

A Convenient Procedure for the Synthesis of 3 β -Hydroxy-6-oxo-5 α -steroids. Application to the Synthesis of Laxogenin

Martín A. Iglesias-Arteaga^{*,a}, Eva M. Símuta-Lopez^b, Sergio Xochihua-Moreno^b, Omar Viñas-Bravo^b, Sara Montiel Smith^b, Socorro Meza Reyes^b and Jesús Sandoval-Ramírez^{*,b}

^a Departamento de Química Orgánica, Facultad de Química, Universidad Nacional Autónoma de México, Ciudad Universitaria, 04510 México - D. F., México

^b Facultad de Ciencias Químicas, Benemérita Universidad Autónoma de Puebla, Ciudad Universitaria, 72570 Puebla - Pue., México

Uma rota conveniente para a síntese de 3 β -hidroxi-6-oxo-5 α -esteroides a partir de Δ^5 -esteroides é descrita tendo sido aplicada para a síntese da laxogenina, substância que apresenta atividade como hormônio de crescimento vegetal. O método é uma alternativa para instalar este grupo funcional importante encontrado em esteróides naturais. O processo descrito utiliza reagentes baratos e pode ser executado em quatro etapas. As etapas de oxidação e de tratamento ácido não afetam a cadeia lateral espirocetálica presente na diosgenina (**16**).

A convenient pathway to obtain 3 β -hydroxy-6-oxo-5 α -steroids from 3 β -acetoxy Δ^5 -steroids is reported; the methodology was applied to the synthesis of laxogenin (**7**), substance that behaves as a plant growth hormone. This is an alternative way to produce an important functionality found in many examples of naturally occurring steroids. The developed procedure uses inexpensive reagents and can be carried out in four steps. The oxidizing and acidic steps used in this methodology did not affect the labile spiroketal side chain present in diosgenin (**16**).

Keywords: laxogenin, diastereoselective epoxidation, 3 β -hydroxy-6-oxo-5 α -steroids

Introduction

The 3 β -hydroxy-6-oxo moiety is present in different naturally occurring steroids.¹ The brassinosteroid teasterone (**1**),² the steroid glycoside osladin (**2**),³ lactones as chiogralactone (**3**)⁴ and dendrosterone (**4**),⁵ and steroidal alkaloids like petisidine (**5**)⁶ and petisine (**6**)⁷ are some examples of naturally occurring steroids bearing such functionality (Figure 1).

Recently, we have synthesized laxogenin (**7**), a steroidal sapogenin isolated from *Smilax sieboldi*⁸ and its C-23 substituted derivatives **8** and **9** and reported that they have shown plant growth promoting activity similar of that of brassinosteroids (Figure 2).⁸ In that report, a protocol based on Brown's hydroboration for the introduction of the 3 β -hydroxy-6-oxo moiety was developed.

The 5 β ,6 β -epoxy moiety has been previously used for the preparation of 6-oxosteroids. In particular, Henbest and Wrigley reported⁹ that treatment of 5,6 β -epoxy-5 β -

cholestan-3 β -ol acetate with BF₃·Et₂O led to the corresponding 3 β -acetoxy-5-fluoro-6 β -hydroxy-5 α -steroid which was oxidized to 3 β -acetoxy-5-fluoro-5 α -cholestan-6-one using Jones reagent. More recently, an extension of this procedure to stigmaterol has been used to obtain brassinosteroid analogues bearing the 5 α -fluoro-6-oxo moiety.⁹

Owing our interest on steroids bearing such functionality, and after some reports on the highly stereoselective β -epoxidation of Δ^5 -steroids using biphasic systems involving potassium permanganate and metal salts,¹⁰ we envisaged the β -epoxidation of the C5-C6 double bond followed by the regioselective oxirane ring opening as the key steps for the introduction of an oxygen atom at position C-6. This led us to an alternative protocol for the synthesis of 3 β -hydroxy-6-oxo steroids.

