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Evaluation on simulative transportation and storage quality of sweet cherry by different varieties based on principal component analysis

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Abstract

To determine the effect of transportation vibration on the nutritional quality and antioxidant activity of sweet cherry fruits in the process of cold chain transportation. Two kinds of sweet cherry "Summit" and "Nanyang" with different storage and transportation resistance were used as experimental materials. The results showed that the content of soluble solids, titratable acids and vitamin C in sweet cherry fruits were significantly reduced, while the content of superoxide anion radical, malondialdehyde and the antioxidant enzyme activities of CAT, APX, POD and PPO increased significantly, At the same time, the content of total phenols and flavonoids were increased under transport vibration stress, but the ability of DPPH and ABTS⁺ free radical scavenging of sweet cherry were reduced significantly. In particular, transportation vibration stress had a greater impact on the nutritional quality and antioxidant activity of "Nanyang", a sweet cherry with poor storage resistance. The results of this study indicated that transportation vibration has a great impact on the nutritional quality of sweet cherry, and that there is a strong correlation between antioxidant activity and nutritional quality. This study confirmed that transportation vibration during cold chain transportation can significantly reduce the nutritional quality and antioxidant activity of sweet cherry.

Keywords: sweet cherry; simulate transport vibration; nutritional quality; antioxidant activity; principal component analysis.

Practical Application: Though simulate transport and storage of sweet cherry, used principal component analysis to analyze the relationship between the nutritional quality and antioxidant activity. Provide a theoretical basis for the storage and preservation technologies, the cultivation of new varieties, and the exploration of softening mechanism of sweet cherries.

1 Introduction

Sweet cherry, which peel is thin, is rich in nutrition, and the flesh is soft and juicy. In recent years, sweet cherry has been welcomed by consumers because of its unique taste and strong antioxidant function (Gao et al., 2017). Sweet cherry is suitable for growing in tropical and subtropical areas with abundant rain, sunny and suitable temperature (Tokatlı & Demirdöven, 2020). In order to eliminate the regional limitation of sweet cherry and improve its commercial value, cold chain transportation and storage has become an inevitable development trend (Alique et al., 2005).

In the whole cold chain process, transportation and storage, interacting parts between the fruit and the box or fruits will be subjected to various pressures, such as squeezing, friction, collision and impact (Wei et al., 2019). When the relative strength of these forces that damage the fruit is relatively high, the fruit will be damaged by the shallow surface, and due to the repeated action of the external force, the connection force between the fruit cells and the strength between the cells will change, resulting in brittleness or plasticity of the fruit injury makes the fruit soft (Wei et al., 2019; Zhou & Wu, 2018). Directly damages the plasma membrane of the damaged area, softens it and reduces its oxidation resistance (Cliff & Toivonen, 2017).

At present, a lot of research has been carried out on post-harvest storage and preservation technologies of sweet cherry, such as modified atmosphere preservation method (Cozzolino et al., 2019), preservation of fungicides, chemical preservation method (Ni, 2018; Yang, 2009; Ma et al., 2019a), and physical preservation method (Michailidis et al., 2019; Tian et al., 2019). However, there are relatively few studies on the effect of different circulation and packaging conditions on fruit quality during transport.

The red cherry "Summit" has small epidermal cells, dense arrangement, thick surface wax and cell walls, and uniform internal cell size, so has good storage and transportation characteristics. The yellow cherry "Nanyang" has a loose epidermis and internal cell arrangement, and the surface layer is not waxy, which is extremely resistant to transportation. In this paper, "Summit" and "Nanyang" were used as experimental samples, studied the effect of transport vibration on the nutritional quality and antioxidant activity of sweet cherry fruit during cold chain transportation. The results will provide a reference for the development of post-harvest long-distance logistics transportation and storage technology development.

2 Materials and methods

2.1 Sample preparation and treatments

The test sweet cherries were picked from orchards in Dalian, Liaoning Province, China. Then they were quickly transported to the laboratory. Sweet cherries with a uniform size $(6 \pm 2 \text{ g})$

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and no obvious mechanical damage was selected as the test sample. Samples of each variety were randomly divided into two groups for processing. Each group was treated as follows: (1) Control group processing: 0 h treatments were used as the control group; (2) Vibration group processing: two layers of paper towels were glued on the bottom and around the incubator. Sweet cherries were put into the incubator gently, and the surface was covered with two layers of paper towels. The sealed insulation box was put into a constant temperature culture oscillator to simulate the mechanical vibration during road transportation. The conditions were set by the constant temperature culture oscillator: vibration frequency (120 r / min), temperature (4 °C), relative humidity (90-95%) and time (4 h). To simplify recording, the "Summit" control and vibration treatment were referred as S-CK and S-V; the "Nanyang" control and vibration treatment were referred as N-CK and N-V.

2.2 Determination of soluble solids content, titratable acid and vitamin C

The content of soluble solids (SSC) was determined according to the method described by NY/ T2637-2014 (China, 2014). Titratable acid (TA) and vitamin C (VC) were determined by Hernández-Muñoz et al. (2008).

2.3 Determination of total phenol content and flavonoids

The total phenol content (TPC) was determined by the Folinphenol colorimetric method (Rodov et al., 2010). Flavonoids content was determined by aluminum nitrate colorimetric method (Jiao et al., 2017).

