

Steroidal Glycoalkaloids and Molluscicidal Activity of *Solanum asperum* Rich. Fruits

Tania M. S. Silva,^{*,a} Celso A. Camara,^b Kristerson R. L. Freire,^c Thiago G. da Silva,^c
Maria de F. Agra^c and Jnanabrata Bhattacharyya^c

^aNúcleo Complexo Produtivo de Saúde, Instituto Multidisciplinar em Saúde, Campus Avançado Anísio Teixeira,
Avenida Olívia Flores, 3000, Candeias, 45055-090 Vitória da Conquista-BA, Brazil

^bDepartamento de Química, Universidade Federal Rural de Pernambuco, R. Dom Manoel de Medeiros, s/n,
Dois Irmãos, 52171-900 Recife-PE, Brazil

^cLaboratório de Tecnologia Farmacêutica, Universidade Federal da Paraíba, CP 5009,
58051-970 João Pessoa-PB, Brazil

O fracionamento bio-monitorado do extrato alcaloídico dos frutos verdes de *Solanum asperum* forneceu um novo alcalóide esteroidal, denominado solanandaina, juntamente com a solasonina e a solamargina. Tanto o extrato alcaloídico como os glicoalcalóides isolados apresentaram potente atividade moluscicida.

Bioassay-guided fractionation of the alkaloidal extract of the green fruits of *Solanum asperum* afforded a new compound, solanandaine along with solasonine and solamargine. The total crude alkaloids as well as the isolated pure alkaloids exhibited significant molluscicidal activity.

Keywords: *Solanum asperum*, glycoalkaloid, solanandaine, solasonine, solamargine, molluscicidal

Introduction

Solanum L. (Solanaceae) is distributed mainly throughout the tropical and subtropical regions of the world and is the largest and most complex genus of the family Solanaceae. In Brazil, *Solanum* is represented by about 350 species.¹ In this region, many species of *Solanum* are widely used in popular medicine and are commonly known as 'yubeba', the word derived from Tupi Guarani that refers to the prickles found on the stems of several of the species.¹ Some common and wide-spread *Solanum* species of Brazil had shown^{2,3} considerable molluscicidal activity demonstrated by several members of the genus investigated earlier.^{4,5} Thus, we have been studying various species of *Solanum* growing in our country^{3,6-10} with the expectation that the extracts of plants of this genus might be useful in the control of *Biomphalaria glabrata*, the intermediate host of *Schistosoma mansoni*, the parasite that causes human schistosomiasis in Brazil. In our previous bioassays, the crude MeOH extract of the unripe fruits of *S. asperum*

Rich showed^{3,11} activity in studies with *Artemia salina* ($LC_{50} = 420.5 \mu\text{g mL}^{-1}$) and *Biomphalaria glabrata* ($LC_{50} = 25.5 \mu\text{g mL}^{-1}$). Bioassay-guided fractionation indicated that the activity was concentrated in the alkaloid fraction. Thus, the alkaloid fraction upon CC on Sephadex LH-20 followed by PTLC on Silica gel afforded a new compound, solanandaine, $\text{C}_{45}\text{H}_{73}\text{O}_{16}\text{N}$ (**1**) mp 262-263 °C, $[\alpha]_D^{29} -60.0$ (MeOH, $c = 1.0 \text{ mg mL}^{-1}$) along with solasonine (**2**) and solamargine (**3**). In this work, we wish to report the isolation, characterization and the molluscicidal activity of the alkaloids of *S. asperum* unripe fruits.

Results and Discussion

Solanum asperum Rich. is popularly known in Brazil as 'jussara' or 'çoça-çoça'. It is a neotropical species belonging to the section *Brevantherum* with wide distribution in South America. Extract of *S. asperum* unripe fruits demonstrated significant molluscicidal activity. With the aid of bioassay-guided fractionation of the crude alkaloid mixture, solanandaine, solasonine and solamargine were isolated from the green fruits of *S. asperum*.

*e-mail: sarmiento@pesquisador.cnpq.br

Solanandaine (**1**) was obtained from MeOH, mp 262-263 °C; $[\alpha]_D^{29}$ -60.0 (MeOH, $c = 1.0 \text{ mg mL}^{-1}$). The structure of (**1**) was determined mainly on the basis of positive ion HREIMS and LC-MS along with one and two dimensional ^1H and ^{13}C NMR spectral analyses. The assignments of the carbon and proton resonances were made on the basis of HBBD, DEPT, ^1H - ^1H COSY, HSQC, HMBC and NOESY experiments. The positive ion HREIMS of solanandaine showed a peak at m/z 884.4964 $[\text{M} + \text{H}]^+$ corresponding to the molecular formula, $\text{C}_{45}\text{H}_{73}\text{O}_{16}\text{N}$ (calculated for $\text{C}_{45}\text{H}_{74}\text{O}_{16}\text{N}$, 884.4929). The positive ion LC-MS showed, in addition to the one at m/z 884 ($\text{M} + 1$), significant peaks at m/z 738 ($\text{M} + \text{H} - 146$), 592 ($738 - 146$), 430 ($592 - 162$) and 154. Solanandaine, therefore, contains three hexose units. The sequential loss of 146, 146 and 162 daltons indicate that solanandaine has a rhamnosyl-rhamnosyl-glucosyl side chain attached to an aglycone moiety. There are three typical anomeric proton signals in the ^1H NMR spectrum of solanandaine. The one at δ_{H} 4.93 (J 7.0 Hz) is certainly due to a β -D-hexose. The ^1H and ^{13}C NMR chemical shifts (Table 1) are fully compatible with a β -D-glucose structure for this hexose unit. The other two anomeric signals are broad singlets at δ_{H} 5.84 and 6.38, typical of α -L-rhamnose unities. The HSQC spectrum shows that the three signals at δ_{C} 100.6, 102.4 and 103.2 are due to the corresponding anomeric carbons of the three sugars unities in the glycoside chain.

^1H NMR spectrum of solanandaine (Table 1) showed among others the presence of five CH_3 signals, two of which are doublets at δ_{H} 1.61 (3H, J 6.1 Hz) and 1.75 (3H, J 6.2 Hz), supporting the presence of two deoxyhexose units like rhamnose and other three are 3H signals at δ_{H} 0.86 (s), 1.04 (s) and 1.08 (d , 7.0 Hz). Thus, the aglycone moiety contains only three CH_3 groups which suggest that it is not solasodine. The ^1H NMR spectrum also showed a signal at δ_{H} 5.30, characteristic of the CH-6 of Δ^5 -spirosolanes. The proton decoupled ^{13}C NMR spectrum of solanandaine (Table 1) shows the presence of 45 signals for 45 carbons in the molecule. In addition to supporting the presence of five CH_3 groups, the spectrum also shows two CH_2 signals (DEPT) at δ_{C} 61.6 and 66.4, typical of two CH_2OH groups. Therefore, apart from the one in the glucose unit, there must be an additional CH_2OH group in the aglycone moiety of solanandaine. The absence of one CH_3 group compared to solasodine and the appearance of a CH_2OH group instead strongly suggests the aglycone moiety of solanandaine to be solaparnaine.¹² The MS peak at m/z 154.16 daltons more than the corresponding peak in solasodine at m/z 138 is characteristic of an oxygenated ring F, like that in solaparnaine. ^{13}C NMR spectrum of solanandaine further shows a signal at δ_{C} 78.4 for C-3, which is considerably downfield relative

to the corresponding shift at ~ 71.50 in solasodine or solaparnaine (Table 1). This suggests the presence of sugar substitution at that position of the aglycone. This is further supported by the resulting upfield shifts of C-2 and C-4 to δ_{C} 30.5 and 39.3, respectively, in solanandaine relative to solasodine or solaparnaine.¹

Table 2 shows HMBC and NOESY correlations of the protons and carbons of solanandaine. The proton signal at δ_{H} 3.87 for H-3 has a cross peak with δ_{C} 100.6 (C-1') and the signal at δ_{H} 4.93 (d , J 7.0 Hz) for H-1' shows a cross peak with δ_{C} 78.4 (C-3) thereby confirming that a β -D-glucose unit is indeed substituted at that position. The NOESY spectrum supports this structure. In addition, it shows cross peaks of glycosidic linkages between δ_{H} 4.21 (H-2') and δ_{H} 6.38 (H-1'') as well as between δ_{H} 4.35 (H-4') and δ_{H} 5.84 (H-1'''). This is also supported by HMBC correlations (Table 2). Thus, there are two α -L-rhamnose units attached to C-2' and C-4' and an inner β -D-glucose unit which, in turn, is attached to C-3 of the aglycone unit of solanandaine.

