions or to cell death. Genomic tools are accelerating the discovery of ROS-responsive genes on a global scale and are expanding our understanding of the oxidative stress response and the pleiotropic roles of ROS in signaling and gene expression.

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