Analysis of cementitious matrices self-healing with bacillus bacteria

Análise de matrizes cimentícias autocicatrizantes com bactéria Bacillus

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Abstract: Bacterial solutions have been studied to promote self-healing of cementitious matrices, however, the concentration of this solutions varied between studies. Consequently, the objective of this study was to evaluate the self-healing potential of different concentrations of Bacillus subtilis AP91 encapsulated in expanded perlite (EP). Visual examination and capillary absorption of water was measured over time. Test samples were also subjected to strength resistance tests. The physicochemical properties of EP and his distribution on the matrix was evaluated. There was no observable trend in the effect of solution concentration on the width of crack healed. However, concentration affected the quantity and length of the fissures healed. Capillary absorption decreased as fissures were healed while no significant changes were measured in strength regardless of the concentration. Results indicated that EP provided suitable encapsulation to the bacterial solution and there is an adequate distribution of the capsules in the cementitious matrix.

Keywords: bacteria, concrete, durability, cracking, mortar, self-healing.

Resumo: Soluções bacterianas têm sido estudadas para promover a autocura de matrizes cimentícias, entretanto, a concentração dessas soluções variou entre os estudos. Consequentemente, o objetivo deste estudo foi avaliar o potencial de autocura de diferentes concentrações de Bacillus subtilis AP91 encapsulado em perlita expandida (PE). O exame visual e a absorção capilar de água foram medidos ao longo do tempo. As amostras de teste também foram submetidas a testes de resistência de força. Foram avaliadas as propriedades físico-químicas do EP e sua distribuição na matriz. Não houve tendência observada no efeito da concentração da solução na largura da fissura cicatrizada. No entanto, a concentração afetou a quantidade e o comprimento das fissuras cicatrizadas. A absorção capilar diminuiu conforme as fissuras foram curadas, enquanto nenhuma mudança significativa foi medida na resistência da força, independentemente da concentração. Os resultados indicaram que a EP proporcionou encapsulamento adequado à solução bacteriana e que há uma distribuição adequada das cápsulas na matriz cimentícia.

Palavras-chave: bactéria, concreto, durabilidade, fissura, argamassas, autorregenerantes.

1 INTRODUCTION

Worldwide, concrete is the material most used material [1] due to several advantages [2]–[4]. However, it has been noted that Portland cement production accounts for 9.5% of global carbon dioxide emissions [5]. Besides these environmental considerations, additional costs are accrued when dealing with interventions in reinforced concrete. In developed countries, costs with repair, maintenance and reconstruction of structures is higher than new ones [6]. In developing countries, concrete structures tend to deteriorate ahead of their expected life cycle [3]. In monetary terms, one-third of the budget of large construction projects in the Netherlands is designated for interventions while in Great Britain, 45% of all activities are repair and maintenance [7]. Complementing, Gardner et al. [8] highlight the problem of maintenance costs, citing the United Kingdom as an example and thus justifying the need for such expenses to be minimized. The demolition/construction cycle of a project results in severe environmental costs [9] which are also believed to be associated with repair work [6].

Fissure formation and the resulting entry of harmful agents is one of the main causes of damage to reinforced concrete [10]–[16]. Thus, mitigating the occurrence and the size of fissures reduces concrete deterioration due to environmental factors [17], [18]. Fissure recovery usually involves the use of chemical products that are expensive and potentially harmful to health and the environment [16]. Consequently, new techniques to extend the durability of structures, such as self-healing concrete, are needed [19]. Self-healing can be obtained from higher cement use [20], [21], addition of pozzolans[21], [22], chemical solution encapsulation in light and porous aggregates [23] and bacterial solutions in concrete [16], [24]–[27].

Bacterial solutions inserted in cementitious compounds promote microbiologically induced calcite precipitation (MICP). Species of the Bacillus genus are capable of metabolizing MICP while at the same time being resistant to aggressive environments [28]. This bacterial genus has been widely studied and cultivated in alkaline mediums and its dormant endospores are able to survive centuries in chemically aggressive environments [29]–[32]. Like other chemoorganotrophic bacteria, this genus extracts energy from the oxidation of organic compounds [29]. The MICP reaction in cementitious compounds occurs through calcium lactate oxidation (Equation 1 and Equation 2), and calcium acetate oxidation (Equation 3 and Equation 4) [33], [34].