Therefore, the structure of solanandaine must be 3-O- $[\alpha$ -L-rhamnosyl-(1 \rightarrow 2)- $[\alpha$ -L-rhamnosyl-(1 \rightarrow 4)- β -D-glucopyranosyl]-solaparnaine (**1**). The key HMBC correlations are shown on structure **1**. Subsequent acid hydrolysis of solanandaine in the usual way furnished solaparnaine (**4**). Solasonine (**2**) and solamargine (**3**) were identified by comparison of their physical and spectral data with those published in the literature.^{13,14}

Solanum species are known to produce a great variety of steroidal saponins and glycoalkaloids. The potato glycoalkaloids may have evolved in nature to protect the plant against phytopathogens and other hostile environments.^{15,16} In our previous studies we had found that several species of *Solanum*, including the unripe fruits of *S. asperum* have potentially significant molluscicidal activity.^{2,3} The crude alkaloid fraction obtained from the active total MeOH extract as well as the pure alkaloids isolated were tested for the molluscicidal activity (Table 2). Individually, the glycoalkaloids solanandaine (**1**) ($\text{LC}_{50} = 73.1 \mu\text{g mL}^{-1}$), solasonine (**2**) ($\text{LC}_{50} = 47.0 \mu\text{g mL}^{-1}$) and solamargine (**3**) ($\text{LC}_{50} = 26.3 \mu\text{g mL}^{-1}$), were found to be less active than the total MeOH ($\text{LC}_{50} = 25.5 \mu\text{g mL}^{-1}$)³ and the crude alkaloidal extracts ($\text{LC}_{50} = 9.7 \mu\text{g mL}^{-1}$) of *S. asperum* (Table 3). The higher activity of the crude extracts may be attributed to synergistic effects. The results of the bioassay show that solasonine and solamargine, the glycosides of the common aglycone solasodine, possess more molluscicidal effect and solamargine, with lesser polarity of the two shows relatively more activity. On the other hand, solanandaine, which has an aglycone different than solasonine or solamargine, presents the least bioactivity of all three. These results

indicate that the activity may be related to both glycosidic and aglycone moieties of the *Solanum* glycoalkaloids. Earlier, the molluscicidal activity of solasonine and solamargine in a mixture was studied¹⁷ against *Lymnaea cubensis* and *Biomphalaria glabrata*. The toxicity was more pronounced for *L. cubensis* (100% mortality, 10 ppm) and *B. glabrata* (100% mortality, 25 ppm). In the present work, we studied the activity of the individual glycoalkaloids in the bioassay with *B. glabrata* upon which the LC₅₀ values were calculated. The molluscicidal activity seen in some *Solanum* species is generally attributed to the presence of glycoalkaloids, with other classes of secondary metabolites, including alkalines, having little if any such activity.¹⁷ Besides solasonine and solamargine, others glycoalkaloids of *Solanum* as tomatine¹⁷ and solamarine¹⁸ presents molluscicidal activity.

Experimental

General

Melting points were determined on a Koeffler hot stage and are uncorrected. Optical rotation was measured with a Bellingham & Stanley Ltd., Model ADP220 polarimeter. The infrared absorption spectra were recorded in KBr pellets, using a Bomem/MB-102 spectrophotometer operating in the 4000–400 cm⁻¹ range. The LC-MS was obtained in positive electrospray mode using a Quattro LC–Micromass (Waters) and HREIMS were obtained by electron impact on a VG Autospec spectrometer. TLC was done using silica gel Kieselgel 60 (E. Merck) and spots were visualized by Dragendorff reagent. ¹H and ¹³C NMR spectra were obtained using a Bruker Advance 500 (500 Hz for ¹H and 125 MHz for ¹³C) Spectrometer as well as a Jeol Eclipse+ 400 spectrometer operating at 400 MHz in pyridine-*d*₅. Sephadex LH-20 (Sigma) was employed for gel permeation chromatography.

Plant material

Fruits of *S. asperum* were collected in the State of Paraíba, Brazil, in September 2005 from a secondary vegetation of the Atlantic forest area at the campus of the Universidade Federal da Paraíba, in the municipality of João Pessoa, at 130 to 160 m elevation. The plant was identified by Dr. Maria de Fátima Agra (LTF-UFPB). Voucher specimen (Agra 1243) is deposited at the Prof. Lauro Pires Xavier (JPB) Herbarium, Universidade Federal da Paraíba, João Pessoa, Brazil.

Extraction and isolation

Fresh fruits of *S. asperum* (740.0 g) were extracted with H₂O:HOAc (8:2) in a blender and filtered through a bed of Celite. The acid aqueous filtrate was basified with NH₄OH and left standing overnight. The gelatinous precipitate (10.1 g) formed was collected by filtration to give a mixture of glycoalkaloids. The alkaloid mixture was then chromatographed over Sephadex LH-20 using MeOH as eluent and fifteen fractions were collected. Fractions 3-10 showed the presence of alkaloids. Fraction 4 (850.0 mg) with three alkaloids was further purified by PTLC in silica gel plates and eluted with CH₂Cl₂:MeOH:NH₃ (9:2:0.5) to furnish solanandaine (**1**, 120.0 mg), solasonine (**2**, 450.0 mg) and solamargine (**3**, 172.0 mg).

Solanandaine (**1**): white powder (MeOH); mp 262–263 °C; [α]_D²⁹ -60.0 (MeOH, c= 1.0 mg mL⁻¹); IR (KBr) 3450, 2940, 1625, 1071, 1045, 980 cm⁻¹; ¹H NMR and ¹³C NMR (pyridine-*d*₅, 500 MHz and 125 MHz, respectively), see Tables 1 and 2. Positive-ion HREIMS *m/z* 884.4964 (calculated for C₄₅H₇₄O₁₆N, [M + H]⁺, 884.4929).

Molluscicidal tests

Molluscicidal activity of the crude alkaloid fraction and the individual glycoalkaloids was measured according to

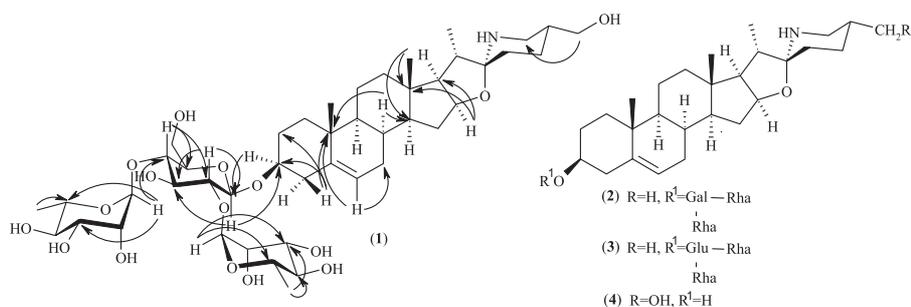


Figure 1. Solanandaine (**1**), Solasonine (**2**), Solamargine (**3**) and Solaparnaine (**4**).

Table 1. ¹H NMR (500 MHz) and ¹³C NMR (125 MHz) spectral data of solanandaine (**1**) in pyridine-*d*₅. Chemical shifts are in δ ppm and coupling constants (*J*= Hz) are given in parenthesis^a

Carbon	δ _c	δ _H	Carbon	δ _c	δ _H
1	37.8	a)1.71; b)0.98	Glu		
2	30.5	a)2.09; b)1.88	1'	100.6	4.93 (d, 7.0)
3	78.4	3.87 (m)	2'	78.1	4.21
4	39.3	a)2.80; b)2.74	3'	78.2	4.19
5	141.1		4'	78.9	4.35
6	122.2	5.30 (br, s)	5'	77.2	3.63 (br, d, 9.2)
7	32.7	1.86	6'	61.6	a)4.21; b)4.07
8	31.9	1.54	Rha		
9	50.6	0.89	1''	102.4	6.38 (s)
10	37.4		2''	72.8	4.83 (br, s)
11	21.4	1.45	3''	73.0	4.62 (dd, 9.3; 3.0)
12	40.4	a)1.69; b)1.11	4''	74.4	4.36 (t, 9.6)
13	41.0		5''	69.9	4.94 (br, d, 9.4)
14	57.0	1.05	6''	19.0	1.75 (d, 6.2)
15	32.9	a)2.08; b)1.45	Rha		
16	79.1	4.43 (m)	1'''	103.2	5.84 (s)
17	63.8	1.81	2'''	72.8	4.68 (br, s)
18	16.8	0.86 (s)	3'''	73.0	4.53 (dd, 9.3, 3.0)
19	19.7	1.04 (s)	4'''	74.2	4.34 (t, 9.4)
20	42.0	1.97 (m)	5'''	70.7	4.89 (m)
21	16.0	1.08 (d, 7.0)	6'''	18.8	1.61 (d, 6.1)
22	99.3				
23	34.6	a)1.78; b)1.67			
24	26.1	1.84			
25	40.5	2.05			
26	44.4	a)3.33; b)3.09			
27	66.4	3.71 (m)			

Superimposed ¹H signals are described without multiplicity.

Table 2. Significant HMBC and NOESY correlations for solanandaine (**1**)

Protons	δ _H	NOESY	Correlated C
H-1	1.71; 0.98	H-3	C-19
H-3	3.87	H-1; H-1'	C-1'
H-4	2.80α; 2.74β	H-3; H-6	C-2; C-3; C-10
H-6	5.30	H-4α; H-7	C-7; C-10
H-8	1.54		C-10; C-14; C-15
H-11	1.45	H-9; H-19	C-9
H-14	1.05	H-15β; H-18	
H-16	4.43	H-15α	C-13; C-17
H-18	0.86	H-20	C-13; C-14
H-19	1.04	H-11	C-1; C-5
H-21	1.08	H-16	C-16; C-22
H-27	3.71		C-26
H-1'	4.93	H-3; H-2'; H-5'	C-3; C-3'
H-2'	4.20	H-1'; H-6β	C-1; C-3'
H-3'	4.19	H-1'''	C-4'
H-4'	4.35	H-2'; H-5'''	C-3'; C-5'; C-1'''
H-1''	6.37	H-2'; H-2'''	C-3'; C-3''; C-5''
1'''	5.84	H-2'''; H-4'; H-6a, H-6b	C-4', C-2'''; C-5'''
4''	4.36	H-1'''	C-3''; C-5''
H-6''	1.75	H-5'''	C-4''; C-5''
H-6'''	1.61	H-5'''; H-4'	C-5'''; C-4'''

Table 3. Lethal concentrations and molluscicidal activity of crude extracts and pure alkaloids to kill 90%, 50% and 10% of *Biomphalaria glabrata* exposed at 24 h experiment

Samples Tested	Concentration μg mL ⁻¹ / (nM)		
	LC ₉₀	LC ₅₀	LC ₁₀
MeOH Extract*	44.1	25.1	7.0
Alkaloidal fraction	17.33	9.7	2.9
Solanandaine (1)	99.7 (112.8)	73.1 (82.7)	46.4 (52.5)
Solasonine (2)	72.0 (82.9)	47.0 (53.1)	22.0 (25.3)
Solamargine (3)	63.6 (73.3)	26.3 (30.3)	8.1 (9.33)