2.4 Determination of the antioxidant enzyme activities

Determination of polyphenol oxidase

The activity of sweet cherry polyphenol oxidase (PPO) was extracted according to the method described by Cao et al. (2007) with slight modifications. Crude enzyme extract (100 μ L) was incubated with a buffered substrate (1.0 mL 50 mmol/ L catechol in 4 mL sodium phosphate buffer pH 7.8), and monitored by measuring the increase in absorbance at 398 nm.

Determination of peroxidase

The activity of cherry peroxidase (POD) was determined by Cao et al. (2007).

Determination of catalase

Catalase (CAT) was extracted according to the method described by Pasquariello et al. (2015) with slight modifications. The reaction solution contained 0.2 mL enzyme solution, 1.5 mL 50 mmol/L phosphate buffers (pH 7.8) and 0.3 mL 10mmol/L H_2O_2 the time was counted immediately and the absorbance was measured at 240 nm.

Determination of ascorbate peroxidase

The activity of ascorbate peroxidase (APX) was determined according to the previously published method (Cao et al., 2007).

2.5 Determination of superoxide anion

The content of superoxide anion radical (O_2^{-1}) was modified according to the previously published method (Zhao et al., 2008).

2.6 Determination of malondialdehyde content

Malondialdehyde content (MDA) was determined by Pasquariello et al. (2015).

2.7 Determination of antioxidant capacity

Determination of DPPH free radical scavenging rate

The DPPH free radical scavenging rate (DPPH) was determined according to the method of Chavan et al. (2013) with slight modifications. The reaction solution contained supernatant (0.2 mL) and DPPH solution (2.8 mL). Thoroughly mix, and leave the water bath at 37 °C for 30 minutes, and then determination of absorbance at 517 nm, which was referred to as A_i . At the same time, the absorbance A_0 of DPPH solution mixed with anhydrous ethanol (1:1) and the absorbance A_j of mixed supernatant solution and anhydrous ethanol (1:1) were determined (Equation 1).

DPPH free radical scavenging rate (%) = $\left[I - \left(A_i - A_j \right) / A_0 \right] \times 100$ (1)

Determination of agreement of basic telecommunications services

The Agreement of Basic Telecommunications Services (ABTS⁺) free radical scavenging rate was determined according to the method of Zhang et al. (2019).

2.8 Statistical analysis

Analyses of data were carried out by one-way ANOVA in SPSS 19.0 for Windows (SPSS Inc., Chicago, IL, USA). Statistic differences were considered to be significant at p < 0.05. All results were expressed as the mean \pm SE. Difference significance among treatments was analyzed by Duncan's multiple range test.

3 Results and discussion

3.1 Effect of the nutritional quality by simulated transport vibration in sweet cherry

SSC, TA and VC

SSC was an important nutritional indicator of fruits and had a strong relationship with the acceptability of consumers, reflecting the sugar, acid, vitamins, minerals and other nutrients in the fruits (Cliff & Toivonen, 2017). As shown in Table 1, SSC first increased slowly and then decreased during the storage and transportation period. The content of SSC in the vibration treatment group of "Summit" or "Nanyang" was lower than

	H				Vibration	n time (d)			
Index	lreatment	0	2	4	6	8	10	12	14
SSC (%)	S-CK	11.8 ± 0.22Aa	$12.1 \pm 0.60 \text{ABb}$	$12.7 \pm 0.41 \text{ABCc}$	$12.9 \pm 0.46 BCc$	13.0 ± 0.49 Cc	13.1 ± 0.15Cc	$13.1 \pm 0.27 Bc$	$12.6 \pm 0.23 \text{ABb}$
	S-V	11.7 ± 0.42Aa	$12.0 \pm 0.36 \mathrm{ABa}$	$12.2 \pm 0.27 \text{ABa}$	$12.7 \pm 0.12Bb$	$12.7 \pm 0.26Bb$	$12.8 \pm 0.33 Bb$	12.7 ± 0.47Cb	12.3 ± 0.42ABCa
	N-CK	11.9 ± 0.19Aa	$12.5 \pm 0.49 \text{ABb}$	$12.6 \pm 0.78 \text{ABb}$	$12.9 \pm 0.22 \text{ABc}$	$13.3 \pm 0.36 Bd$	$13.3 \pm 0.46 Bd$	$13.1 \pm 0.45 \mathrm{Ac}$	$12.8 \pm 0.41 \mathrm{Ac}$
	N-V	11.9 ± 0.12 Aa	$12.2 \pm 0.47 \mathrm{Ab}$	12.2 ± 0.46 Aa	12.4 ± 0.26 Aa	12.5 ± 0.13 Aa	12.5 ± 0.43Aa	$12.4 \pm 0.54 \text{ABa}$	12.2 ± 0.23ABa
TA (%)	S-CK	0.30 ± 0.01 Ca	$0.29 \pm 0.02 BCb$	$0.28 \pm 0.03 BCb$	$0.28 \pm 0.03 Bb$	$0.27 \pm 0.02 \text{ABb}$	$0.26 \pm 0.02 \text{Ab}$	$0.24 \pm 0.02 \text{Ab}$	$0.23 \pm 0.01 \text{Ab}$
	S-V	$0.28 \pm 0.02 Ba$	$0.26\pm0.03\mathrm{ABa}$	$0.25 \pm 0.03 \mathrm{ABa}$	$0.24 \pm 0.04 \text{ABa}$	$0.23 \pm 0.04 \mathrm{ABa}$	$0.21 \pm 0.03 \mathrm{ABa}$	$0.21 \pm 0.01 \text{ABa}$	$0.19 \pm 0.03 \text{Aa}$
	N-CK	$0.31 \pm 0.01 \text{Cb}$	0.32 ± 0.03 Cd	$0.31 \pm 0.03 BCd$	$0.30 \pm 0.04 BCc$	$0.29 \pm 0.07 BCc$	0.28 ± 0.02 BCd	$0.28 \pm 0.01 \text{ACd}$	$0.26 \pm 0.02 \text{Ad}$
	N-V	0.31 ± 0.03 Bb	$0.30 \pm 0.01 Bc$	$0.29 \pm 0.01 Bc$	$0.28 \pm 0.02 Bb$	$0.29 \pm 0.04 \mathrm{Ac}$	$0.27 \pm 0.02 \mathrm{Ac}$	$0.25 \pm 0.02 \mathrm{Ac}$	$0.24 \pm 0.03 \mathrm{Ac}$
VC (mg/kg)	S-CK	390.0 ± 34.11 Fa	351.1 ± 34.11 Fb	342.8 ± 39.77Ea	312.2 ± 49.12Dc	271.1 ± 44.87Cb	262.2 ± 46.15 Cc	$255.0 \pm 36.41 Bc$	$197.8 \pm 29.37 Ac$
	S-V	384.4 ± 43.30Ga	342.8 ± 45.14Ea	$322.2 \pm 51.87 \text{Eb}$	262.2 ± 47.51Da	232.2 ± 19.65Ca	228.9 ± 43.84BCa	197.2 ± 49.53Ba	136.7 ± 25.43Aa
	N-CK	387.2 ± 27.32Da	$381.1 \pm 16.54Eb$	$363.9 \pm 35.69 \text{Ec}$	$323.3 \pm 35.36 \text{Dd}$	284.4 ± 41.04Cc	$280.6\pm15.30BCd$	$272.8\pm35.98\mathrm{Bd}$	206.7 ± 32.60Ad
	N-V	381.7 ± 25.00Fa	$345.0 \pm 48.59 \text{Db}$	$329.4 \pm 37.09 \text{Db}$	292.8 ± 34.11Cb	$261.1 \pm 41.74Bb$	249.4 ±4 2.29Bb	$240.6 \pm 34.28Bb$	169.4 ± 35.19 Ab
All means in the sar	ne row followed	1 by different letters (A-D)	are significantly (p < 0.05)) different by Duncan's mul	tiple range test. All mean	s in the same column follow	ved by different letters (a-d	1) are significantly $(p < 0.0)$	05) different by Duncan's
multiple range test.	Values are mea	n ± SE of triplicate sample	s. Bars represent SE (p < 0).05). All samples were prej	vared in triplicates.				