\[
\begin{align*}
CaC_6H_{12}O_6 + 6O_2 &\rightarrow CaCO_3 + 5CO_2 + 5H_2O \\
5CO_2 + 5Ca(OH)_2 &\rightarrow 5CaCO_3 + 5H_2O \\
Ca(C_4H_6O_4) + 4O_2 &\rightarrow CaCO_3 + 3CO_2 + 3H_2O \\
3CO_2 + 3Ca(OH)_2 &\rightarrow 3CaCO_3 + 3H_2O
\end{align*}
\]

In Equation 1, calcium lactate (\(CaC_6H_{12}O_6\)) oxidates into calcium carbonate (\(CaCO_3\)) with carbon dioxide (\(CO_2\)) and water (\(H_2O\)) as leftover products. As shown in Equation 2, \(CO_2\) produced from Equation 1 reacts with calcium hydroxide (\(Ca(OH)_2\)) present in large quantities in concrete as a product of cement hydration to precipitate additional \(CaCO_3\). Similarly, Equation 3 shows that calcium acetate (\(Ca(C_4H_6O_4)\)) oxidates into \(CaCO_3\), \(H_2O\) and \(CO_2\), with the latter undergoing the same reaction with (\(Ca(OH)_2\)) as seen in Equation 4. It has been noted that MICP through oxidation of organic compounds occurs primarily with bacteria of the genera Bacillus pseudofirmus, Bacillus cohnii, Bacillus alkalinitrilicus and Bacillus sp., which may be further complemented with other MICP-inducing metabolic processes [31].

As noted in Equation 1 and Equation 3, calcic nutrients are needed for the bacteria to perform MICP. Most studies make use of calcium lactate [10], [15], [27], [33], [35]–[37] calcium nitrate [25], [38], [39] calcium chloride [40], [41] and calcium acetate [33], [40], [42]. In addition to calcic nutrients, other more complex nutrients have been used such as yeast extract [15], [25], [33], [38], [39], [43], [44], meat extract [45] and peptone [33], [45]. However, these nutrients are known to have lower efficiency since their exact composition is unknown, especially with respect to calcium content [31]. The effect of different nutrients on strength resistance has been tested in cement pastes aged over time [33]. Results demonstrated that calcium lactate yielded the best results, with a strength increase comparing with control samples.

Studies conducted in Brazil have obtained positive results of MICP with Bacillus subtilis AP91 [46]. Surface treatment of pre-existing concretes has also been performed with Bacillus sphaericus and resulted in water absorption, gas permeability and chromatic aspect values similar to the application of commercial hydrophobic additives [40].
Surface treatment also reduced water absorption with beneficial effects on concrete durability [47]. Visual investigation of healed fissures confirmed the great potential of this technique with healing of fissures as wide as 0.79 mm [15]. Testing variations in general, each with distinct concentrations, led to an overall increase in strength resistance of cementitious compounds [45], [30]. *Bacillus, Sporosarcina* and *Shewanella* genera were tested in concentrations varying from $5 \times 10^6$ cells/mL to $2.8 \times 10^5$ cells/mL and a concentration of $3 \times 10^5$ measured in colony-forming units/mL (CFU/mL).

Bacterial solution used in substitution to mixing water resulted in an increase of 13% in strength resistance and 8.5% reduction in permeability [47]. However, bacteria may be damaged over time, cause its metabolic activity decreases in environments with pH above 12 [43]. In extreme cases, bacteria may not survive the alkaline environment, mechanical stresses of concrete mixing or cement hydration since matrix pores have dimensions of less than 0.5 μm while bacteria have dimensions between 1 μm and 3 μm [10], [25], [33], [36]. Consequently, micro-organisms encapsuled in light or porous aggregates have been tested [15], [27], [34]. Fissuring ruptures the capsules and releases the healing agent, which seeps into the matrix through capillary penetration [48].

Bacteria of the specie *B. cohnnii* were tested encapsulated in expanded clay (EC) and expanded perlite (EP) in the 2 mm to 4 mm granulometric range [15] and found healing to be more effective with EP. A bacterial solution of *B. pseudofirnis* DSM 8715 encapsulated in EP was successfully used in 20% substitution by volume of sand in concrete [34]. In order to prevent premature release of the healing agent, higher resistance coatings are recommended for the capsules in order to ensure light aggregate integrity during the mixing process [48]. Some types of coatings commonly used are geopolymers [15] and cement [49]. As for the effects of curing, self-healing mortars were tested in submerged curing and wet/dry cycles [37]. It was determined that the underwater cure form of curing yielded better results, probably due to the higher availability of oxygen.