Results and Discussion

Treatment of cholesteryl acetate, in a mixture of CH₂Cl₂, *tert*-butyl alcohol and H₂O, with KMnO₄ and

* e-mail: martin.iglesias@servidor.unam.mx, jsandova@siu.buap.mx

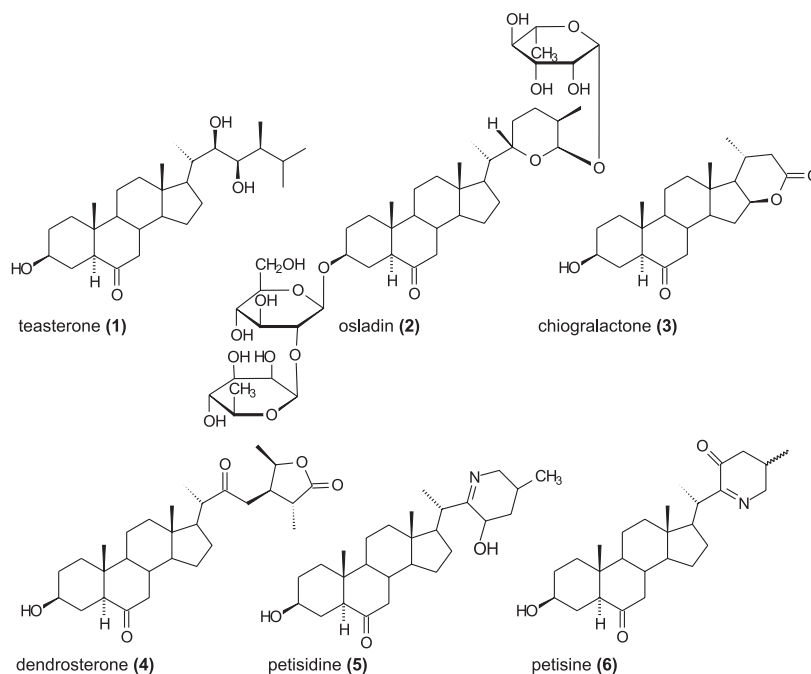


Figure 1. Some naturally occurring 3β -hydroxy-6-oxo- 5α -steroids.

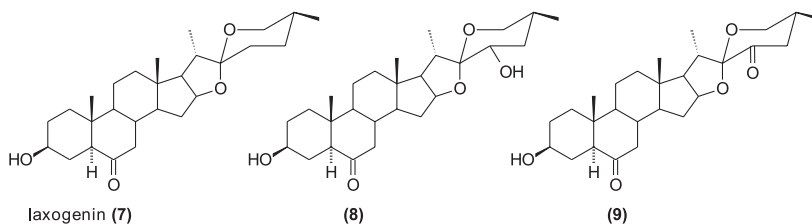
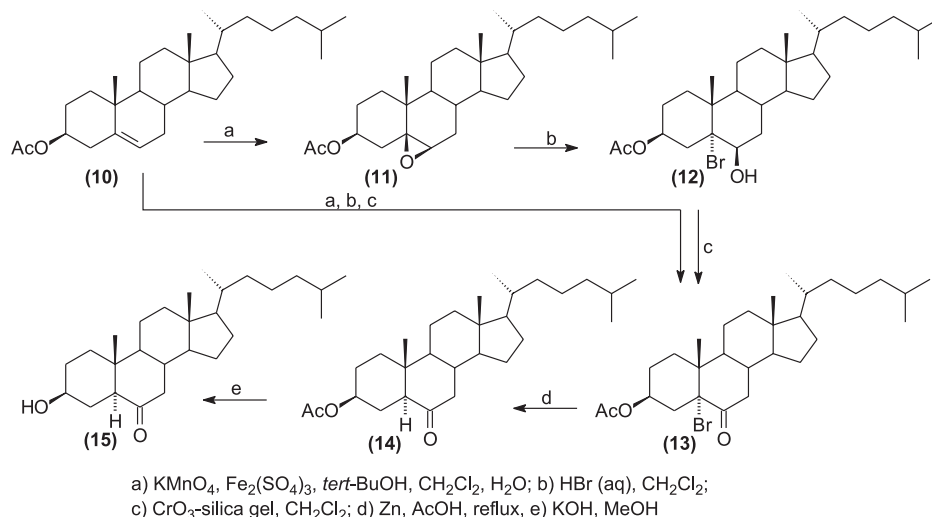


Figure 2. Laxogenin and 23-oxygenated derivatives with plant growth-promoting activity.