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that in the control treatment groups. The vibration treatment group reached the maximum value on the 10th day, while the control treatment group on the 12th day. At the 14th day, SSC of the S-V and N-V were 2.44% and 4.92% lower than that of S-CK and N-CK respectively. SSC included soluble proteins and soluble sugars. During storage period, basic physiological activities were maintained and the substances such as soluble solids were accumulated. However, with the ripening and aging of sweet cherry, the fruit itself cannot make up for consumption, so the soluble solid content gradually decreased.

TA is also an important part of the quality of the fruits (Yang, 2009). The organic acid content has an important influence on the taste, flavor, sugar-acid ratio and pH of fruits (Fan et al., 2014). As shown in Table 1, TA content of sweet cherry in each treatment group showed a downward trend. Whether it was "Summit" or "Nanyang", the TA content of the vibration treatment group was always lower than that of the control treatment group. At the 14th day, the decline trend of the S-V was the most obvious, with a decline of 32.14%. From this, it could be seen that the simulated transport vibration treatment promoted the consumption of the TA content of the sweet cherry.

VC is an important nutrients component, but it is easily decomposed during storage. The change of VC content reflects the freshness of fruits, and it is also one of the indicators for judging the freshness preservation effect (Xu et al., 2017). As shown in Table 1, the VC content of all treatment groups gradually decreased. The simulated transport vibration treatment accelerated the decomposition of VC in sweet cherry, and the VC content of the S-V decreased more rapidly than the N-V. At the 14th day, the decline rates of S-CK, S-V, N-CK and N-V were 48.55%, 64.96%, 45.85% and 56.24%, respectively.

TPC and flavonoids content

After mechanical damage, fruits often produce a series of secondary metabolites, such as phenols, flavonoids, and alkaloids. These substances are mainly concentrated in the wound and its adjacent parts, and participate in the formation of callus and resist the invasion of insects or germ (Ma et al., 2018). As shown in Figure 1a, TPC of each treatment group increased first and then decreased, and the content of the vibration treatment group was significantly higher than that of the control treatment group. In the early storage period, the TPC of S-V was at a relatively high level, and in the later storage period, the total phenol content of N-V was significantly higher than that of other treatment groups. On the 14th day, the total phenol contents of S-CK, S-V, N-CK and N-V were 1.45, 1.75, 1.50 and 2.08, U, respectively. This indicated that simulated transport vibration accelerated the TPC of sweet cherries.

As shown in Figure 1b, the flavonoids content of sweet cherry in each treatment group first increased and then decreased, and its peak value appeared on the 8th day of storage. The peak values of S-CK, S-V, N-CK, and N-V were 30.08, 36.59, 39.32 and 42.38, mg/g, respectively. On the 14th day, the flavonoid content of S-V was 18.39% higher than that of S-CK, and the flavonoid content of N-V was 3.59% higher than that of N-CK. The results indicated that simulated transport vibration increased the secondary metabolites of sweet cherry. The increase of secondary metabolites is also a stress response of fruits to injury stress, and a self-protection effect of fruits against adversity.