*Ref 3.

the method described³ using laboratory-bred *B. glabrata* as the target snails. The samples were dissolved in three drops of Cremophor emulsifier (BASF, Ludwigshafen, Germany) and dechlorinated water to give stock solutions containing 100 μg mL⁻¹ of each of the samples of crude alkaloid extracts and pure glycoalkaloids. For the preliminary bio-assays, each stock solution was either left undiluted

or further diluted with dechlorinated water to give test solutions containing 100, 50 and 10 $\mu\text{g mL}^{-1}$ of the samples. Subsequently, five different test solutions, ranging in concentration from 10 to 100 $\mu\text{g mL}^{-1}$ were prepared. For each assay, 10 adult snails (measuring 8–12 mm diameter) were exposed to 250 mL of each test solution in a glass beaker for 24 h at room temperature. After this period, each test solution was replaced with dechlorinated water. Snail mortality was then recorded over the following 24 h period, and compared with the positive controls (cupric carbonate at 50 $\mu\text{g mL}^{-1}$) and negative controls (extract-free dechlorinated water containing the same amount of Cremophor as the stock solutions). All assays were run in duplicate. The concentrations that kill 90% (LC_{90}), 50% (LC_{50}) and 10% (LC_{10}) of the treated snails (survived in the negative control cultures) were estimated by probit analysis, using the Origin 6.0

Acknowledgments

The authors thank IMSEAR-CNPq, CAPES and PIBIC-UFPB for financial support. JB thanks CAPES for generous support of Visiting Professorship. TMSS thanks Prof. Edilberto R. Silveira and Daniel E. Uchoa (CENAUREM – Centro Nordestino de Aplicação e Uso de RMN) and Prof. Raimundo Braz-Filho (UENF-RJ) for kindly recording the NMR data, and Socrates Golzio (LTF-UFPB) for kindly recording the LC-MS data.

Supplementary Information

Supplementary data of the isolated compounds as ^{13}C and ^1H NMR spectra are available free of charge at <http://jbcs.sbq.org.br>, as PDF file.

References

1. Agra, M. F.; Bhattacharyya, J. In *Solanaceae. IV, Advances in Biology and Utilization*; Nee, M.; Symon, D. E.; Lester, R. N.; Jessop, J. P. eds.; *Royal Botanic Gardens*: Kew, U.K., 1999, pp. 341–343.
2. Silva, T. M. S.; Camara, C. A.; Agra, M. F.; Carvalho, M. G.; Frana, M. T.; Brandolini, S. V. P. B.; Paschoal, L. S.; Braz-Filho, R.; *Fitoterapia* **2006**, *77*, 449.
3. Silva, T. M. S.; Batista, M. M.; Camara, C. A.; Agra, M. F.; *Ann. Trop. Med. Parasitol.* **2005**, *4*, 419.
4. Hostettmann, K.; Kizu, H.; Tomimori, T.; *Planta Medica* **1982**, *44*, 34.
5. Marston, A.; Hostettmann, K.; *Phytochemistry* **1985**, *24*, 639.
6. Silva, T. M. S.; Braz-Filho, R.; Carvalho, M. G.; Agra, M. F.; *Biochem. System. Ecol.* **2002**, *30*, 1083.
7. Silva, T. M. S.; Braz-Filho, R.; Carvalho, M. G.; Agra, M. F.; *Biochem. System. Ecol.* **2002**, *30*, 479.
8. Silva, T. M. S.; Nascimento, R. J. B.; Camara, C. A.; Agra, M. F.; Braz-Filho, R.; Carvalho, M. G.; *Biochem. System. Ecol.* **2004**, *32*, 513.
9. Esteves-Souza, A.; Silva, T. M. S.; Alves, C. C. F.; Carvalho, M. G.; Braz-Filho, R.; Echevarria, A.; *J. Braz. Chem. Soc.* **2002**, *13*, 838.
10. Silva, T. M. S.; Costa, R. A.; Oliveira, E. J.; Barbosa-Filho, J. M.; Agra, M. F.; Camara, C. A.; *J. Braz. Chem. Soc.* **2005**, *16*, 1467.
11. Silva, T. M. S.; Nascimento, R. J. B.; Batista, M. M.; Agra, M. F.; Barbosa-Filho, J. M.; Camara, C. A.; *Rev. Bras. Farmacogn.* **2006**, *17*, 35.
12. Bhattacharyya, J.; *Heterocycles* **1985**, *23*, 3111.
13. Puri, R.; Wong, T. C.; *J. Nat. Prod.* **1994**, *57*, 587.
14. Fukuhara, K.; Kubo, I.; *Phytochemistry* **1991**, *30*, 685.
15. Friedman, M.; Rayburn, J. R.; Bantle, J. A.; *Food Chem. Toxicol.* **1991**, *29*, 537.
16. Friedman, M.; *J. Agric. Food. Chem.* **2006**, *54*, 8655.
17. Alzerreca, A.; Hart, G.; *Toxicol. Lett.* **1982**, *12*, 151.
18. Wanyonyi, A. W.; Chlabra, S. C.; Mkoji, G.; Eilert, U.; Njue, W. M.; *Phytochemistry* **2002**, *59*, 79.

Received: June 21, 2006

Web Release Date: April 29, 2008

The Steroidal Glycoalkaloids and Molluscicidal Activity of *Solanum asperum* Rich. Fruits

Tania M. S. Silva,^{*,a} Celso A. Camara,^b Kristerson R. L. Freire,^c Thiago G. da Silva,^c
Maria de F. Agra^c and Jnanabrata Bhattacharyya^c

^aNúcleo Complexo Produtivo de Saúde, Instituto Multidisciplinar em Saúde, Campus Avançado Anísio Teixeira,
Avenida Olívia Flores, 3000, Candeias, 45055-090 Vitória da Conquista-BA, Brazil

^bDepartamento de Química, Universidade Federal Rural de Pernambuco, R. Dom Manoel de Medeiros, s/n,
Dois Irmãos, 52171-900 Recife-PE, Brazil

^cLaboratório de Tecnologia Farmacêutica, Universidade Federal da Paraíba, CP 5009,
58051-970 João Pessoa-PB, Brazil

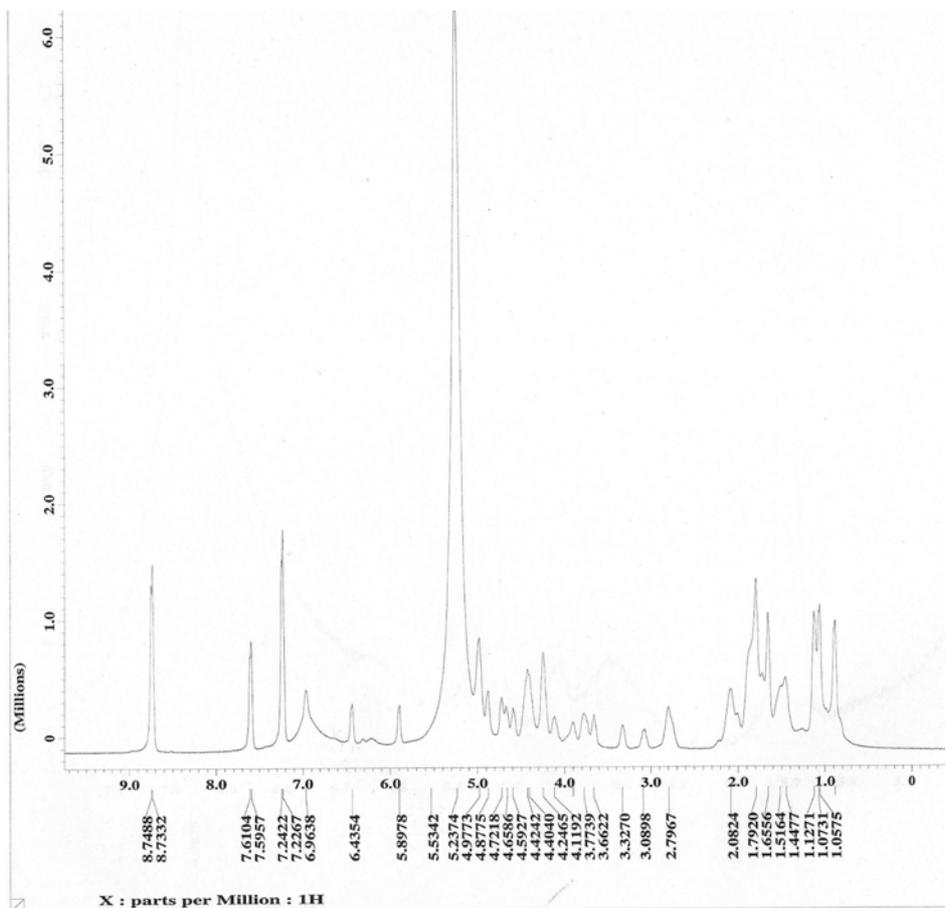


Figure S1. ¹H NMR (400 MHz, Pyridine-*d*₅) spectrum of solanandaine (1).