Despite the use of coatings and the use of capsules for bacterial solutions, as mentioned before, it is also noted that studies evaluated the bacterial solution concentration, considering its relevance in the process of healing. Vijay, Murmu and Deo [30], for example, presented several studies with the authors choice for the bacterial concentration. Sidiq, Gravina and Giustozzi [50], considering specifically the specie *B. subtilis* pointed out the use of $10^5$ cell/ml. It is noted that the concentration must be selected for each specie, considering the characteristics of the microorganisms.

This study conducted mechanical and visual analyses of mortars with the addition of *B. subtilis* AP91 solutions in CFU/mL concentrations of $3 \times 10^6$, $3 \times 10^7$ and $3 \times 10^8$ encapsulated in EP, aiming to analyze if this concentration has effect on mechanical performance of the samples and its potential to promote self-healing. The encapsulating potential of the EP aggregate was also evaluated through void space measurements in a scanning electronic microscope (SEM) and the EP distribution in the mortar matrix was evaluated with micro-computed tomography (μCT).

EP capsules were chosen among other options due to their availability in the study region, and for their compatibility with concrete matrix in self-healing concretes [15]. Other systems can be used, such as superabsorbent polymers [51] or vascular networks however [52], with complex preparation procedures.

Several research have been carried out in the study of self-healing of cementitious materials, however, when it comes to self-healing promoted by biological agents, little is discussed about the influence of the concentration of these agents in this process. Still, there is a lack of published studies using microorganisms cataloged in Brazil. It is hoped that this study will contribute to the development of self-healing Brazilian cementitious materials, as well as bringing to discussion not only the different varieties of microorganisms used to promote self-healing, but also the concentration in which they are inserted in matrices.

## 2 EXPERIMENTAL PROCEDURES

### 2.1 Mixing ratio and characterization of materials

The bacteria used in this study was the same as Schwantes-Cezario, Nogueira and Toralles [53] and Schwantes-Cezario et al. [54]. Consequently, the same mixing ratio of 1:1 (cement: sand) in mass was used in the mortar. Expanded perlite (EP) with grain sizes between 2.4 mm and 4.8 mm was used to encapsulate the micro-organisms as in Zhang et al. [15]. It should be noted that granulometry of the EP used in Zhang et al. [15] yielded sizes between 2 mm to 4 mm. Since the sieve sizes were defined according to ASTM C33 [55], an adaptation to the diameters was necessary for this study. Fine EP aggregate was used in a substitution of 30% in volume for sand so that the final adapted mixing ratio was defined as 1:0.7:0.084 (cement: sand: EP). While the amount of EP substitute remained constant, the concentrations of bacterial solution varied throughout this study. Table 1 presents the mixing ratio of materials for each sample with respect to a 1 kg consumption of cement.
Table 1 – Materials and mixing ratio of samples studied

<table>
<thead>
<tr>
<th>Sample</th>
<th>Cement, kg</th>
<th>Sand, kg</th>
<th>EP, kg</th>
<th>B. subtilis AP91 concentration, UFC/ml</th>
<th>Water/cement ratio</th>
</tr>
</thead>
<tbody>
<tr>
<td>BAC.6</td>
<td>1</td>
<td>0.7</td>
<td>0.084</td>
<td>$3 \times 10^6$</td>
<td>0.37</td>
</tr>
<tr>
<td>BAC.7</td>
<td>1</td>
<td>0.7</td>
<td>0.084</td>
<td>$3 \times 10^5$</td>
<td>0.37</td>
</tr>
<tr>
<td>BAC.8</td>
<td>1</td>
<td>0.7</td>
<td>0.084</td>
<td>$3 \times 10^6$</td>
<td>0.37</td>
</tr>
</tbody>
</table>

Cement used was of Type IL(10) as defined in ASTM C595 [56] with no pozzolanic addition. Laser granulometry, surface area and density tests were conducted on the cement. Laser granulometry was performed in a Microtrac model S3500 apparatus with isopropyl alcohol as fluid. Surface area tests were conducted with Brunauer, Emmett and Teller (BET) isotherms and MicroActive TriStar II Plus 2.02 software. Density was measured in a helium gas pycnometer following ASTM D5550 [57] procedures. Results were a surface area of $2.0474 \pm 0.0085$ m$^2$/g and density of $3.0049$ g/cm$^3$, while granulometry results are presented in Figure 1. Sand granulometry was obtained with ASTM C33 [55] and density was obtained with ASTM C128 [58] and ASTM C29 [59]. The sand used was quartz-based and locally sourced from a river. Similar procedures were followed to obtain EP granulometry and apparent density. The apparent density of EP was measured for the natural original as well as the bacterial solution-impregnated and coated materials. Sand and EP characterizations are shown in Table 2 while granulometries are shown in Figure 2.