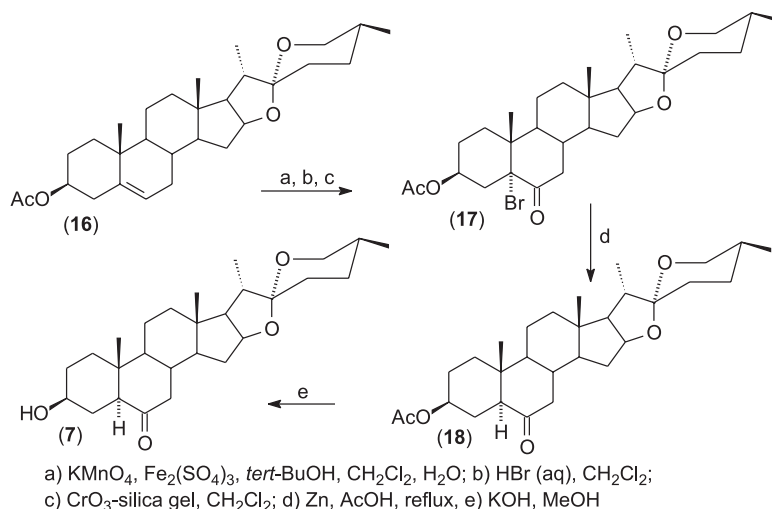
$\text{Fe}_2(\text{SO}_4)_3$ resulted in the highly diastereoselective β -epoxidation of the C5-C6 double bond; only traces of the α -epoxide could be detected in the ^1H NMR spectra. Regioselective opening of the β -oxirane ring with aq. HBr led to the bromohydrin **11** as sole product.

Treatment of the bromohydrin **12** with Jones reagent, supported on silica gel, led the bromoketone **13** (Scheme 1). In this way, manipulation of the reaction was rapidly effectuated and permitted a better treatment for wastes.

We also studied this three-reaction sequence without



Scheme 1.



Scheme 2.

isolation of the intermediate epoxide **11** and bromohydrin **12**; this resulted in a very fast and convenient protocol for the conversion of cholesteryl acetate into the bromoketone **13** (75% overall yield for the consecutive three steps). Treatment of **13** with zinc in refluxing acetic acid yielded the acetylated ketone **14**, which on saponification afforded the desired β -hydroxy- 5α -cholestan-6-one (**15**).

The same one-pot synthetic sequence was applied to diosgenin acetate (**16**) to produce laxogenin (**7**). Under these strong oxidizing and acidic media, the labile spiroketal side chain of diosgenin resulted unchanged (Scheme 2).

Experimental

NMR spectra were registered in CDCl_3 on a Varian Mercury spectrometer at 400 MHz for ^1H or 100 MHz for ^{13}C . Chemical shifts (δ) are expressed on ppm downfield from TMS. Melting points were obtained on a Gallenkamp MFB 595 apparatus and were not corrected.

5,6β-Epoxy-5β-cholestan-3β-ol acetate (**11**)

KMnO_4 (2 g) and $\text{Fe}_2(\text{SO}_4)_3 \cdot n\text{H}_2\text{O}$ (1 g) were finely grounded in a mortar, H_2O (0.2 mL) was added and the mixture was placed in a round bottom flask containing CH_2Cl_2 (5 mL). A solution of cholesteryl acetate (**10**) (428 mg, 1 mmol) in CH_2Cl_2 (5 mL) was added followed by addition of *tert*-butyl alcohol (0.5 mL). After 20 min of stirring at room temperature, the mixture was filtered through a pad of celite, and eluted with 30 mL of CH_2Cl_2 . The crude filtrate was washed with H_2O (5x15) mL, dried (anhydrous Na_2SO_4) and evaporated to afford 427 mg (96%)