*e-mail: sarmento@pesquisador.cnpq.br

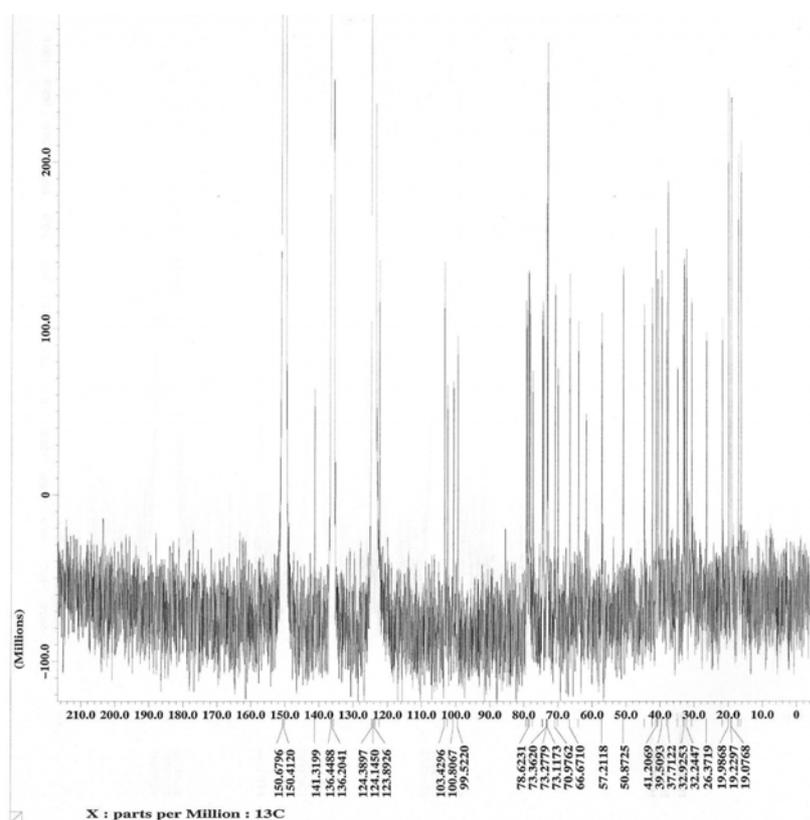


Figure S2. ¹³C NMR (100 MHz, Pyridine-d₅) spectrum of solanandaine (1).

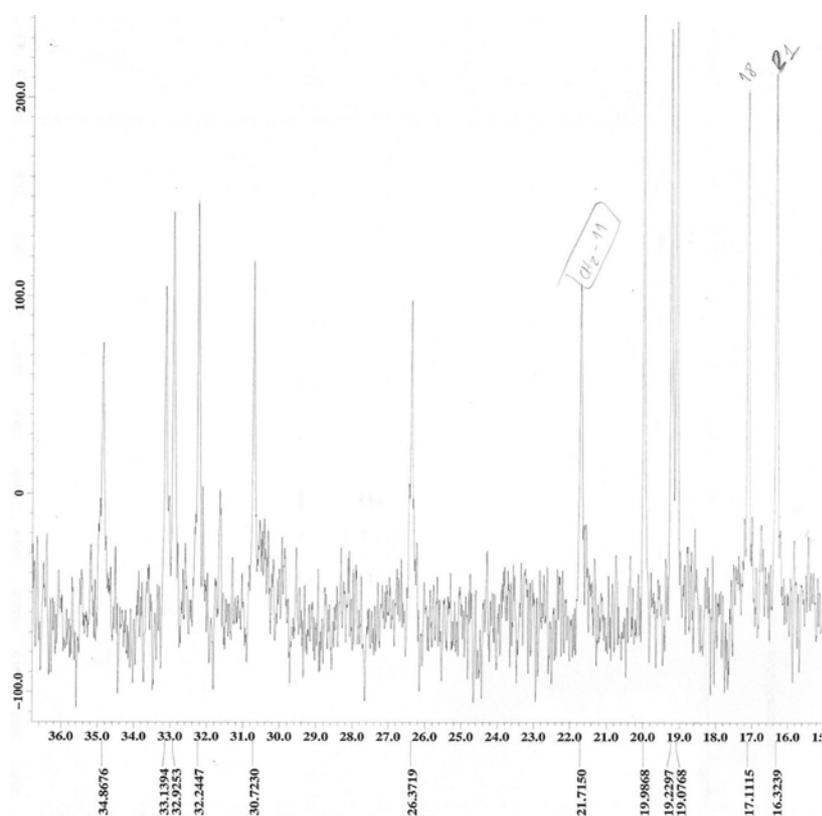


Figure S3. ¹³C NMR spectrum (100 MHz, Pyridine-d₅) expansion upfield of solanandaine (1).

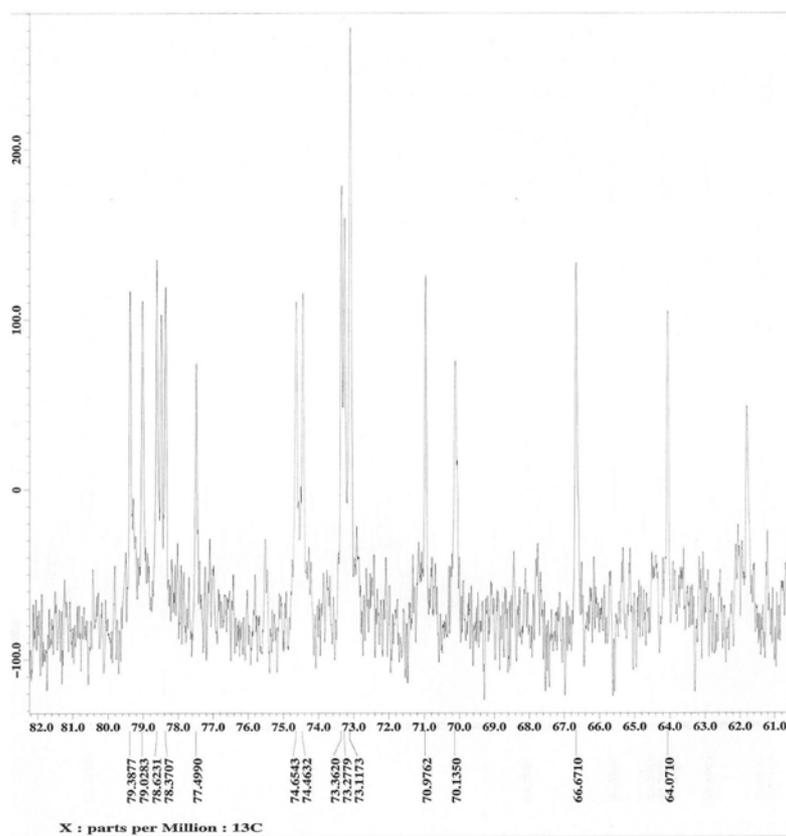


Figure S4. ^{13}C NMR spectrum (100 MHz, Pyridine- d_5) expansion downfield of solanandaine.

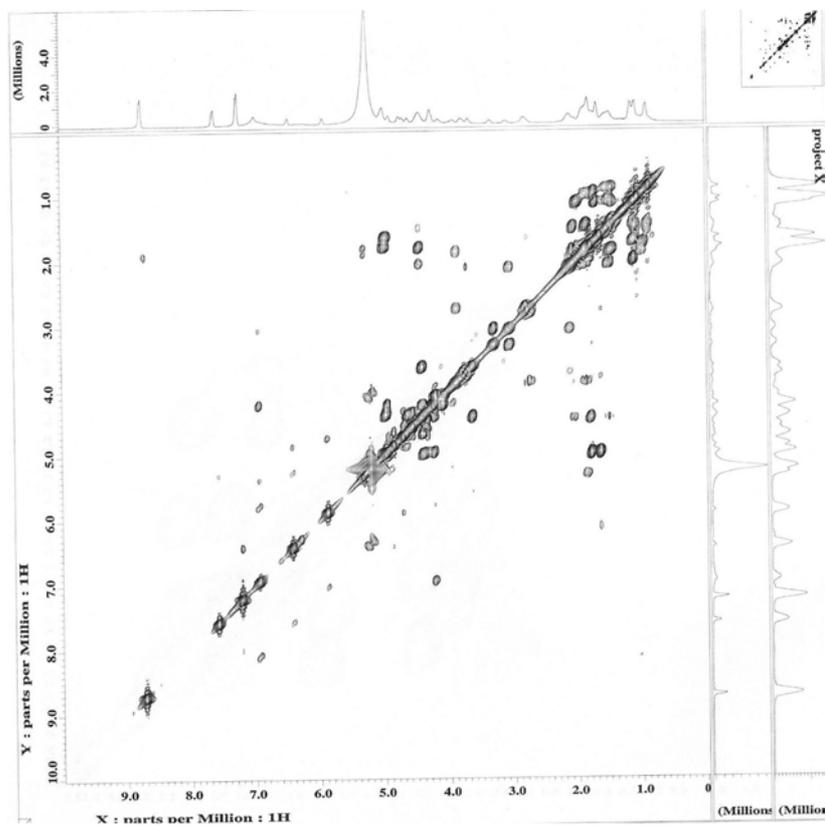


Figure S5. COSY spectrum (^1H NMR: 400 MHz, Pyridine- d_5) of solanandaine (1).