![Cement particle size distribution](image)

**Figure 1** – Cement particle size distribution.

Table 2 – Density and apparent density of aggregates

<table>
<thead>
<tr>
<th>Aggregate</th>
<th>Apparent Density kg/m$^3$</th>
<th>Density kg/m$^3$</th>
<th>Maximum characteristic dimension, mm</th>
<th>Fineness modulus</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sand</td>
<td>1,734.33</td>
<td>2,700</td>
<td>3.26</td>
<td>3.01</td>
</tr>
<tr>
<td>EP</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Natural</td>
<td>94.36</td>
<td>-</td>
<td>4.75</td>
<td>4.14</td>
</tr>
<tr>
<td>Treated</td>
<td>475.50</td>
<td>-</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>
It should be noted that despite sand not having the same granulometric distribution, a limited EP granulometric range was used in this study. This was defined in order to match sizes used in other studies [15], [27], and to allow direct comparison of the results between them. The chemical composition of EP is shown in Table 3, according to data provided by the producer.

Table 3 – Chemical composition of EP

<table>
<thead>
<tr>
<th></th>
<th>SiO₂</th>
<th>Al₂O₃</th>
<th>Fe₂O₃</th>
<th>TiO₂</th>
<th>CaO</th>
<th>MgO</th>
<th>Na₂O</th>
<th>K₂O</th>
<th>SO₃</th>
<th>P₂O₅</th>
<th>MnO</th>
<th>Loss of Ignition</th>
</tr>
</thead>
<tbody>
<tr>
<td>Min</td>
<td>71</td>
<td>78</td>
<td>8.5</td>
<td>15</td>
<td>0.3</td>
<td>1.2</td>
<td>0.01</td>
<td>0.01</td>
<td>0.3</td>
<td>1.2</td>
<td>0.1</td>
<td>0.3</td>
</tr>
<tr>
<td>Max</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
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</tr>
</tbody>
</table>

2.2 Growth curve, cultivation and bacterial encapsulation

The *B. subtilis* AP91 strain used was provided by the Campinas branch of Empresa Brasileira de Pesquisa Agropecuária – EMBRAPA. Cultivation in vitro was conducted in Luria Bertani (LB) broth containing 10g/L of tryptone 5g/L of yeast extract and 5g/L sodium chloride in deionized water. This broth has been used to adequately cultivate *B. subtilis* [48], [53]. A volume of 100 mL of LB broth was placed in two 250 mL Erlenmeyers flasks each and sterilized in an autoclave at 121° C for 15 min. Following sterilization, each flask was inoculated, and the cultures were incubated in an orbital shaker at 165 rpm and 37° C (98.6°F) for 24 h. The growth curve was obtained by taking samples over the incubation period. The cultivation absorbance was analyzed in a Kazuaki model IL-226-NM spectrophotometer of 600 nm wavelength. Simultaneously to the absorption measurement, each sample was ten-fold serially diluted in saline solution (0.9% sodium chloride) and each dilution was pour-plated in agar LB for colony forming unit (CFU) counting. It should be noted that not all bacteria exist as isolated cells and bacterial agglomerates and groups also generate a single colony. Due to this fact, colony-counting cannot be taken as the absolute number of cells and, thus, CFU was selected [60]. Plated samples were placed in an incubator at 36 °C for 48 h, at which point colonies were counted for the corresponding absorption value. The procedure adopted is shown in Figure 3 and described in detail in Swanson et al. [60]. Figure 4 shows the species spectrophotometric behavior relating the absorbance and concentration of each sample.
Figure 3 – Procedures for spectrophotometry of *B. subtilis* AP91: (a) sterile LB Broth, (b) LB broth with bacterial growth and (c) colonies 48h after in an incubator.

The figure allows the determination of the cultivation level of *B. subtilis* AP91 based on its absorbance. In the equation, shown overlaid in Figure 4, the absorbance ($A_{600}$) is dimensionless and concentration ($C$) is given in CFU/mL. The figure allows the determination of the cultivation level of *B. subtilis* AP91 based on its absorbance. In the equation, shown overlaid in Figure 4, the absorbance ($A_{600}$) is dimensionless and concentration ($C$) is given in CFU/mL.