3.2 Effect of the active oxygen metabolism by simulated transport vibration in sweet cherry

PPO and POD

PPO is a copper-based enzyme that exists in the free form of plant cytoplasm or bounds to mitochondria, chloroplasts and other subcellular organelles (Ma et al., 2019a). It catalyzes the reaction of various phenols with O_2 to form hydrazine, which is also a major part of terminal oxidase. PPO is a key enzyme that promotes enzymatic browning of fruits. As shown in Table 2, the PPO activity of sweet cherry showed an upward trend. On the 12th day, the PPO activity of each treatment group increased significantly. The activity of the S-V and N-V were higher than that of the corresponding control treatment group. Among them, the activity of the S-V was always at the highest level.



Figure 1. Effect of simulated transports vibration on TPC (a) and flavonoids (b) of sweet cherry. Bars represent SE (p < 0.05). All samples were prepared in triplicates.

T J					Vibratio	n time (d)			
Index	Ireaument	0	2	4	6	8	10	12	14
PPO	S-CK	$0.36 \pm 0.04 Ac$	$0.41 \pm 0.03 BCbc$	$0.48 \pm 0.08 \text{Cb}$	$0.57 \pm 0.04 Ac$	$0.64 \pm 0.01 \text{ABbc}$	$0.72 \pm 0.05 \text{Db}$	0.82 ± 0.07Dab	$1.05 \pm 0.01 \text{Eb}$
(U/g·min FW)	S-V	0.57 ± 0.02 Aa	$0.64 \pm 0.01 \text{ABa}$	$0.69 \pm 0.05 Ba$	0.76 ± 0.02 Ca	$0.79 \pm 0.05 \text{CDa}$	$0.83 \pm 0.05 \text{DEa}$	$0.89 \pm 0.01 \text{Ea}$	1.23 ± 0.02 Fa
	N-CK	$0.30 \pm 0.05 \mathrm{Ac}$	$0.37 \pm 0.03 \mathrm{ABc}$	$0.46 \pm 0.03 BCb$	0.54 ± 0.06 Cc	$0.59 \pm 0.04 \text{Dc}$	$0.63 \pm 0.03 \mathrm{Ec}$	0.74 ± 0.04 Fc	$0.91 \pm 0.02 \text{Gc}$
	N-V	$0.43 \pm 0.04 \text{Ab}$	$0.46 \pm 0.03 Bb$	$0.52 \pm 0.05 \text{Cb}$	$0.68 \pm 0.03 \text{Eb}$	$0.71 \pm 0.05 \text{Eb}$	$0.79 \pm 0.03 \text{Dab}$	$0.82 \pm 0.01 \text{Ebc}$	0.98 ± 0.02 Fb
	S-CK	0.33 ± 0.02Aa	$0.49 \pm 0.03 Bb$	$0.65 \pm 0.03 Ca$	0.69 ± 0.07Cab	0.71 ± 0.04 Ca	1.07 ± 0.01Da	$1.13 \pm 0.03 \text{DEa}$	$1.20 \pm 0.02 \text{Ea}$
POD	S-V	0.36 ± 0.09 Aa	$0.55 \pm 0.01 Ba$	$0.68 \pm 0.02 Bb$	0.73 ± 0.03 Ca	0.76 ± 0.09 Ca	$1.11 \pm 0.06 \text{Da}$	$1.15 \pm 0.01 \text{ABa}$	1.26 ± 0.11 Ea
(U/g·min FW)	N-CK	0.27 ± 0.08 Aa	$0.32 \pm 0.03 \mathrm{Ac}$	$0.55 \pm 0.07 Bab$	$0.58 \pm 0.06Bb$	$0.59 \pm 0.03 Bb$	$0.83 \pm 0.07 \text{Cb}$	$0.93 \pm 0.09 \text{Db}$	$0.99 \pm 0.04 \text{Db}$
	N-V	0.31 ± 0.05 Aa	$0.45 \pm 0.03 Bb$	$0.64 \pm 0.07 \text{Cb}$	0.64 ± 0.04Cab	$0.64 \pm 0.04 \text{Cb}$	$1.09 \pm 0.06 \text{Da}$	$1.13 \pm 0.07 \text{DEa}$	$1.20 \pm 0.02 \text{Ea}$
	S-CK	$95.8 \pm 7.45 \text{Db}$	$96.6 \pm 5.48 \text{Dc}$	$96.0 \pm 8.26 \text{DEb}$	$92.0 \pm 7.34 \text{Eb}$	$85.8 \pm 9.37 \text{Cb}$	$74.0 \pm 43.84 Bb$	$74.0 \pm 49.53 Ac$	$70.2 \pm 25.43 \mathrm{Ab}$
CAT	S-V	110.4 ± 7.03 Aa	112.7 ± 4.16Aa	113.7 ± 9.97Ba	$120.1 \pm 9.65 Ba$	93.1 ± 4.85Ba	82.5 ± 6.51 Ca	82.5 ± 36.41Da	79.3 ± 29.37Da
(U/g·min FW)	N-CK	$85.4 \pm 3.32 Dc$	$91.8 \pm 8.96 \text{Dd}$	88.5 ± 7.43Dc	83.4 ± 4.18Dc	75.5 ± 4.56Cc	$64.3 \pm 42.29 Bc$	64.3 ± 34.28Ad	$63.6 \pm 35.19 \mathrm{Ac}$
	N-V	$98.6 \pm 3.23 \text{Cb}$	101.0 ± 6.44 CDb	$99.6 \pm 5.74 \text{CDb}$	92.6 ± 5.76Db	$87.0 \pm 4.67 BCb$	$78.2 \pm 15.30Bb$	$78.2 \pm 35.98 \text{Ab}$	$74.6 \pm 32.60 \mathrm{Ab}$
	S-CK	$7.78 \pm 0.24 \text{Eb}$	$8.56\pm0.10\mathrm{Fc}$	9.78 ± 0.25 Fc	$7.92 \pm 0.22 Ec$	$6.31 \pm 0.32 \text{Dc}$	5.87 ± 0.26 Cc	$5.11 \pm 0.26Bb$	$4.06 \pm 0.36 \mathrm{Ac}$
APX	S-V	8.32 ± 0.26Da	12.30 ± 0.12 Fa	$12.32\pm0.23\mathrm{ABa}$	9.45 ± 0.26Ea	8.50 ± 0.27Da	7.39 ± 0.16Ca	$5.96 \pm 0.15 Ba$	4.70 ± 0.28 Aa
(U/g·min FW)	N-CK	5.80 ± 0.18 Cc	8.19 ± 0.28 Fd	9.37 ± 0.32 Gd	$7.26 \pm 0.24 Ed$	$6.05 \pm 0.25 \text{Dd}$	5.59 ± 0.25 Cd	$4.40 \pm 0.31 Bc$	$3.69 \pm 0.26 \text{Ad}$
	N-V	$6.10 \pm 0.23 Cc$	10.32 ± 0.26 Fb	$10.65 \pm 0.37 \text{Gb}$	$8.51 \pm 0.29 \text{Eb}$	$7.36 \pm 0.35 \text{Db}$	$6.31 \pm 0.30 \text{Cb}$	$5.20 \pm 0.16Bb$	$4.43 \pm 0.45 \text{Ab}$