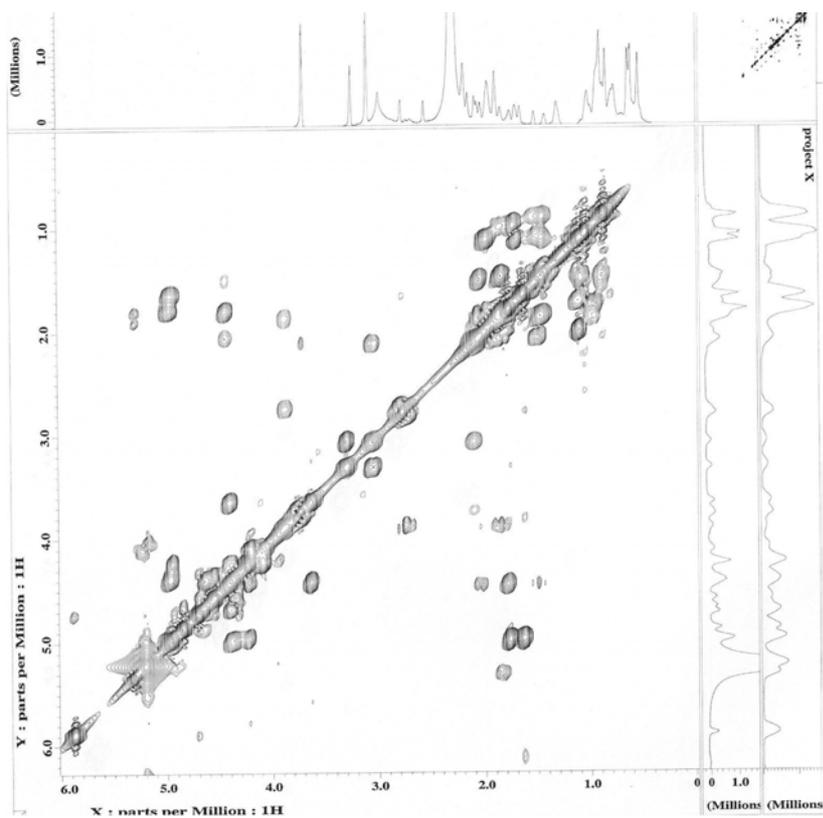


Figure S6. COSY spectrum (^1H NMR: 400 MHz, Pyridine- d_5) expansion upfield of solanandaine (**1**).

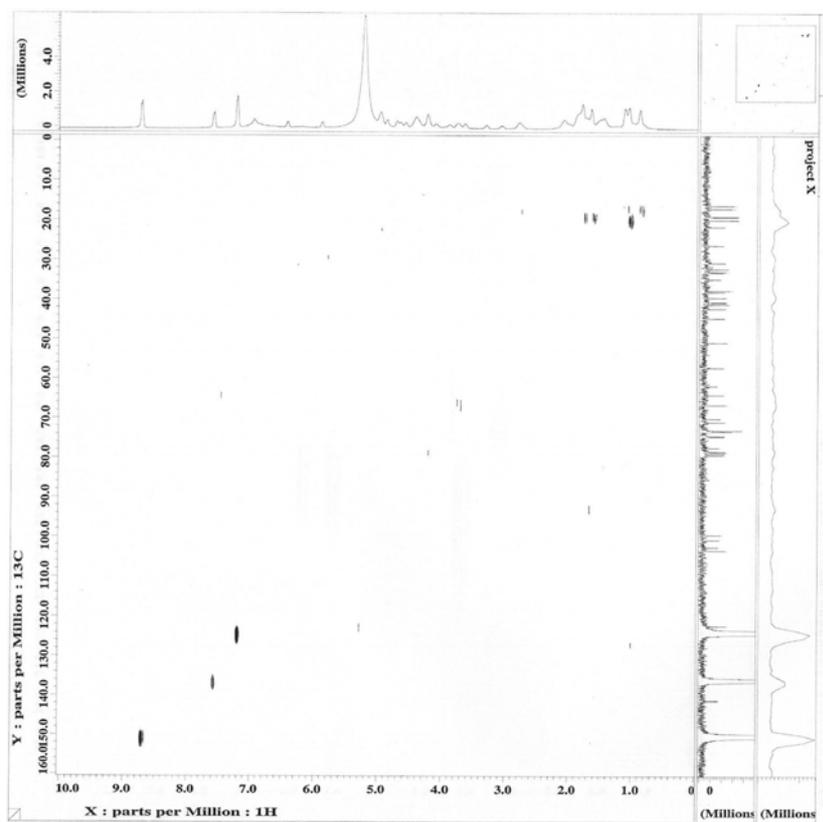


Figure S7. HMBC spectrum (^1H NMR: 400 MHz, ^{13}C NMR: 100 MHz, Pyridine- d_5) of solanandaine (**1**).

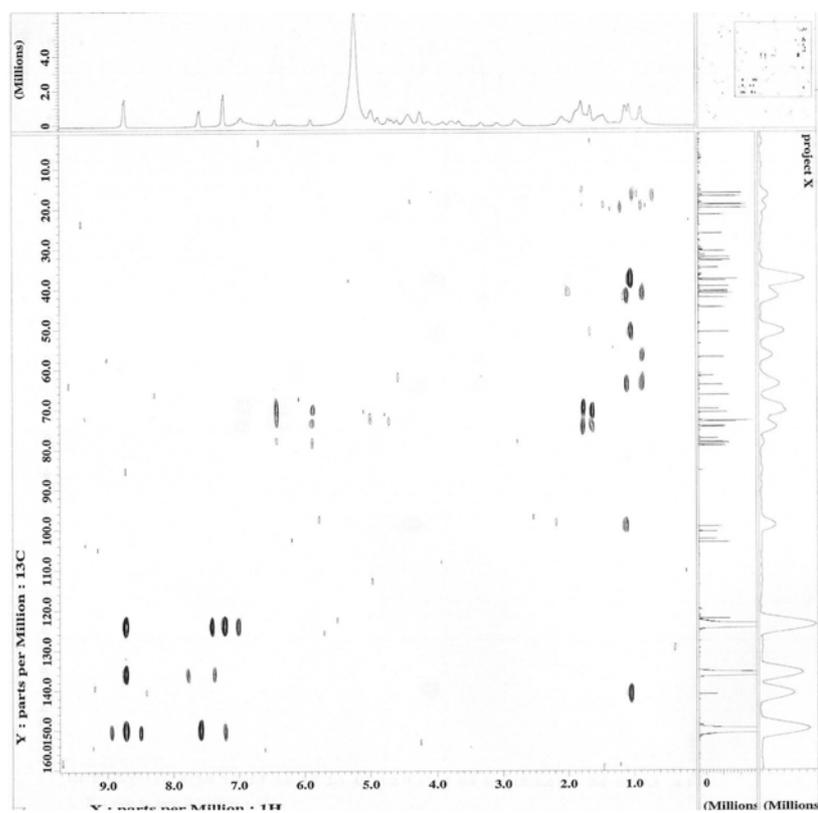


Figure S8. HMBC spectrum (¹H NMR: 400 MHz, ¹³C NMR: 100 MHz, Pyridine-*d*₃) of solanandaine (1).

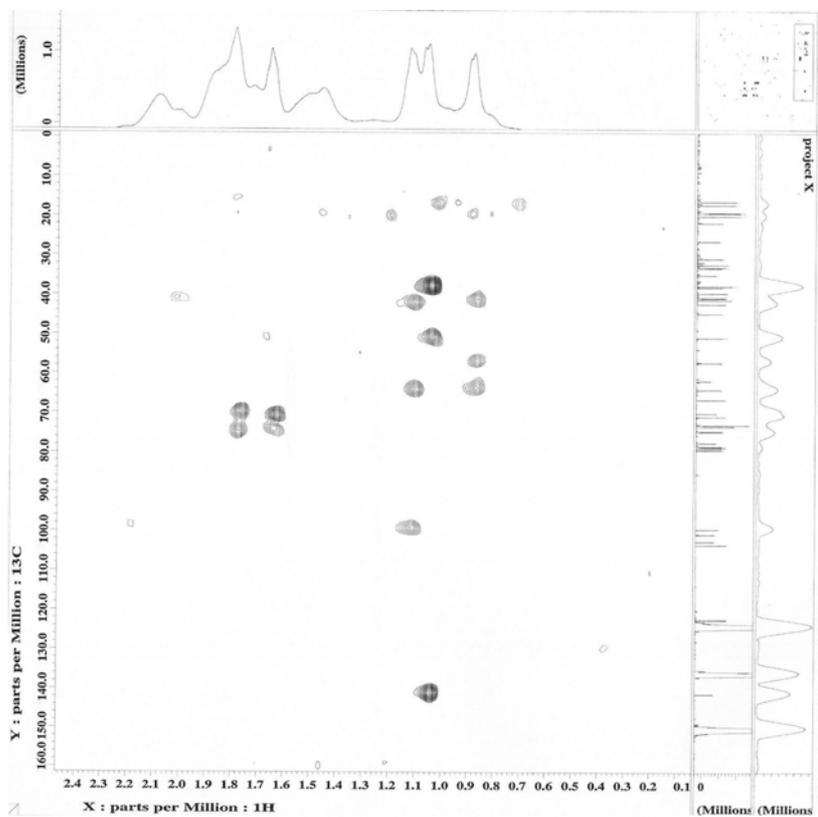


Figure S9. HMBC spectrum (¹H NMR: 400 MHz, ¹³C NMR: 100 MHz, Pyridine-*d*₃) expansion upfield of solanandaine (1).