![Graph showing spectrophotometric behavior of B. subtilis AP91](image)

$A_{600} = 2 \times 10^{-5}C + 0.0228$

$R^2 = 0.9847$

Figure 4 – Spectrophotometric behavior of *B. subtilis* AP91

A sterile buffer solution containing 1.06 g/L of sodium phosphate (anhydrous dibasic), 0.36 g/L sodium phosphate (monobasic) and 8.17 g/L sodium chloride in deionized water was used as diluant for the bacteria for the purpose of encapsulation and later insertion in the cementitious mixture. In parallel, the bacteria were inoculated in 2 L Erlenmayer flasks with 1 L of sterile BL broth and incubated in an orbital shaker for 24 h. Following the 24 h growth period, the solutions were centrifuged in an Eppendorf AG model 5430 R centrifuge at 4,000 rpm and 23 °C for 3 min to precipitate the cells. The supernatant was discarded, and the cells washed 4 more times with the sterile phosphate buffer solution, each time following the same previous centrifugation and supernatant discarding procedures. This was done in order to remove any organic matter present in the cultivation solution. The progressive changes in visual appearance of the solution are shown in Figure 5. The buffer solution and washing procedures followed the methodology of Schwantes-Cezario et al. [53].

Figure 5 – LB broth (a) before (b) and after centrifugation; (c) buffer solution before (d) and after centrifugation
After washing, cells were re-suspended in phosphate buffer solution and its absorbance was measured to define the initial concentration. Then, the solution was diluted to the target concentration. The solutions were refrigerated at 8 ºC for 2 days in order to promote sporulation [53] and slows down cellular multiplication [29], ensuring that concentrations are kept constant.

Impregnation of EP with bacterial solution was accomplished by submerging the aggregate for 3 hours in different concentration solutions. After soaking, the EP was placed in a vacuum desiccator for the 30 min necessary to achieve maximum impregnation [49]. Following impregnation, excessive liquid was drained, and EP dried in an incubator at 45 ºC for 48 h. After drying, EP was sprayed with a calcium lactate solution containing 8 g/L of calcium lactate, 1 g/L of yeast extract and deionized water and further incubated at 45 ºC for 48 h. This procedure followed the methodology of Zhang et al. [15]. As a protective coating, the same cement used in the mortar was sprinkled on the aggregate following the procedure of Sisomphon et al. [49]. Although not mentioned in other studies, a further spraying of the calcium lactate solution prior to coating with cement was necessary to promote hydration and improve adhesiveness. Once the encapsulation process was completed, the aggregate was kept at rest for 7 days to cure the cement coating prior to molding the test samples as per the methodology of Zhang et al. [15].

2.3 Molding of samples

Cylindrical (50 mm x 100 mm or 1.96in x 3.93in) and prismatic (40 mm x 40 mm x 160 mm) samples were molded and cured in a climate-controlled chamber (humidity and temperature) for 7 days [61]. Fissures were induced in prismatic samples with a bending test following ASTM C348 [62]. To prevent total sample rupture, a 5 mm diameter CA 60 steel bar was inserted ¼ into the base of the samples in a similar procedure as Alghamri et al. [63]. After fissuring, samples were returned to the chamber to promote oxygenation to the bacteria and induce the bio-mineralization reaction.

2.4 Visual analysis of healing, capillary absorption of water and strength resistance tests

After fissuring, visual analyses were conducted at 7, 14, 28 and 42 days to inspect for signs of healing and monitor its evolution. A Motic SMZ-168 stereo zoom microscope was used for inspection and measurements were taken with AutoCAD™ software. A capillary absorption test was conducted at 42 days following the procedure of RILEM TC 116-PCD [64] to obtain the mass of absorbed water relative to exposed area. The sample was leached with hydrochloric acid to remove any calcium carbonate precipitation and return the samples to their pre-healing state. The absorptivity test was then repeated, and the results allowed comparison of absorption pre and post-healing. Strength resistance tests were conducted following ASTM C39 [65] procedures at 7 days, 28 days and 54 days after demolding.

2.5 Complementary analyses

Scanning electronic microscopy (SEM) with a Zeiss brand apparatus was used in EP samples embedded in the cementitious matrix in order to determine their internal microstructure and suitability as capsule for B. subtilis. Along with SEM, an energy dispersion spectroscopy (EDS) was performed to evaluate EP chemical composition before and after calcium lactate spraying in order to determine its effective presence on the surface of the structure. Since EP is extremely porous, density variations allowed micro-computed tomography (µCT) to determine its distribution in the cementitious matrix. For these complementary analyses, fragments of sample BAC.7 were used from which planes were sliced with a circular saw. Exceptionally, pre-spraying samples of EP were taken for EDS analysis from the natural aggregate as supplied by the manufacturer. The samples were dried at 50ºC until mass constant and then metallized with gold.