of the desired epoxide **11**; mp 110-111 °C, lit.¹¹ 111-112 °C; ^1H NMR δ 4.76 (m, H-3), 3.07 (d, J 2.4 Hz, H-6), 2.03 (s, CH_3COO -3), 1.0 (s, CH_3 -19), 0.89 (d, J_{21-20} 6.8 Hz, CH_3 -21), 0.86 (d, J 6.4 Hz, CH_3 -26 and CH_3 -27), 0.64 (s, CH_3 -18); ^{13}C NMR δ 29.76 C-1, 27.25 C-2, 71.28 C-3, 38.02 C-4, 62.47 C-5, 63.55 C-6, 36.71 C-7, 32.49 C-8, 56.13 C-9, 35.03 C-10, 21.99 C-11, 39.78 C-12, 42.28 C-13, 50.97 C-14, 24.23 C-15, 28.21 C-16, 56.13 C-17, 11.85 C-18, 17.11 C-19, 35.76 C-20, 18.7 C-21, 36.15 C-22, 23.85 C-23, 39.51 C-24, 28.06 C-25, 22.64 C-26, 22.90 C-27, 21.41 CH_3COO -3, 170.24 CH_3COO -3.

5-Bromo-5α-cholestan-3β,6β-diol 3-acetate (**12**)

A solution of the epoxide **11** (445 mg, 1 mmol) in CH_2Cl_2 (15 mL) was vigorously shaken for 5 min in a separatory funnel with 5 mL of 48% HBr . The organic layer was washed with H_2O (5x 15 mL), dried (anh. Na_2SO_4) and evaporated to afford 510 mg (97%) of the bromohydrin **12**; mp 174-175 °C, lit.¹² 177-179 °C; ^1H NMR δ 5.47 (m, H-3), 4.18 (s, H-6), 2.03 (s, CH_3COO -3), 1.31 (s, CH_3 -19), 0.90 (d, J_{21-20} 6.6 Hz, CH_3 -21), 0.86 (d, J 6.4 Hz, CH_3 -26 and CH_3 -27), 0.67 (s, CH_3 -18); ^{13}C NMR δ 35.16 C-1, 26.43 C-2, 72.17 C-3, 38.42 C-4, 86.73 C-5, 75.68 C-6, 34.62 C-7, 30.64 C-8, 47.44 C-9, 41.30 C-10, 21.39 C-11, 39.78 C-12, 42.72 C-13, 55.73 C-14, 24.16 C-15, 28.29 C-16, 56.10 C-17, 12.32 C-18, 18.15 C-19, 35.83 C-20, 18.78 C-21, 36.18 C-22, 23.90 C-23, 39.51 C-24, 28.10 C-25, 22.67 C-26, 22.93 C-27, 21.50 CH_3COO -3, 170.30 CH_3COO -3.

β-Acetoxy-5-bromo-5α-cholestan-6-one (**13**)

A mixture of Jones reagent (2.5 mL) and silica gel (5 g)