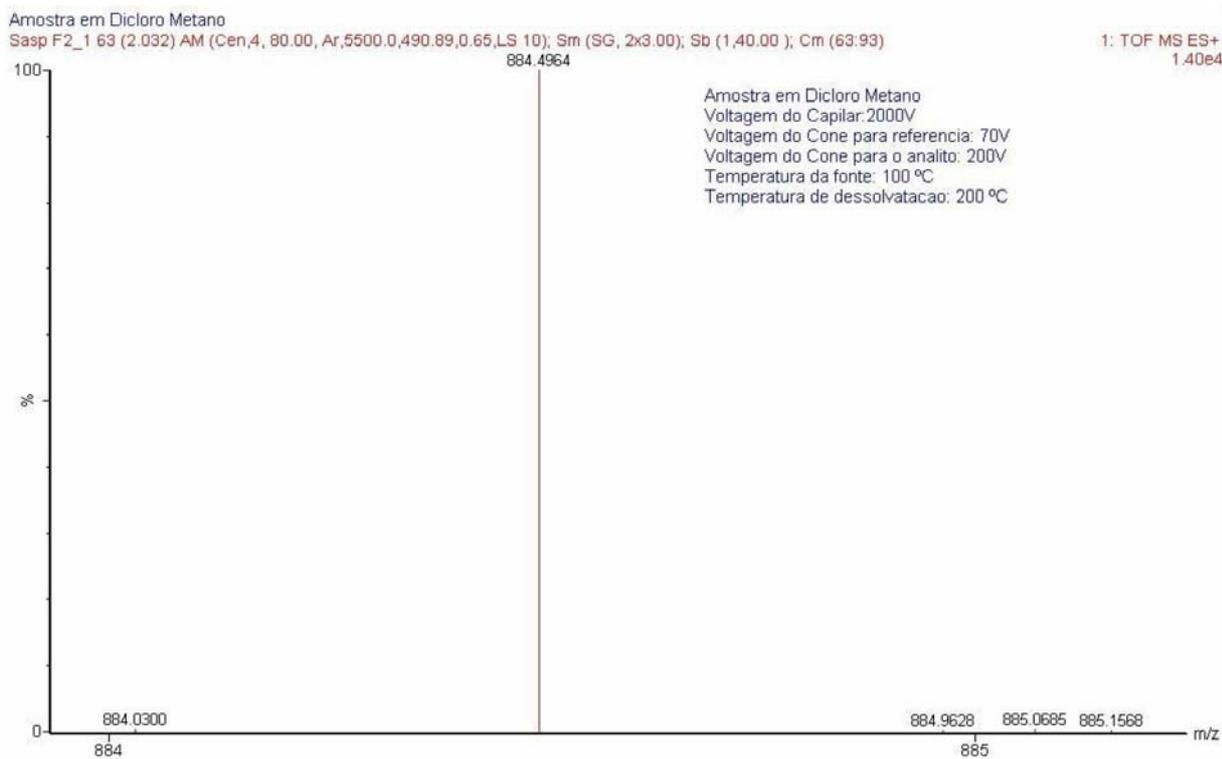


Figure S10. HREIMS (70 eV) spectrum of solanandaine (1).

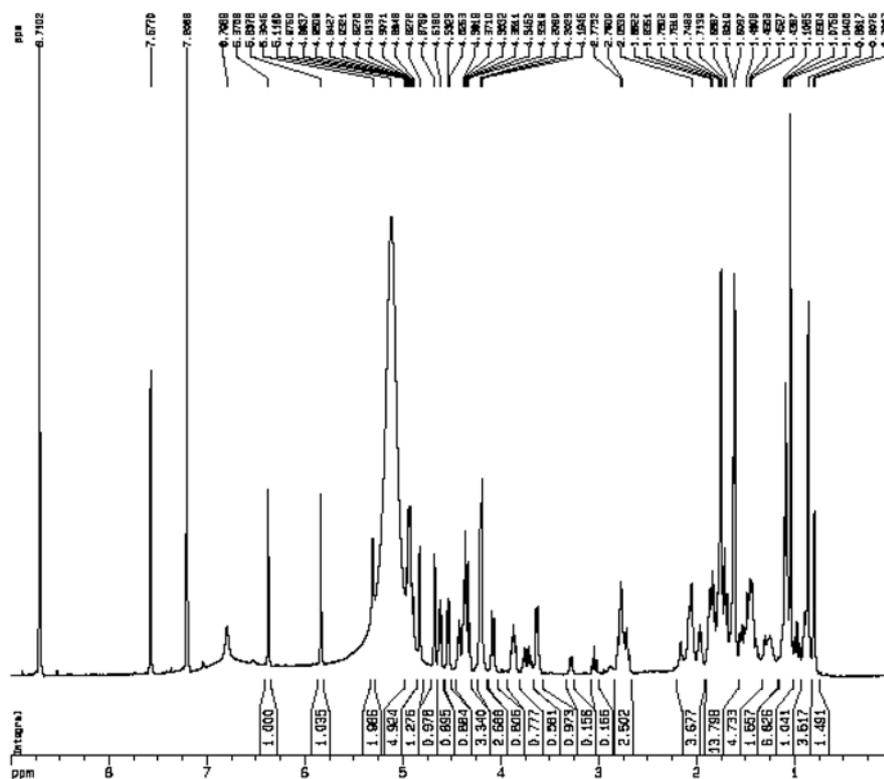


Figure S11. ^1H NMR (500 MHz, Pyridine- d_3) spectrum of solanandaine (1) and solamargine (3) mixture.

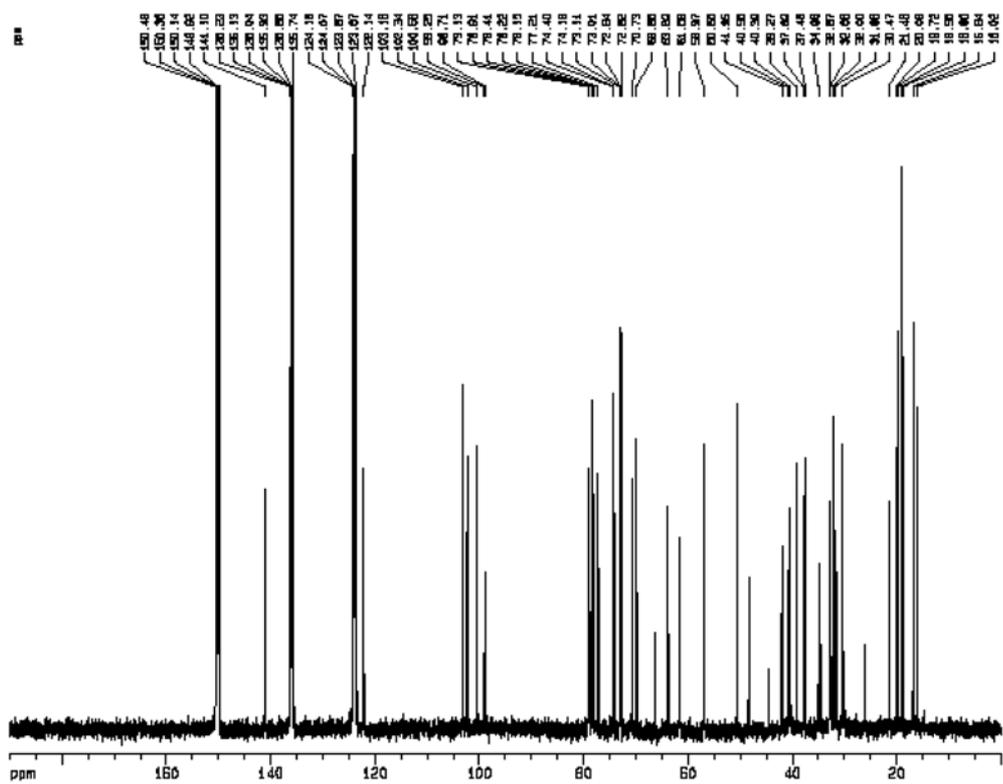


Figure S12. ^{13}C NMR (125 MHz, Pyridine- d_5) spectrum of solanandaine (1) and solamargine (3) mixture.

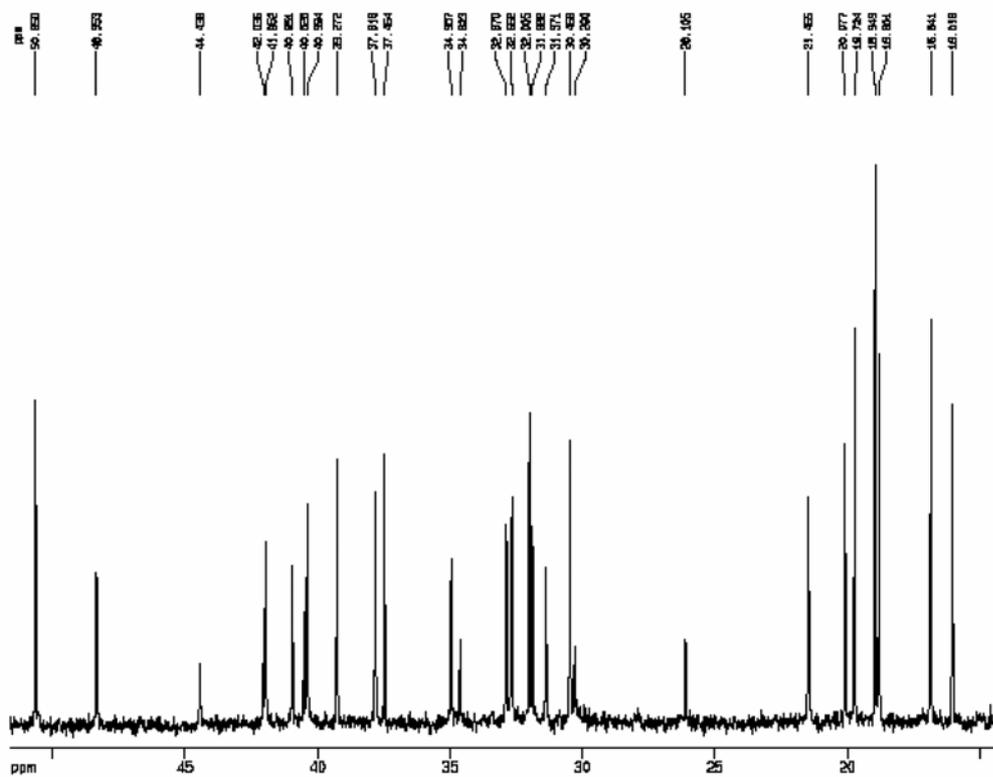


Figure S13. ^{13}C NMR spectrum (125 MHz, Pyridine- d_5) expansion upfield of solanandaine and solamargine mixture.