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This indicates that simulated transport vibration significantly increased the PPO activity of sweet cherry. Moreover, compared with "Nanyang", the simulate transport vibration treatment made the PPO activity of the "Summit" easier to rise. Studies showed that PPO activity could increase significantly when plant tissue was susceptible or causes damage under other adverse conditions, and it played a protective role.

POD is one of the important enzymes in the active oxygen free radical scavenging system (Ma et al., 2019a). It plays a role in the decomposition of H₂O₂ in the last step of lignin biosynthesis and has a certain effect on the color change of fruits. The increase of POD activity is an important sign of the ripening and aging of fruit tissues (Chen et al., 2019). POD can be activated by H₂O₂, which triggers a series of toxic effects on tissues. The mechanism is that POD degrades the indoleacetic acid in fruit tissues by catalytic oxidation (Zhong & Zhu, 2004). As shown in Table 2, the POD activity showed an increasing trend. At 8 days after storage, the POD activity of each treatment group changed slowly, and the activity of POD increased rapidly on the 10th day. And during the whole storage period, the POD activity of S-V and N-V were always higher than the corresponding control treatment group. It indicated that simulated transport vibration promotes an increase in POD activity of sweet cherry.

CAT and APX

CAT is a protective enzyme in the ripening and aging process of fruits, and one of the key enzymes in the biological defense system. It can decompose H₂O₂ into oxygen and water, remove H₂O₂ and prevent cells from being poisoned by H_2O_2 , so as to reduce damage to the membrane and achieve the purpose of delaying cell senescence (Wang et al., 2018). As shown in Table 2, the CAT activity of each treatment group showed a slow increase first, and reached the highest on the 6th day, then decreased rapidly. This may be due to the accumulation of H₂O₂ during the early storage period of sweet cherries, which induced the increase of CAT enzyme activity. When the accumulated amount is not cleaned in time, it is possible to inhibit the CAT enzyme activity. Meanwhile, The CAT activity of S-V and N-V were always higher than S-CK and N-CK, respectively. It indicated that simulated transportation vibration promoted the expression of CAT activity of sweet cherry.

APX is a highly specific peroxidase that exists in the fruit. It can catalyze the oxidation-reduction reaction of ascorbic acid with H_2O_2 to achieve the purpose of removing H_2O_2 and thus limit the reaction between O_2^- and H_2O_2 to produce peroxides such as \cdot OH (Liu et al., 2017). As shown in Table 2, the APX activity increased first and then decreased. The content of each treatment group reached the maximum value on the 4th day of storage. The APX activity of the S-CK, S-V, N-CK and N-V were 12.80, 9.78, 10.60 and 9.40, U/g·min FW. The activity levels of S-V and the N-V were lower than the corresponding control treatment group, and the activity of S-CK was always at the highest level, while the activity of N-V was always at the lowest level.

3.3 O_2^{-} production rate

During the storage and transportation of fruits, the intracellular free radical metabolism balance is often disrupted, accumulating a large amount of free radicals and triggering or aggravating lipid peroxidation, resulting in the loss of the cell plasma membrane system. Free radicals are atoms or groups of atoms with unpaired valence electrons. Oxygen molecules that enter the body through respiration receive only one electron and are converted into O_2^- . O_2^- acts with other oxygen free radicals on biofilms, attacks membrane proteins, and causes the unsaturated fatty acids on the membrane undergo peroxidation, which destroys the membrane system (Nan et al., 2017).

As shown in Table 3, the O_2^- tended to rise as a whole. At the beginning of storage, the content of O_2^- in each treatment group increased slowly. From the 6th day, the content of O_2^- increased rapidly. The content of O_2^- in vibration treatment groups were always higher than that in control treatment groups, and the increase of N-V was the most obvious. This indicates that simulated transport vibration increased the rise of O_2^- of sweet cherry.