was stirred until an homogeneous orange powder was formed. CH_2Cl_2 (15 mL) was added followed by the addition of a solution of the bromohydrin **12** (526 mg, 1 mmol) in CH_2Cl_2 (10 mL); the mixture was stirred for 20 min, filtered through a small pad of silica gel and the eluent was evaporated to afford 414 mg (79%) of the bromoketone **13**; mp 157-158 °C (decomp.), lit.¹³ 162 °C (decomposition); ^1H NMR δ 5.32 (m, H-3), 3.15 (dd, $J_{7\text{ax}-7\text{eq}}, 7\text{ax}-8\text{ax}}$ 14.85, 12.1 Hz, H-7a), 2.38 (ddd, $J_{4\text{eq}-4\text{ax}}$ 14.3, $J_{4\text{eq}-3\text{ax}}$ 4.2, $J_{4\text{eq}-2\text{eq}}$ 1.5 Hz, H-4eq), 2.28 (dd, $J_{7\text{ax}-7\text{eq}}$ 15, $J_{7\text{eq}-8\text{ax}}$ 5.5 Hz, H-7e), 2.03 (s, CH_3COO -3), 0.99 (s, CH_3 -19), 0.90 (d, J_{21-20} 6.2 Hz, CH_3 -21), 0.86 and 0.85 (d, J 6.6 Hz, CH_3 -26 and 27), 0.65 (s, CH_3 -18); ^{13}C NMR δ 30.43 C-1, 26.09 C-2, 70.92 C-3, 34.84 C-4, 79.61 C-5, 203.67 C-6, 40.45 C-7, 36.24 C-8, 47.26 C-9, 42.69 C-10, 21.79 C-11, 39.51 C-12, 43.03 C-13, 55.93 C-14, 23.84 C-15, 28.10 C-16, 56.16 C-17, 12.18 C-18, 14.59 C-19, 35.71 C-20, 18.72 C-21, 36.10 C-22, 23.84 C-23, 39.28 C-24, 28.07 C-25, 22.65 C-26, 22.90 C-27, 21.36 CH_3COO -3, 170.04 CH_3COO -3.

3 β -Acetoxy-5-bromo-5 α -cholestan-6-one (13) via a three steps procedure: β -epoxidation, oxirane-opening, alcohol-oxidation

KMnO_4 (6 g) and $\text{Fe}_2(\text{SO}_4)_3 \cdot n\text{H}_2\text{O}$ (3 g) were finely grounded in a mortar, H_2O (0.6 mL) was added and the mixture was placed in a round bottom flask containing CH_2Cl_2 (45 mL). A solution of cholesteryl acetate (**10**) (1.284 g, 3 mmol) in CH_2Cl_2 (45 mL) was added followed by addition of *tert*-butyl alcohol (1.5 mL). After 20 min of stirring at room temperature, the mixture was filtered through a pad of celite and eluted with 30 mL of CH_2Cl_2 . The crude filtrate was washed with H_2O (5x25 mL) and vigorously shaken for 5 min in a separatory funnel with 15 mL of 48% HBr, washed with H_2O (5x 30 mL) and added to a stirred mixture of silica gel-supported Jones Reagent (prepared from 2.5 mL of Jones reagent and 5 g silica gel as described for the oxidation of **12**) and CH_2Cl_2 (30 mL). The mixture was stirred for 20 min, filtered through a small pad of silica gel and the eluent was evaporated to afford 1.178 mg (75%) of the bromoketone **13**, identical as described above.

3 β -Acetoxy-5 α -cholestan-6-one (14)

A mixture of the bromoketone (524, 1 mmol), zinc powder (262 mg, 4 mmol) and acetic acid (10 mL) was stirred under reflux for 2 h. AcOEt (50 mL) was added and the mixture was washed with saturated NaCl solution (5x15 mL), 10% NaHCO_3 solution (3x20 mL) and H_2O (5x15 mL). The organic layer was dried with Na_2SO_4 and evaporated to afford 436 mg (98%) of the acetylated ketone

14; mp 126-127 °C, lit.¹⁴ 128-129 °C; ^1H NMR δ 4.67 (m, H-3), 2.03 (s, CH_3COO -3), 0.91 (d, J_{21-20} 6.6 Hz, CH_3 -21), 0.87 and 0.86 (d, J 6.6 Hz, CH_3 -26 and CH_3 -27), 0.77 (s, CH_3 -19), 0.66 (s, CH_3 -18); ^{13}C NMR δ 36.32 C-1, 26.75 C-2, 72.78 C-3, 26.04 C-4, 56.01 C-5, 210.40 C-6, 46.58 C-7, 37.85 C-8, 53.73 C-9, 40.87 C-10, 21.28 C-11, 39.38 C-12, 42.89 C-13, 56.59 C-14, 23.89 C-15, 27.97 C-16, 56.38 C-17, 11.94 C-18, 12.96 C-19, 35.62 C-20, 18.56 C-21, 36.00 C-22, 23.73 C-23, 39.38 C-24, 27.93 C-25, 22.48 C-26, 22.74 C-27, 21.39 CH_3COO -3, 170.56 CH_3COO -3.