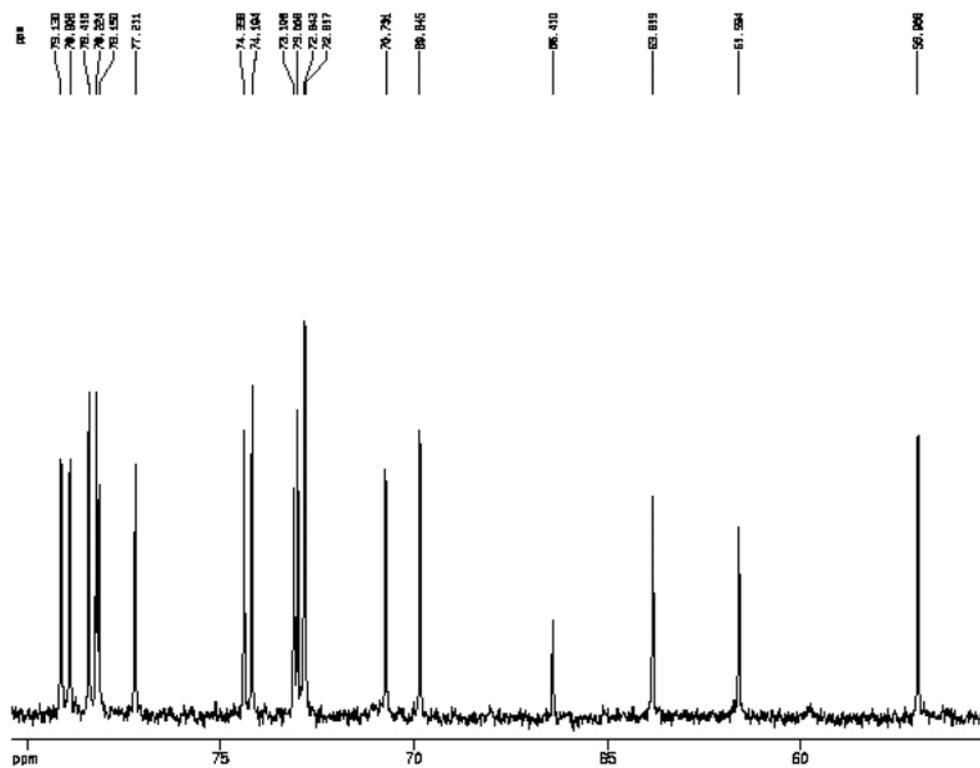


Figure S14. ^{13}C NMR spectrum (125 MHz, Pyridine- d_5) expansion downfield of solanandaine (1) and solamargine (3) mixture.

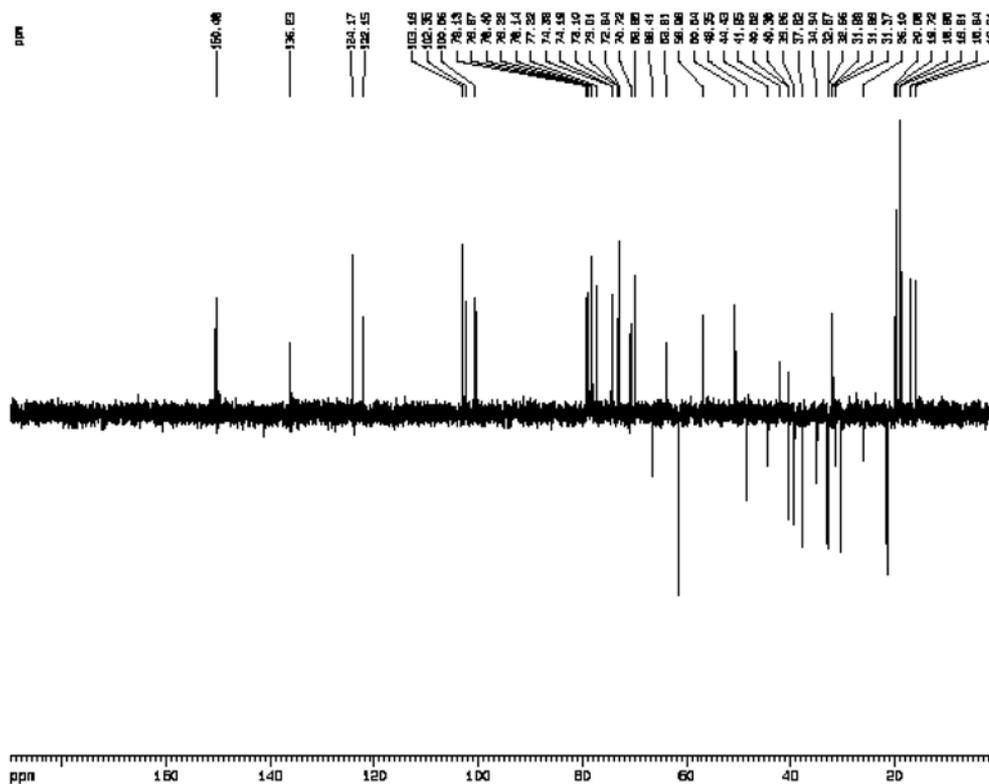


Figure S15. DEPT ($\theta = 135^\circ$, 125 MHz, Pyridine- d_5) spectrum of solanandaine (1) and solamargine (3) mixture.

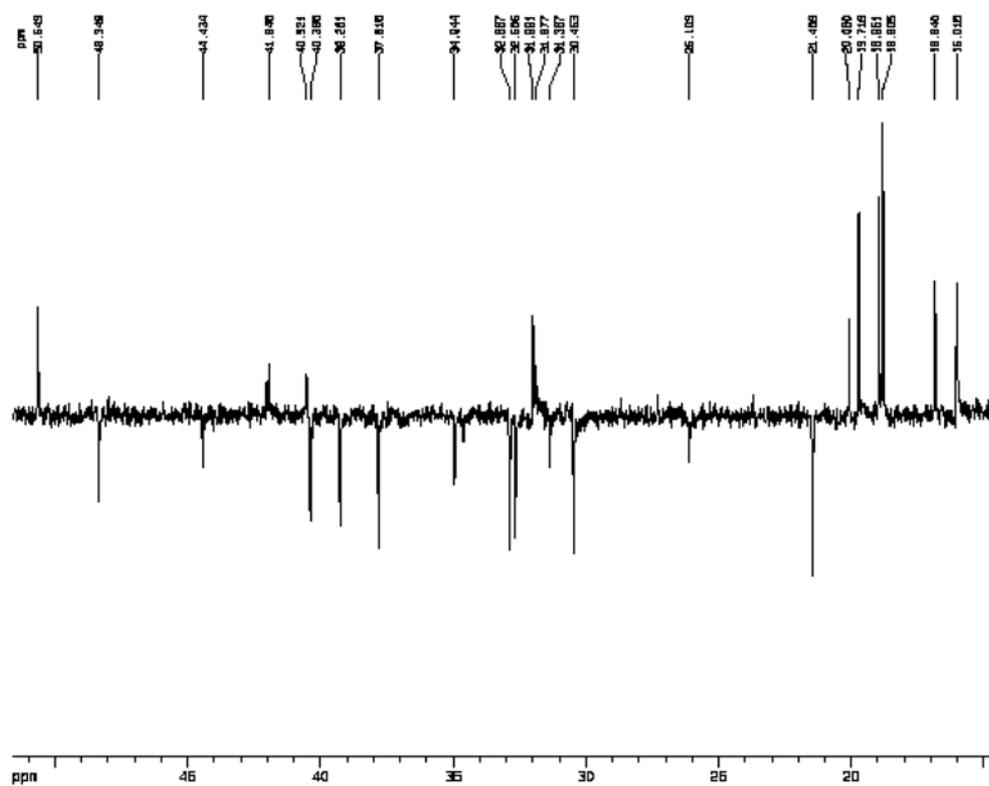


Figure S16. DEPT spectrum ($\theta = 135^\circ$, 125 MHz, Pyridine- d_5) expansion upfield of solanandaine (1) and solamargine (3) mixture.

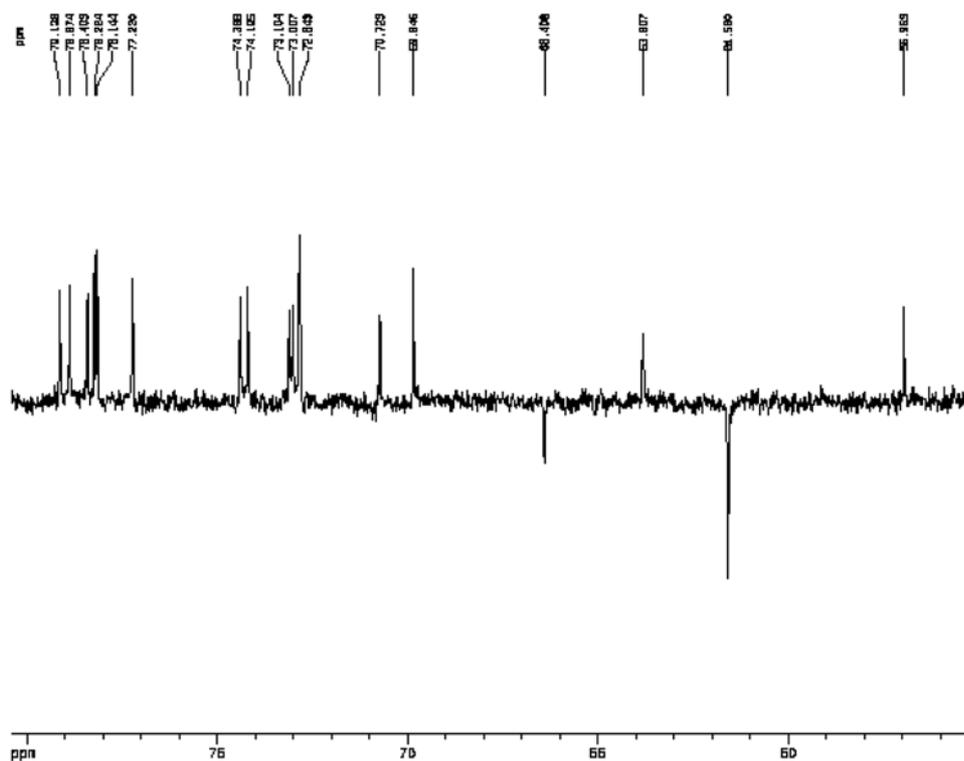


Figure S17. DEPT spectrum ($\theta = 135^\circ$, 125 MHz, Pyridine- d_5) expansion downfield of solanandaine (1) and solamargine (3) mixture.