3. RESULTS AND DISCUSSION

3.1 Compressive strength and capillary absorption of water

Figure 6 demonstrates that the average strength resistance remained at around the same level for the samples over the time periods of this study.
Figure 6 – Compression resistance results at 7 days, 28 days and 56 days

The largest variation was measured at 4.4 MPa, or 10.26% with respect to the smallest value, for sample BAC.8 between 7 days and 28 days. The fact that there was little variance from 7 days to 56 days could be attributed to the type of cement used, which contained an elevated percentage of clinker [56] and resulted in considerable strength from an early age. The largest difference in strength between mixing ratios for the same period was measured between BAC.7 and BAC.8 at 28 days: 7.9 MPa, or 20% with respect to the smallest value. The relatively low variance in strength between mixing ratios could be attributed to the identical cement: sand: EP ratio and water/cement relation between samples, of which only bacterial concentration varied. These results indicated that bacterial addition had no impact in the uncracked specimens despite the possibility of naturally occurring micro-fissures being formed in the cementitious matrices. Even if these micro-fissures were present, they were likely in small quantities and size since the samples themselves were of small size and have a high degree of freedom, thus minimizing fissuring from retractions.

Comparisons with Schwantes-Cezario et al. [53], which provided the mixing ratio, denoted that there was a decrease in strength. While strengths for the samples of this study were measured in the approximate range of 35 MPa to 50 MPa, Schwantes-Cezario et al. [53] obtained strengths in the range of 55 MPa to 75 MPa. This decrease was likely due to inserted EP being a lighter aggregate in the cementitious matrix as noted by Jonkers [35]. On the other hand, Afifudin et al. [24] observed that the same mixing ratio but with *B. subtilis* inserted directly in the matrix in different concentrations yielded more significant differences. It should be noticed that the authors did not use capsules, and thus, the bacterial solution may have contributed with the formation of the crystals along the sample, achieving better strength results.

The low variation in strength due to concentration observed in the samples of this study were likely the result of EP preventing bacterial activity in the uncracked samples. This was in agreement with the expectation that healing agents would only become active after the capsules were ruptured [63], [66].

Figure 7 shows the average water absorption of all samples before and after leaching with hydrochloric acid. It should be noted that it was not possible to compare absorption between samples because the fissures did not have the same geometry. At the 24 h mark, it could be seen that all samples had higher absorption after leaching with increases of 18.76%, 8.62% and 43.45% for samples BAC.6, BAC.7 and BAC.8, respectively. Although there was no proportional relationship between bacterial concentration and decrease in absorption, it could be stated that BAC.8 presented a higher efficiency in healing.

Figure 7 – Average absorptivity of all samples before and after acid leaching
Results obtained in this study, especially with BAC.8, were in-line with Alghamri et al. [63], which obtained a decrease of about 50% in absorption with a sodium silicate solution encapsulated when compared to control samples. Additionally, Siddique et al. [62] [67] inserted a bacterial solution in the matrix with no encapsulation and obtained absorption reductions between 50% and 70%. In this case, the direct insertion of bacteria had a more efficient impact in absorption since their action was not conditioned to capsule rupturing.

3.2 Visual surface analysis, SEM analysis and Micro-computed tomography

A total of 9 samples were molded for visual analysis and all presented evidence of at least 1 or more surface healing locations.

Figure 8 shows the results for sample BAC.6 at several ages, designated fissure 1 through 3. In Figure 8, fissures 1 and 2 were healed at 7 days and no further major healing products were formed at older ages. However, fissure 3 presented a different coloration healing which healed further between 28 days and 42 days. For sample BAC.6, the maximum healing width was of 0.46 mm and healing occurred in spots rather than along the extent of the fissure. Spotted healing similar to fissure 1 and fissure 2 was also observed in others studies [63] [68]. Wall-bound healing presented in fissure 3, which could further evolve and result in complete healing at later ages, was also annotated by Khaliq and Ehsan [69]. Whether spotted or wall-bound, healing could be attributed to the local availability of EP releasing products for self-healing. It should be noted that, regardless of the type of healing, the visual aspect observed in the samples were of a whitish product, referred to in other works as calcite formed from bacterial solutions [21], [70]. Figure 9 shows the results from sample BAC.7, designated fissure 4 through 7.
Figure 9 – Self-healing activity of sample BAC.7 at several ages

Fissure 4 did not present evidence of healing at 7 days and 14 days so there are no images associated with these ages. The healing in sample BAC.7 occurred in a similar fashion to sample BAC.6: namely, there were observable appearances of products in the fissures but, after a certain age, no further healing evolution. Figure 10 shows the results from sample BAC.8, designated fissure 8 through 11.