3 β -Hydroxy-5 α -cholestan-6-one (15)

A mixture of the acetylated ketone **14** (445 mg, 1 mmol), KOH (0.5 g) and MeOH (50 mL) was gently warmed for 15 min, H_2O (15 mL) was added, most of the MeOH was evaporated and the mixture was extracted with AcOEt (2x20 mL). The organic layer was washed with brine (3x15 mL) and H_2O , dried with MgSO_4 and evaporated to afford 412 mg (98%) of the hydroxylated ketone **15**; mp 141-142 °C, lit.¹⁵ 142-143 °C; ^1H NMR δ 3.57 (m, H-3), 0.91 (d, J_{21-20} 6.6 Hz, CH_3 -21), 0.87 and 0.86 (d, J 6.6 Hz, CH_3 -26 and CH_3 -27), 0.75 (s, CH_3 -19), 0.66 (s, CH_3 -18); ^{13}C NMR δ 36.65 C-1, 30.61 C-2, 70.44 C-3, 29.97 C-4, 56.72 C-5, 210.87 C-6, 46.70 C-7, 37.92 C-8, 53.83 C-9, 40.96 C-10, 21.53 C-11, 39.44 C-12, 42.93 C-13, 56.66 C-14, 23.99 C-15, 28.03 C-16, 56.02 C-17, 12.06 C-18, 13.21 C-19, 35.68 C-20, 18.67 C-21, 36.07 C-22, 23.84 C-23, 39.44 C-24, 28.02 C-25, 22.60 C-26, 22.86 C-27.

(25R)-3 β -Acetoxy-5-bromo-5 α -spirostan-6-one (17)

Application of the described *one-pot* β -epoxidation, oxirane-opening, alcohol-oxidation procedure to diosgenin acetate (**16**) (1.370 g, 3 mmol) afforded 1.208 mg (73%) of the bromoketone **17**; mp 229-230 °C (petroleum ether/AcOEt), lit.¹⁶ 230-233 °C; ^1H NMR δ 5.31 (m, H-3), 4.42 (ddd, J_{16-17} , J_{16-15} 7.3 Hz, H-16), 3.48 (ddd, $J_{26\text{ax}-26\text{eq}}$ 11, $J_{26\text{eq}-25\text{ax}}$ 4.8, $J_{26\text{eq}-24\text{eq}}$ 1.8 Hz, H-26eq), 3.59 (dd, $J_{26\text{ax}-26\text{eq}}$, $J_{26\text{ax}-25\text{ax}}$ 11 Hz, H-26ax), 3.18 (dd, $J_{7\text{ax}-7\text{eq}}$ 11.7, $J_{7\text{ax}-8\text{ax}}$ 15 Hz, H-7 α), 2.40 (ddd, $J_{4\text{ax}-4\text{eq}}$ 14.7, $J_{4\text{eq}-3\text{ax}}$ 5.3, $J_{4\text{eq}-2\text{eq}}$ 1.8 Hz, H-4 α), 2.31 (dd, $J_{7\text{ax}-7\text{eq}}$ 15, $J_{7\text{eq}-8\text{ax}}$ 5.1 Hz, H-7 β), 2.03 (s, CH_3COO -3), 1.01 (s, CH_3 -19), 0.97 (d, J_{20-21} 7 Hz, CH_3 -21), 0.79 (d, J_{25-27} 7 Hz, CH_3 -27), 0.76 (s, CH_3 -18); ^{13}C NMR δ 30.40 C-1, 36.05 C-2, 70.82 C-3, 34.80 C-4, 79.37 C-5, 203.21 C-6, 40.50 C-7, 35.74 C-8, 47.29 C-9, 42.68 C-10, 21.62 C-11, 39.25 C-12, 40.99 C-13, 55.88 C-14, 31.38 C-15, 80.34 C-16, 61.90 C-17, 16.55 C-18, 14.68 C-19, 41.60 C-20, 14.58 C-21, 109.14 C-22, 31.44 C-23, 28.80 C-24, 30.29 C-25, 66.83 C-26, 17.22 C-27, 170.02 CH_3COO -3, 21.35 CH_3COO -3