3.4 MDA content

Lipids exist on the cell membrane, which are related to the stress resistance of the fruit during storage. During the storage process, with the content of fatty acids decreases, the peroxidation of internal membrane lipids and the increase of active oxygen and free radicals will destroy the structure and function of the membrane, thereby destroying the structure of the cell membrane, enhancing permeability and graduality oxidation and aging of fruits (Nan et al., 2017). MDA is an important product of membrane peroxidation, which can directly produce toxic effects on cells, caused cross-linking and inactivation of enzyme proteins in biofilms, lead to membrane quality index of peroxidation.

With the extension of storage time, MDA content of sweet cherry increased gradually. At the same time, MDA contents of simulated transportation vibration treatments S-V and N-V were higher than their control treatments, and MDA content of N-V was always at the highest level during the later stage of storage (Table 3). It showed that simulated transportation vibration increased the degree of peroxidation of sweet cherry membrane lipids.

3.5 Effect of the antioxidant capacity by simulated transport vibration in sweet cherry.

According to the DPPH and ABTS⁺ methods which determine the antioxidant capacity, the higher clearance value indicated higher antioxidant capacity (Ma et al., 2019b; Pasquariello et al., 2015). As shown in Table 3, their antioxidant capacity decreased generally, and the antioxidant capacities of S-V and N-V treatment were lower than that of its control treatment, respectively. In the early storage period, the antioxidant capacity of "Summit" was always higher than that of "Nanyang". In the later stage, the result is just the opposite.

-	E				Vibratio	n time (d)			
Index	Ireatment	0	2	4	6	8	10	12	14
O_2^{-1} (mol/g min)	S-CK	0.6 ± 0.04 Aa	$1.9 \pm 0.03 Bb$	2.2 ± 0.08Bb	$6.0 \pm 0.04 Cc$	6.3 ± 0.01CDd	7.3 ± 0.05Dd	$10.9 \pm 0.07 \text{Ec}$	12.0 ± 0.01 Fd
	S-V	$0.7 \pm 0.02 \text{Aa}$	$2.5 \pm 0.01 Ba$	3.5 ± 0.05Ca	$7.4 \pm 0.02 \text{Db}$	$7.7 \pm 0.05 \text{DEc}$	$8.2 \pm 0.05 Ec$	13.1 ± 0.01 Fb	14.5 ± 0.02 Gb
	N-CK	$0.6 \pm 0.05 \text{Aa}$	$2.7 \pm 0.03 Ba$	$3.8 \pm 0.03 Ba$	$7.3 \pm 0.06 \text{Cb}$	$8.5 \pm 0.04 \text{Cb}$	$11.5 \pm 0.03 \text{Db}$	$13.2 \pm 0.04 \text{Eb}$	15.6 ± 0.02 Fc
	N-N	0.7 ± 0.04 Aa	$3.0 \pm 0.03 Ba$	$4.1 \pm 0.05 Ba$	8.0 ± 0.03 Ca	10.0 ± 0.05 Ca	$14.1\pm0.03\mathrm{Da}$	15.7 ± 0.01DEa	17.1 ± 0.02Ea
MDA (µnnol/g Fw)	S-CK	$0.42 \pm 0.02 \text{Ab}$	$0.42 \pm 0.05 Bb$	$0.48 \pm 0.01 Bc$	$0.55 \pm 0.03 Cc$	$0.72 \pm 0.05 \text{Dc}$	$0.89\pm0.04\mathrm{Eb}$	0.97 ± 0.04 Fd	$1.05 \pm 0.02 \text{Gc}$
	S-V	0.51 ± 0.01 Aa	0.55 ± 0.01 Aa	$0.57 \pm 0.03 Ba$	0.71 ± 0.02 Ca	$0.80 \pm 0.03 \text{Db}$	$0.98 \pm 0.05 \text{Eb}$	1.08 ± 0.03 Fc	$1.14 \pm 0.05 \text{Gb}$
	N-CK	$0.41 \pm 0.02 \text{Ab}$	$0.43 \pm 0.01 Bb$	$0.52 \pm 0.02 \text{Cb}$	$0.54 \pm 0.02 Dv$	$0.78 \pm 0.02Eb$	$0.92 \pm 0.03 Fc$	$1.11 \pm 0.04 Gb$	$1.15 \pm 0.06 \text{Hb}$
	N-V	$0.50 \pm 0.03 \mathrm{Aa}$	$0.52 \pm 0.03 \mathrm{Aa}$	$0.55 \pm 0.03 Ba$	$0.62 \pm 0.03 Bb$	0.95 ± 0.02 Ca	$1.04 \pm 0.02 \text{Da}$	$1.22 \pm 0.05 Ea$	$1.29 \pm 0.04 \text{Ea}$
DPPH (%)	S-CK	64.1 ± 7.45Da	57.0 ± 5.48 Ca	56.3 ± 9.97Ba	55.7 ± 9.65Ba	$53.0 \pm 4.85 Ba$	$52.3 \pm 6.51 Ba$	49.0 ± 36.41Aa	48.4 ± 29.37Aa
	S-V	$62.9 \pm 7.03 \text{Da}$	56.4 ± 4.16Ca	52.5 ± 8.26Cc	$52.5 \pm 7.34 \text{Cb}$	$51.0 \pm 9.37Bb$	$50.9\pm43.84\mathrm{Bb}$	46.3 ± 49.53Ac	$46.0\pm25.43\mathrm{Ab}$
	N-CK	$57.9 \pm 3.23 \text{Eb}$	57.0 ± 5.48Ca	$54.1 \pm 5.74 \text{DEb}$	$52.9 \pm 5.76 \text{Db}$	52.2 ± 4.67Cab	$51.5\pm15.30BCb$	$49.5 \pm 35.98 \text{ABa}$	47.7 ± 32.60Aa
	N-V	$54.9 \pm 3.32 \text{Ec}$	56.2 ± 6.44Ea	$53.0 \pm 7.43 \text{Dc}$	$51.9 \pm 4.18 \text{CDb}$	50.0 ± 4.56CDb	$49.0 \pm 42.29 BCb$	$47.6 \pm 34.28 \text{ABb}$	$46.6 \pm 35.19 \text{Ab}$
$ABTS^{+}$ (%)	S-CK	99.8 ± 0.39 Ea	98.9 ± 1.21Ea	98.9 ± 1.00Ea	98.7 ± 1.42Ea	$90.8 \pm 1.59 \text{Da}$	88.1 ± 0.90Ca	$72.4 \pm 0.47 Bc$	$62.7 \pm 0.97 \text{Ac}$
	S-V	98.9 ± 0.09 Fa	$93.4 \pm 0.25 \text{Eb}$	$93.0 \pm 0.09 \text{Eb}$	$91.9 \pm 0.06 \text{Eb}$	$89.2 \pm 0.26 \text{Db}$	79.4 ± 1.40 Cb	$68.4 \pm 3.17Bb$	56.6 ± 1.36 Ad
	N-CK	98.5 ± 0.61 Ea	92.7 ± 3.19Db	$90.9 \pm 2.26 \text{Db}$	85.5 ± 1.68 Cc	$83.6 \pm 3.81 \text{Cc}$	$82.1 \pm 5.33 BCc$	77.6 ± 1.37Ba	72.3 ± 3.53Aa
	N-V	$97.8 \pm 0.12 \text{Da}$	92.9 ± 3.79Db	85.1 ± 4.02 Cc	$81.9 \pm 3.69 BCd$	$79.0 \pm 3.20 \text{Bd}$	$76.6 \pm 1.82 Bd$	$71.3 \pm 1.91 \text{Ac}$	$69.5 \pm 3.06 \text{Ab}$
All means in the same row foll multiple range test. Values are	owed by different l mean ± SE of tripli	etters (A-E) are significa. icate samples. Bars repre.	ntly ($p < 0.05$) different b sent SE ($p < 0.05$). All sa	y Duncan's multiple rang mples were prepared in 1	ge test. All means in the s triplicates.	ame column followed by	y different letters (a-d) a	re significantly (p < 0.05)	different by Duncan's