- 0.07 mm, partially continuous cicatrization
- 0.20 mm, spotted cicatrization
- 0.22 mm, spotted cicatrization
- Healing on the intern walls
Unlike previous samples, BAC.8 presented an expressive healing evolution from 7 days to 42 days. Similar evolutions were observed in other studies [15], [27]. Healing was also observed to appear along the entire length of the fissure, which made the BAC.8 sample the most efficient of the samples tested in this study, with a maximum width of fissure healed of 0.20 mm. It should also be noted that not only concentrations of healing product were detected in the fissure but further calcite deposits were observed on the surface – a behavior observed by Zhang et al. [15]. Table 4 presents the results of the visual analysis.

Table 4 – Synthesis of visual analysis results

<table>
<thead>
<tr>
<th>Sample</th>
<th>Fissure</th>
<th>Maximum healed width, mm</th>
<th>Type of healing</th>
</tr>
</thead>
<tbody>
<tr>
<td>BAC.6</td>
<td>1</td>
<td>0.15</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>2</td>
<td>0.22</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>3</td>
<td>0.46</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>4</td>
<td>0.07</td>
<td>Partially continuous</td>
</tr>
<tr>
<td></td>
<td>5</td>
<td>0.22</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>6</td>
<td>0.22</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>7</td>
<td>0.00</td>
<td>Starting healing on the walls</td>
</tr>
<tr>
<td>BAC.7</td>
<td>8</td>
<td>0.19</td>
<td>Partially continuous</td>
</tr>
<tr>
<td></td>
<td>9</td>
<td>0.20</td>
<td>Continuous</td>
</tr>
<tr>
<td></td>
<td>10</td>
<td>0.11</td>
<td>Spotted</td>
</tr>
<tr>
<td></td>
<td>11</td>
<td>0.11</td>
<td>Spotted*</td>
</tr>
</tbody>
</table>

*There was no total closure of the fissure width. The reported value is the difference between the initial and final widths.
The maximum fissure width healed was 0.46 mm for BAC.6, which was an exceptional value with respect to the other samples. The remaining samples had maximum healed fissure widths between 0.20 mm and 0.22 mm regardless of the composition. Comparing the results of this study with reference works, encapsulated in EP and coated in cement obtained a maximum healing width of 0.4 mm \( B. \textit{cohnii} \) but most healed fissures were below 0.3 mm [27]. Similarly, \( B. \textit{cohnii} \) encapsulated in EP but coated in geopolymer obtained a maximum healing width of 0.8 mm while also presenting higher efficiency for fissures up to 0.3 mm in width [15]. The differences observed between the studies may be related to the genus of bacteria or the type of coating. In the case of Jiang et al. [27], an alternative coating of magnesium potassium phosphate cement was able to heal a fissure 1.24 mm wide.

Besides differences in composition between samples, an aspect of extreme importance for healing to occur was the EP distribution inside the matrix. Accordingly, visual analysis of this study did not indicate a relation between the aspect of the fissure and the type of healing, further reinforcing that aggregate dispersion may be the single most relevant factor. This consideration was also postulated in De Koster et al. [71]. Scanning electron microscopy (SEM) was used to obtain the EP and micro-structure shown in Figure 11.

![Figure 11 – SEM of a slice of EP at magnifications of (a) 100x and (b) 2,000x](image)

It could be observed that a great quantity of void spaces was present inside the aggregate, in opposite contrast to a compact cementitious matrix. Pores were measured to have dimensions in the order of 10 µm to 40 µm. Since \( B. \textit{subtilis} \) AP91 spores had known approximate diameters of 1 µm [35] this demonstrated that the aggregate had sufficient space for encapsulation. However, bacterial spores were not seen inside EP in Figure 11. This could be explained as: (a) the EP cut used for SEM was too far from the surface of the aggregate at a depth at which bacteria might not have been absorbed (b) sample preparation required cutting and water from the cutting disk inevitably reached the test sample and may washing away of the micro-organisms from the EP.