(25R)-3 β -Acetoxy-5 α -spirostan-6-one (18)

1.103 g (2 mmol) of the bromoketone **17** were reduced as described for **13** to afford 756 mg (80%) of the ketone **20**; mp 223-224° (petroleum ether/AcOEt), lit.¹⁶ 222-224 °C; ¹H NMR δ 4.66 (m, H-3), 4.41 (m, H-16), 3.47 (ddd, $J_{26ax-26eq}$ 11, $J_{26eq-25ax}$ 4.2, $J_{26eq-24eq}$ 1.8 Hz, H-26eq), 3.36 (dd, $J_{26ax-26e}$ $J_{26ax-25ax}$ 11 Hz, H-26ax.), 2.03 (s, CH₃COO-3), 0.97 (d, J_{21-20} 6.97 Hz, CH₃-21), 0.79 (d, J_{27-25} 6.97 Hz, CH₃-27), 0.78 (s, CH₃-19), 0.78 (s, CH₃-18). ¹³C NMR δ 36.35 C-1, 26.84 C-2, 72.66 C-3, 26.16 C-4, 56.35 C-5, 209.63 C-6, 46.68 C-7, 37.38 C-8, 53.71 C-9, 40.95 C-10, 21.36 C-11, 39.44 C-12, 40.92 C-13, 56.42 C-14, 31.58 C-15, 80.32 C-16, 61.96 C-17, 16.49 C-18, 13.18 C-19, 41.60 C-20, 14.56 C-21, 109.14 C-22, 31.34 C-23, 28.77 C-24, 30.29 C-25, 66.81 C-26, 17.21 C-27, 170.31 CH₃COO-3, 21.41 CH₃COO-3.

(25R)-3 β -Hydroxy-5 α -spirostan-6-one, laxogenin (7)

473 mg (1 mmol) of the ketone **18** were hydrolyzed as described for **14** to afford 422 mg (98%) of laxogenin (**7**); mp 210-211 °C (petroleum ether/AcOEt), $[\alpha]_D^{25}$ - 83 (c 1.1, CHCl₃); lit.¹⁷ 210-212 °C; $[\alpha]_D^{25}$ - 86 (c = 1.0, CHCl₃); ¹H NMR δ 4.41 (m, H-16), 3.57 (m, H-3), 3.47 (ddd, $J_{26ax-26eq}$ 11, $J_{26e-25ax}$ 4.2 and $J_{26eq-24eq}$ 1.8 Hz, H-26e), 3.36 (dd, $J_{26ax-26eq}$ and $J_{26ax-25ax}$ 11 Hz, H-26ax.), 0.97 (d, J_{21-20} 6.97 Hz, CH₃-21), 0.79 (d, J_{27-25} 6.97 Hz, CH₃-27), 0.78 (s, CH₃-19), 0.78 (s, CH₃-18); ¹³C NMR δ 36.59 C-1, 30.56 C-2, 70.27 C-3, 29.91 C-4, 56.70 C-5, 210.39 C-6, 46.68 C-7, 37.33 C-8, 53.79 C-9, 40.90 C-10, 21.36 C-11, 39.46 C-12, 40.90 C-13, 56.39 C-14, 31.53 C-15, 80.31 C-16, 61.91 C-17, 16.43 C-18, 13.26 C-19, 41.55 C-20, 14.50 C-21, 109.11 C-22, 31.29 C-23, 28.71 C-24, 30.21 C-25, 66.75 C-26, 17.14 C-27.

Conclusions

A convenient procedure to generate 3β -hydroxy-6-oxo- 5α -steroids from Δ^5 -unsaturated steroids has been accomplished through a four steps process; the three first steps can be effectuated in a one-pot manner. Each step has been performed obtaining a good overall yield.

Acknowledgments

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