3.6 Correlation ability

In order to determine the effects of antioxidant components and simulated transportation vibration on the antioxidant capacity of sweet cherry, the correlation of between antioxidant indexes and antioxidant capacity were analyzed (see Figure 2). DPPH free radical scavenging ability was positively related to TA, flavonoids, PPO, POD, CAT and APX, but negatively correlated with O_2^- , MDA and TPC. ABTS⁺ free radical scavenging ability was positively correlated with TA, VC, flavonoids and O_2^- and MDA.

The correlation results of 13 indicators showed that TA, VC, TPC, flavonoids, PPO, POD, CAT, APX, O_2^{-} and MDA of sweet cherry were important antioxidants. They can directly affect the antioxidant capacity of sweet cherry. The antioxidant capacity

of vibration groups was become weaker, which may be due to their lower retention of antioxidant components.

3.7 Principal component analysis

The principal component analysis (PCA) showed that the cumulative variance contribution rate of the first three principal components reached 88.97% (see Table 4), which indicated that these three principal components could accurately and objectively reflect the similarity between samples, so a total of three principal components were extracted (Nascimento et al. 2020; Fernandes et al. 2019).

The first main component included TPC, PPO, POD, CAT, APX, and O_2^- and MDA. Among them, PPO, POD, CAT, and



Figure 2. Correlation Analysis of the measured parameters of sweet cherry.

Table 4. Explain the total variance of principal components

		Initial eigenvalue		1	Rotational square sum loa	ding
Ingredient	Total	Variance	Cumulative variance	Total	Variance	Cumulative variance
	Iotai	contribution rate%	contribution %	Totai	contribution rate%	contribution%
1	6.559	43.729	43.729	6.434	42.896	42.896
2	5.365	35.766	79.496	5.254	35.027	77.923
3	1.421	9.474	88.970	1.657	11.047	88.970
4	.596	3.973	92.943			
5	.379	2.527	95.470			
6	.232	1.544	97.014			
7	.206	1.373	98.387			
8	.154	1.026	99.413			
9	.055	.366	99.779			
10	.022	.147	99.926			
11	.011	.074	100.000			
12	.000	.000	100.000			
13	.000	.000	100.000			

APX were positively related to the first main component, while TPC, O_2^{-} and MDA were negatively correlated with the first principal component. The second main component including TA, V_C , flavonoids, and ABTS⁺ free radical scavenging rate were all positively correlated with the second main component. The third main component including SSC and DPPH free radical scavenging rates were positively correlated with the third main component (see Table 5 and Table 6)

Table 5. Component Matrix of principal component analysis.