Scanning electronic microscopy of EP impregnated with \( B. \textit{cohnii} \) was also conducted by Jiang et al. [27] and bacterial spore were observed. Spore distribution, however, was not homogeneous within the aggregate. This result highlighted the need of more studies to determine the total bacterial impregnation effectiveness. Otherwise, bacteria would be limited to void spaces located around the external part of the aggregate with considerable impact on the chosen EP granulometry to be used. To this end, an energy dispersion spectroscopy (EDS) was performed in the samples of this study before and after spraying of calcium lactate. Results are shown in Figures 12a and 12b and the percentage of elements detected is shown in Table 5.
Table 5 – Percentage of elements found through EDS in the samples

<table>
<thead>
<tr>
<th>Element</th>
<th>EP with calcium lactate, %</th>
<th>EP without bacterial solution or coating, %</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Weight</td>
<td>Atomic</td>
</tr>
<tr>
<td>O</td>
<td>41.91</td>
<td>57.85</td>
</tr>
<tr>
<td>Na</td>
<td>1.96</td>
<td>1.88</td>
</tr>
<tr>
<td>Al</td>
<td>5.79</td>
<td>4.74</td>
</tr>
<tr>
<td>Si</td>
<td>33.41</td>
<td>26.27</td>
</tr>
<tr>
<td>K</td>
<td>8.57</td>
<td>4.84</td>
</tr>
<tr>
<td>Ca</td>
<td>7.09</td>
<td>3.91</td>
</tr>
<tr>
<td>Fe</td>
<td>1.26</td>
<td>0.50</td>
</tr>
<tr>
<td>Total</td>
<td>100.00</td>
<td></td>
</tr>
</tbody>
</table>

Table 5 allows a chemical comparison between samples. It was clear that the EP sample sprayed with calcium lactate yielded a percentage of calcium 11x higher in its composition compared with the untreated EP grain. This indicated a successful absorption of the bacterial solution in the light aggregate.

Figure 13 displays images from micro-computed tomography (µCT) at the sample.

It could be seen that EP distribution inside the cementitious matrix was fairly homogeneous. Despite the differences in mass percentage of each material, there were no large accumulations or lack of aggregates throughout the planes. This EP distribution was significant since, in the case of a fissure, there should be a higher probability of rupture of capsules and subsequent release of healing product [6], [72], [73].
4. CONCLUSIONS

Strength resistance of the samples of this study denoted that the concentration of the bacterial solution encapsulated in the cementitious matrix did not affect performance. This was probably due to the lack of micro-organism activity since the samples were not fissured and no capsule rupture likely. Capillary absorption of water comparison across samples with varying levels of healing agent yielded a reduction in absorption after 42 days of healing. Sample BAC.8 obtained the largest reduction in absorption, 43.45%, while also presenting the highest visual efficiency with healing along the length of fissures, and thus, it is noticed its superior performance among others, considering the durability potential due to low absorption and its efficiency related to healing product formation. However, the width of fissure healed was equivalent in all samples with a maximum value of 0.22 mm and a single outlier result of 0.46 mm. Microanalysis of EP inserted in the cementitious matrix observed void spaces with adequate dimensions to hold bacterial spores. Further SEM/EDS analysis confirmed effective EP impregnation with calcium lactate spraying. Samples presented a uniform EP distribution within the matrix which increased the probability that, upon formation of a fissure, capsule rupturing would be likely and healing agent released.

Overall, it could be concluded that the concentration of \textit{B. subtilis} AP91 bacterial solution encapsulated in EP and coated in Portland cement did not affect the width of fissure healed. However, concentration did appear to have an effect in the length of fissure healed. It was unclear if the lower effectiveness of lower concentration samples was a result of the number of cells present lack of nutrients for bio-mineralization of calcium carbonate or uneven EP distribution in the cementitious matrix.

For this material to become viable on a large scale, both in financial and executive terms, conditions of use of bacterial solutions with less preparation must be evaluated, for example, invest in encapsulating the solution only by immersion, or analyze its use directly in the mixture water.

REFERENCES


**Author contributions:** VM: experimental procedure, materials characterization, writing; FP: experimental procedure, writing, revising; CMC: experimental procedure, materials characterization; FF: experimental procedure, materials characterization; VHV: writing, revising; RCEM: writing, revising; HZE: writing, revising; BFT: supervision, revising, formal analysis.

**Editors:** Mark Alexander, Guilherme Aris Parsekian.