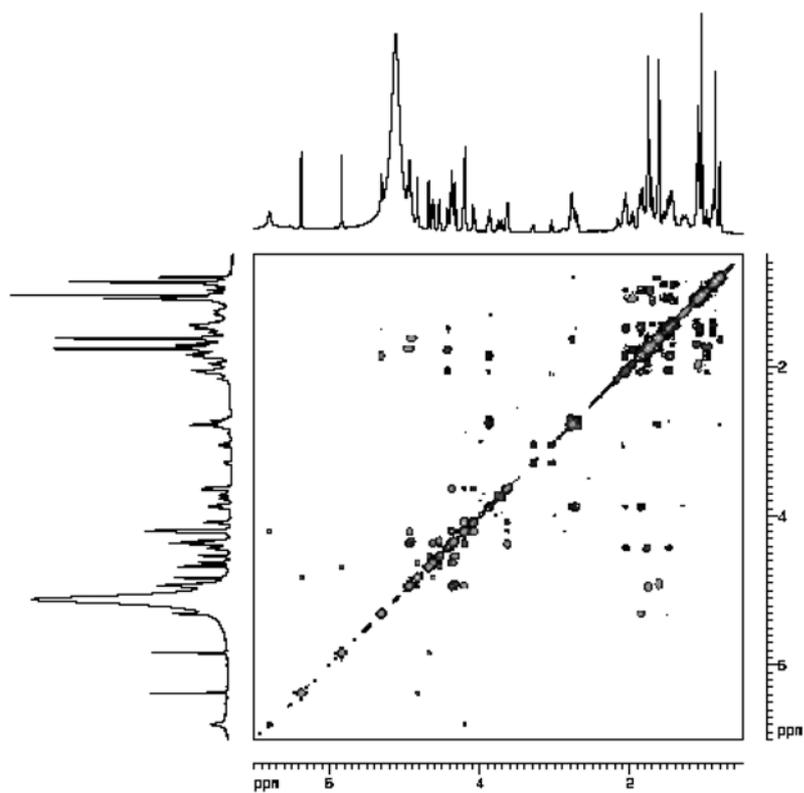


Figure S18. COSY spectrum (¹H NMR: 500 MHz, Pyridine-*d*₃) of solanandaine (1) and solamargine (3) mixture.

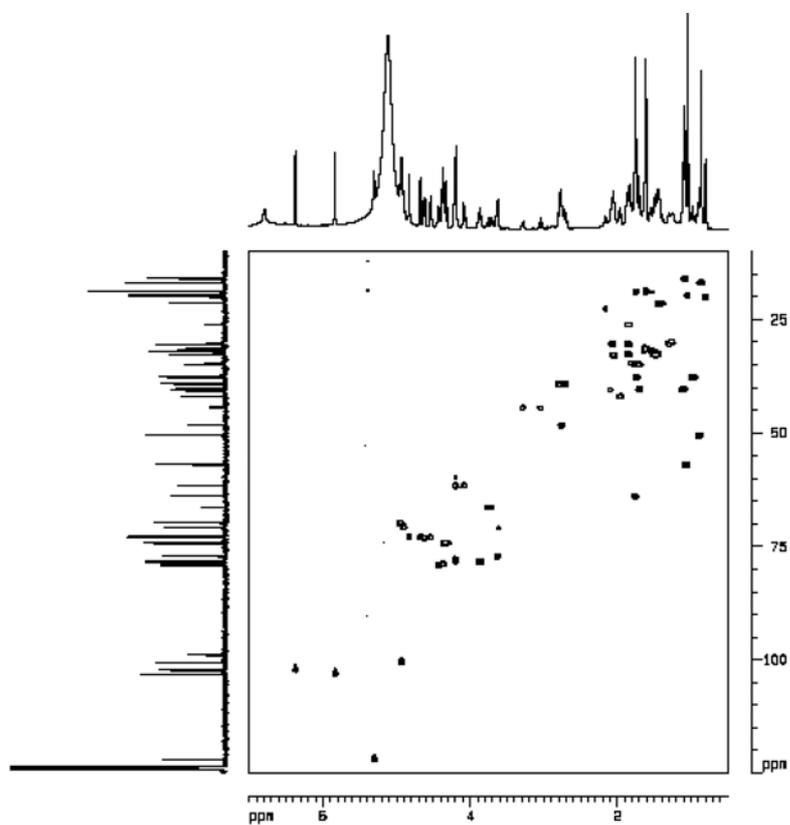


Figure S19. HSQC spectrum (¹H NMR: 500 MHz, ¹³C NMR: 125 MHz, Pyridine-*d*₃) of solanandaine (1) and solamargine (3) mixture.

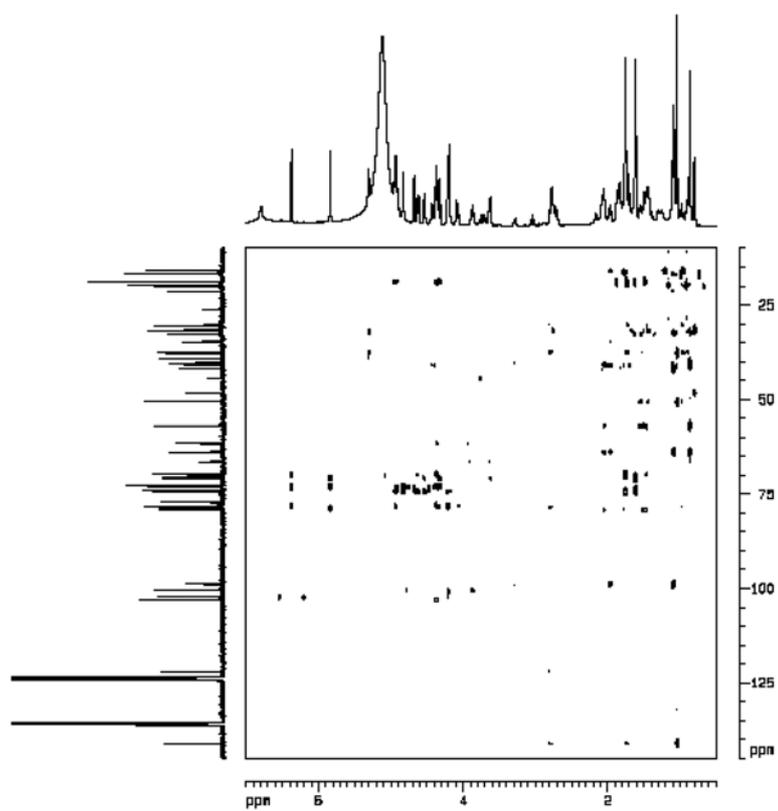


Figure S20. HMBC spectrum (^1H NMR: 500 MHz, ^{13}C NMR: 125 MHz, Pyridine- d_3) of solanandaine (1) and solamargine (3) mixture.

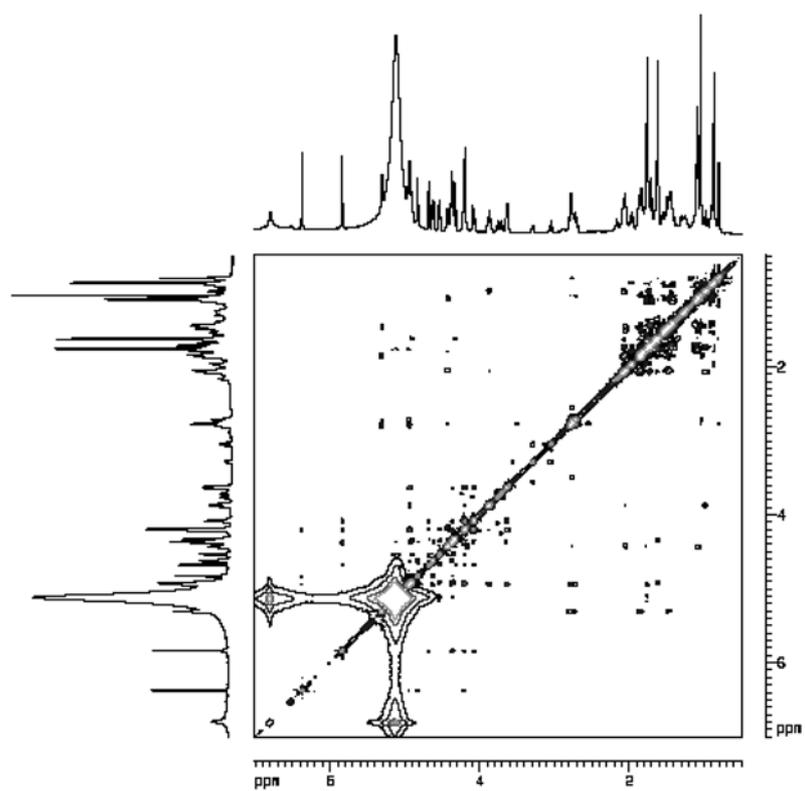


Figure S21. NOESY spectrum (^1H NMR: 500 MHz, Pyridine- d_3) of solanandaine (1) and solamargine (3) mixture.