In gradiant -		Component	
Ingredient	1	2	3
SSC	013	.061	.940
TA	.220	.963	011
VC	246	.853	.035
TPC	940	104	107
Flavonoids	164	.910	.321
PPO	.878	.235	.209
POD	.959	007	161
CAT	.883	.024	.134
APX	.951	.140	.125
O2-	864	.286	.082
MDA	-9.37	.284	017
DPPH	.521	.266	.625
ABTS+	189	.933	.118

Table 6. Component load matrix of principal component analysis

Tu and i and		composition	
Ingredient	1	2	3
SSC	013	.061	.940
TA	.220	.963	011
VC	246	.853	.035
TPC	940	104	107
Flavonoids	164	.910	.321
PPO	.878	.235	.209
POD	.959	007	161
CAT	.883	.024	.134
APX	.951	.140	.125
O2-	864	.286	.082
MDA	-9.37	.284	017
DPPH	.521	.266	.625
ABTS+	189	.933	.118

3.8 PCA analysis of shelf quality of sweet cherries with different treatments

According to the results of PCA, the contribution rate of the first, second and third principal component was 42.896%, 35.027% and 11.047%, respectively (Table 4), and the cumulative contribution rate was 88.970%. Three principal components were used to evaluate the different treatments of sweet cherry, and the comprehensive score (F) was calculated (Equation 2).

$$F = (42.896 \% * F_1 + 35.027 \% * F_2 + 11.047 \% * F_3) / 88.970\%$$
(2)

where: F_1 , F_2 , F_3 and F were the score of the first, second, third principal component and a comprehensive score. The higher the F score, the better the quality (Ma et al., 2018).

As shown in Table 7, the F of each group showed a downward trend with the prolongation of storage time. After 14 days storage, the comprehensive score relationship of each group was: S-CK> N-CK> S-V> N-V. This showed that the simulated transport vibration accelerated the quality degradation of sweet cherry during simulated transport storage. Meanwhile, transportation vibration stress had a greater impact on the nutritional quality and antioxidant activity of "Nanyang", a sweet cherry with poor storage resistance.

4 Conclusions

Transport vibration often leads to the damage of fruit surface tissue and physiological abnormalities inside the fruit, thus accelerating the softening and aging of fruit. Among them, apparent damage of the fruit which is visible to the naked eye is easy to find and deal with in time. The damage of internal tissue caused by transportation vibration is most easily overlooked and the damage of fruit is more serious. The results of this study indicated that transportation vibration has a great impact on the nutritional quality of sweet cherry, and that there is a strong correlation between antioxidant activity and nutritional quality. In order to maintain the original quality of the sweet cherry fruit, the logistics time should be shorten, and it is necessary that vibration reduction packaging measures should be taken to reduce the damage of vibration to the fruit.

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 Table 7. Comprehensive quality change of sweet cherry in simulated transportation.

Treatment				Storage	time(d)			
freatment	0	2	4	6	8	10	12	14
S-CK	1.18 ± 0.08Ha	1.03 ± 0.09Ga	0.98 ± 0.05 Fa	0.85 ± 0.09 Ea	$0.67 \pm 0.07 \text{Da}$	0.23 ± 0.05 Ca	-0.01 ± 0.09Ba	$-0.15\pm0.08\mathrm{Aa}$
S-V	$1.09\pm0.09\mathrm{Ha}$	$0.95\pm0.08Gb$	$0.88\pm0.08Fc$	$0.62 \pm 0.05 \text{Ec}$	$0.43\pm0.09 Dc$	$0.15\pm0.08Cb$	$-0.12\pm0.06\mathrm{Bc}$	$-0.24\pm0.09\mathrm{Ac}$
N-CK	$1.12 \pm 0.03 \text{Hb}$	1.01 ± 0.02 Ga	$0.92\pm0.06Fb$	$0.81\pm0.06\mathrm{Eb}$	$0.59\pm0.06\text{Db}$	$0.21\pm0.07Ca$	$-0.06\pm0.07Bb$	$-0.18\pm0.05 Ab$
N-V	$1.01\pm0.08 Hb$	$0.89 \pm 0.06 Gc$	$0.76\pm0.04 \mathrm{Fd}$	$0.59\pm0.09\text{Ed}$	$0.38\pm0.07 \text{Dd}$	$0.11\pm0.06Cc$	$-0.15\pm0.08Bd$	$-0.31\pm0.08\mathrm{Ad}$
S-V N-CK N-V	1.09 ± 0.09Ha 1.12 ± 0.03Hb 1.01 ± 0.08Hb	0.95 ± 0.08Gb 1.01 ± 0.02Ga 0.89 ± 0.06Gc	0.88 ± 0.08Fc 0.92 ± 0.06Fb 0.76 ± 0.04Fd	0.62 ± 0.05Ec 0.81 ± 0.06Eb 0.59 ± 0.09Ed	0.43 ± 0.09Dc 0.59 ± 0.06Db 0.38 ± 0.07Dd	0.15 ± 0.08Cb 0.21 ± 0.07Ca 0.11 ± 0.06Cc	$-0.12 \pm 0.06Bc$ $-0.06 \pm 0.07Bb$ $-0.15 \pm 0.08Bd$	-0.24 ± 0.09 -0.18 ± 0.05 -0.31 ± 0.08

All means in the same row followed by different letters (A-H) are significantly (p < 0.05) different by Duncan's multiple range test. All means in the same column followed by different letters (a-d) are significantly (p < 0.05) different by Duncan's multiple range test. Values are mean \pm SE of triplicate samples. Bars represent SE (p < 0.05). All samples were prepared in triplicates.

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